List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/5805029/publications.pdf Version: 2024-02-01



#	Article	IF	CITATIONS
1	Critical micelle concentration of surfactants in aqueous buffered and unbuffered systems. Analytica Chimica Acta, 2005, 548, 95-100.	5.4	317
2	Dissociation constants of neutral and charged acids in methyl alcohol. The acid strength resolution. Analytica Chimica Acta, 1998, 374, 309-324.	5.4	223
3	Solute-solvent and solvent-solvent interactions in binary solvent mixtures. Part 7. Comparison of the enhancement of the water structure in alcohol-water mixtures measured by solvatochromic indicators. Journal of Physical Organic Chemistry, 1998, 11, 185-192.	1.9	199
4	Comparison of the acidity of residual silanol groups in several liquid chromatography columns. Journal of Chromatography A, 2003, 986, 33-44.	3.7	196
5	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 1. A comparison of several preferential solvation models for describing ET(30) polarity of bipolar hydrogen bond acceptor-cosolvent mixtures. Journal of the Chemical Society Perkin Transactions II, 1995, , 1607-1615.	0.9	195
6	Retention of Ionizable Compounds on HPLC. pH Scale in Methanolâ^'Water and the pK and pH Values of Buffers. Analytical Chemistry, 1996, 68, 3651-3657.	6.5	195
7	Determination of dissociation constants of flavonoids by capillary electrophoresis. Electrophoresis, 2005, 26, 1886-1895.	2.4	194
8	Retention of Ionizable Compounds on HPLC. 5. pH Scales and the Retention of Acids and Bases with Acetonitrileâ~'Water Mobile Phases. Analytical Chemistry, 2000, 72, 5193-5200.	6.5	166
9	Retention of Ionizable Compounds on HPLC. 4. Mobile-Phase pH Measurement in Methanol/Water. Analytical Chemistry, 2000, 72, 1802-1809.	6.5	162
10	Influence of mobile phase acid–base equilibria on the chromatographic behaviour of protolytic compounds. Journal of Chromatography A, 2002, 982, 1-30.	3.7	144
11	Acidity in methanol–water. Analytica Chimica Acta, 2001, 439, 315-333.	5.4	141
12	Potentiometric and spectrophotometric pKa determination of water-insoluble compounds: Validation study in a new cosolvent system. Analytica Chimica Acta, 2007, 583, 418-428.	5.4	141
13	Static Dielectric Constants of Acetonitrile/Water Mixtures at Different Temperatures and Debyeâ~'HückelAanda0BParameters for Activity Coefficients. Journal of Chemical & Engineering Data, 2007, 52, 1103-1107.	1.9	139
14	Setup and validation of shake-flask procedures for the determination of partition coefficients (logD) from low drug amounts. European Journal of Pharmaceutical Sciences, 2015, 76, 181-191.	4.0	139
15	Relationship between ETpolarity and composition in binary solvent mixtures. Journal of the Chemical Society, Faraday Transactions, 1992, 88, 3541-3546.	1.7	123
16	Retention of Ionizable Compounds on HPLC. 2. Effect of pH, Ionic Strength, and Mobile Phase Composition on the Retention of Weak Acids. Analytical Chemistry, 1996, 68, 4094-4100.	6.5	123
17	Retention of ionizable compounds in high-performance liquid chromatography. Journal of Chromatography A, 2002, 964, 55-66.	3.7	115
18	On the Effect of Organic Solvent Composition on the pH of Buffered HPLC Mobile Phases and the p <i>K</i> <sub>a</sub> of Analytes—A Review. Separation and Purification Reviews, 2007, 36, 231-255.	5.5	104

#	Article	IF	CITATIONS
19	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 6. A quantitative measurement of the enhancement of the water structure in 2-methylpropan-2-ol–water and propan-2-ol–water mixtures by solvatochromic indicators. Journal of the Chemical Society Perkin Transactions II, 1997, , 1341-1348.	0.9	97
20	Retention of ionizable compounds on high-performance liquid chromatography. Journal of Chromatography A, 1998, 824, 137-146.	3.7	93
21	Retention of ionizable compounds on HPLC. 6. pH measurements with the glass electrode in methanol–water mixtures. Journal of Chromatography A, 2001, 911, 191-202.	3.7	91
22	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 3. The ET(30) polarity of binary mixtures of hydroxylic solvents. Journal of the Chemical Society Perkin Transactions II, 1996, , 1497-1503.	0.9	85
23	Solute-solvent and solvent-solvent interactions in binary solvent mixtures. 2. Effect of temperature on theET(30) polarity parameter of dipolar hydrogen bond acceptor-hydrogen bond donor mixtures. Journal of Physical Organic Chemistry, 1996, 9, 403-410.	1.9	85
24	Retention of Ionizable Compounds on HPLC. 12. The Properties of Liquid Chromatography Buffers in Acetonitrileâ^'Water Mobile Phases That Influence HPLC Retention. Analytical Chemistry, 2002, 74, 3809-3818.	6.5	85
25	Solute–solvent interactions in micellar electrokinetic chromatography. Journal of Chromatography A, 2002, 942, 237-248.	3.7	85
26	Retention of Ionizable Compounds on HPLC. 8. Influence of Mobile-Phase pH Change on the Chromatographic Retention of Acids and Bases during Gradient Elution. Analytical Chemistry, 2001, 73, 4937-4945.	6.5	80
27	Equilibrium solubility measurement of ionizable drugs – consensus recommendations for improving data quality. ADMET and DMPK, 2016, 4, 117.	2.1	78
28	Linear solvation energy relationships in reversed-phase liquid chromatography. Prediction of retention from a single solvent and a single solute parameter. Analytica Chimica Acta, 1993, 274, 147-162.	5.4	76
29	Acidâ^'Base Equilibria in Nonpolar Media. Absolute pKa Scale of Bases in Tetrahydrofuran. Journal of Organic Chemistry, 2006, 71, 9062-9067.	3.2	76
30	δConversion Parameter between pH Scales ( and ) in Acetonitrile/Water Mixtures at Various Compositions and Temperatures. Analytical Chemistry, 2007, 79, 3180-3187.	6.5	74
31	Molecular interactions between warfarin and human (HSA) or bovine (BSA) serum albumin evaluated by isothermal titration calorimetry (ITC), fluorescence spectrometry (FS) and frontal analysis capillary electrophoresis (FA/CE). Journal of Pharmaceutical and Biomedical Analysis, 2018, 150, 452-459.	2.8	72
32	Linear description of solute retention in reversed-phase liquid chromatography by a new mobile phase polarity parameter. Analytica Chimica Acta, 1994, 299, 219-229.	5.4	68
33	Densities, Refractive Indices, Absolute Viscosities, and Static Dielectric Constants of 2-Methylpropan-2-ol + Hexane, + Benzene, + Propan-2-ol, + Methanol, + Ethanol, and + Water at 303.2 K. Journal of Chemical & Engineering Data, 1995, 40, 1111-1114.	1.9	67
34	Prediction of the retention in reversed-phase liquid chromatography using solute–mobile phase–stationary phase polarity parameters. Journal of Chromatography A, 2002, 955, 19-34.	3.7	66
35	Retention of ionizable compounds in high-performance liquid chromatography. Journal of Chromatography A, 2002, 947, 47-58.	3.7	65
36	Solute–solvent interactions in micellar electrokinetic chromatography. Journal of Chromatography A, 1999, 845, 217-226.	3.7	63

#	Article	IF	CITATIONS
37	Dissociation constants of phenols in methanol–water mixtures. Journal of Chromatography A, 2000, 867, 45-56.	3.7	63
38	A comparison between different approaches to estimate the aqueous pKa values of several non-steroidal anti-inflammatory drugs. Analytica Chimica Acta, 1997, 338, 127-134.	5.4	62
39	Solute–solvent interactions in normal-phase liquid chromatography: a linear free-energy relationships study. Analytica Chimica Acta, 1999, 382, 301-308.	5.4	62
40	Retention models for ionizable compounds in reversed-phase liquid chromatography. Journal of Chromatography A, 2009, 1216, 1756-1775.	3.7	62
41	Effect of temperature on pH measurements and acid–base equilibria in methanol–water mixtures. Journal of Chromatography A, 2003, 1002, 41-53.	3.7	59
42	Molecular interactions between some non-steroidal anti-inflammatory drugs (NSAID׳s) and bovine (BSA) or human (HSA) serum albumin estimated by means of isothermal titration calorimetry (ITC) and frontal analysis capillary electrophoresis (FA/CE). Talanta, 2014, 130, 241-250.	5.5	59
43	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 4. Preferential solvation of solvatochromic indicators in mixtures of 2-methylpropan-2-ol with hexane, benzene, propan-2-ol, ethanol and methanol. Journal of the Chemical Society Perkin Transactions II, 1996, , 2177-2184.	0.9	57
44	Analysis of a solute polarity parameter in reversed-phase liquid chromatography on a linear solvation relationship basis. Analytica Chimica Acta, 2004, 515, 209-227.	5.4	56
45	Unique selectivity of perfluorinated stationary phases with 2,2,2-trifluoroethanol as organic mobile phase modifier. Journal of Chromatography A, 2001, 933, 73-81.	3.7	54
46	Critical evaluation of buffering solutions for p <b><i>K</i></b> <sub>a</sub> determination by capillary electrophoresis. Electrophoresis, 2008, 29, 2841-2851.	2.4	54
47	Potentiometric determination of aqueous dissociation constants of flavonols sparingly soluble in water. Talanta, 2008, 74, 1008-1013.	5.5	54
48	Solubility–pH profiles of some acidic, basic and amphoteric drugs. European Journal of Pharmaceutical Sciences, 2013, 48, 291-300.	4.0	54
49	Variation of acidity constants and pH values of some organic acids in water—2-propanol mixtures with solvent composition. Effect of preferential solvation. Analytica Chimica Acta, 1995, 302, 109-119.	5.4	52
50	Retention of ionizable compounds on high-performance liquid chromatography. Journal of Chromatography A, 2002, 945, 83-96.	3.7	52
51	Selectivity of single, mixed, and modified pseudostationary phases in electrokinetic chromatography. Electrophoresis, 2006, 27, 1900-1914.	2.4	51
52	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 5. Preferential solvation of solvatochromic indicators in mixtures of propan-2-ol with hexane, benzene, ethanol and methanol. Journal of the Chemical Society Perkin Transactions II, 1997, , 243-248.	0.9	50
53	Effect of temperature on the chromatographic retention of ionizable compounds. Journal of Chromatography A, 2004, 1042, 23-36.	3.7	47
54	Interpretive optimisation strategy applied to the isocratic separation of phenols by reversed-phase liquid chromatography with acetonitrile–water and methanol–water mobile phases. Journal of Chromatography A, 2000, 886, 31-46.	3.7	46

#	Article	IF	CITATIONS
55	Solute-solvent interactions in micellar electrokinetic chromatography: IV. Characterization of electroosmotic flow and micellar markers. Electrophoresis, 2002, 23, 56.	2.4	45
56	Acid–base constants of neutral bases in acetonitrile–water mixtures. Analytica Chimica Acta, 2002, 454, 157-166.	5.4	45
57	A potentially simpler approach to measure aqueous pKa of insoluble basic drugs containing amino groups. Journal of Pharmaceutical Sciences, 2003, 92, 1473-1481.	3.3	44
58	Hammett–Taft and Drago models in the prediction of acidity constant values of neutral and cationic acids in methanol â€. Journal of the Chemical Society Perkin Transactions II, 1999, , 1953-1958.	0.9	43
59	Acidity and Hydrophobicity of Several New Potential Antitubercular Drugs: Isoniazid and Benzimidazole Derivatives. Journal of Chemical & Engineering Data, 2012, 57, 330-338.	1.9	43
60	Considerations on the modelling and optimisation of resolution of ionisable compounds in extended pH-range columns. Journal of Chromatography A, 2005, 1089, 170-186.	3.7	41
61	Chromatographic Determination of Aqueous Dissociation Constants of Some Water-Insoluble Nonsteroidal Antiinflammatory Drugs. Journal of Pharmaceutical Sciences, 2002, 91, 991-999.	3.3	40
62	Retention of ionisable compounds on high-performance liquid chromatography. Journal of Chromatography A, 2004, 1059, 33-42.	3.7	40
63	Autoprotolysis in aqueous organic solvent mixtures. Analytical Chemistry, 1993, 65, 2294-2299.	6.5	39
64	Acidity constants in methanol/water mixtures of polycarboxylic acids used in drug salt preparations. European Journal of Pharmaceutical Sciences, 2006, 28, 118-127.	4.0	39
65	Fast high-throughput method for the determination of acidity constants by capillary electrophoresis. Journal of Chromatography A, 2009, 1216, 3646-3651.	3.7	39
66	A QSPR Study of thepSolute Polarity Parameter to Estimate Retention in HPLC. Journal of Chemical Information and Computer Sciences, 2003, 43, 1240-1247.	2.8	38
67	Characterization of the Solvation Properties of Sodiumn-Dodecyl Sulfate Micelles in Buffered and Unbuffered Aqueous Phases by Solvatochromic Indicators. Langmuir, 2003, 19, 55-62.	3.5	36
68	Retention of ionisable compounds on high-performance liquid chromatography XVII. Journal of Chromatography A, 2007, 1138, 203-215.	3.7	35
69	Inorganic salts as hold-up time markers in C18 columns. Talanta, 2000, 53, 667-677.	5.5	34
70	Critical validation of a new simpler approach to estimate aqueous pKa of drugs sparingly soluble in water. Analytica Chimica Acta, 2005, 550, 210-221.	5.4	34
71	Solute–solvent interactions in micellar electrokinetic chromatography. Journal of Chromatography A, 2001, 907, 257-265.	3.7	33
72	Determination of the hydrophobicity of organic compounds measured as logPo/w through a new chromatographic method. Journal of Chromatography A, 2010, 1217, 3026-3037.	3.7	33

#	Article	IF	CITATIONS
73	lonic equilibria in aqueous organic solvent mixtures the dissociation constants of acids and salts in tetrahydrofuran/water mixtures. Analytica Chimica Acta, 1997, 340, 133-141.	5.4	31
74	Dissociation constants of several non-steroidal anti-inflammatory drugs in isopropyl alcohol/water mixtures. Analytica Chimica Acta, 1997, 350, 249-255.	5.4	31
75	The determination of solvation descriptors for terpenes, and the prediction of nasal pungency thresholds. Journal of the Chemical Society Perkin Transactions II, 1998, , 2405-2412.	0.9	31
76	Retention of ionisable compounds on high-performance liquid chromatography. Journal of Chromatography A, 2006, 1121, 170-177.	3.7	31
77	Variation ofE T(30) polarity and the Kamlet-Taft solvatochromic parameters with composition in alcohol-alcohol mixtures. Journal of Solution Chemistry, 1995, 24, 51-63.	1.2	30
78	Effect of temperature on the chromatographic retention of ionizable compounds. Journal of Chromatography A, 2005, 1077, 159-169.	3.7	30
79	Characterization of the acidity of residual silanol groups in microparticulate and monolithic reversed-phase columns. Journal of Chromatography A, 2004, 1060, 135-145.	3.7	29
80	Lipophilicity assessment of basic drugs (logPo/w determination) by a chromatographic method. Journal of Chromatography A, 2011, 1218, 6356-6368.	3.7	29
81	Prediction of the separation of phenols by capillary zone electrophoresis. Analytica Chimica Acta, 2002, 458, 355-366.	5.4	28
82	Determination of flavonoid aglycones in several food samples by mixed micellar electrokinetic chromatography. Journal of Separation Science, 2007, 30, 2493-2500.	2.5	28
83	Modeling the Retention of Neutral Compounds in Gradient Elution RP-HPLC by Means of Polarity Parameter Models. Analytical Chemistry, 2009, 81, 9135-9145.	6.5	28
84	Retention of ionizable compounds on HPLC. Modelling retention in reversed-phase liquid chromatography as a function of pH and solvent composition with methanol-water mobile phases. Chromatographia, 2002, 55, 565-571.	1.3	27
85	Background electrolytes in 50% methanol/water for the determination of acidity constants of basic drugs by capillary zone electrophoresis. Journal of Chromatography A, 2006, 1123, 113-120.	3.7	27
86	Resolution of acid strength in tetrahydrofuran of substituted benzoic acids. Analytica Chimica Acta, 1992, 265, 157-165.	5.4	26
87	Autoprotolysis in aqueous organic solvent mixtures. Water/alcohol binary systems. Analytica Chimica Acta, 1996, 335, 291-302.	5.4	26
88	Simultaneous effect of pH, temperature and mobile phase composition in the chromatographic retention of ionizable compounds. Journal of Chromatography A, 2011, 1218, 4995-5009.	3.7	26
89	Retention of ionizable compounds on high-performance liquid chromatography. Journal of Chromatography A, 2001, 910, 187-194.	3.7	24
90	Change of mobile phase pH during gradient reversed-phase chromatography with 2,2,2-trifluoroethanol–water as mobile phase and its effect on the chromatographic hydrophobicity index determination. Journal of Chromatography A, 2002, 954, 77-87.	3.7	24

#	Article	IF	CITATIONS
91	Ionic equilibria in neutral amphiprotic solvents; resolution of acid strength in tert-butyl alcohol. Talanta, 1989, 36, 627-632.	5.5	23
92	Modeling Retention and Selectivity as a Function of pH and Column Temperature in Liquid Chromatography. Analytical Chemistry, 2006, 78, 5858-5867.	6.5	23
93	Polarity parameters of the Symmetry C18 and Chromolith Performance RP-18 monolithic chromatographic columns. Journal of Chromatography A, 2006, 1107, 96-103.	3.7	22
94	Retention of ionisable compounds on high-performance liquid chromatography XVIII: pH variation in mobile phases containing formic acid, piperazine, tris, boric acid or carbonate as buffering systems and acetonitrile as organic modifier. Journal of Chromatography A, 2009, 1216, 2491-2498.	3.7	22
95	Prediction of retention in reversed-phase liquid chromatography by means of the polarity parameter model. Journal of Chromatography A, 2009, 1216, 5214-5227.	3.7	22
96	Extension of the liquid chromatography/quantitative structure–property relationship method to assess the lipophilicity of neutral, acidic, basic and amphotheric drugs. Journal of Chromatography A, 2012, 1240, 113-122.	3.7	22
97	The Ca2+–EDTA chelation as standard reaction to validate Isothermal Titration Calorimeter measurements (ITC). Talanta, 2016, 154, 354-359.	5.5	22
98	Ionic equilibria in neutral amphiprotic solvents: variation of electrolyte dissociation constants in tert-butyl alcohol with addition of a second solvent. Analytical Chemistry, 1988, 60, 2008-2013.	6.5	21
99	Standardization of potentiometric sensors in tetrahydrofuran. Analytica Chimica Acta, 1992, 264, 229-239.	5.4	21
100	Dissociation constants and preferential solvation in some 2-methylpropan-2-ol–alcohol mixtures. Journal of the Chemical Society, Faraday Transactions, 1993, 89, 1723-1728.	1.7	21
101	Empirical treatment of solvent-solute interactions: medium effects on the electronic absorption spectrum of ?-carotene. Journal of Physical Organic Chemistry, 1998, 11, 193-200.	1.9	21
102	Micellar electrokinetic chromatography estimation of size and composition of procyanidins after thiolysis with cysteine. Electrophoresis, 2003, 24, 1404-1410.	2.4	21
103	Physicochemical Properties of a New Multicomponent Cosolvent System for the p <i>K</i> <sub>a</sub> Determination of Poorly Soluble Pharmaceutical Compounds. Helvetica Chimica Acta, 2007, 90, 1538-1553.	1.6	21
104	Critical comparison of shake-flask, potentiometric and chromatographic methods for lipophilicity evaluation (log P o/w ) of neutral, acidic, basic, amphoteric, and zwitterionic drugs. European Journal of Pharmaceutical Sciences, 2018, 122, 331-340.	4.0	21
105	Ionic equilibria in neutral amphiprotic solvents: Structural effects on dissociation constants of several substituted phenols and mercaptopyrimidines in isopropyl alcohol. Talanta, 1989, 36, 1227-1231.	5.5	20
106	Ionic equilibria in neutral amphiprotic solvents: relationships between electrolyte pK values and solvent polarity and composition for several binary isopropyl alcohol mixtures. Analytical Chemistry, 1990, 62, 102-107.	6.5	20
107	Soluteâ~'Solvent Interactions in Micellar Electrokinetic Chromatography. 6. Optimization of the Selectivity of Lithium Dodecyl Sulfateâ~'Lithium Perfluorooctanesulfonate Mixed Micellar Buffers. Analytical Chemistry, 2002, 74, 4447-4455.	6.5	20
108	Lipophilicity of amphoteric and zwitterionic compounds: A comparative study of determination methods. Talanta, 2017, 162, 293-299.	5.5	20

#	Article	IF	CITATIONS
109	Color changes in screened indicators. Analytical Chemistry, 1984, 56, 1422-1428.	6.5	19
110	A comparative study of some hydroxyanthraquinones as acid-base indicators. Talanta, 1985, 32, 1077-1081.	5.5	19
111	Neutralisation indicators in 2-methylpropan-2-ol: their pKavalues and chromatic parameters of transition ranges. Analyst, The, 1987, 112, 179-184.	3.5	18
112	Solute-solvent interactions in micellar electrokinetic chromatography: V. Factors that produce peak splitting. Electrophoresis, 2002, 23, 2408-2416.	2.4	18
113	Determination of the chromatographic hydrophobicity index for ionisable solutes. Journal of Chromatography A, 2007, 1173, 110-119.	3.7	18
114	Acidity of Several Anilinium Derivatives in Pure Tetrahydrofuran. Journal of Solution Chemistry, 2008, 37, 689-700.	1.2	18
115	Effect of vinylpyrrolidone polymers on the solubility and supersaturation of drugs; a study using the Cheqsol method. European Journal of Pharmaceutical Sciences, 2018, 117, 227-235.	4.0	18
116	Conductometric determination of dissociation constants of several acids and their tetrabutylammonium salts in propan-2-ol/water mixtures. Analytica Chimica Acta, 1996, 333, 241-247.	5.4	17
117	Mixed micellar electrokinetic capillary chromatography separation of depolymerized grape procyanidins. Electrophoresis, 2003, 24, 707-713.	2.4	17
118	A Fast Method for p <i>K</i> <sub>a</sub> Determination by Capillary Electrophoresis. Chemistry and Biodiversity, 2009, 6, 1822-1827.	2.1	17
119	Chromatographic models to predict the elution of ionizable analytes by organic modifier gradient in reversed phase liquid chromatography. Journal of Chromatography A, 2012, 1247, 71-80.	3.7	17
120	Evaluation of log Po/w values of drugs from some molecular structure calculation softwares. ADMET and DMPK, 2014, 2, .	2.1	17
121	The contribution of the hydrogen bond acidity on the lipophilicity of drugs estimated from chromatographic measurements. European Journal of Pharmaceutical Sciences, 2013, 48, 484-493.	4.0	16
122	Study of 1,4-dihydroxyanthraquinone as an acid-base indicator in isopropyl alcohol medium Evaluation of colour-change limits through complementary chromaticity parameters. Talanta, 1984, 31, 279-282.	5.5	15
123	Enthalpies and constants of dissociation of several neutral and cationic acids in aqueous and methanol/water solutions at various temperatures. Journal of Pharmaceutical and Biomedical Analysis, 2009, 49, 923-930.	2.8	15
124	Phenothiazines solution complexity – Determination of pKa and solubility-pH profiles exhibiting sub-micellar aggregation at 25 and 37°C. European Journal of Pharmaceutical Sciences, 2016, 93, 163-176.	4.0	15
125	Multiwavelength Spectrophotometric Determination of Dissociation Constants of Mercaptopyrimidines. Analytical Letters, 1988, 21, 1273-1284.	1.8	14
126	Effect of temperature on the chromatographic retention of ionizable compounds. III. Modeling retention of pharmaceuticals as a function of eluent pH and column temperature in RPLC. Journal of Separation Science, 2008, 31, 969-980.	2.5	14

#	Article	IF	CITATIONS
127	Ionic equilibria in neutral amphiprotic solvents of low dielectric constant: Buffer solutions. Talanta, 1989, 36, 615-621.	5.5	13
128	Autoprotolysis in aqueous organic solvent mixtures. Water/dipolar protophilic solvent binary systems. Analytica Chimica Acta, 1997, 349, 367-376.	5.4	13
129	Hydrophobic and cation exchange mechanisms in the retention of basic compounds in a polymeric column. Journal of Chromatography A, 2004, 1028, 139-148.	3.7	13
130	Acidity of several polyprotic acids, amiodarone and quetiapine hemifumarate in pure methanol. Talanta, 2007, 73, 115-120.	5.5	13
131	Acidâ~'Base Dissociation Constants of <i>o</i> -Phthalic Acid in Acetonitrile/Water Mixtures over the (15 to 50) °C Temperature Range and Related Thermodynamic Quantities. Journal of Chemical & Engineering Data, 2010, 55, 85-91.	1.9	12
132	Molecular Details of INH-C <sub>10</sub> Binding to <i>wt</i> KatG and Its S315T Mutant. Molecular Pharmaceutics, 2015, 12, 898-909.	4.6	12
133	Study of neutralisation indicators in propan-2-ol: their pKavalues and chromatic parameters of transition ranges. Analyst, The, 1985, 110, 1473-1476.	3.5	10
134	Chromatic characterisation and applicability of neutralisation indicator series in anhydrous acetic acid. Analyst, The, 1987, 112, 1717.	3.5	10
135	Dissociation constants of sore anti-inflammatory agents (α-phenylpropionic acids) in Isopropyl andtert-butyl alcohol media. Electroanalysis, 1991, 3, 365-370.	2.9	10
136	Dissociation constants of organophosphinic acid compounds. Talanta, 1993, 40, 1339-1343.	5.5	10
137	Interpretation of Hydroxylic Solvent Effects Based on Correlations with Solvent Parameters. Reaction of Et3N with Etl. Collection of Czechoslovak Chemical Communications, 1994, 59, 898-904.	1.0	10
138	lonic equilibria in aqueous organic solvent mixtures The equilibria of HF in an ethanol + water mixture used for cleaning up semiconductors. Journal of Electroanalytical Chemistry, 1997, 433, 77-83.	3.8	10
139	Retention of ionisable compounds on high-performance liquid chromatography. Journal of Chromatography A, 2009, 1216, 5445-5448.	3.7	10
140	Ionic equilibria in neutral amphiprotic solvents of low dielectric constant: Titration curves. Talanta, 1989, 36, 623-626.	5.5	9
141	Gradient retention prediction of acid–base analytes in reversed phase liquid chromatography: A simplified approach for acetonitrile–water mobile phases. Journal of Chromatography A, 2014, 1370, 129-134.	3.7	9
142	Prediction of the chromatographic retention of acid–base compounds in pH buffered methanol–water mobile phases in gradient mode by a simplified model. Journal of Chromatography A, 2015, 1385, 42-48.	3.7	9
143	High-throughput logPo/w determination from UHPLC measurements: Revisiting the chromatographic hydrophobicity index. Journal of Pharmaceutical and Biomedical Analysis, 2016, 127, 26-31.	2.8	9
144	Ionizable Drug Self-Associations and the Solubility Dependence on pH: Detection of Aggregates in Saturated Solutions Using Mass Spectrometry (ESI-Q-TOF-MS/MS). Molecular Pharmaceutics, 2021, 18, 2311-2321.	4.6	9

#	Article	IF	CITATIONS
145	Ionic equilibria in amphiprotic solvents of low dielectric constant. Analytica Chimica Acta, 1992, 256, 211-220.	5.4	8
146	Autoprotolysis in aqueous organic solvent mixtures. Water-amide and water-amine binary systems. Analytica Chimica Acta, 1995, 302, 355-363.	5.4	8
147	Chromatographic hydrophobicity index: pH profile for polyprotic compounds. Journal of Chromatography A, 2009, 1216, 7798-7805.	3.7	8
148	Polarographic behaviour of mercaptopyrimidines. Electrochimica Acta, 1982, 27, 1465-1467.	5.2	7
149	Linear solvation energy relationships between electrolyte pKvalues and solvent properties for several 2-methylpropan-2-Ol-cosolvent mixtures. Journal of Physical Organic Chemistry, 1994, 7, 696-704.	1.9	7
150	Retention of ionizable compounds on HPLC. Modelling retention for neutral and ionizable compounds by linear solvation energy relationships. Chromatographia, 2002, 56, 431-437.	1.3	7
151	Erratum to "Solute–solvent interactions in micellar electrokinetic chromatography. III. Characterization of the selectivity of micellar electrokinetic chromatography systems―[J. Chromatogr. A 942 (2002) 237–248]. Journal of Chromatography A, 2009, 1216, 6877-6879.	3.7	7
152	Molecular characteristics of several drugs evaluated from solvent/water partition measurements: Solvation parameters and intramolecular hydrogen bond indicator. European Journal of Pharmaceutical Sciences, 2022, 168, 106066.	4.0	7
153	Study of semicarbazones and thiosemicarbazones derived from 1,2-naphthoquinone, as acid—base indicators: Evaluation of their transition limits through the chromaticity co-ordinates. Talanta, 1982, 29, 1125-1129.	5.5	6
154	Dissociation constants, neutralization enthalpies and reactions of 3-styryl-2-mercaptopropenoic and 3-(1-naphthyl)-2-mercaptopropenoic acids. Talanta, 1984, 31, 475-478.	5.5	6
155	Polarographic determination of chlorquinaldol in pharmaceutical preparations. Microchemical Journal, 1987, 35, 133-136.	4.5	6
156	Polarographic determination of clioquinol in pharmaceutical preparations. Journal of Pharmaceutical and Biomedical Analysis, 1988, 6, 983-986.	2.8	6
157	Standardization of potentiometric cells in propan-2-ol-water. Analytica Chimica Acta, 1993, 280, 75-83.	5.4	6
158	Solvent extraction of alkylphosphoric acid derivatives between water and immiscible organic solvents. Analytica Chimica Acta, 1997, 350, 197-202.	5.4	6
159	Kinetic and Thermodynamic Solubility Values of Some Bioactive Compounds. Chemistry and Biodiversity, 2009, 6, 1789-1795.	2.1	6
160	Isothermal titration calorimetry of Ni(II) binding to histidine and to N-2-aminoethylglycine. Talanta, 2011, 84, 347-354.	5.5	6
161	Chromatographic Hydrophobicity Index (CHI). Advances in Chromatography, 2012, 50, 377-414.	1.0	6
162	Ionic equilibria in amphiprotic solvents of low dielectric constant. Analytica Chimica Acta, 1992, 256, 203-210.	5.4	5

#	Article	IF	CITATIONS
163	Potentiometric and thermometric determination of mercaptopyrimidines. Mikrochimica Acta, 1985, 86, 339-346.	5.0	4
164	1,2-Naphthoquinone-2-thiosemicarbazone as a new acid-base indicator in isopropyl and tert-butyl alcohol media. Talanta, 1988, 35, 419-423.	5.5	4
165	Variation of some microscopic properties with composition in 2-methoxyethanol and 1,2-ethanediol mixtures. Journal of Solution Chemistry, 1994, 23, 735-746.	1.2	4
166	Application of a polarity parameter model to the separation of fatâ€soluble vitamins by reversedâ€phase HPLC. Journal of Separation Science, 2008, 31, 3170-3181.	2.5	4
167	A comparative study of the behaviour of Crystal Violet, alizarin-9-imine and quinalizarin-9-imine as indicators in anhydrous acetic acid medium. Mikrochimica Acta, 1986, 90, 13-26.	5.0	3
168	Improvement of the titrimetric method for the determination of total basicity and available lysine residues in proteinaceous samples in anhydrous acetic acid. Analytica Chimica Acta, 1992, 256, 177-181.	5.4	3
169	Binding thermodynamics of paromomycin, neomycin, neomycinâ€dinucleotide and â€diPNA conjugates to bacterial and human rRNA. Journal of Molecular Recognition, 2016, 29, 142-150.	2.1	3
170	Solute–solvent and solvent–solvent interactions in binary solvent mixtures. Part 7. Comparison of the enhancement of the water structure in alcohol–water mixtures measured by solvatochromic indicators. Journal of Physical Organic Chemistry, 1998, 11, 185-192.	1.9	3
171	Screened indicators in anhydrous acetic acid medium: Chromatic evaluation of their transitions. Mikrochimica Acta, 1991, 105, 89-99.	5.0	1
172	Soluteâ€solvent and solventâ€solvent interactions in binary solvent mixtures. 2. Effect of temperature on the ET(30) polarity parameter of dipolar hydrogen bond acceptorâ€hydrogen bond donor mixtures. Journal of Physical Organic Chemistry, 1996, 9, 403-410.	1.9	1
173	A syllabus for a two-semester chemistry course for health professions. Journal of Chemical Education, 1990, 67, 539.	2.3	Ο
174	A QSPR Study of the p Solute Polarity Parameter to Estimate Retention of HPLC ChemInform, 2003, 34, no.	0.0	0
175	Ionic equilibria in aqueous organic solvent mixtures. Speciation of hydrofluoric acid in several ethanol/water solutions. Journal of Electroanalytical Chemistry, 2019, 848, 113318.	3.8	О