Hein W J P Neomagus

List of Publications by Year in descending order

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147566 143772 3,478 91 31 57 citations h-index g-index papers 93 93 93 3238 docs citations times ranked citing authors all docs

#	Article	IF	CITATIONS
1	Comparing the porosity and surface areas of coal as measured by gas adsorption, mercury intrusion and SAXS techniques. Fuel, 2015, 141, 293-304.	3.4	360
2	Chemical–structural properties of South African bituminous coals: Insights from wide angle XRD–carbon fraction analysis, ATR–FTIR, solid state 13 C NMR, and HRTEM techniques. Fuel, 2015, 158, 779-792.	3.4	262
3	High-temperature membrane reactors: potential and problems. Chemical Engineering Science, 1999, 54, 1997-2017.	1.9	230
4	Reaction kinetics of pulverized coal-chars derived from inertinite-rich coal discards: Gasification with carbon dioxide and steam. Fuel, 2006, 85, 1076-1082.	3.4	168
5	Reactor technology options for distributed hydrogen generation via ammonia decomposition: A review. International Journal of Hydrogen Energy, 2013, 38, 14968-14991.	3.8	131
6	Copper(II) removal from polluted water with alumina/chitosan composite membranes. Journal of Membrane Science, 2002, 197, 147-156.	4.1	108
7	Co-pyrolysis of coal and raw/torrefied biomass: A review on chemistry, kinetics and implementation. Renewable and Sustainable Energy Reviews, 2021, 135, 110189.	8.2	101
8	The influence of the degree of cross-linking on the adsorption properties of chitosan beads. Bioresource Technology, 2008, 99, 7377-7382.	4.8	93
9	Influence of maceral composition on the structure, properties and behaviour of chars derived from South African coals. Fuel, 2015, 142, 9-20.	3.4	89
10	Assessing the catalytic effect of coal ash constituents on the CO2 gasification rate of high ash, South African coal. Fuel Processing Technology, 2011, 92, 2048-2054.	3.7	87
11	Salt rejection in nanofiltration for single and binary salt mixtures in view of sulphate removal. Desalination, 2005, 171, 205-215.	4.0	86
12	Properties of high ash coal-char particles derived from inertinite-rich coal: II. Gasification kinetics with carbon dioxide. Fuel, 2008, 87, 3403-3408.	3.4	85
13	The carbon dioxide gasification characteristics of biomass char samples and their effect on coal gasification reactivity during co-gasification. Bioresource Technology, 2018, 258, 70-78.	4.8	83
14	X-ray diffraction parameters and reaction rate modeling for gasification and combustion of chars derived from inertinite-rich coals. Fuel, 2013, 109, 148-156.	3.4	67
15	CFD modeling of particle charging and collection in electrostatic precipitators. Journal of Electrostatics, 2016, 84, 10-22.	1.0	60
16	Structural and chemical modifications of typical South African biomasses during torrefaction. Bioresource Technology, 2016, 202, 192-197.	4.8	59
17	Density functional theory molecular modelling and experimental particle kinetics for CO2–char gasification. Carbon, 2015, 93, 295-314.	5.4	58
18	Experimentation and CFD modelling of a microchannel reactor for carbon dioxide methanation. Chemical Engineering Journal, 2017, 313, 847-857.	6.6	57

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19	Improved reactivity of large coal particles by K2CO3 addition during steam gasification. Fuel Processing Technology, 2013, 114, 75-80.	3.7	52
20	The transient swelling behaviour of large (\hat{a}^2 20 + 16 mm) South African coal particles during low-temperature devolatilisation. Fuel, 2014, 136, 79-88.	3.4	48
21	Hydrogen production from ammonia decomposition over a commercial Ru/Al2O3 catalyst in a microchannel reactor: Experimental validation and CFD simulation. International Journal of Hydrogen Energy, 2016, 41, 3774-3785.	3.8	48
22	Properties of high ash char particles derived from inertinite-rich coal: 1. Chemical, structural and petrographic characteristics. Fuel, 2008, 87, 3082-3090.	3.4	42
23	The adsorption of copper in a packed-bed of chitosan beads: Modeling, multiple adsorption and regeneration. Journal of Hazardous Materials, 2009, 167, 1242-1245.	6.5	40
24	Recent Advances in Membrane-Based Electrochemical Hydrogen Separation: A Review. Membranes, 2021, 11, 127.	1.4	39
25	A modelling evaluation of an ammonia-fuelled microchannel reformer for hydrogen generation. International Journal of Hydrogen Energy, 2014, 39, 11390-11402.	3.8	38
26	Chemical and structural characterization of char development during lignocellulosic biomass pyrolysis. Bioresource Technology, 2017, 243, 941-948.	4.8	38
27	The characterisation of slow-heated inertinite- and vitrinite-rich coals from the South African coalfields. Fuel, 2015, 158, 591-601.	3.4	36
28	The effect of acid demineralising bituminous coals and de-ashing the respective chars on nitrogen functional forms. Journal of Analytical and Applied Pyrolysis, 2017, 125, 127-135.	2.6	35
29	Transformation of nitrogen functional forms and the accompanying chemical-structural properties emanating from pyrolysis of bituminous coals. Applied Energy, 2018, 216, 414-427.	5.1	34
30	Centrifugal casting of ceramic membrane tubes and the coating with chitosan. Separation and Purification Technology, 2001, 25, 407-413.	3.9	33
31	Performance evaluation of a high-throughput microchannel reactor for ammonia decomposition over a commercial Ru-based catalyst. International Journal of Hydrogen Energy, 2015, 40, 2921-2926.	3.8	33
32	Pore development during gasification of South African inertinite-rich chars evaluated using small angle X-ray scattering. Carbon, 2015, 95, 250-260.	5.4	32
33	Particle size influence on the pore development of nanopores in coal gasification chars: From micron to millimeter particles. Carbon, 2017, 112, 37-46.	5.4	32
34	Hydrogen Separation and Purification from Various Gas Mixtures by Means of Electrochemical Membrane Technology in the Temperature Range 100–160 °C. Membranes, 2021, 11, 282.	1.4	32
35	Elucidation of the Structural and Molecular Properties of Typical South African Coals. Energy & Camp; Fuels, 2013, 27, 3161-3172.	2.5	31
36	The catalytic combustion of natural gas in a membrane reactor with separate feed of reactants. Chemical Engineering Journal, 2000, 77, 165-177.	6.6	30

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37	The random pore model with intraparticle diffusion for the description of combustion of char particles derived from mineral- and inertinite rich coal. Fuel, 2011, 90, 2347-2352.	3.4	30
38	Evaluation and prediction of slow pyrolysis products derived from coals of different rank. Journal of Analytical and Applied Pyrolysis, 2017, 128, 156-167.	2.6	30
39	The effect of acid washing on the pyrolysis products derived from a vitrinite-rich bituminous coal. Journal of Analytical and Applied Pyrolysis, 2015, 116, 142-151.	2.6	28
40	Pervaporation separation of methanol from methanol/tert-amyl methyl ether mixtures with a commercial membrane. Journal of Membrane Science, 2002, 209, 353-362.	4.1	26
41	Reaction kinetics of pulverized coal-chars derived from inertinite-rich coal discards: Characterisation and combustion. Fuel, 2006, 85, 1067-1075.	3.4	26
42	A perspective on South African coal fired power station emissions. Journal of Energy in Southern Africa, 2015, 26, 27-40.	0.5	26
43	Kinetic analysis of non-isothermal thermogravimetric analyser results using a new method for the evaluation of the temperature integral and multi-heating rates. Fuel, 2006, 85, 418-422.	3.4	25
44	Influence of Chemical Pretreatment on the Internal Structure and Reactivity of Pyrolysis Chars Produced from Sugar Cane Bagasse. Energy & Energy & 2012, 26, 4497-4506.	2.5	24
45	The carbon dioxide, methane and nitrogen high-pressure sorption properties of South African bituminous coals. International Journal of Coal Geology, 2019, 209, 40-53.	1.9	22
46	The effect of added minerals on the pyrolysis products derived from a vitrinite-rich demineralised South African coal. Journal of Analytical and Applied Pyrolysis, 2016, 121, 41-49.	2.6	21
47	A comparison of glycans and polyglycans using solid-state NMR and X-ray powder diffraction. Solid State Nuclear Magnetic Resonance, 2006, 30, 150-161.	1.5	20
48	The influence of design parameters on the occurrence of shielding in multi-electrode ESPs and its effect on performance. Journal of Electrostatics, 2018, 93, 17-30.	1.0	19
49	Transport properties of chitosan membranes for zinc (II) removal from aqueous systems. Separation and Purification Technology, 2017, 179, 428-437.	3.9	18
50	Experimental performance evaluation of an ammonia-fuelled microchannel reformer for hydrogen generation. International Journal of Hydrogen Energy, 2014, 39, 7225-7235.	3.8	17
51	An evaluation of a new automated duplicate-sample Fischer Assay setup according to ISO/ASTM standards and analysis of the tar fraction. Journal of Analytical and Applied Pyrolysis, 2014, 106, 190-196.	2.6	17
52	The catalytic oxidation of H2S in a stainless steel membrane reactor with separate feed of reactants. Journal of Membrane Science, 1998, 148, 147-160.	4.1	16
53	Co-pyrolysis of torrefied biomass and coal: Effect of pressure on synergistic reactions. Journal of Analytical and Applied Pyrolysis, 2022, 161, 105363.	2.6	15
54	Release of Nitrogenous Volatile Species from South African Bituminous Coals during Pyrolysis. Energy &	2.5	14

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55	Mineralogy and Petrology of Chars Produced by South African Caking Coals and Density-Separated Fractions during Pyrolysis and Their Effects on Caking Propensity. Energy & Energy & Energy & South States of Caking Propensity. Energy & Ener	2.5	14
56	The modeling of the combustion of high-ash coal–char particles suitable for pressurised fluidized bed combustion: shrinking reacted core model. Fuel, 2005, 84, 1136-1143.	3.4	13
57	SO ₂ Solubility in 50 wt % H ₂ SO ₄ at Elevated Temperatures and Pressures. Journal of Chemical & Solution (1997) and Pressures. Journal of Chemical & Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution (1997) are substituted in the Solution (1997) and Pressures (1997) are substituted in the Solution	1.0	12
58	Modeling the Nonisothermal Devolatilization Kinetics of Typical South African Coals. Energy & Energy & Fuels, 2014, 28, 920-933.	2.5	12
59	The effect of carbon dioxide partial pressure on the gasification rate and pore development of Highveld coal chars at elevated pressures. Fuel Processing Technology, 2018, 179, 1-9.	3.7	12
60	Sulphur trioxide decomposition with supported platinum/palladium on rutile catalysts: 1. Reaction kinetics of catalyst pellets. International Journal of Hydrogen Energy, 2015, 40, 85-94.	3.8	11
61	Coal-derived low smoke fuel assessment through coal stove combustion testing. Journal of Analytical and Applied Pyrolysis, 2017, 126, 158-168.	2.6	11
62	An experimentally validated computational model to predict the performance of a single-channel laboratory-scale electrostatic precipitator equipped with spiked and wire discharge electrodes. Journal of Electrostatics, 2021, 112, 103595.	1.0	10
63	Thermal reduction of barium sulphate with carbon monoxide—A thermogravimetric study. Thermochimica Acta, 2010, 498, 67-70.	1.2	9
64	Dissolution kinetics of sorbents and effect of additives in wet flue gas desulfurization. Reviews in Chemical Engineering, 2014, 30, .	2.3	9
65	The properties of large coal particles and reaction kinetics of corresponding chars. Fuel, 2015, 140, 17-26.	3.4	9
66	The influence of particle size on the thermal performance of coal and its derived char in a Union stove. Energy Geoscience, 2021, 2, 148-159.	1.3	9
67	Manufacturing and testing of briquettes from inertinite-rich low-grade coal fines using various binders. Journal of the South African Institute of Mining and Metallurgy, 2018, 118, 83-88.	0.5	9
68	The use of thermomechanical analysis to characterise Söderberg electrode paste raw materials. Minerals Engineering, 2013, 46-47, 167-176.	1.8	8
69	Lumped chemical kinetic modelling of raw and torrefied biomass under pressurized pyrolysis. Energy Conversion and Management, 2022, 253, 115199.	4.4	8
70	A kinetic expression for the pyrolytic decomposition of polytetrafluoroethylene. Journal of Fluorine Chemistry, 2008, 129, 314-318.	0.9	7
71	Sulphur trioxide decomposition with supported platinum/palladium on rutile catalyst: 2. Performance of a laboratory fixed bed reactor. International Journal of Hydrogen Energy, 2015, 40, 2493-2499.	3.8	7
72	Dissolution kinetics of South African coal fly ash and the development of a semi-empirical model to predict dissolution. Chemical Industry and Chemical Engineering Quarterly, 2015, 21, 319-330.	0.4	6

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73	Reactivity study of fine discard coal agglomerates. Journal of Analytical and Applied Pyrolysis, 2015, 113, 723-728.	2.6	5
74	Effect of Relative Humidity and Temperature on the Mechanical Properties of PFSA Nafionâ,,¢-cation-exchanged membranes for Electrochemical Applications. International Journal of Electrochemical Science, 2017, 12, 2573-2582.	0.5	5
75	Organicâ€Organic Separation by Pervaporation. II. Separation of Methanol from Tame by an αâ€Alumina Supported Nayâ€Zeolite Membrane. Separation Science and Technology, 2005, 40, 1047-1065.	1.3	4
76	Leaching kinetics of bottom ash waste as a source of calcium ions. Journal of the Air and Waste Management Association, 2015, 65, 126-132.	0.9	4
77	Dataset on the carbon dioxide, methane and nitrogen high-pressure sorption properties of South African bituminous coals. Data in Brief, 2019, 25, 104248.	0.5	4
78	A FIXED BED BARRIER REACTOR WITH SEPARATE FEED OF REACTANTS. Chemical Engineering Communications, 2001, 184, 49-69.	1.5	3
79	Influence of Potassium Carbonate on the Swelling Propensity of South African Large Coal Particles. Energy & Energy & Ene	2.5	3
80	A laboratory scale fixed-bed coal conversion reactor part 1: Operation, reaction zone identification and industrial representativeness. Journal of Analytical and Applied Pyrolysis, 2015, 115, 428-436.	2.6	3
81	The effect of particle size on the pollution reduction potential of a South African coal-derived low-smoke fuel. Energy Geoscience, 2020, 1, 165-173.	1.3	3
82	Influence of additives on the devolatilization product yield of typical South African coals, and effect on tar composition. Journal of the Southern African Institute of Mining and Metallurgy, 2018, 118, 395-407.	0.1	3
83	A Predictive Model for Permeation Flux in a Membrane Reactor: Aspects of Esterification. Separation Science and Technology, 2005, 40, 433-452.	1.3	2
84	Reduction of Caking Propensity in Large (Millimeter-Sized) South African Coal Particles with Potassium Carbonate Impregnation To Expand Fixed- and Fluidized-Bed Gasification Feedstock Suitability. Energy & Dels, 2015, 29, 4255-4263.	2.5	2
85	Effect of Fly Ash as an Additive on the Limestone Dissolution Rate Constant. Energy & Emp; Fuels, 2015, 29, 3284-3291.	2.5	2
86	Coal reactivity and selection for solid-based pre-reduction of sponge iron. International Journal of Coal Preparation and Utilization, 2020, 40, 233-246.	1.2	2
87	Significance of coal properties on the caking degree of coarse coal particles mined at Limpopo Province, Republic of South Africa. International Journal of Coal Preparation and Utilization, 2020, 40, 297-319.	1.2	2
88	The CO Tolerance of Pt/C and Pt-Ru/C Electrocatalysts in a High-Temperature Electrochemical Cell Used for Hydrogen Separation. Membranes, 2021, 11, 670.	1.4	2
89	Effect of the V5+ to V4+ Molar Ratio on H2S Absorption and Conversion to Elemental Sulfur. Industrial & Engineering Chemistry Research, 2021, 60, 1505-1516.	1.8	1
90	A Comparative Study of the Processing Scheme of Chitosan and Nafion 117 in Membrane Electrode Assembly. Petroleum Science and Technology, 2013, 31, 121-128.	0.7	0

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91	1	Semiâ€Empirical Model for Limestone Dissolution in Adipic Acid for Wet Flue Gas Desulfurization. Chemical Engineering and Technology, 2014, 37, 1919-1928.	0.9	O