

# Elizabeth Castillo-Martinez

## List of Publications by Year in descending order

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59  
papers

4,358  
citations

168829

31  
h-index

169272

56  
g-index

65  
all docs

65  
docs citations

65  
times ranked

8039  
citing authors

#	ARTICLE	IF	CITATIONS
1	Cellulose Nanocrystals in Sustainable Energy Systems. <i>Advanced Sustainable Systems</i> , 2022, 6, .	2.7	15
2	Sequential Fe Reduction, Involving Two Different Fe <sup>+</sup> Intermediates, in the Conversion Reaction of Prussian Blue in Lithium-Ion Batteries. <i>Chemistry of Materials</i> , 2022, 34, 4660-4671.	3.2	0
3	Electrochemical synthesis of MnO <sub>2</sub> / NiO / ZnO trijunction coated stainless steel substrate as a supercapacitor electrode and cyclic voltammetry behavior modeling using artificial neural network. <i>International Journal of Energy Research</i> , 2022, 46, 17163-17179.	2.2	11
4	High Conductivity in a Fluorine-Free K-Ion Polymer Electrolyte. <i>ACS Applied Energy Materials</i> , 2022, 5, 9009-9019.	2.5	9
5	Sustainable materials for off-grid battery applications: advances, challenges and prospects. <i>Sustainable Energy and Fuels</i> , 2021, 5, 310-331.	2.5	14
6	Revisiting metal fluorides as lithium-ion battery cathodes. <i>Nature Materials</i> , 2021, 20, 841-850.	13.3	109
7	Lithiation phase behaviors of metal oxide anodes and extra capacities. <i>Cell Reports Physical Science</i> , 2021, 2, 100543.	2.8	6
8	Lithium ion storage in 1D and 2D redox active metal-organic frameworks. <i>Electrochimica Acta</i> , 2020, 341, 136063.	2.6	6
9	Understanding LiOH Formation in a Li-O <sub>2</sub> Battery with LiI and H <sub>2</sub> O Additives. <i>ACS Catalysis</i> , 2019, 9, 66-77.	5.5	57
10	Hybrid biopolymer electrodes for lithium- and sodium-ion batteries in organic electrolytes. <i>Sustainable Energy and Fuels</i> , 2018, 2, 836-842.	2.5	23
11	Polymer-Templated LiFePO <sub>4</sub> /C Nanonetworks as High-Performance Cathode Materials for Lithium-Ion Batteries. <i>ACS Applied Materials &amp; Interfaces</i> , 2018, 10, 1646-1653.	4.0	71
12	Polymeric Redox-Active Electrodes for Sodium-Ion Batteries. <i>ChemSusChem</i> , 2018, 11, 311-319.	3.6	19
13	Understanding Fluoroethylene Carbonate and Vinylene Carbonate Based Electrolytes for Si Anodes in Lithium Ion Batteries with NMR Spectroscopy. <i>Journal of the American Chemical Society</i> , 2018, 140, 9854-9867.	6.6	219
14	Temperature effect on the synthesis of lignin-derived carbons for electrochemical energy storage applications. <i>Journal of Power Sources</i> , 2018, 397, 296-306.	4.0	34
15	Electrochemical performance of novel O3 layered Al,Mg doped titanates as anode materials for Na-ion batteries. <i>Materials Research Bulletin</i> , 2017, 94, 199-207.	2.7	8
16	Advanced anode materials for sodium ion batteries: carbodiimides. <i>MRS Advances</i> , 2017, 2, 1165-1176.	0.5	11
17	Metal-Organic Nanosheets Formed via Defect-Mediated Transformation of a Hafnium Metal-Organic Framework. <i>Journal of the American Chemical Society</i> , 2017, 139, 5397-5404.	6.6	224
18	Electrochemical performance of CuNCN for sodium ion batteries and comparison with ZnNCN and lithium ion batteries. <i>Journal of Power Sources</i> , 2017, 367, 130-137.	4.0	37

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19	Identifying the Structural Basis for the Increased Stability of the Solid Electrolyte Interphase Formed on Silicon with the Additive Fluoroethylene Carbonate. <i>Journal of the American Chemical Society</i> , 2017, 139, 14992-15004.	6.6	176
20	Na-ion Batteries for Large Scale Applications: A Review on Anode Materials and Solid Electrolyte Interphase Formation. <i>Advanced Energy Materials</i> , 2017, 7, 1700463.	10.2	261
21	Electrochemical characterization of NaFe <sub>2</sub> (CN) <sub>6</sub> Prussian Blue as positive electrode for aqueous sodium-ion batteries. <i>Electrochimica Acta</i> , 2016, 210, 352-357.	2.6	62
22	Response to Comment on "Cycling Li-O <sub>2</sub> batteries via LiOH formation and decomposition". <i>Science</i> , 2016, 352, 667-667.	6.0	11
23	Response to Comment on "Cycling Li-O <sub>2</sub> batteries via LiOH formation and decomposition". <i>Science</i> , 2016, 352, 667-667.	6.0	32
24	Highly water-soluble three-redox state organic dyes as bifunctional analytes. <i>Energy and Environmental Science</i> , 2016, 9, 3521-3530.	15.6	66
25	Towards environmentally friendly Na-ion batteries: Moisture and water stability of Na <sub>2</sub> Ti <sub>3</sub> O <sub>7</sub> . <i>Journal of Power Sources</i> , 2016, 324, 378-387.	4.0	39
26	Higher voltage plateau cubic Prussian White for Na-ion batteries. <i>Journal of Power Sources</i> , 2016, 324, 766-773.	4.0	91
27	Identification of the critical synthesis parameters for enhanced cycling stability of Na-ion anode material Na <sub>2</sub> Ti <sub>3</sub> O <sub>7</sub> . <i>Acta Materialia</i> , 2016, 104, 125-130.	3.8	27
28	Optimizing the electrolyte and binder composition for Sodium Prussian Blue, Na <sub>1-x</sub> Fe <sub>x+(1/3)</sub> (CN) <sub>6</sub> ·yH <sub>2</sub> O, as cathode in sodium ion batteries. <i>Electrochimica Acta</i> , 2016, 200, 123-130.	2.6	42
29	Carbodiimides: new materials applied as anode electrodes for sodium and lithium ion batteries. <i>Journal of Materials Chemistry A</i> , 2016, 4, 1608-1611.	5.2	69
30	Response to Comment on "Cycling Li-O <sub>2</sub> , batteries via LiOH formation and decomposition". <i>Science</i> , 2016, 352, 667.	6.0	0
31	Structure of H <sub>2</sub> Ti <sub>3</sub> O <sub>7</sub> and its evolution during sodium insertion as anode for Na ion batteries. <i>Physical Chemistry Chemical Physics</i> , 2015, 17, 6988-6994.	1.3	46
32	Composition and Evolution of the Solid-Electrolyte Interphase in Na <sub>2</sub> Ti <sub>3</sub> O <sub>7</sub> Electrodes for Na-Ion Batteries: XPS and Auger Parameter Analysis. <i>ACS Applied Materials &amp; Interfaces</i> , 2015, 7, 7801-7808.	4.0	164
33	Oligomeric-Schiff bases as negative electrodes for sodium ion batteries: unveiling the nature of their active redox centers. <i>Energy and Environmental Science</i> , 2015, 8, 3233-3241.	15.6	97
34	Tunneling phenomena in aligned multi-walled carbon nanotube sheets: conductivity and Raman correlations. <i>Materials Research Express</i> , 2014, 1, 045603.	0.8	2
35	Polymeric Schiff Bases as Low Voltage Redox Centers for Sodium-ion Batteries. <i>Angewandte Chemie - International Edition</i> , 2014, 53, 5341-5345.	7.2	170
36	High temperature structural transformations of few layer graphene nanoribbons obtained by unzipping carbon nanotubes. <i>Journal of Materials Chemistry A</i> , 2014, 2, 221-228.	5.2	32

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37	K1 $\sim$ Fe <sub>2+3</sub> (CN) <sub>6</sub> ·yH <sub>2</sub> O, Prussian Blue as a displacement anode for lithium ion batteries. Journal of Power Sources, 2014, 271, 489-496.	4.0	43
38	Update on Na-based battery materials. A growing research path. Energy and Environmental Science, 2013, 6, 2312.	15.6	886
39	Comprehensive Insights into the Structural and Chemical Changes in Mixed-Anion FeOF Electrodes by Using Operando PDF and NMR Spectroscopy. Journal of the American Chemical Society, 2013, 135, 4070-4078.	6.6	124
40	Reconstructed Ribbon Edges in Thermally Reduced Graphene Nanoribbons. Journal of Physical Chemistry C, 2012, 116, 24006-24015.	1.5	20
41	Oriented Graphene Nanoribbon Yarn and Sheet from Aligned Multi-Walled Carbon Nanotube Sheets. Advanced Materials, 2012, 24, 5695-5701.	11.1	67
42	Photoinduced Optical Transparency in Dye-Sensitized Solar Cells Containing Graphene Nanoribbons. Journal of Physical Chemistry C, 2011, 115, 25125-25131.	1.5	35
43	Highly Stable Cooperative Distortion in a Weak Jahn-Teller d <sup>2</sup> Cation: Perovskite-Type ScVO <sub>3</sub> Obtained by High-Pressure and High-Temperature Transformation from Bixbyite. Journal of the American Chemical Society, 2011, 133, 8552-8563.	6.6	31
44	Biscrolling Nanotube Sheets and Functional Guests into Yarns. Science, 2011, 331, 51-55.	6.0	338
45	Thermal actuation of graphene oxide nanoribbon mats. Chemical Physics Letters, 2011, 505, 31-36.	1.2	15
46	Magneto-thermal and dielectric properties of biferroic YCrO <sub>3</sub> prepared by combustion synthesis. Journal of Solid State Chemistry, 2010, 183, 1863-1871.	1.4	88
47	Spinel to CaFe <sub>2</sub> O <sub>4</sub> Transformation: Mechanism and Properties of $\text{[}^{2-}\text{CdCr}_2\text{O}_4\text{]}$ . Inorganic Chemistry, 2010, 49, 2827-2833.	1.9	29
48	The A(II)Cr(IV)O <sub>3</sub> (A=Sr, Ca, Pb) $\text{A}^{\text{simple}}\text{TM}$ perovskites. Structure and properties: magnetic structure of CaCrO <sub>3</sub> . High Pressure Research, 2009, 29, 254-260.	0.4	14
49	Structure, microstructure and magnetic properties of Sr <sub>1-x</sub> Ca <sub>x</sub> CrO <sub>3</sub> (0 $\leq$ x $\leq$ 1). Journal of Solid State Chemistry, 2008, 181, 895-904.	1.4	37
50	Structure and microstructure of the high pressure synthesised misfit layer compound [Sr <sub>2</sub> O <sub>2</sub> ][CrO <sub>2</sub> ] <sub>1.85</sub> . Journal of Solid State Chemistry, 2008, 181, 1840-1847.	1.4	10
51	Electron energy loss spectroscopy in ACrO <sub>3</sub> (A = Ca, Sr and Pb) perovskites. Journal of Physics Condensed Matter, 2008, 20, 505207.	0.7	17
52	Increasing the Structural Complexity of Chromium(IV) Oxides by High-Pressure and High-Temperature Reactions of CrO <sub>2</sub> . Inorganic Chemistry, 2008, 47, 8526-8542.	1.9	22
53	A Study of [Cr-O <sub>6</sub> ]-based rutile analogues by means of EELS. Materials Research Society Symposia Proceedings, 2008, 1148, 1.	0.1	1
54	Revisiting the High Pressure Ternary Oxides of Cr(IV): Structures and Microstructures. , 2008, , .		1

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55	Revisiting the Sr <sup>2+</sup> /Cr(IV)O system at high pressure and temperature with special reference to Sr <sub>3</sub> Cr <sub>2</sub> O <sub>7</sub> . Solid State Sciences, 2007, 9, 564-573.	1.5	27
56	High-temperature neutron diffraction study of the cation ordered perovskites TbBaMn <sub>2</sub> O <sub>5+x</sub> and TbBaMn <sub>2</sub> O <sub>5.5</sub> . Journal of Solid State Chemistry, 2006, 179, 3505-3510.	1.4	16
57	Optical and morphological study of disorder in opals. Journal of Applied Physics, 2005, 97, 063502.	1.1	53
58	Optical and morphological study of compound polymer opals. , 2004, , .		0
59	Optical study of the pseudogap in thickness and orientation controlled artificial opals. Physical Review B, 2003, 68, .	1.1	188