Fuat Gzel

List of Publications by Year in Descending Order

Source: https://exaly.com/author-pdf/6561301/fuat-guzel-publications-by-year.pdf

Version: 2024-04-28

This document has been generated based on the publications and citations recorded by exaly.com. For the latest version of this publication list, visit the link given above.

The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

45
papers

2,057
citations

19
h-index

45
g-index

46
ext. papers

2,250
ext. citations

5.1
avg, IF

L-index

#	Paper	IF	Citations
45	Performance of wild plants-derived biochar in the remediation of water contaminated with lead: sorption optimization, kinetics, equilibrium, thermodynamics and reusability studies. <i>International Journal of Phytoremediation</i> , 2021 , 1-10	3.9	
44	Use of new nanoporous carbon produced from Mandarin (Citrus reticulata) industrial processing waste to remove anionic and cationic dyes. <i>Separation Science and Technology</i> , 2021 , 56, 1001-1013	2.5	1
43	Performance of grape (Vitis vinifera L.) industrial processing solid wastederived nanoporous carbon in copper(II) removal. <i>Biomass Conversion and Biorefinery</i> , 2021 , 11, 1363-1373	2.3	4
42	Synthesis, characterization, and lead (II) sorption performance of a new magnetic separable composite: MnFe2O4@wild plants-derived biochar. <i>Journal of Environmental Chemical Engineering</i> , 2021 , 9, 104567	6.8	10
41	Sorptive removal of copper(II) from water by biochar produced from a novel sustainable feedstock: wild herbs. <i>Environmental Science and Pollution Research</i> , 2021 , 28, 995-1005	5.1	2
40	Using of magnetized and non-magnetized tomato industrial processing solid waste in remediation of Reactive Blue 19 dye aqueous solution. <i>International Journal of Phytoremediation</i> , 2020 , 22, 1420-143	i ∂ ·9	3
39	Use of a novel bio-magnetic nanocomposite synthesized from industrial tomato processing waste for methylene blue removal: sorption optimization, kinetic and isotherm studies. <i>Cellulose</i> , 2020 , 27, 9577-9591	5.5	
38	Lead recovery from aqueous environment by using porous carbon of citrus fruits waste: equilibrium, kinetics and thermodynamic studies. <i>Separation Science and Technology</i> , 2020 , 55, 2699-271	1 2 ·5	5
37	Uptake of anionic and cationic dyes by highly effective porous carbon adsorber based on industrial processing residues. <i>Separation Science and Technology</i> , 2018 , 53, 1465-1475	2.5	5
36	Novel and sustainable precursor for high-quality activated carbon preparation by conventional pyrolysis: Optimization of produce conditions and feasibility in adsorption studies. <i>Advanced Powder Technology</i> , 2018 , 29, 726-736	4.6	10
35	Usability of activated carbon with well-developed mesoporous structure for the decontamination of malachite green from aquatic environments: kinetic, equilibrium and regeneration studies. Journal of Porous Materials, 2018 , 25, 477-488	2.4	13
34	Surface modification of black tea waste using bleaching technique for enhanced biosorption of Methylene blue in aqueous environment. <i>Separation Science and Technology</i> , 2018 , 53, 2882-2895	2.5	7
33	Role of optimization parameters in the production of nanoporous carbon from mandarin shells by microwave-assisted chemical activation and utilization as dye adsorbent. <i>Advanced Powder Technology</i> , 2018 , 29, 2108-2118	4.6	19
32	Chemical modification of a cellulose-based material to improve its adsorption capacity for anionic dyes. <i>Journal of Dispersion Science and Technology</i> , 2017 , 38, 381-392	1.5	19
31	Optimal oxidation with nitric acid of biochar derived from pyrolysis of weeds and its application in removal of hazardous dye methylene blue from aqueous solution. <i>Journal of Cleaner Production</i> , 2017 , 144, 260-265	10.3	97
30	Behavior of mesoporous activated carbon used as a remover in Congo red adsorption process. Water Science and Technology, 2017 , 2017, 170-183	2.2	3
29	High surface area mesoporous activated carbon from tomato processing solid waste by zinc chloride activation: process optimization, characterization and dyes adsorption. <i>Journal of Cleaner Production</i> , 2016 , 113, 995-1004	10.3	247

(2007-2016)

28	Adsorptive efficacy analysis of novel carbonaceous sorbent derived from grape industrial processing wastes towards tetracycline in aqueous solution. <i>Journal of the Taiwan Institute of Chemical Engineers</i> , 2016 , 60, 236-240	5.3	24
27	Effective removal of tetracycline from aqueous solution using activated carbon prepared from tomato (Lycopersicon esculentum Mill.) industrial processing waste. <i>Ecotoxicology and Environmental Safety</i> , 2016 , 131, 22-9	7	86
26	New low-cost nanoporous carbonaceous adsorbent developed from carob (Ceratonia siliqua) processing industry waste for the adsorption of anionic textile dye: Characterization, equilibrium and kinetic modeling. <i>Journal of Molecular Liquids</i> , 2015 , 206, 244-255	6	56
25	Development and physicochemical characterization of a new magnetic nanocomposite as an economic antibiotic remover. <i>Chemical Engineering Research and Design</i> , 2015 , 94, 441-451	5.5	23
24	Performance of new mesoporous carbon sorbent prepared from grape industrial processing wastes for malachite green and congo red removal. <i>Chemical Engineering Research and Design</i> , 2015 , 100, 27-38	5.5	48
23	Conversion of grape industrial processing waste to activated carbon sorbent and its performance in cationic and anionic dyes adsorption. <i>Journal of Cleaner Production</i> , 2015 , 93, 84-93	10.3	156
22	Evaluation of Pomegranate (Punica Granatum L.) Pulps for the Removal of Copper(II) Ions: Kinetic, Equilibrium, and Desorption Studies. <i>Journal of Dispersion Science and Technology</i> , 2014 , 35, 482-493	1.5	7
21	Elimination of anionic dye by using nanoporous carbon prepared from an industrial biowaste. Journal of Molecular Liquids, 2014 , 194, 130-140	6	52
20	APPLICATION OF SOME DOMESTIC WASTES AS NEW LOW-COST BIOSORBENTS FOR REMOVAL OF METHYLENE BLUE: KINETIC AND EQUILIBRIUM STUDIES. <i>Chemical Engineering Communications</i> , 2014 , 201, 557-578	2.2	45
19	Decolorisation of aqueous crystal violet solution by a new nanoporous carbon: Equilibrium and kinetic approach. <i>Journal of Industrial and Engineering Chemistry</i> , 2014 , 20, 3375-3386	6.3	34
18	Using grape pulp as a new alternative biosorbent for removal of a model basic dye. <i>Asia-Pacific Journal of Chemical Engineering</i> , 2014 , 9, 214-225	1.3	6
17	Bioremoval and recovery of Cu(II) and Pb(II) from aqueous solution by a novel biosorbent watermelon (Citrullus lanatus) seed hulls: Kinetic study, equilibrium isotherm, SEM and FTIR analysis. <i>Desalination and Water Treatment</i> , 2013 , 51, 7311-7322		20
16	Optimization of Copper and Lead Removal by a Novel Biosorbent: Cucumber (Cucumis Sativus) Peelskinetic, Equilibrium, and Desorption Studies. <i>Journal of Dispersion Science and Technology</i> , 2013 , 34, 1295-1307	1.5	12
15	Determination of lead separated selectively with ion exchange method from solution onto BCW in Sirnak, East Anatolia of Turkey. <i>Microchemical Journal</i> , 2011 , 98, 246-253	4.8	14
14	Adsorption of some highly toxic dyestuffs from aqueous solution by chitin and its synthesized derivatives. <i>Desalination</i> , 2009 , 249, 1115-1123	10.3	27
13	Kinetics and thermodynamics of the adsorption of some dyestuffs from aqueous solution by poplar sawdust. <i>Bioresource Technology</i> , 2008 , 99, 2009-17	11	36
12	Determination of kinetic and equilibrium parameters of the batch adsorption of Mn(II), Co(II), Ni(II) and Cu(II) from aqueous solution by black carrot (Daucus carota L.) residues. <i>Journal of Hazardous Materials</i> , 2008 , 153, 1275-87	12.8	132
11	Kinetics of the adsorption of reactive dyes by chitin. <i>Dyes and Pigments</i> , 2007 , 73, 168-177	4.6	94

10	Rate studies on the adsorption of some dyestuffs and p-nitrophenol by chitosan and monocarboxymethylated(mcm)-chitosan from aqueous solution. <i>Journal of Hazardous Materials</i> , 2005 , 118, 141-54	12.8	47
9	Kinetics and thermodynamics of the adsorption of some dyestuffs and p-nitrophenol by chitosan and MCM-chitosan from aqueous solution. <i>Journal of Colloid and Interface Science</i> , 2004 , 274, 398-412	9.3	77
8	Removal of copper, nickel, cobalt and manganese from aqueous solution by kaolinite. <i>Water Research</i> , 2003 , 37, 948-52	12.5	593
7	Characterization of the Micropore Structure of Activated Carbons by Adsorptions of Nitrogen and Some Hydrocarbons. <i>Separation Science and Technology</i> , 1999 , 34, 587-597	2.5	3
6	Preparation and Characterization of Polyoxometallates of Molybdenum, Tungsten and Their Salts. <i>Synthesis and Reactivity in Inorganic, Metal Organic, and Nano Metal Chemistry</i> , 1997 , 27, 1399-1415		5
5	The Effect of Surface Acidity upon the Adsorption Capacities of Activated Carbons. <i>Separation Science and Technology</i> , 1996 , 31, 283-290	2.5	13
4	Determination of Surface Areas of Active Carbons by Retention of Ethylene Glycol and Ethylene Glycol Monoethyl Ether. <i>Separation Science and Technology</i> , 1996 , 31, 1687-1693	2.5	1
3	On the performance of wild plant-derived biochar@MnFe2O4 composite in remediation of synthetic aqueous copper solution. <i>Biomass Conversion and Biorefinery</i> ,1	2.3	
2	Conversion of citrus industrial processing solid residues to well-developed mesoporous powder-activated carbon and its some water pollutant removal performance. <i>Biomass Conversion and Biorefinery</i> ,1	2.3	1
1	Adsorptive removal of diclofenac sodium from aqueous solution via industrial processed citrus solid wasteBased activated carbon: optimization, kinetics, equilibrium, thermodynamic, and reusability analyses. <i>Biomass Conversion and Biorefinery</i> ,1	2.3	О