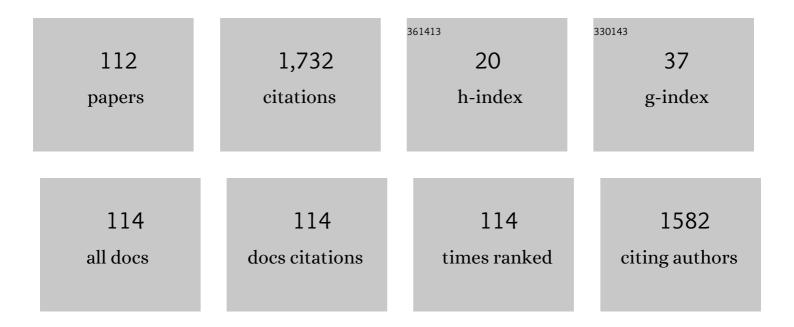
List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Sorption of Se(-II) on illite, MX-80 bentonite, shale, and limestone in Na–Ca–Cl solutions. Nuclear Engineering and Technology, 2022, 54, 1616-1622.	2.3	2
2	Sorption of Pd on illite, MX-80 bentonite and shale in Na–Ca–Cl solutions. Nuclear Engineering and Technology, 2021, 53, 894-900.	2.3	2
3	Contribution of Ternary Reaction to Pd Sorption on MX-80 in Na-Ca-Cl Solution at High Ionic Strength. Science and Technology of Nuclear Installations, 2019, 2019, 1-6.	0.8	1
4	System dynamics simulation of the thermal dynamic processes in nuclear power plants. Nuclear Engineering and Technology, 2019, 51, 1540-1553.	2.3	25
5	Optical Properties of Trinuclear Citrate Complexes Containing 4f and 5f Block Metals. Bulletin of the Chemical Society of Japan, 2018, 91, 882-890.	3.2	1
6	A study of background on why Ontarians support nuclear energy. Journal of Nuclear Fuel Cycle and Environment, 2018, 25, 15-20.	0.1	0
7	Comparison of proliferation resistance among natural uranium, thorium–uranium, and thorium–plutonium fuels used in CANada Deuterium Uranium in deep geological repository by combining multiattribute utility analysis with transport model. Nuclear Engineering and Technology, 2018. 50. 794-800.	2.3	4
8	Sorption behaviour of Np(IV) on illite, shale and MX-80 in high ionic strength solutions. Journal of Radioanalytical and Nuclear Chemistry, 2017, 313, 1-11.	1.5	13
9	Interaction of <mml:math <br="" altimg="si1.gif" xmlns:mml="http://www.w3.org/1998/Math/MathML">overflow="scroll"><mml:mrow><mml:msubsup><mml:mrow><mml:mtext>NpO</mml:mtext></mml:mrow><r with Clâ€" in Na Ca Cl-type solutions at ionic strength of 6M: Effect of presence of Ca ion on interaction. Nuclear Engineering and Technology, 2017, 49, 1778-1782.</r </mml:msubsup></mml:mrow></mml:math>	nml:ŋŋ>2<	/mml:mn> <a< td=""></a<>
10	Cause–effect analysis on Fukushima accident reports – What did McMaster undergraduate students learn?. Journal of Nuclear Science and Technology, 2016, 53, 756-765.	1.3	1
11	Sorption behavior of Np(V) on illite, shale and MX-80 in high ionic strength solutions. Journal of Radioanalytical and Nuclear Chemistry, 2016, 308, 143-153.	1.5	12
12	Integrating Social-Scientific Literacy in Nuclear Engineering Education. , 2015, , 1-17.		0
13	Burning of MOX fuels in LWRs; fuel history effects on thermal properties of hull and end piece wastes and the repository performance. Journal of Nuclear Science and Technology, 2012, 49, 310-319.	1.3	0
14	Speciation of Eu3+ bound to humic substances by time-resolved laser fluorescence spectroscopy (TRLFS) and parallel factor analysis (PARAFAC). Geochimica Et Cosmochimica Acta, 2012, 88, 199-215.	3.9	25
15	Surface speciation of Eu3+ adsorbed on kaolinite by time-resolved laser fluorescence spectroscopy (TRLFS) and parallel factor analysis (PARAFAC). Journal of Colloid and Interface Science, 2012, 374, 258-266.	9.4	31
16	Isotopic ratio and vertical distribution of radionuclides in soil affected by the accident of Fukushima Dai-ichi nuclear power plants. Journal of Environmental Radioactivity, 2012, 113, 37-44.	1.7	45
17	Application of Parallel Factor Analysis for Time-Resolved Laser Fluorescence Spectroscopy: Implication for Metal Speciation Study. Environmental Science & Technology, 2010, 44, 5055-5060.	10.0	36
18	Study on the high level radioactive wastes geological isolation seen from the social acceptance of science and technology;. Journal of Nuclear Fuel Cycle and Environment, 2010, 17, 3-22.	0.1	0

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19	Electrostatic potentials of humic acid: Fluorescence quenching measurements and comparison with model calculations. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2009, 347, 27-32.	4.7	9
20	The structure of monomeric and dimeric uranyl adsorption complexes on gibbsite: A combined DFT and EXAFS study. Geochimica Et Cosmochimica Acta, 2009, 73, 5975-5988.	3.9	73
21	LWR High Burn-Up Operation and MOX Introduction; Fuel Cycle Performance from the Viewpoint of Waste Management. Journal of Nuclear Science and Technology, 2009, 46, 677-689.	1.3	17
22	Social Acceptance Process Model for Ensuring the High-Level Radioactive Waste Disposal Site. Transactions of the Atomic Energy Society of Japan, 2009, 8, 19-33.	0.3	4
23	Thermal Impact on Geological Disposal of Hull and End Piece Wastes Resulting from High-Burn-up Operation of LWR and Introduction of MOX Fuels into LWR. Journal of Nuclear Science and Technology, 2009, 46, 443-452.	1.3	4
24	Adsorption of Heterogeneously Charged Nanoparticles on a Variably Charged Surface by the Extended Surface Complexation Approach:  Charge Regulation, Chemical Heterogeneity, and Surface Complexation. Journal of Physical Chemistry B, 2008, 112, 1339-1349.	2.6	18
25	Moderation of Uncertainty in a Database of Sorption Coefficients (II) Analysis of Data Scatter by Adopting Ion Exchange Model. Transactions of the Atomic Energy Society of Japan, 2008, 7, 210-220.	0.3	Ο
26	Moderation of Uncertainty in a Database of Sorption Coefficients (I) Factors Drawing the Scatter in Database and Data Selection. Transactions of the Atomic Energy Society of Japan, 2008, 7, 194-209.	0.3	0
27	Metal Sorption toPseudomonas fluorescens: Influence of pH, Ionic Strength and Metal Concentrations. Geomicrobiology Journal, 2007, 24, 205-210.	2.0	7
28	A fast and sensitive method for evaluating nuclides migration characteristics in rock medium by using micro-channel reactor concept. Physics and Chemistry of the Earth, 2007, 32, 463-468.	2.9	6
29	Polynuclear complex formation of trivalent lanthanides by 5-sulfosalicylate in an aqueous system—potentiometric, 1H NMR, and TRLIFS studies. Inorganica Chimica Acta, 2007, 360, 1575-1583.	2.4	8
30	Evaluation of Particle Release from Montmorillonite Gel by Flowing Groundwater Based on the DLVO Theory. Transactions of the Atomic Energy Society of Japan, 2007, 6, 205-213.	0.3	1
31	A theoretical study on molecular structure of Eu(III)-salicylate complexes in aqueous system. Computational and Theoretical Chemistry, 2005, 757, 87-97.	1.5	18
32	Change in pore structure and composition of hardened cement paste during the process of dissolution. Cement and Concrete Research, 2005, 35, 943-950.	11.0	72
33	Effects of porosity on leaching of Ca from hardened ordinary Portland cement paste. Cement and Concrete Research, 2005, 35, 1764-1775.	11.0	193
34	Electrostatic interaction models for ion binding to humic substances. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2005, 265, 104-113.	4.7	43
35	Direct Quantitative Analysis of Particulate Aluminum Suspended in Water Using Laser-Induced Breakdown Spectroscopy. Soil Science and Plant Nutrition, 2005, 51, 911-916.	1.9	3
36	Selective and in-situ determination of carbonate and oxide particles in aqueous solution using laser-induced breakdown spectroscopy (LIBS) for wearable information equipment. Microsystem Technologies, 2005, 11, 974-979.	2.0	10

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37	Analysis of Copper Binding in the Ternary System Cu2+/Humic Acid/Goethite at Neutral to Acidic pH. Environmental Science & Technology, 2005, 39, 4886-4893.	10.0	63
38	Chemical Exchange Reaction of Glycinatocopper(II) Complex in Water: A Theoretical Study. Journal of Physical Chemistry A, 2005, 109, 10403-10409.	2.5	19
39	Reduction rate of neptunium(V) in heterogeneous solution with magnetite. Radiochimica Acta, 2004, 92, 145-150.	1.2	38
40	Application of the NICA-Donnan model for proton, copper and uranyl binding to humic acid. Radiochimica Acta, 2004, 92, 567-574.	1.2	32
41	Fluorescence characteristics of complex formation of europium(III)-salicylate. Radiochimica Acta, 2004, 92, 589-593.	1.2	17
42	Stoichiometry, stability constants and coordination geometry of Eu(iii) 5-sulfosalicylate complex in aqueous system—A TRLIFS study. Dalton Transactions, 2004, , 3495-3502.	3.3	11
43	Adsorption of Humic Acid on Goethite:Â Isotherms, Charge Adjustments, and Potential Profiles. Langmuir, 2004, 20, 689-700.	3.5	134
44	Modeling of the Metal Binding to Humic Substances: Comparison between the Discrete and Continuous Affinity Distribution Models. Transactions of the Atomic Energy Society of Japan, 2004, 3, 215-232.	0.3	3
45	Sorption of nonylphenol on Na-Montmorillonite. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2003, 230, 131-139.	4.7	20
46	STUDY ON THE LEACHING BEHAVIOR OF Ca FROM THE ALITE HYDRATE. Doboku Gakkai Ronbunshu, 2003, 2003, 141-149.	0.2	2
47	P-PW-08 SELECTIVE AND IN-SITU DETERMINATION OF CARBONATE AND OXIDE PARTICLES IN AQUEOUS SOLUTION USING LASER-INDUCED BREAKDOWN SPECTROSCOPY (LIBS) FOR WEARABLE INFORMATION EQUIPMENT. Proceedings of JSME-IIP/ASME-ISPS Joint Conference on Micromechatronics for Information and Precision Equipment IIP/ISPS Joint MIPE, 2003, 2003, 415-416.	0.0	0
48	Evaluation of the complexation behavior between humic acid and UO2 2+ with fluorescence spectroscopy and its mixture analysis. Radiochimica Acta, 2002, 90, 27-33.	1.2	15
49	Molecular fluorescence spectroscopy and mixture analysis for the evaluation of the complexation between humic acid and UO2 2+. Radiochimica Acta, 2002, 90, 545-548.	1.2	7
50	Sorption and reduction of neptunium(V) on the surface of iron oxides. Radiochimica Acta, 2002, 90, 665-669.	1.2	53
51	Silicate Anion Structural Change in Calcium Silicate Hydrate Gel on Dissolution of Hydrated Cement. Journal of Nuclear Science and Technology, 2002, 39, 540-547.	1.3	45
52	Influence of Heterogeneity of Binding Sites of Hnmic Acid on its Complexation with Actinyl Ions. Journal of Nuclear Science and Technology, 2002, 39, 466-472.	1.3	0
53	Adsorption behavior of IO3â~' by CO32â~'- and NO3â~'-hydrotalcite. Applied Clay Science, 2002, 22, 17-23.	5.2	67
54	Efficiency of Time Resolved Laser Induced Breakdown Spectroscopy for Particle Detection and Energy Transfer of Breakdown Plasma. Transactions of the Atomic Energy Society of Japan, 2002, 1, 472-476.	0.3	1

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55	Silicate Anion Structural Change in Calcium Silicate Hydrate Gel on Dissolution of Hydrated Cement Journal of Nuclear Science and Technology, 2002, 39, 540-547.	1.3	10
56	Migration of humic acid through silicate-packed columns consideringfiltration effect. Journal of Nuclear Fuel Cycle and Environment, 2002, 9, 15-20.	0.1	0
57	Sorption behavior of strontium onto C-S-H (calcium silicate hydrated phases). Studies in Surface Science and Catalysis, 2001, , 901-904.	1.5	6
58	Effect of Flow Field on Colloid Deposition in Filtration Process of Polystyrene Latex Particles through Columns Packed with Glass Beads. Journal of Nuclear Science and Technology, 2001, 38, 645-654.	1.3	3
59	Deposition Behavior of Polystyrene Latex Particles on Solid Surfaces during Migration through an Artificial Fracture in a Granite Rock Sample. Journal of Nuclear Science and Technology, 2001, 38, 439-443.	1.3	12
60	<i>Ab Initio</i> Calculations for the Substitutions of Al(OH) ⁻ ₄ and SO ²⁻ ₄ with Si Tetrahedra. Journal of Nuclear Science and Technology, 2001, 38, 533-541.	1.3	7
61	Sorption of uranium(VI) on Na-montmorillonite colloids – Effect of humic acid and its migration –. Studies in Surface Science and Catalysis, 2001, 132, 829-832.	1.5	0
62	Effect of Flow Field on Colloid Deposition in Filtration Process of Polystyrene Latex Particles through Columns Packed with Glass Beads Journal of Nuclear Science and Technology, 2001, 38, 645-654.	1.3	1
63	Dissolution Phenomena of Ca0-Si02-H20 Gel at Ca/Si>l Coexisting with Ettringite System. Journal of Nuclear Science and Technology, 2000, 37, 793-801.	1.3	3
64	Speciation analysis on europium(III) using laser-induced breakdown spectroscopy. Radiochimica Acta, 2000, 88, 645-650.	1.2	16
65	Sorption equilibrium and kinetics of NpO2 + on dispersed particles of Na-montmorillonite. Radiochimica Acta, 2000, 88, 705-710.	1.2	16
66	Sorption and desorption kinetics of Np(V) on magnetite and hematite. Radiochimica Acta, 2000, 88, 453-458.	1.2	13
67	Sorption study of strontium onto hydrated cement phases using a sequential desorption method. Radiochimica Acta, 2000, 88, 483-487.	1.2	22
68	放射性å»f棄物処å^†ã,∙ã,¹ãf†ãfã«ãҌã҈яã┥ã,»ãf¡ãf³ãf^ã«æœŸå¾…ã•ã,Œã,‹å½¹å‰²,(II) 朖™è	è∽㮜¢−ç,¹	ã∙ã,‰. Nippo
69	Dissolution Phenomena of CaO-SiO2-H2O Gel at Ca/Si>1 Coexisting with Ettringite System Journal of Nuclear Science and Technology, 2000, 37, 793-801.	1.3	1
70	Sorption of neptunium on bentonite and its migration in geosphere. Colloids and Surfaces A: Physicochemical and Engineering Aspects, 1999, 155, 137-143.	4.7	15
71	A model for dissolution of CaO-SiO2-H2O gel at Ca/Si > 1. Cement and Concrete Research, 1999, 29, 1091-1097.	11.0	32
72	A Model for Dissolution of Cao-Sio2-H2O Gel at Ca/Si<1 by Considering Disordered Structure. Materials Research Society Symposia Proceedings, 1999, 556, 1237.	0.1	6

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73	(SiO4)4- structure change in C-S-H on dissolution processes of tricalcium silicate hydrate. Journal of Nuclear Fuel Cycle and Environment, 1999, 5, 43-50.	0.1	3
74	Diffusion in binary suspensions of charged colloids. Journal of Contaminant Hydrology, 1998, 35, 277-284.	3.3	0
75	Geochemical behavior of actinides in high-level radioactive waste disposal. Progress in Nuclear Energy, 1998, 32, 141-161.	2.9	12
76	Surface sorption and surface diffusion of NpO2+ with poorly crystallized ferric oxide. Journal of Alloys and Compounds, 1998, 271-273, 252-256.	5.5	13
77	A Raman Spectroscopic Study of Uranyl Species Adsorbed onto Colloidal Particles. Journal of Physical Chemistry B, 1998, 102, 9029-9032.	2.6	31
78	Adsorption/Desorption of Lanthanides on Metal Oxides Interfaces. Radiochimica Acta, 1998, 82, 239-242.	1.2	11
79	Sorption Equilibrium and Kinetics of NpO ⁺ ₂ Uptake onto Illite. Radiochimica Acta, 1998, 82, 263-268.	1.2	11
80	Filtration in Colloid Migration through Porous Media. Journal of Nuclear Fuel Cycle and Environment, 1998, 5, 59-66.	0.1	0
81	Separation and Coprecipitation of Lanthanides and Americium by Photolysis. Nuclear Technology, 1997, 118, 42-48.	1.2	0
82	Impact of Colloid Generation on Actinide Migration in High-Level Radioactive Waste Disposal: Overview and Laboratory Analysis. Nuclear Technology, 1997, 118, 58-68.	1.2	18
83	Diffusion of Charged Spherical Particles in Binary System. Journal of Nuclear Fuel Cycle and Environment, 1997, 4, 3-7.	0.1	0
84	Affinity of finely dispersed montmorillonite colloidal particles for americium and lanthanides. Journal of Nuclear Materials, 1997, 244, 29-35.	2.7	16
85	Interfacial behavior of actinides with colloids in the geosphere. Journal of Nuclear Materials, 1997, 248, 323-327.	2.7	17
86	Photochemical separation and co-precipitation of lanthanides in nitric acid solution. Journal of Photochemistry and Photobiology A: Chemistry, 1997, 106, 57-60.	3.9	7
87	The Role of Cement to be Expected in Radioactive Waste Disposal System Nippon Genshiryoku Gakkaishi/Journal of the Atomic Energy Society of Japan, 1997, 39, 1008-1018.	0.0	3
88	Photochemical Oxidation of Americium in Dilute Nitric Acid Solution with the Addition of Ozone. Separation Science and Technology, 1996, 31, 2443-2453.	2.5	7
89	Sorption behavior of Np (IV), Np (V) and Am (III) in the disturbed zone between engineered and natural barriers. Journal of Radioanalytical and Nuclear Chemistry, 1996, 214, 381-389.	1.5	10
90	Quantitative Performance Allocation of Multi-Barrier System for High-Level Radioactive Waste Disposal Nippon Genshiryoku Gakkaishi/Journal of the Atomic Energy Society of Japan, 1995, 37, 59-77.	0.0	15

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91	Separation of Lanthanides and Oxidation of Americium in Nitric Acid Solution by Photolysis. Journal of Nuclear Science and Technology, 1995, 32, 154-156.	1.3	4
92	Separation of Lanthanides and Oxidation of Americium in Nitric Acid Solution by Photolysis Journal of Nuclear Science and Technology, 1995, 32, 154-156.	1.3	2
93	Colloid Formation and Sorption of Americium in the Water/Bentonite System. Radiochimica Acta, 1994, 66-67, 207-212.	1.2	4
94	Influence of Fe(III) Colloids on Np(V) Migration through Quartz-Packed Columns. Journal of Nuclear Science and Technology, 1994, 31, 143-150.	1.3	17
95	THIRD PHASE FORMATION IN NITRIC ACID EXTRACTION BYn-OCTYL(PHENYL)-N,N-DIISOBUTYLCARBAMOYL-METHYLPHOSPHINE OXIDE. Solvent Extraction and Ion Exchange, 1994, 12, 459-473.	2.0	6
96	Impact of Pseudocolloid Formation on Migration of Nuclides within Fractures. Journal of Nuclear Science and Technology, 1994, 31, 623-625.	1.3	4
97	Facilitated Transport of Charged Colloids in Geologic Media. Materials Research Society Symposia Proceedings, 1994, 353, 157.	0.1	2
98	Influence of Fe(III) Colloids on Np(V) Migration through Quartz-Packed Columns Journal of Nuclear Science and Technology, 1994, 31, 143-150.	1.3	3
99	Impact of Pseudocolloid Formation on Migration of Nuclides within Fractures Journal of Nuclear Science and Technology, 1994, 31, 623-625.	1.3	3
100	Colloid Formation and Sorption of Americium in the Water/Bentonite System. Radiochimica Acta, 1994, 66-67, 207-212.	1.2	9
101	TEMPERATURE EFFECT ON THE EXTRACTION OF Np(V) BYn-(OCTYLPHENYL)-N,N- DIISOBUTYLCARBAMOYLMETHYLPHOSPHINE OXIDE. Solvent Extraction and Ion Exchange, 1993, 11, 377-387.	2.0	3
102	Fast Transport of Colloidal Particles through Quartz-Packed Columns. Journal of Nuclear Science and Technology, 1993, 30, 1136-1144.	1.3	22
103	Analysis on Evolving Environments of Engineered Barriers of High-Level Radioactive Waste Repositories during the First 1,000 Years Nippon Genshiryoku Gakkaishi/Journal of the Atomic Energy Society of Japan, 1993, 35, 420-437.	0.0	5
104	Fast Transport of Colloidal Particles through Quartz-Packed Columns Journal of Nuclear Science and Technology, 1993, 30, 1136-1144.	1.3	13
105	Effect of TBP on Solvent Extraction of Np(V) with M-Octyl(phenyl)-, N-N Diisobutylcarbamoylmethylphosphine Oxide. Journal of Nuclear Science and Technology, 1992, 29, 263-268.	1.3	14
106	Oxidation of Pentavalent Neptunium by Nitrous Acid in CMPO-TBP-n-Dodecane Organic Solution. Journal of Nuclear Science and Technology, 1992, 29, 671-676.	1.3	12
107	Geochemical Behavior of Neptunium. Journal of Nuclear Science and Technology, 1992, 29, 706-718.	1.3	11
108	Solvent Extraction of Np (V) with CMPO from Nitric Acid Solutions Containing U (VI). Journal of Nuclear Science and Technology, 1992, 29, 1100-1107.	1.3	17

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109	Solvent Extraction of Np(V) with CMPO from Nitric Acid Solutions Containing U(VI) Journal of Nuclear Science and Technology, 1992, 29, 1100-1107.	1.3	2
110	Effect of TBP on Solvent Extraction of Np(V) with n-Octyl(pheny1)-N, N-Diisobutylcarbamoylmethylphosphine Oxide. Journal of Nuclear Science and Technology, 1992, 29, 263-283.	1.3	6
111	Oxidation of Pentavalent Neptunium by Nitrous Acid in CMPO-TBP-n-Dodecane Organic Solution Journal of Nuclear Science and Technology, 1992, 29, 671-676.	1.3	4
112	Speciation and solubility of neptunium in underground environments by paper electrophoresis. Journal of Radioanalytical and Nuclear Chemistry, 1988, 124, 383-395.	1.5	18