

# Franz Peter Lang

## List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/6477537/publications.pdf>

Version: 2024-02-01

15

papers

227

citations

1478505

6

h-index

996975

15

g-index

15

all docs

15

docs citations

15

times ranked

271

citing authors

#	ARTICLE	IF	CITATIONS
1	Ionic radii for Group 1 and Group 2 halide, hydride, fluoride, oxide, sulfide, selenide and telluride crystals. <i>Dalton Transactions</i> , 2010, 39, 7786.	3.3	80
2	Ionization Energies of Atoms and Atomic Ions. <i>Journal of Chemical Education</i> , 2003, 80, 938.	2.3	38
3	Ionization Energies of Lanthanides. <i>Journal of Chemical Education</i> , 2010, 87, 875-881.	2.3	29
4	Electronegativity effects and single covalent bond lengths of molecules in the gas phase. <i>Dalton Transactions</i> , 2014, 43, 8016-8025.	3.3	27
5	An equation to calculate internuclear distances of covalent, ionic and metallic lattices. <i>Physical Chemistry Chemical Physics</i> , 2015, 17, 3355-3369.	2.8	17
6	Fermi energy, metals and the drift velocity of electrons. <i>Chemical Physics Letters</i> , 2021, 770, 138447.	2.6	8
7	An investigation into some important properties of transition metals (densities, resistivities,) Tj ETQq1 1 0.784314 rgBT /Overlock 10 690, 5-13.	2.6	5
8	Is a Metal à€œlons in a Sea of Delocalized Electrons?â€¢ <i>Journal of Chemical Education</i> , 2018, 95, 1787-1793.	2.3	5
9	Ionisation potentials of one-electron atoms. <i>Inorganic and Nuclear Chemistry Letters</i> , 1981, 17, 27-29.	0.7	3
10	A simple formula to calculate the ionization energies of two-, three-, and four-electron atomic ions. <i>Die Naturwissenschaften</i> , 2010, 97, 689-696.	1.6	3
11	Methods of Calculating Ionization Energies of Multielectron (Five or More) Isoelectronic Atomic Ions. <i>Scientific World Journal</i> , The, 2013, 2013, 1-10.	2.1	3
12	Relativistic Corrections for Calculating Ionization Energies of One- to Five-Electron Isoelectronic Atomic Ions. <i>ISRN Inorganic Chemistry</i> , 2013, 2013, 1-10.	0.2	3
13	Applying the soft sphere model to improve the understanding of bonding in transition metals. <i>Heliyon</i> , 2020, 6, e03110.	3.2	3
14	Ionisation energies of two-electron atoms. <i>Journal of the Chemical Society, Faraday Transactions 2</i> , 1984, 80, 1089.	1.1	2
15	Calculation of the Fermi energy and bulk modulus of metals. <i>Bulletin of Materials Science</i> , 2022, 45, .	1.7	1