List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Effect of Nitrogen Impurity on Diamond Crystal Growth Processes. Crystal Growth and Design, 2010, 10, 3169-3175.	3.0	197
2	Mantle–slab interaction and redox mechanism of diamond formation. Proceedings of the National Academy of Sciences of the United States of America, 2013, 110, 20408-20413.	7.1	163
3	Germanium: a new catalyst for diamond synthesis and a new optically active impurity in diamond. Scientific Reports, 2015, 5, 14789.	3.3	145
4	Optical and microwave control of germanium-vacancy center spins in diamond. Physical Review B, 2017, 96, .	3.2	125
5	Germanium-Vacancy Color Center in Diamond as a Temperature Sensor. ACS Photonics, 2018, 5, 765-770.	6.6	105
6	The role of mantle ultrapotassic fluids in diamond formation. Proceedings of the National Academy of Sciences of the United States of America, 2007, 104, 9122-9127.	7.1	97
7	The effect of composition of mantle fluids/melts on diamond formation processes. Lithos, 2009, 112, 690-700.	1.4	79
8	Fluid regime and diamond formation in the reduced mantle: Experimental constraints. Geochimica Et Cosmochimica Acta, 2009, 73, 5820-5834.	3.9	79
9	The system K2CO3-MgCO3 at 6 GPa and 900-1450 ÂC. American Mineralogist, 2013, 98, 1593-1603.	1.9	79
10	High-temperature calibration of a multi-anvil high pressure apparatus. High Pressure Research, 2015, 35, 139-147.	1.2	71
11	High-Pressure Synthesis and Characterization of Ge-Doped Single Crystal Diamond. Crystal Growth and Design, 2016, 16, 3510-3518.	3.0	68
12	Diamond Growth and Morphology under the Influence of Impurity Adsorption. Crystal Growth and Design, 2013, 13, 5411-5419.	3.0	58
13	Conditions of diamond formation through carbonate-silicate interaction. European Journal of Mineralogy, 2005, 17, 207-214.	1.3	57
14	Partitioning of H2O between olivine and carbonate–silicate melts at 6.3 GPa and 1400 °C: Implications for kimberlite formation. Earth and Planetary Science Letters, 2013, 383, 58-67.	4.4	57
15	Effect of H ₂ O on Diamond Crystal Growth in Metal–Carbon Systems. Crystal Growth and Design, 2012, 12, 5571-5578.	3.0	55
16	High-pressure crystallization and properties of diamond from magnesium-based catalysts. CrystEngComm, 2017, 19, 4459-4475.	2.6	54
17	Carbonatite melt–peridotite interaction at 5.5–7.0 GPa: Implications for metasomatism in lithospheric mantle. Lithos, 2016, 248-251, 66-79.	1.4	49
18	Revealing of dislocations in diamond crystals by the selective etching method. Journal of Crystal Growth, 2006, 293, 469-474.	1.5	48

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19	Melting and subsolidus phase relations in the system Na2CO3-MgCO3ÂH2O at 6 GPa and the stability of Na2Mg(CO3)2 in the upper mantle. American Mineralogist, 2013, 98, 2172-2182.	1.9	47
20	Melting experiments on the Udachnaya kimberlite at 6.3–7.5GPa: Implications for the role of H2O in magma generation and formation of hydrous olivine. Geochimica Et Cosmochimica Acta, 2013, 101, 133-155.	3.9	47
21	Carbon and nitrogen speciation in nitrogen-rich C–O–H–N fluids at 5.5–7.8 GPa. Earth and Planetary Science Letters, 2017, 460, 234-243.	4.4	45
22	New experimental data on phase relations for the system Na2CO3-CaCO3 at 6 GPa and 900-1400 ÂC. American Mineralogist, 2013, 98, 2164-2171.	1.9	42
23	Incorporation of Large Impurity Atoms into the Diamond Crystal Lattice: EPR of Split-Vacancy Defects in Diamond. Crystals, 2017, 7, 237.	2.2	41
24	Monitoring diamond crystal growth, a combined experimental and SIMS study. European Journal of Mineralogy, 2008, 20, 365-374.	1.3	40
25	Diamond Growth from a Phosphorus–Carbon System at High Pressure High Temperature Conditions. Crystal Growth and Design, 2011, 11, 2599-2605.	3.0	38
26	Phase relations in the system FeCO3-CaCO3 at 6 GPa and 900-1700 ÂC and its relation to the system CaCO3-FeCO3-MgCO3. American Mineralogist, 2014, 99, 773-785.	1.9	38
27	EPR study of Si―and Geâ€related defects in HPHT diamonds synthesized from Mgâ€based solventâ€catalysts. Physica Status Solidi (A) Applications and Materials Science, 2016, 213, 2623-2628.	1.8	35
28	High-pressure synthesis and characterization of Sn-doped single crystal diamond. Carbon, 2019, 143, 769-775.	10.3	31
29	Effect of oxygen fugacity on the H2O storage capacity of forsterite in the carbon-saturated systems. Geochimica Et Cosmochimica Acta, 2010, 74, 4793-4806.	3.9	30
30	Phase relations on the K ₂ CO ₃ -CaCO ₃ -MgCO ₃ join at 6 GPa and 900–1400 °C: Implications for incipient melting in carbonated mantle domains. American Mineralogist, 2016, 101, 437-447.	1.9	28
31	Na-Ca carbonates synthesized under upper-mantle conditions: Raman spectroscopic and X-ray diffraction studies. European Journal of Mineralogy, 2015, 27, 175-184.	1.3	27
32	Crystal Growth of Diamond. , 2015, , 671-713.		27
33	Phase relationships in the system K2CO3-CaCO3 at 6 GPa and 900-1450 ÂC. American Mineralogist, 2015, 100, 223-232.	1.9	26
34	Revealing of planar defects and partial dislocations in large synthetic diamond crystals by the selective etching. Journal of Crystal Growth, 2007, 306, 458-464.	1.5	25
35	Graphitization of 13C enriched fine-grained graphitic material under high-pressure annealing. Carbon, 2019, 141, 323-330.	10.3	24
36	Crystal growth and perfection of large octahedral synthetic diamonds. Journal of Crystal Growth, 2011, 317, 32-38.	1.5	23

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37	The system Na ₂ CO ₃ – CaCO ₃ – MgCO ₃ < 6â€CPa and 900–1250°C and its relation to the partial melting of carbonated mantle. High Pressure Research, 2016, 36, 23-41.	nt 1.2	23
38	Carbon and Nitrogen Speciation in N-poor C-O-H-N Fluids at 6.3 GPa and 1100–1400 °C. Scientific R 2017, 7, 706.	eports,	23
39	Morphology of diamond crystals grown in magnesium-based systems at high temperatures and high pressures. Journal of Crystal Growth, 2015, 426, 276-282.	1.5	22
40	A DFT calculation of EPR parameters of a germanium-vacancy defect in diamond. Diamond and Related Materials, 2017, 76, 86-89.	3.9	22
41	Thermal expansion of iron carbides, Fe7C3 and Fe3C, at 297–911 K determined by in situ X-ray diffraction. Journal of Alloys and Compounds, 2015, 628, 102-106.	5.5	21
42	Effect of CO 2 on crystallization and properties of diamond from ultra-alkaline carbonate melt. Lithos, 2016, 265, 339-350.	1.4	21
43	Effect of crystal defects on diamond morphology during dissolution in the mantle. American Mineralogist, 2015, 100, 1528-1532.	1.9	19
44	Stability of phlogopite in ultrapotassic kimberlite-like systems at 5.5–7.5 GPa. Contributions To Mineralogy and Petrology, 2017, 172, 1.	3.1	19
45	Formation of various types of graphite inclusions in diamond: Experimental data. Lithos, 2009, 112, 683-689.	1.4	18
46	Distribution of OK1, N3 and NU1 defects in diamond crystals of different habits. European Journal of Mineralogy, 2012, 24, 645-650.	1.3	17
47	The system Na2CO3-FeCO3 at 6 GPa and its relation to the system Na2CO3-FeCO3-MgCO3. American Mineralogist, 2015, 100, 130-137.	1.9	17
48	Sulfidation of silicate mantle by reduced S-bearing metasomatic fluids and melts. Geology, 2016, 44, 271-274.	4.4	17
49	Iron carbide as a source of carbon for graphite and diamond formation under lithospheric mantle P-T parameters. Lithos, 2017, 286-287, 151-161.	1.4	17
50	Aluminum Nitride Crystal Growth from an Alâ^'N System at 6.0 GPa and 1800 °C. Crystal Growth and Design, 2010, 10, 2563-2570.	3.0	16
51	Effect of nitrogen impurity on the dislocation structure of large HPHT synthetic diamond crystals. Journal of Crystal Growth, 2014, 386, 162-167.	1.5	16
52	Diamond Crystallization from an Antimony–Carbon System under High Pressure and Temperature. Crystal Growth and Design, 2015, 15, 2539-2544.	3.0	16
53	HPHT growth and characterization of diamond from a copper-carbon system. Diamond and Related Materials, 2016, 69, 198-206.	3.9	16
54	Synthesis of diamonds with mineral, fluid and melt inclusions. Lithos, 2016, 265, 292-303.	1.4	16

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55	Phase relations in the Fe-Fe 3 C-Fe 3 N system at 7.8 GPa and 1350 °C: Implications for carbon and nitrogen hosts in Fe 0 -saturated upper mantle. Physics of the Earth and Planetary Interiors, 2017, 265, 43-53.	1.9	16
56	HPHT Diamond Crystallization in the Mg-Si-C System: Effect of Mg/Si Composition. Crystals, 2017, 7, 119.	2.2	16
57	Phase relations in the K2CO3–FeCO3 and MgCO3–FeCO3 systems at 6 GPa and 900–1700° C. European Journal of Mineralogy, 2015, 27, 487-499.	1.3	15
58	Experimental Petrology Applied to Natural Diamond Growth. Reviews in Mineralogy and Geochemistry, 2022, 88, 755-808.	4.8	15
59	Diamond crystallization from a tin–carbon system at HPHT conditions. Diamond and Related Materials, 2015, 58, 40-45.	3.9	13
60	The dislocation structure of diamond crystals grown on seeds in the Mg-C system. Diamond and Related Materials, 2016, 70, 1-6.	3.9	13
61	Multiscale characterization of 13C-enriched fine-grained graphitic materials for chemical and electrochemical applications. Carbon, 2017, 124, 161-169.	10.3	13
62	Carbon isotope fractionation during experimental crystallisation of diamond from carbonate fluid at mantle conditions. Contributions To Mineralogy and Petrology, 2015, 170, 1.	3.1	11
63	Sulfide Formation as a Result of Sulfate Subduction into Silicate Mantle (Experimental Modeling) Tj ETQq1 1 0.784	4314 rgBT 2.0	/Qverlock]
64	The Fe–C–O–H–N system at 6.3–7.8 GPa and 1200–1400°C: implications for deep carbon and nit cycles. Contributions To Mineralogy and Petrology, 2018, 173, 1.	rogen 3.1	11
65	Effect of Oxygen on Diamond Crystallization in Metal–Carbon Systems. ACS Omega, 2020, 5, 18376-18383.	3.5	11
66	Effect of sulfur on diamond growth and morphology in metal–carbon systems. CrystEngComm, 2020, 22, 5497-5508.	2.6	11
67	The role of water in generation of group II kimberlite magmas: Constraints from multiple saturation experiments. American Mineralogist, 2014, 99, 2292-2302.	1.9	10
68	Siliconâ€containing defects in HPHT diamond synthetized in Mg–Si–C system. Physica Status Solidi (A) Applications and Materials Science, 2015, 212, 2460-2462.	1.8	10
69	Wüstite stability in the presence of a CO 2 -fluid and a carbonate-silicate melt: Implications for the graphite/diamond formation and generation of Fe-rich mantle metasomatic agents. Lithos, 2016, 244, 20-29.	1.4	10
70	Experimental and Theoretical Evidence for Surface-Induced Carbon and Nitrogen Fractionation during Diamond Crystallization at High Temperatures and High Pressures. Crystals, 2017, 7, 190.	2.2	10
71	Effect of the solvent-catalyst composition on diamond crystallization in the Mg-Ge-C system. Diamond and Related Materials, 2018, 89, 1-9.	3.9	10
72	Decarbonation Reactions Involving Ankerite and Dolomite under upper Mantle P,T-Parameters: Experimental Modeling. Minerals (Basel, Switzerland), 2020, 10, 715.	2.0	10

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73	EPR study of the hydrogen center in HPHT diamonds grown in carbonate medium. Physica Status Solidi (A) Applications and Materials Science, 2014, 211, 2274-2278.	1.8	9
74	The influence of HTHP treatment on the OK1 and N3 centers in natural diamond crystals. Physica Status Solidi (A) Applications and Materials Science, 2015, 212, 2474-2479.	1.8	9
75	Effect of nitrogen impurity on etching of synthetic diamond crystals. Journal of Crystal Growth, 2015, 430, 71-74.	1.5	9
76	Effect of Rare-Earth Element Oxides on Diamond Crystallization in Mg-Based Systems. Crystals, 2019, 9, 300.	2.2	9
77	X-ray topography of diamond using forbidden reflections: which defects do we really see?. Journal of Applied Crystallography, 2011, 44, 65-72.	4.5	8
78	Photoluminescence of HPHT diamonds synthesized in the Mg-Ge-C system. Diamond and Related Materials, 2017, 79, 145-149.	3.9	8
79	Specific Internal Structure of Diamonds from Zarnitsa Kimberlite Pipe. Crystals, 2017, 7, 133.	2.2	8
80	Step Patterns on {100} Faces of Diamond Crystals As-Grown in Mg-Based Systems. Crystal Growth and Design, 2018, 18, 152-158.	3.0	8
81	Dislocation etching of diamond crystals grown in Mg-C system with the addition of silicon. Diamond and Related Materials, 2018, 88, 67-73.	3.9	8
82	Hydrogenation of carbon at 5.5–7.8†GPa and 1100–1400†°C: Implications to formation of hydrocarbon in reduced mantles of terrestrial planets. Physics of the Earth and Planetary Interiors, 2019, 291, 12-23.	ns 1.9	8
83	Solubility of carbon and nitrogen in a sulfur-bearing iron melt: Constraints for siderophile behavior at upper mantle conditions. American Mineralogist, 2019, 104, 1857-1865.	1.9	8
84	Diamonds from the Mir Pipe (Yakutia): Spectroscopic Features and Annealing Studies. Crystals, 2021, 11, 366.	2.2	8
85	Fate of fluids at the base of subcratonic lithosphere: Experimental constraints at 5.5–7.8â€ ⁻ GPa and 1150–1350 deg C. Lithos, 2018, 318-319, 419-433.	1.4	7
86	Diamond formation in an electric field under deep Earth conditions. Science Advances, 2021, 7, .	10.3	7
87	Unusual growth macrolayers on {100} faces of diamond crystals from magnesium-based systems. Journal of Crystal Growth, 2016, 455, 76-82.	1.5	6
88	Diamond crystallization in a CO 2 -rich alkaline carbonate melt with a nitrogen additive. Journal of Crystal Growth, 2016, 449, 119-128.	1.5	6
89	An effect of reduced S-rich fluids on diamond formation under mantle-slab interaction. Lithos, 2019, 336-337, 27-39.	1.4	6
90	Cymrite as mineral clathrate: An overlooked redox insensitive transporter of nitrogen in the mantle. Gondwana Research, 2020, 79, 70-86.	6.0	6

YURY N PALYANOV

#	Article	IF	CITATIONS
91	Effect of HPHT Treatment on Spectroscopic Features of Natural Type Ib-IaA Diamonds Containing Y Centers. Crystals, 2020, 10, 378.	2.2	6
92	Rare-earth metal catalysts for high-pressure synthesis of rare diamonds. Scientific Reports, 2021, 11, 8421.	3.3	6
93	<scp>EPR</scp> study of impurity defects in diamonds grown in carbonate medium. Physica Status Solidi (A) Applications and Materials Science, 2013, 210, 2074-2077.	1.8	5
94	Graphite and diamond formation via the interaction of iron carbide and Fe,Ni-sulfide under mantle P–T parameters. Doklady Earth Sciences, 2016, 471, 1144-1148.	0.7	5
95	Phase Relations in the Harzburgite–Hydrous Carbonate Melt at 5.5–7.5 GPa and 1200–1350°Ðj. Petrolog 2018, 26, 575-587.	у _{0.9}	5
96	Spin Relaxation of the Neutral Germaniumâ€Vacancy Center in Diamond. Physica Status Solidi (A) Applications and Materials Science, 2018, 215, 1800193.	1.8	5
97	Experimental Modeling of Silicate and Carbonate Sulfidation under Lithospheric Mantle P,T-Parameters. Minerals (Basel, Switzerland), 2019, 9, 425.	2.0	5
98	Crystallization of Diamond from Melts of Europium Salts. Crystals, 2020, 10, 376.	2.2	5
99	EPR of synthetic diamond heavily doped with phosphorus. Physica Status Solidi (A) Applications and Materials Science, 2015, 212, 2568-2571.	1.8	4
100	Stability of methane in reduced C–O–H fluid at 6.3 GPa and 1300–1400°C. Doklady Earth Sciences, 2017 474, 680-683.	0.7	4
101	Optical and electrical properties of synthetic single-crystal diamond under high-fluence ion irradiation. Journal of Surface Investigation, 2017, 11, 619-624.	0.5	4
102	Formation of Water-Bearing Defects in Olivine in the Presence of Water–Hydrocarbon Fluid at 6.3 GPa and 1200°C. Doklady Earth Sciences, 2018, 483, 1451-1453.	0.7	4
103	Correction to Germanium-Vacancy Color Center in Diamond as a Temperature Sensor. ACS Photonics, 2018, 5, 4710-4710.	6.6	4
104	High-pressure synthesis and characterization of diamond from europium containing systems. Carbon, 2021, 182, 815-824.	10.3	4
105	Crystallomorphological and Crystallochemical Indicators of Diamond Formation Conditions. Crystallography Reports, 2021, 66, 142-155.	0.6	4
106	Nitrogen and hydrogen aggregation in natural octahedral and cuboid diamonds. Geochemical Journal, 2017, 51, 181-192.	1.0	4
107	Distribution of light alkanes in the reaction of graphite hydrogenation at pressure of 0.1–7.8â€GPa and temperatures of 1000–1350°C. High Pressure Research, 2018, 38, 468-481.	1.2	3

Graphite and Diamond Formation in the Carbide $\hat{a}\in$ "Oxide $\hat{a}\in$ "Carbonate Interactions (Experimental) Tj ETQq0 0 0 rgBT /Overlock 10 Tf 5 2.0 rds)

#	Article	IF	CITATIONS
109	Carbon Isotope Composition of Diamond Crystals Grown Via Redox Mechanism. Geochemistry International, 2018, 56, 1398-1404.	0.7	3
110	Manifestation of diamond sectoriality during dissolution and graphitization. Journal of Crystal Growth, 2018, 502, 1-6.	1.5	3
111	Processes and Conditions of the Origin for Fe3+-Bearing Magnesiowüstite under Lithospheric Mantle Pressures and Temperatures. Minerals (Basel, Switzerland), 2019, 9, 474.	2.0	3
112	Magnetic Properties of 1D Iron–Sulfur Compounds Formed Inside Singleâ€Walled Carbon Nanotubes. Physica Status Solidi - Rapid Research Letters, 2020, 14, 2000291.	2.4	3
113	Formation of Spessartine and CO2 via Rhodochrosite Decarbonation along a Hot Subduction P-T Path. Minerals (Basel, Switzerland), 2020, 10, 703.	2.0	3
114	Ranges of 10–350 keV H and H 2 ions in (1 1 1) diamond. Nuclear Instruments & Methods in Physics Research B, 2017, 406, 634-637.	1.4	2
115	Phases of the Fe–C–N system as hosts of mantle carbon and nitrogen: Experimental studies at 7.8 GPa and 1350°C. Doklady Earth Sciences, 2017, 475, 780-783.	0.7	2
116	Influence of a silicon impurity on growth of diamond crystals in the Mg-C system. Diamond and Related Materials, 2018, 87, 27-34.	3.9	2
117	Phase Relations in the FeO-Fe3C-Fe3N System at 7.8 GPa and 1350 °C: Implications for Oxidation of Native Iron at 250 km. Minerals (Basel, Switzerland), 2020, 10, 984.	2.0	2
118	Diamond formation during sulfidation of metal–carbon melts. Diamond and Related Materials, 2021, 120, 108660.	3.9	2
119	Phase relations in the Fe-Fe ₃ C-Fe ₃ N system at 7.8â€GPa and 1150°C: implications for C and N hosts in metal-saturated mantle. High Pressure Research, 2021, 41, 392-404.	1.2	2
120	Experimental Modeling of Ankerite–Pyrite Interaction under Lithospheric Mantle P–T Parameters: Implications for Graphite Formation as a Result of Ankerite Sulfidation. Minerals (Basel, Switzerland), 2021, 11, 1267.	2.0	2
121	Raman scattering in the submicrometer diamond membrane formed by the lift-off technique. Bulletin of the Lebedev Physics Institute, 2017, 44, 210-214.	0.6	1
122	Formation of the Fe,Mg-Silicates, Fe0, and Graphite (Diamond) Assemblage as a Result of Cohenite Oxidation under Lithospheric Mantle Conditions. Doklady Earth Sciences, 2018, 479, 335-338.	0.7	1
123	Experimental Modeling of C0-Forming Processes Involving Cohenite and CO2-Fluid in a Silicate Mantle. Doklady Earth Sciences, 2018, 483, 1427-1430.	0.7	1
124	Conditions of Formation of Iron–Carbon Melt Inclusions in Garnet and Orthopyroxene under P-T Conditions of Lithospheric Mantle. Petrology, 2018, 26, 565-574.	0.9	1
125	The Many Facets of Diamond Crystals. Crystals, 2018, 8, 72.	2.2	1