## Nicolae Viorel Pavel

List of Publications by Year in descending order

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111 papers 3,606 citations

35 h-index 55 g-index

113 all docs

113 does citations

113 times ranked 2780 citing authors

#	Article	IF	CITATIONS
1	Hydrogen and Higher Shell Contributions in Zn2+, Ni2+, and Co2+Aqueous Solutions:Â An X-ray Absorption Fine Structure and Molecular Dynamics Study. Journal of the American Chemical Society, 2002, 124, 1958-1967.	6.6	175
2	Size and shape of sodium deoxycholate micellar aggregates. The Journal of Physical Chemistry, 1987, 91, 356-362.	2.9	141
3	An extended xâ€ray absorption fine structure study of aqueous solutions by employing molecular dynamics simulations. Journal of Chemical Physics, 1994, 100, 985-994.	1.2	133
4	Evidence of distorted fivefold coordination of the Cu2+aqua ion from an x-ray-absorption spectroscopy quantitative analysis. Physical Review B, 2002, 65, .	1.1	131
5	Double-electron excitation channels at the BrKedge of HBr andBr2. Physical Review A, 1993, 47, 2055-2063.	1.0	101
6	Nuclear magnetic resonance and x-ray studies on micellar aggregates of sodium deoxycholate. The Journal of Physical Chemistry, 1984, 88, 5720-5724.	2.9	100
7	Development and Validation of an Integrated Computational Approach for the Study of Ionic Species in Solution by Means of Effective Two-Body Potentials. The Case of Zn2 +, Ni2 +, and Co2 +in Aqueous Solutions. Journal of the American Chemical Society, 2002, 124, 1968-1976.	6.6	92
8	Combined XANES and EXAFS analysis of Co2+, Ni2+, and Zn2+aqueous solutions. Physical Review B, 2002, 66, .	1.1	88
9	Bile salts and derivatives: Rigid unconventional amphiphiles as dispersants, carriers and superstructure building blocks. Current Opinion in Colloid and Interface Science, 2015, 20, 170-182.	3.4	87
10	Evidence for Sevenfold Coordination in the First Solvation Shell of Hg(II) Aqua Ion. Journal of the American Chemical Society, 2007, 129, 5430-5436.	6.6	78
11	Supramolecular Structures Generated by ap-tert-Butylphenyl-amide Derivative of Cholic Acid: From Vesicles to Molecular Tubes. Advanced Materials, 2007, 19, 1752-1756.	11.1	78
12	Multiple-scattering effects in the K-edge x-ray-absorption near-edge structure of crystalline and amorphous silicon. Physical Review B, 1987, 36, 6426-6433.	1.1	73
13	EXAFS: a new approach to the structure of micellar aggregates. The Journal of Physical Chemistry, 1988, 92, 2858-2862.	2.9	70
14	About the albumin structure in solution: cigar Expanded form versus heart Normal shape. Physical Chemistry Chemical Physics, 2008, 10, 6741.	1.3	70
15	Bile Salts: Natural Surfactants and Precursors of a Broad Family of Complex Amphiphiles. Langmuir, 2019, 35, 6803-6821.	1.6	64
16	Triplet correlations in the hydration shell of aquaions. Chemical Physics Letters, 1994, 225, 150-155.	1.2	59
17	Structural investigation of copper(II) chloride solutions using x-ray absorption spectroscopy. Journal of Chemical Physics, 1997, 107, 2807-2812.	1.2	59
18	Evidence for multielectron resonances at the SrKedge. Physical Review A, 1996, 53, 798-805.	1.0	57

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19	Catanionic Tubules with Tunable Charge. Angewandte Chemie - International Edition, 2010, 49, 6604-6607.	7.2	55
20	X-ray Absorption Study of Copper(II)â^'Glycinate Complexes in Aqueous Solution. Journal of Physical Chemistry B, 1998, 102, 3114-3122.	1.2	49
21	From crystal to micelle: A new approach to the micellar structure. Journal of Inclusion Phenomena and Macrocyclic Chemistry, 1989, 7, 391-400.	1.6	48
22	Multielectron excitations at theLedges of barium in aqueous solution. Physical Review B, 1996, 54, 12129-12138.	1.1	48
23	An extended xâ€ray absorption fine structure study by employing molecular dynamics simulations: Bromide ion in methanolic solution. Journal of Chemical Physics, 1996, 104, 1779-1790.	1.2	48
24	Synthesis and Characterization of a New Gemini Surfactant Derived from $3\hat{1}\pm,12\hat{1}\pm-Dihydroxy-5\hat{1}^2$ -cholan-24-amine (Steroid Residue) and Ethylenediamintetraacetic Acid (Spacer). Langmuir, 2008, 24, 6060-6066.	1.6	47
25	Urea-Induced Denaturation Process on Defatted Human Serum Albumin and in the Presence of Palmitic Acid. Journal of Physical Chemistry B, 2009, 113, 12590-12602.	1.2	46
26	Kinetics of formation of supramolecular tubules of a sodium cholate derivative. Soft Matter, 2009, 5, 3018.	1.2	46
27	Multiple scattering x-ray absorption analysis of simple brominated hydrocarbon molecules. The Journal of Physical Chemistry, 1993, 97, 5486-5494.	2.9	44
28	New Lamellar Structure Formed by an Adamantyl Derivative of Cholic Acid. Journal of Physical Chemistry B, 2006, 110, 13679-13681.	1.2	43
29	An Integrated Study of Small-Angle X-ray Scattering and Dynamic Light Scattering on Cylindrical Micelles of Sodium Glycodeoxycholate. Journal of Physical Chemistry B, 2004, 108, 3078-3085.	1.2	41
30	Human serum albumin binding ibuprofen: A 3D description of the unfolding pathway in urea. Biophysical Chemistry, 2010, 147, 111-122.	1.5	40
31	Intermolecular interactions in sodium deoxycholate micelles: an NMR study involving a spin-labeled cholestane. The Journal of Physical Chemistry, 1987, 91, 83-89.	2.9	39
32	An x-ray and conformational study of Kapton H. Journal of Polymer Science, Polymer Physics Edition, 1976, 14, 1553-1560.	1.0	38
33	Structural Resolution of the Complex between a Fungal Polygalacturonase and a Plant Polygalacturonase-Inhibiting Protein by Small-Angle X-Ray Scattering Â. Plant Physiology, 2011, 157, 599-607.	2.3	38
34	Study of Intermicellar Interactions and Micellar Sizes in Ionic Micelle Solutions by Comparing Collective Diffusion and Self-Diffusion Coefficients. Journal of Physical Chemistry B, 2004, 108, 4799-4805.	1.2	37
35	pH sensitive tubules of a bile acid derivative: a tubule opening by release of wall leaves. Physical Chemistry Chemical Physics, 2013, 15, 7560.	1.3	37
36	Tailoring Supramolecular Nanotubes by Bile Salt Based Surfactant Mixtures. Angewandte Chemie - International Edition, 2015, 54, 7018-7021.	7.2	37

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37	Characterization of Carbon Nanotube Dispersions in Solutions of Bile Salts and Derivatives Containing Aromatic Substituents. Journal of Physical Chemistry B, 2014, 118, 1012-1021.	1.2	35
38	The crystal structure of the inclusion compound between cycloveratril, benzene and water. Acta Crystallographica Section B: Structural Crystallography and Crystal Chemistry, 1979, 35, 2605-2609.	0.4	34
39	Amino acid–bile acid based molecules: extremely narrow surfactant nanotubes formed by a phenylalanine-substituted cholic acid. Chemical Communications, 2012, 48, 12011.	2.2	34
40	Nanoparticles with a Bicontinuous Cubic Internal Structure Formed by Cationic and Non-ionic Surfactants and an Anionic Polyelectrolyte. Langmuir, 2012, 28, 16536-16546.	1.6	34
41	Between Peptides and Bile Acids: Self-Assembly of Phenylalanine Substituted Cholic Acids. Journal of Physical Chemistry B, 2013, 117, 9248-9257.	1.2	33
42	Catanionic Gels Based on Cholic Acid Derivatives. Langmuir, 2013, 29, 12342-12351.	1.6	33
43	The molecular and crystal structure of an allylpalladium- (II) triazenido complex: [(1-3-ÎC3H5)Pd(II)(p-CH3C6H4NNNC6H4CH3-p)]2. Journal of Organometallic Chemistry, 1976, 108, 409-421.	0.8	32
44	EXAFS study of probe molecules in micellar solutions. The Journal of Physical Chemistry, 1991, 95, 7880-7886.	2.9	32
45	Small-Angle X-ray Scattering and Light Scattering on Lysozyme and Sodium Glycocholate Micelles. Journal of Physical Chemistry B, 2005, 109, 23857-23869.	1.2	32
46	Human Serum Albumin Unfolding: A Small-Angle X-ray Scattering and Light Scattering Study. Journal of Physical Chemistry B, 2008, 112, 15460-15469.	1.2	32
47	Double-Electron Excitation Channels at the Ca2+ K-Edge of Hydrated Calcium Ion. Journal of Physical Chemistry B, 2004, 108, 11857-11865.	1.2	28
48	Sugar–Bile Acid-Based Bolaamphiphiles: From Scrolls to Monodisperse Single-Walled Tubules. Langmuir, 2014, 30, 6358-6366.	1.6	27
49	Multi stimuli response of a single surfactant presenting a rich self-assembly behavior. RSC Advances, 2015, 5, 37800-37806.	1.7	27
50	EXAFS Study of Micellar Aggregates of Bile Acid Rubidium Salts. The Journal of Physical Chemistry, 1995, 99, 5471-5480.	2.9	26
51	<i>Arabidopsis</i> and <i>Chlamydomonas</i> phosphoribulokinase crystal structures complete the redox structural proteome of the Calvin–Benson cycle. Proceedings of the National Academy of Sciences of the United States of America, 2019, 116, 8048-8053.	3.3	25
52	Evidence of three-body correlation functions in Rb+ and Sr2+ acetonitrile solutions. Journal of Chemical Physics, 1999, 111, 5107-5115.	1.2	24
53	Formation of tubules by p-tert-butylphenylamide derivatives of chenodeoxycholic and ursodeoxycholic acids in aqueous solution. Steroids, 2012, 77, 1205-1211.	0.8	23
54	A Single Amino-Acid Substitution Allows Endo-Polygalacturonase of Fusarium verticillioides to Acquire Recognition by PGIP2 from Phaseolus vulgaris. PLoS ONE, 2013, 8, e80610.	1.1	23

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55	Collective diffusion and self-diffusion coefficients comparison to separate interactions and micellar size effects on ionic micelle diffusivities: Cylindrical micelles of sodium taurodeoxycholate. Journal of Chemical Physics, 2003, 118, 2865.	1.2	22
56	Study on the Structure of Hostâ^'Guest Supramolecular Polymers. Macromolecules, 2007, 40, 5899-5906.	2.2	22
57	Structure of Sodium Glycodeoxycholate Micellar Aggregates from Small-Angle X-ray Scattering and Light-Scattering Techniques. Journal of Physical Chemistry B, 2006, 110, 12351-12359.	1.2	20
58	Supramolecular Structures Generated by a <i>p</i> - <i>tert</i> -Butylphenylamide Derivative of Deoxycholic Acid. From Planar Sheets to Tubular Structures through Helical Ribbons. Langmuir, 2010, 26, 7768-7773.	1.6	20
59	Time-Dependent pH Scanning of the Acid-Induced Unfolding of Human Serum Albumin Reveals Stabilization of the Native Form by Palmitic Acid Binding. Journal of Physical Chemistry B, 2017, 121, 4388-4399.	1.2	20
60	On the self-assembly of a tryptophan labeled deoxycholic acid. Physical Chemistry Chemical Physics, 2014, 16, 19492.	1.3	19
61	Interaction between bile salt sodium glycodeoxycholate and PEO–PPO–PEO triblock copolymers in aqueous solution. RSC Advances, 2016, 6, 69313-69325.	1.7	19
62	Wormlike reverse micelles in lecithin/bile salt/water mixtures in oil. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2017, 532, 411-419.	2.3	19
63	Supramolecular assembly of a thermoresponsive steroidal surfactant with an oppositely charged thermoresponsive block copolymer. Physical Chemistry Chemical Physics, 2017, 19, 1504-1515.	1.3	19
64	Drug-loaded nanoparticles and supramolecular nanotubes formed from a volatile microemulsion with bile salt derivatives. Physical Chemistry Chemical Physics, 2013, 15, 6016.	1.3	18
65	Block copolymers as bile salt sequestrants: intriguing structures formed in a mixture of an oppositely charged amphiphilic block copolymer and bile salt. Physical Chemistry Chemical Physics, 2019, 21, 12518-12529.	1.3	18
66	K- and L-edge XAFS determination of the local structure of aqueous Nd(III) and Eu(III). Journal of Synchrotron Radiation, 2001, 8, 666-668.	1.0	17
67	Bile acid derivative-based catanionic mixtures: versatile tools for superficial charge modulation of supramolecular lamellae and nanotubes. Physical Chemistry Chemical Physics, 2018, 20, 18957-18968.	1.3	17
68	Sodium Taurodeoxycholate Structure from Solid to Liquid Phase. Langmuir, 2002, 18, 2812-2816.	1.6	16
69	Spherical wave exafs analysis of the silicon K-edge X-ray absorption spectrum. Solid State Communications, 1987, 61, 635-639.	0.9	15
70	Minimization programs for potential energy calculations in crystals and isolated molecules and macromolecules. Zeitschrift Fýr Kristallographie, 1976, 144, 64-75.	1.1	14
71	Structure of the 4:1 inclusion compound between deoxycholic acid and (E)-p-dimethylaminoazobenzene. Acta Crystallographica Section B: Structural Crystallography and Crystal Chemistry, 1982, 38, 2615-2620.	0.4	14
72	Early Stages of Formation of Branched Hostâ^'Guest Supramolecular Polymers. Journal of Physical Chemistry B, 2008, 112, 8536-8541.	1.2	14

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73	Correlation between Small-Angle X-ray Scattering Spectra and Apparent Diffusion Coefficients in the Study of Structure and Interaction of Sodium Taurodeoxycholate Micelles. Journal of Physical Chemistry B, 2005, 109, 6111-6120.	1.2	13
74	Aggregation Behavior of Tetracarboxylic Surfactants Derived from Cholic and Deoxycholic Acids and Ethylenediaminetetraacetic Acid. Langmuir, 2009, 25, 9037-9044.	1.6	13
75	Unravelling the shape and structural assembly of the photosynthetic GAPDH–CP12–PRK complex from <i>Arabidopsis thaliana</i> by small-angle X-ray scattering analysis. Acta Crystallographica Section D: Biological Crystallography, 2015, 71, 2372-2385.	2.5	13
76	Structural Response of Human Serum Albumin to Oxidation: Biological Buffer to Local Formation of Hypochlorite. Journal of Physical Chemistry B, 2016, 120, 12261-12271.	1.2	13
77	On the stability of lithocholate derivative supramolecular tubules. RSC Advances, 2017, 7, 512-517.	1.7	13
78	Potential energy calculations of ethylene–butadiene copolymers. Journal of Polymer Science: Polymer Chemistry Edition, 1975, 13, 125-131.	0.8	12
79	Crystal structure and van der Waals energy study of the 2:1 inclusion compound between deoxycholic acid and norbornadiene. Acta Crystallographica Section B: Structural Crystallography and Crystal Chemistry, 1981, 37, 368-372.	0.4	12
80	Diffusivity Study of Dihydroxyâ^Trihydroxy Bile Salt Systems. Langmuir, 2003, 19, 1319-1323.	1.6	12
81	Ibuprofen and Propofol Cobinding Effect on Human Serum Albumin Unfolding in Urea. Journal of Physical Chemistry B, 2014, 118, 10043-10051.	1.2	11
82	Twisted nanoribbons from a RGD-bearing cholic acid derivative. Colloids and Surfaces B: Biointerfaces, 2017, 159, 183-190.	2.5	11
83	A Stereochemically Driven Supramolecular Polymerisation. Chemistry - A European Journal, 2018, 24, 8195-8204.	1.7	11
84	Chain folding in crystalline syndiotactic poly(vinyl chloride). Journal of Polymer Science, Polymer Physics Edition, 1979, 17, 753-762.	1.0	10
85	XAS Study of Solubilization Loci of Brominated Molecules in Aqueous Micellar Solutions. The Journal of Physical Chemistry, 1994, 98, 2982-2990.	2.9	10
86	EXAFS and molecular dynamics studies of ionic solutions. Journal of Synchrotron Radiation, 2001, 8, 173-177.	1.0	10
87	Complete spectrum of multielectron excitations at the Br- K edge x-ray absorption spectra. Physical Review B, 2001, 64, .	1.1	10
88	QELS and X-ray study of two dihydroxy bile salt aqueous solutions. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2004, 248, 79-84.	2.3	10
89	Sodium Glycodeoxycholate and Glycocholate Mixed Aggregates in Gas and Solution Phases. Journal of Physical Chemistry B, 2009, 113, 7162-7169.	1.2	10
90	A tryptophan-substituted cholic acid: Expanding the family of labelled biomolecules. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2015, 483, 142-149.	2.3	9

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91	Generation and best fitting of molecular models. Acta Crystallographica Section B: Structural Crystallography and Crystal Chemistry, 1972, 28, 1968-1969.	0.4	8
92	Crystal structure of the 2:1 inclusion compound between deoxycholic acid and quadricyclane. Journal of Inclusion Phenomena, 1984, 1, 329-337.	0.6	8
93	Measurement of x-ray multielectron photoexcitations at the <mml:math display="inline" xmlns:mml="http://www.w3.org/1998/Math/MathML"><mml:mrow><mml:msup><mml:mtext>I</mml:mtext><mml:mo>â^'</mml:mo></mml:msup><mr 2008.="" 78<="" b.="" physical="" review="" td=""><td>nl:mtext&gt;</td><td> </td></mr></mml:mrow></mml:math>	nl:mtext>	
94	Crystal structure of head-to-head dimers of cholic and deoxycholic acid derivatives with different symmetric bridges. Steroids, 2013, 78, 247-254.	0.8	8
95	Effect of temperature on the association behavior in aqueous mixtures of an oppositely charged amphiphilic block copolymer and bile salt. Polymer, 2020, 206, 122871.	1.8	8
96	Tailoring Supramolecular Nanotubes by Bile Salt Based Surfactant Mixtures. Angewandte Chemie, 2015, 127, 7124-7127.	1.6	7
97	Self-Assembly of Model Amphiphilic Peptides in Nonaqueous Solvents: Changing the Driving Force for Aggregation Does Not Change the Fibril Structure. Langmuir, 2020, 36, 8451-8460.	1.6	7
98	Polymorphic Self-Organization of Lauroyl Peptide in Response to pH and Concentration. Langmuir, 2020, 36, 3941-3951.	1.6	7
99	Comment on "Two-Dimensional NMR Study on the Structures of Micelles of Sodium Taurocholateâ€. Journal of Physical Chemistry B, 2005, 109, 9849-9850.	1.2	6
100	Potential energy calculations about the chain folding of $\hat{l}_{\pm}(\hat{a}^{*})$ poly(L-alanine). Polymer, 1976, 17, 257-259.	1.8	5
101	Determination of two- and three-body correlation functions in ionic solutions by means of MD and EXAFS investigations. Journal of Synchrotron Radiation, 1999, 6, 281-283.	1.0	5
102	Formation of host-guest and sandwich complexes by a $\hat{l}^2$ -cyclodextrin derivative. Journal of Inclusion Phenomena and Macrocyclic Chemistry, 2011, 69, 245-253.	1.6	5
103	Physicochemical study of crystals of ethylene–butadiene copolymers. Journal of Polymer Science: Polymer Chemistry Edition, 1978, 16, 115-127.	0.8	4
104	Chain folding in single crystals of polytetrafluoroethylene. Journal of Polymer Science, Polymer Physics Edition, 1983, 21, 321-328.	1.0	4
105	The structure of the 2/1 'channel' inclusion compound between deoxycholic acid and pinacolone, 2C24H40O4.C6H12O. Acta Crystallographica Section C: Crystal Structure Communications, 1985, 41, 229-232.	0.4	4
106	Crystal structure of a lithium salt of a glucosyl derivative of lithocholic acid. Steroids, 2016, 113, 87-94.	0.8	4
107	Possible models for the polyethylene hexagonal phase. Polymer, 1980, 21, 973-974.	1.8	3
108	The effect of fatty acid binding in the acid isomerizations of albumin investigated with a continuous acidification method. Colloids and Surfaces B: Biointerfaces, 2018, 168, 109-116.	2.5	3

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109	SPHERICAL WAVES EXAFS AND MULTIPLE SCATTERING EFFECTS IN XANES OF THE K-EDGE SPECTRUM OF SILICON. Journal De Physique Colloque, 1986, 47, C8-71-C8-74.	0.2	3
110	Study of the 4:1 inclusion compound between deoxycholic acid and (E)-p-dimethylaminoazobenzene by vapour pressure measurements. Thermochimica Acta, 1985, 87, 231-238.	1.2	2
111	Differential anomalous scattering study of probe molecules. Nuclear Instruments & Methods in Physics Research B, 1995, 97, 539-542.	0.6	0