

# Heon Kang

## List of Publications by Year in descending order

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57  
papers

1,135  
citations

394421

19  
h-index

454955

30  
g-index

59  
all docs

59  
docs citations

59  
times ranked

786  
citing authors

#	ARTICLE	IF	CITATIONS
1	Chemistry of Ice Surfaces. Elementary Reaction Steps on Ice Studied by Reactive Ion Scattering. <i>Accounts of Chemical Research</i> , 2005, 38, 893-900.	15.6	86
2	FORMATION OF GLYCINE ON ULTRAVIOLET-IRRADIATED INTERSTELLAR ICE-ANALOG FILMS AND IMPLICATIONS FOR INTERSTELLAR AMINO ACIDS. <i>Astrophysical Journal</i> , 2009, 697, 428-435.	4.5	66
3	H/D isotopic exchange between water molecules at ice surfaces. <i>Journal of Chemical Physics</i> , 2004, 121, 2765.	3.0	59
4	Some fundamental properties and reactions of ice surfaces at low temperatures. <i>Physical Chemistry Chemical Physics</i> , 2010, 12, 12000.	2.8	59
5	Adsorption, Ionization, and Migration of Hydrogen Chloride on Ice Films at Temperatures between 100 and 140 K. <i>Journal of Physical Chemistry B</i> , 2005, 109, 5124-5132.	2.6	49
6	Mechanistic study of proton transfer and H <sup>+</sup> /D exchange in ice films at low temperatures (100–140K). <i>Journal of Chemical Physics</i> , 2007, 127, 084701.	3.0	47
7	Protons at Ice Surfaces. <i>Angewandte Chemie - International Edition</i> , 2006, 45, 5529-5533.	13.8	38
8	DIRECT EVIDENCE FOR AMMONIUM ION FORMATION IN ICE THROUGH ULTRAVIOLET-INDUCED ACID-BASE REACTION OF NH <sub>3</sub> WITH H <sub>3</sub> O <sup>+</sup> . <i>Astrophysical Journal</i> , 2010, 713, 906-911.	4.5	34
9	Generation of strong electric fields in an ice film capacitor. <i>Journal of Chemical Physics</i> , 2013, 139, 074201.	3.0	34
10	Reactive Ion Scattering of Low Energy Cs <sup>+</sup> from Surfaces. A Technique for Surface Molecular Analysis. <i>Bulletin of the Korean Chemical Society</i> , 2011, 32, 389-398.	1.9	30
11	Solvation and Reaction of Ammonia in Molecularly Thin Water Films. <i>Journal of Physical Chemistry C</i> , 2015, 119, 23052-23058.	3.1	28
12	Proton mobility in thin ice films: a revisit. <i>Physical Chemistry Chemical Physics</i> , 2008, 10, 4814.	2.8	27
13	Interaction of NaF, NaCl, and NaBr with Amorphous Ice Films. Salt Dissolution and Ion Separation at the Ice Surface. <i>Journal of Physical Chemistry C</i> , 2007, 111, 8030-8036.	3.1	25
14	Energy barrier of proton transfer at ice surfaces. <i>Journal of Chemical Physics</i> , 2010, 133, 044709.	3.0	25
15	Asymmetric Transport Efficiencies of Positive and Negative Ion Defects in Amorphous Ice. <i>Physical Review Letters</i> , 2012, 108, 226103.	7.8	25
16	Effect of Electric Field on Condensed-Phase Molecular Systems. II. Stark Effect on the Hydroxyl Stretch Vibration of Ice. <i>Journal of Physical Chemistry C</i> , 2015, 119, 15596-15603.	3.1	25
17	Zundel-like and Eigen-like Hydrated Protons on a Platinum Surface. <i>Angewandte Chemie - International Edition</i> , 2015, 54, 7626-7630.	13.8	24
18	Effect of Electric Field on Condensed-Phase Molecular Systems. I. Dipolar Polarization of Amorphous Solid Acetone. <i>Journal of Physical Chemistry C</i> , 2015, 119, 15588-15595.	3.1	22

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19	Electric Field Effect on Condensed-Phase Molecular Systems. III. The Origin of the Field-Induced Change in the Vibrational Frequency of Adsorbed CO on Pt(111). <i>Journal of Physical Chemistry C</i> , 2016, 120, 17579-17587.	3.1	21
20	Asymmetric Transport Mechanisms of Hydronium and Hydroxide Ions in Amorphous Solid Water: Hydroxide Goes Brownian while Hydronium Hops. <i>Journal of Physical Chemistry Letters</i> , 2014, 5, 2568-2572.	4.6	19
21	Hydrolysis of Sodium Atoms on Water/Ice Films. Characterization of Reaction Products and Interfacial Distribution of Sodium and Hydroxide Ions. <i>Journal of Physical Chemistry C</i> , 2009, 113, 321-327.	3.1	18
22	Comparative Proton Transfer Efficiencies of Hydronium and Hydroxide in Aqueous Solution: Proton Transfer vs Brownian Motion. <i>Journal of Physical Chemistry B</i> , 2014, 118, 13671-13678.	2.6	18
23	Efficient Conversion of Nitrogen Dioxide into Nitrous Acid on Ice Surfaces. <i>Journal of Physical Chemistry Letters</i> , 2010, 1, 3085-3089.	4.6	17
24	Brute Force Orientation of Matrix-Isolated Molecules: Reversible Reorientation of Formaldehyde in an Argon Matrix toward Perfect Alignment. <i>Angewandte Chemie - International Edition</i> , 2017, 56, 1046-1049.	13.8	17
25	Adsorption structure of 2-butyne on Si(100)-(2 $\times$ 1). <i>Journal of Chemical Physics</i> , 2003, 118, 6083-6088.	3.0	16
26	Segregation of hydroxide ions to an ice surface. <i>Journal of Chemical Physics</i> , 2011, 135, 074703.	3.0	16
27	Transport and Surface Accumulation of Hydroniums and Chlorides in an Ice Film. A High Temperature (140-180 K) Study. <i>Journal of Physical Chemistry C</i> , 2012, 116, 21828-21835.	3.1	15
28	Real-Space Investigation of Electrical Double Layers. Potential Gradient Measurement with a Nanometer Potential Probe. <i>Journal of Physical Chemistry C</i> , 2011, 115, 17384-17391.	3.1	14
29	Electric Field Effect on Condensed-Phase Molecular Systems. VI. Field-Driven Orientation of Hydrogen Chloride in an Argon Matrix. <i>Journal of Physical Chemistry A</i> , 2018, 122, 2871-2876.	2.5	14
30	The frequency-domain infrared spectrum of ammonia encodes changes in molecular dynamics caused by a DC electric field. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2019, 116, 23444-23447.	7.1	14
31	UV-induced protonation of molecules adsorbed on ice surfaces at low temperature. <i>Journal of Chemical Physics</i> , 2008, 128, 191101.	3.0	13
32	Proton transfer and H/D isotopic exchange of water molecules mediated by hydroxide ions on ice film surfaces. <i>Journal of Chemical Physics</i> , 2009, 131, 044705.	3.0	13
33	Kinetic Isolation of Reaction Intermediates on Ice Surfaces. Precursor States of SO <sub>2</sub> Hydrolysis. <i>Journal of Physical Chemistry C</i> , 2009, 113, 16863-16865.	3.1	13
34	Comment on "HCl adsorption on ice at low temperature: a combined X-ray absorption, photoemission and infrared study" by P. Parent, J. Lasne, G. Marcotte and C. Laffon, <i>Phys. Chem. Chem. Phys.</i> , 2011, 13, 7142. <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 1048-1049.	2.8	13
35	The Nature of Hydrated Protons on Platinum Surfaces. <i>Chemistry - A European Journal</i> , 2017, 23, 17566-17575.	3.3	13
36	Acid-Promoted Crystallization of Amorphous Solid Water. <i>Journal of Physical Chemistry C</i> , 2018, 122, 24164-24170.	3.1	13

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37	Electric Field Effect on Condensed-Phase Molecular Systems. IV. Conformational Change of 1,2-Dichloroethane in a Frozen Molecular Solid. <i>Journal of Physical Chemistry C</i> , 2017, 121, 25342-25346.	3.1	12
38	Electric Field Effect on Condensed-Phase Molecular Systems. VII. Vibrational Stark Sensitivity of Spatially Oriented Water Molecules in an Argon Matrix. <i>Journal of Physical Chemistry C</i> , 2019, 123, 9868-9874.	3.1	12
39	Proton Transport and Related Chemical Processes of Ice. <i>Journal of Physical Chemistry B</i> , 2021, 125, 8270-8281.	2.6	11
40	Organic light emitting diodes using NaCl:N,Nâ€²-bis(naphthalene-1-yl)-N,Nâ€²-bis(phenyl)benzidine composite as a hole injection buffer layer. <i>Journal of Applied Physics</i> , 2010, 108, 103703.	2.5	9
41	Metastable hydronium ions in UV-irradiated ice. <i>Journal of Chemical Physics</i> , 2012, 137, 204704.	3.0	9
42	Surface Charge Layer of Amorphous Solid Water with Adsorbed Acid or Base: Asymmetric Depth Distributions of H <sup>+</sup> and OH <sup>-</sup> Ions. <i>Journal of Physical Chemistry C</i> , 2016, 120, 12051-12058.	3.1	9
43	Efficient Thermal Reactions of Sulfur Dioxide on Ice Surfaces at Low Temperature: A Combined Experimental and Theoretical Study. <i>ACS Earth and Space Chemistry</i> , 2017, 1, 503-510.	2.7	9
44	Reaction of Nitrogen Dioxide with Ice Surface at Low Temperature (â‰ˆ170 K). <i>Journal of Physical Chemistry C</i> , 2015, 119, 22016-22024.	3.1	8
45	Dissociation of Trifluoroacetic Acid in Amorphous Solid Water: Charge-Delocalized Hydroniums and Zundel Continuum Absorption. <i>Journal of Physical Chemistry C</i> , 2017, 121, 12842-12848.	3.1	6
46	Electric Field Effect on Condensed-Phase Molecular Systems. VIII. Vibrational Stark Effect and Dipolar Inversion in a Carbon Monoxide Crystal. <i>Journal of Physical Chemistry C</i> , 2019, 123, 31262-31271.	3.1	6
47	Electric Field Effect on Condensed-Phase Molecular Systems. IX. Control of Proton Displacement in Matrix-Isolated Hydrogen Chlorideâ€“Water Complexes. <i>Journal of Physical Chemistry C</i> , 2020, 124, 1129-1134.	3.1	6
48	Electric Field Effect on Condensed-Phase Molecular Systems: V. Acidâ€“Base Proton Transfer at the Interface of Molecular Films. <i>Journal of Physical Chemistry C</i> , 2018, 122, 4901-4907.	3.1	5
49	Entropy-Driven Spontaneous Reaction in Cryogenic Ice: Dissociation of Fluoroacetic Acids. <i>Journal of Physical Chemistry Letters</i> , 2018, 9, 4282-4286.	4.6	5
50	Recent Progress in the Manipulation of Molecules with DC Electric Fields. <i>Accounts of Chemical Research</i> , 2021, 54, 323-331.	15.6	5
51	Electronic and Nuclear Contributions to Vibrational Stark Shifts of Hydroxyl Stretching Frequencies of Water Clusters. <i>Journal of Physical Chemistry C</i> , 2018, 122, 12970-12974.	3.1	4
52	Tunneling Diffusion of Excess Protons in Amorphous Solid Water at 10 and 80 K. <i>Journal of Physical Chemistry C</i> , 2019, 123, 3657-3663.	3.1	4
53	Transmission and Trapping of Low-Energy (1â€“10 eV) Electrons in Crystalline Ice Films. <i>Journal of Physical Chemistry C</i> , 2020, 124, 15862-15869.	3.1	4
54	Electric Field Effect on Condensed-Phase Molecular Systems. X. Interconversion Dynamics and Vibrational Stark Effect of Hydrogen Chloride Clusters in an Argon Matrix. <i>Journal of Physical Chemistry B</i> , 2020, 124, 4581-4589.	2.6	2

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55	Preparation and Characterization of Metastable <i>trans</i> -Dinitrogen Tetroxide. Journal of Physical Chemistry A, 2022, 126, 2353-2360.	2.5	2
56	Scalable energetic-impact deposition of a mixture composed of multi-walled carbon nanotubes and silver nanoparticles. Journal of the Korean Physical Society, 2013, 62, 980-982.	0.7	0
57	Brute Force Orientation of Matrix-Isolated Molecules: Reversible Reorientation of Formaldehyde in an Argon Matrix toward Perfect Alignment. Angewandte Chemie, 2017, 129, 1066-1069.	2.0	0