

# Sarah A Saslow

## List of Publications by Year in descending order

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45  
papers

984  
citations

430874

18  
h-index

454955

30  
g-index

60  
all docs

60  
docs citations

60  
times ranked

1096  
citing authors

#	ARTICLE	IF	CITATIONS
1	Metallic technetium sequestration in nickel core/shell microstructure during Fe(OH) <sub>2</sub> transformation with Ni doping. <i>Journal of Hazardous Materials</i> , 2022, 425, 127779.	12.4	3
2	The behavior of iodine in stabilized granular activated carbon and silver mordenite in cementitious waste forms. <i>Journal of Environmental Radioactivity</i> , 2022, 244-245, 106824.	1.7	2
3	Effect of Temperature on Local Hydration of Zn in Hematite. <i>ACS Earth and Space Chemistry</i> , 2022, 6, 551-557.	2.7	6
4	The evolution of hydrated lime-based cementitious waste forms during leach testing leading to enhanced technetium retention. <i>Journal of Hazardous Materials</i> , 2022, 430, 128507.	12.4	4
5	Review and experimental comparison of the durability of iodine waste forms in semi-dynamic leach testing. <i>Chemical Engineering Journal Advances</i> , 2022, 11, 100300.	5.2	7
6	An exploration of benchtop X-ray emission spectroscopy for precise characterization of the sulfur redox state in cementitious materials. <i>X-Ray Spectrometry</i> , 2022, 51, 151-162.	1.4	2
7	Behavior of iodate substituted ettringite during aqueous leaching. <i>Applied Geochemistry</i> , 2021, 125, 104863.	3.0	6
8	Influences on Subsurface Plutonium and Americium Migration. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 279-294.	2.7	4
9	A Focused Ion Beam-Scanning Transmission Electron Microscopy with Energy-Dispersive X-ray Spectroscopy Study on Technetium Incorporation within Iron Oxides through Fe(OH) <sub>2</sub> (s) Mineral Transformation. <i>ACS Earth and Space Chemistry</i> , 2021, 5, 525-534.	2.7	5
10	Resolving Configurational Disorder for Impurities in a Low-Entropy Phase. <i>Journal of Physical Chemistry Letters</i> , 2021, 12, 5689-5694.	4.6	6
11	Micrometer-sized magnetite synthesis using Fe(OH) <sub>2</sub> (s) as a precursor for technetium sequestration from liquid nuclear waste streams. <i>Journal of Nuclear Materials</i> , 2021, 552, 152964.	2.7	2
12	Competitive TcO <sub>4</sub> <sup>2-</sup> , IO <sub>3</sub> <sup>-</sup> , and CrO <sub>4</sub> <sup>2-</sup> Incorporation into Ettringite. <i>Environmental Science &amp; Technology</i> , 2021, 55, 1057-1066.	10.0	11
13	Vanadium Oxidation States and Structural Role in Aluminoborosilicate Glasses: An Integrated Experimental and Molecular Dynamics Simulation Study. <i>Journal of Physical Chemistry B</i> , 2021, 125, 12365-12377.	2.6	8
14	Iodine immobilization by materials through sorption and redox-driven processes: A literature review. <i>Science of the Total Environment</i> , 2020, 716, 132820.	8.0	59
15	Technetium immobilization by materials through sorption and redox-driven processes: A literature review. <i>Science of the Total Environment</i> , 2020, 716, 132849.	8.0	19
16	Evaluation of materials for iodine and technetium immobilization through sorption and redox-driven processes. <i>Science of the Total Environment</i> , 2020, 716, 136167.	8.0	16
17	Impact of Cr and Co on <sup>99</sup> Tc retention in magnetite: A combined study of ab initio molecular dynamics and experiments. <i>Journal of Hazardous Materials</i> , 2020, 387, 121721.	12.4	3
18	Spectroscopic and first-principles investigations of iodine species incorporation into ettringite: Implications for iodine migration in cement waste forms. <i>Journal of Hazardous Materials</i> , 2020, 389, 121880.	12.4	39

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19	Immobilizing Pertechnetate in Ettringite via Sulfate Substitution. <i>Environmental Science &amp; Technology</i> , 2020, 54, 13610-13618.	10.0	20
20	Hybrid Sorbents for <sup>129</sup> I Capture from Contaminated Groundwater. <i>ACS Applied Materials &amp; Interfaces</i> , 2020, 12, 26113-26126.	8.0	19
21	Kinetics of Co-Mingled <sup>99</sup> Tc and Cr Removal during Mineral Transformation of Ferrous Hydroxide. <i>ACS Earth and Space Chemistry</i> , 2020, 4, 218-228.	2.7	5
22	Association of Defects and Zinc in Hematite. <i>Environmental Science &amp; Technology</i> , 2019, 53, 13687-13694.	10.0	20
23	Chromate Effect on Iodate Incorporation into Calcite. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 1624-1630.	2.7	16
24	Technetium and iodine aqueous species immobilization and transformations in the presence of strong reductants and calcite-forming solutions: Remedial action implications. <i>Science of the Total Environment</i> , 2018, 636, 588-595.	8.0	17
25	Incorporation Modes of Iodate in Calcite. <i>Environmental Science &amp; Technology</i> , 2018, 52, 5902-5910.	10.0	31
26	Cr(VI) Effect on Tc-99 Removal from Hanford Low-Activity Waste Simulant by Ferrous Hydroxide. <i>Environmental Science &amp; Technology</i> , 2018, 52, 11752-11759.	10.0	11
27	Characterizing Technetium in Subsurface Sediments for Contaminant Remediation. <i>ACS Earth and Space Chemistry</i> , 2018, 2, 1145-1160.	2.7	8
28	Facile incorporation of technetium into magnetite, magnesioferrite, and hematite by formation of ferrous nitrate <i>in situ</i> : precursors to iron oxide nuclear waste forms. <i>Dalton Transactions</i> , 2018, 47, 10229-10239.	3.3	15
29	Enhanced <sup>99</sup> Tc retention in glass waste form using Tc(IV)-incorporated Fe minerals. <i>Journal of Nuclear Materials</i> , 2017, 495, 455-462.	2.7	21
30	Experimental determination of partitioning in the Fe-Ni system for applications to modeling meteoritic metals. <i>Meteoritics and Planetary Science</i> , 2017, 52, 1133-1145.	1.6	34
31	Aqueous Synthesis of Technetium-Doped Titanium Dioxide by Direct Oxidation of Titanium Powder, a Precursor for Ceramic Nuclear Waste Forms. <i>Chemistry of Materials</i> , 2017, 29, 10369-10376.	6.7	12
32	Reduction and Simultaneous Removal of <sup>99</sup> Tc and Cr by Fe(OH) <sub>2</sub> (s) Mineral Transformation. <i>Environmental Science &amp; Technology</i> , 2017, 51, 8635-8642.	10.0	68
33	Phase-referenced nonlinear spectroscopy of the $\hat{\pm}$ -quartz/water interface. <i>Nature Communications</i> , 2016, 7, 13587.	12.8	130
34	A General Chemistry Assignment Analyzing Environmental Contamination for the DePue, IL, National Superfund Site. <i>Journal of Chemical Education</i> , 2015, 92, 638-642.	2.3	4
35	Precipitates of Al(III), Sc(III), and La(III) at the Muscovite-Water Interface. <i>Journal of Physical Chemistry A</i> , 2014, 118, 10974-10981.	2.5	11
36	Zinc Ion-Hydroxyl Interactions at Undecanol-Functionalized Fused Silica/Water Interfaces Using the Eisenthal $\hat{\pm}$ (3) Technique. <i>Journal of Physical Chemistry C</i> , 2012, 116, 7016-7020.	3.1	15

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37	Uranyl Adsorption at the Muscovite (Mica)/Water Interface Studied by Second Harmonic Generation. <i>Environmental Science &amp; Technology</i> , 2012, 46, 11154-11161.	10.0	25
38	U(VI) Adsorption and Speciation at the Acidic Silica/Water Interface Studied by Resonant and Nonresonant Second Harmonic Generation. <i>Journal of Physical Chemistry C</i> , 2011, 115, 13353-13360.	3.1	24
39	Exponential Sensitivity and Speciation of Al(III), Sc(III), Y(III), La(III), and Gd(III) at Fused Silica/Water Interfaces. <i>Journal of Physical Chemistry A</i> , 2011, 115, 14438-14445.	2.5	19
40	Partitioning behavior at 9GPa in the Fe-S system and implications for planetary evolution. <i>Earth and Planetary Science Letters</i> , 2011, 305, 425-434.	4.4	16
41	The deep water abundance on Jupiter: New constraints from thermochemical kinetics and diffusion modeling. <i>Icarus</i> , 2010, 209, 602-615.	2.5	78
42	The iron-nickel-phosphorus system: Effects on the distribution of trace elements during the evolution of iron meteorites. <i>Geochimica Et Cosmochimica Acta</i> , 2009, 73, 2674-2691.	3.9	35
43	An investigation of the behavior of Cu and Cr during iron meteorite crystallization. <i>Meteoritics and Planetary Science</i> , 2009, 44, 505-519.	1.6	34
44	The Fe-C system at 5GPa and implications for Earth's core. <i>Geochimica Et Cosmochimica Acta</i> , 2008, 72, 4146-4158.	3.9	48
45	The effect of Ni on element partitioning during iron meteorite crystallization. <i>Meteoritics and Planetary Science</i> , 2007, 42, 1735-1750.	1.6	26