

Olga Lobacheva

List of Publications by Citations

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The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

25
papers

127
citations

7
h-index

10
g-index

27
ext. papers

135
ext. citations

1.2
avg, IF

2.75
L-index

#	Paper	IF	Citations
25	Recovery and separation of Ce ³⁺ and Y ³⁺ ions from aqueous solutions by ion flotation. <i>Russian Journal of Applied Chemistry</i> , 2009 , 82, 1370-1374	0.8	18
24	Ion flotation of lanthanum(III) and holmium(III) from nitrate and nitrate-chloride media. <i>Russian Journal of Applied Chemistry</i> , 2012 , 85, 25-28	0.8	16
23	The thermodynamic properties of hydroxo compounds and the mechanism of ion flotation for cerium, europium, and yttrium. <i>Russian Journal of Physical Chemistry A</i> , 2009 , 83, 2022-2027	0.7	12
22	THE METHOD OF REMOVAL YTTRIUM (III) AND YTTERBIUM (III) FROM DILUTE AQUEOUS SOLUTIONS. <i>Journal of Ecological Engineering</i> , 2016 , 17, 38-42	2	10
21	Rare-Earth Elements Recovery on the Example of Europium (III) from Lean Technogenic Raw Materials. <i>Journal of Ecological Engineering</i> , 2017 , 18, 122-126	2	10
20	Gibbs energies of formation of hydroxides of lanthanides and yttrium. <i>Russian Journal of Physical Chemistry A</i> , 2010 , 84, 2047-2050	0.7	9
19	Synthesis of aluminum-based scandium-yttrium master alloys. <i>Russian Metallurgy (Metally)</i> , 2015 , 2015, 516-520	0.5	7
18	Ion flotation of rare-earth metals with sodium dodecyl sulfate. <i>Russian Journal of Applied Chemistry</i> , 2011 , 84, 1476-1482	0.8	7
17	Ion flotation of cations of cerium group. <i>Russian Journal of Applied Chemistry</i> , 2013 , 86, 1862-1866	0.8	5
16	Solvent sublation and ion flotation in aqueous salt solutions containing Ce(III) and Y(III) in the presence of a surfactant. <i>Russian Journal of Applied Chemistry</i> , 2014 , 87, 1863-1867	0.8	5
15	Effect of chlorides on cerium(III) and samarium(III) ionic flotation. <i>Russian Journal of Applied Chemistry</i> , 2011 , 84, 341-344	0.8	5
14	Foam Separation of Nickel and Copper Ions from Dilute Aqueous Solutions. <i>Russian Journal of Applied Chemistry</i> , 2005 , 78, 1869-1871	0.8	5
13	Experimental Liquid-Liquid Equilibrium and Solubility Study of an Acetic Acid-Propyl Alcohol-Propyl Acetate-Water System at 323.15 and 333.15 K. <i>Journal of Chemical & Engineering Data</i> , 2020 , 65, 5352-5359	2.8	4
12	Thermodynamic properties of lanthanum(III) and holmium(III) hydroxo compounds. <i>Russian Journal of Physical Chemistry A</i> , 2011 , 85, 1872-1875	0.7	3
11	Thermodynamics of complexation in an aqueous solution of Tb(III) nitrate at 298 K. <i>Russian Journal of Physical Chemistry A</i> , 2017 , 91, 67-69	0.7	2
10	Sorptive separation of yttrium and cerium on a weakly basic anionite. <i>Russian Journal of Physical Chemistry A</i> , 2015 , 89, 119-124	0.7	2
9	Thermodynamics of formation of lanthanide hydroxo complexes in aqueous solutions. <i>Russian Journal of Inorganic Chemistry</i> , 2012 , 57, 605-609	1.5	2

8	Solvent sublation of cerium ions from dilute aqueous solutions. <i>Russian Chemical Bulletin</i> , 2012 , 61, 962-965	1.9	1
7	Foam Separation of Nickel. <i>Mendeleev Communications</i> , 1994 , 4, 215-216	1.9	1
6	Solvent extraction of europium(III) from technogenic solutions with the use of surfactants. <i>Fine Chemical Technologies</i> , 2020 , 15, 51-58	0.5	1
5	Reduction in Technogenic Burden on the Environment by Flotation Recovery of Rare Earth Elements from Diluted Industrial Solutions. <i>Applied Sciences (Switzerland)</i> , 2021 , 11, 7452	2.6	1
4	The Experimental Study of Innovative Methods Regarding the Removal of Sm(III). <i>Applied Sciences (Switzerland)</i> , 2021 , 11, 7726	2.6	1
3	Ion Flotation of Ytterbium Water-Salt Systems—An Innovative Aspect of the Modern Industry. <i>Water (Switzerland)</i> , 2021 , 13, 3493	3	0
2	Research of extraction in system Ln(NO ₃) ₃ -sodium dodecyl sulfate-isoctyl alcohol. <i>Russian Journal of Applied Chemistry</i> , 2013 , 86, 293-296	0.8	
1	Solvent sublation of yttrium ions from dilute aqueous solutions by use of sodium dodecyl sulfate. <i>Russian Journal of Applied Chemistry</i> , 2012 , 85, 1153-1156	0.8	