

Karen Chan

List of Publications by Citations

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The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

91
papers

10,382
citations

46
h-index

101
g-index

125
ext. papers

13,358
ext. citations

13
avg, IF

6.84
L-index

#	Paper	IF	Citations
91	Progress and Perspectives of Electrochemical CO Reduction on Copper in Aqueous Electrolyte. <i>Chemical Reviews</i> , 2019 , 119, 7610-7672	68.1	1244
90	Designing an improved transition metal phosphide catalyst for hydrogen evolution using experimental and theoretical trends. <i>Energy and Environmental Science</i> , 2015 , 8, 3022-3029	35.4	671
89	Transition-metal doped edge sites in vertically aligned MoS ₂ catalysts for enhanced hydrogen evolution. <i>Nano Research</i> , 2015 , 8, 566-575	10	478
88	Electrochemical Ammonia Synthesis—The Selectivity Challenge. <i>ACS Catalysis</i> , 2017 , 7, 706-709	13.1	442
87	Theoretical Insights into a CO Dimerization Mechanism in CO ₂ Electroreduction. <i>Journal of Physical Chemistry Letters</i> , 2015 , 6, 2032-7	6.4	419
86	Electrochemical Activation of CO through Atomic Ordering Transformations of AuCu Nanoparticles. <i>Journal of the American Chemical Society</i> , 2017 , 139, 8329-8336	16.4	392
85	Metal ion cycling of Cu foil for selective C ₂ coupling in electrochemical CO ₂ reduction. <i>Nature Catalysis</i> , 2018 , 1, 111-119	36.5	383
84	Promoter Effects of Alkali Metal Cations on the Electrochemical Reduction of Carbon Dioxide. <i>Journal of the American Chemical Society</i> , 2017 , 139, 11277-11287	16.4	381
83	Understanding trends in electrochemical carbon dioxide reduction rates. <i>Nature Communications</i> , 2017 , 8, 15438	17.4	369
82	Active edge sites in MoSe ₂ and WSe ₂ catalysts for the hydrogen evolution reaction: a density functional study. <i>Physical Chemistry Chemical Physics</i> , 2014 , 16, 13156-64	3.6	364
81	Electric Field Effects in Electrochemical CO ₂ Reduction. <i>ACS Catalysis</i> , 2016 , 6, 7133-7139	13.1	275
80	Theoretical insights into the hydrogen evolution activity of layered transition metal dichalcogenides. <i>Surface Science</i> , 2015 , 640, 133-140	1.8	256
79	Transition-Metal Single Atoms in a Graphene Shell as Active Centers for Highly Efficient Artificial Photosynthesis. <i>CheM</i> , 2017 , 3, 950-960	16.2	249
78	Understanding cation effects in electrochemical CO ₂ reduction. <i>Energy and Environmental Science</i> , 2019 , 12, 3001-3014	35.4	231
77	Machine-Learning Methods Enable Exhaustive Searches for Active Bimetallic Facets and Reveal Active Site Motifs for CO ₂ Reduction. <i>ACS Catalysis</i> , 2017 , 7, 6600-6608	13.1	224
76	Electrochemical Barriers Made Simple. <i>Journal of Physical Chemistry Letters</i> , 2015 , 6, 2663-8	6.4	211
75	pH effects on the electrochemical reduction of CO towards C products on stepped copper. <i>Nature Communications</i> , 2019 , 10, 32	17.4	207

74	Molybdenum Sulfides and Selenides as Possible Electrocatalysts for CO ₂ Reduction. <i>ChemCatChem</i> , 2014 , 6, 1899-1905	5.2	194
73	How Doped MoS ₂ Breaks Transition-Metal Scaling Relations for CO ₂ Electrochemical Reduction. <i>ACS Catalysis</i> , 2016 , 6, 4428-4437	13.1	193
72	Electrochemical Carbon Monoxide Reduction on Polycrystalline Copper: Effects of Potential, Pressure, and pH on Selectivity toward Multicarbon and Oxygenated Products. <i>ACS Catalysis</i> , 2018 , 8, 7445-7454	13.1	175
71	CO-CO coupling on Cu facets: Coverage, strain and field effects. <i>Surface Science</i> , 2016 , 654, 56-62	1.8	142
70	Potential Dependence of Electrochemical Barriers from ab Initio Calculations. <i>Journal of Physical Chemistry Letters</i> , 2016 , 7, 1686-90	6.4	138
69	Rational design of MoS ₂ catalysts: tuning the structure and activity via transition metal doping. <i>Catalysis Science and Technology</i> , 2015 , 5, 246-253	5.5	128
68	Double layer charging driven carbon dioxide adsorption limits the rate of electrochemical carbon dioxide reduction on Gold. <i>Nature Communications</i> , 2020 , 11, 33	17.4	107
67	Confined local oxygen gas promotes electrochemical water oxidation to hydrogen peroxide. <i>Nature Catalysis</i> , 2020 , 3, 125-134	36.5	106
66	pH in atomic scale simulations of electrochemical interfaces. <i>Physical Chemistry Chemical Physics</i> , 2013 , 15, 10321-5	3.6	100
65	Challenges in Modeling Electrochemical Reaction Energetics with Polarizable Continuum Models. <i>ACS Catalysis</i> , 2019 , 9, 920-931	13.1	100
64	Barriers of Electrochemical CO ₂ Reduction on Transition Metals. <i>Organic Process Research and Development</i> , 2016 , 20, 1424-1430	3.9	98
63	Catalytic Polysulfide Conversion and Physiochemical Confinement for Lithium Sulfur Batteries. <i>Advanced Energy Materials</i> , 2020 , 10, 1904010	21.8	94
62	Synergistic enhancement of electrocatalytic CO reduction to C oxygenates at nitrogen-doped nanodiamonds/Cu interface. <i>Nature Nanotechnology</i> , 2020 , 15, 131-137	28.7	92
61	Is There Anything Better than Pt for HER?. <i>ACS Energy Letters</i> , 2021 , 6, 1175-1180	20.1	83
60	A Pore-Scale Model of Oxygen Reduction in Ionomer-Free Catalyst Layers of PEFCs. <i>Journal of the Electrochemical Society</i> , 2011 , 158, B18	3.9	73
59	Influence of Atomic Surface Structure on the Activity of Ag for the Electrochemical Reduction of CO ₂ to CO. <i>ACS Catalysis</i> , 2019 , 9, 4006-4014	13.1	72
58	Theoretical Investigations into Defected Graphene for Electrochemical Reduction of CO ₂ . <i>ACS Sustainable Chemistry and Engineering</i> , 2017 , 5, 11080-11085	8.3	68
57	On the pH dependence of electrochemical proton transfer barriers. <i>Catalysis Today</i> , 2016 , 262, 36-40	5.3	68

56	Trends in the Catalytic Activity of Hydrogen Evolution during CO ₂ Electroreduction on Transition Metals. <i>ACS Catalysis</i> , 2018 , 8, 3035-3040	13.1	67
55	pH Effects on Hydrogen Evolution and Oxidation over Pt(111): Insights from First-Principles. <i>ACS Catalysis</i> , 2019 , 9, 6194-6201	13.1	64
54	Mechanistic insights into nitrogen fixation by nitrogenase enzymes. <i>Physical Chemistry Chemical Physics</i> , 2015 , 17, 29541-7	3.6	62
53	Understanding the Reactivity of Layered Transition-Metal Sulfides: A Single Electronic Descriptor for Structure and Adsorption. <i>Journal of Physical Chemistry Letters</i> , 2014 , 5, 3884-9	6.4	57
52	Solvation at metal/water interfaces: An ab initio molecular dynamics benchmark of common computational approaches. <i>Journal of Chemical Physics</i> , 2020 , 152, 144703	3.9	56
51	Unified Approach to Implicit and Explicit Solvent Simulations of Electrochemical Reaction Energetics. <i>Journal of Chemical Theory and Computation</i> , 2019 , 15, 6895-6906	6.4	51
50	Role of Subsurface Oxygen on Cu Surfaces for CO ₂ Electrochemical Reduction. <i>Journal of Physical Chemistry C</i> , 2018 , 122, 16209-16215	3.8	49
49	Dipole-Field Interactions Determine the CO ₂ Reduction Activity of 2D FeN ₄ Single-Atom Catalysts. <i>ACS Catalysis</i> , 2020 , 10, 7826-7835	13.1	48
48	The Predominance of Hydrogen Evolution on Transition Metal Sulfides and Phosphides under CO ₂ Reduction Conditions: An Experimental and Theoretical Study. <i>ACS Energy Letters</i> , 2018 , 3, 1450-1457	20.1	48
47	Guiding Electrochemical Carbon Dioxide Reduction toward Carbonyls Using Copper Silver Thin Films with Interphase Miscibility. <i>ACS Energy Letters</i> , 2018 , 3, 2947-2955	20.1	47
46	Electric Field Effects in Oxygen Reduction Kinetics: Rationalizing pH Dependence at the Pt(111), Au(111), and Au(100) Electrodes. <i>Journal of Physical Chemistry C</i> , 2020 , 124, 14581-14591	3.8	44
45	Selective reduction of CO to acetaldehyde with CuAg electrocatalysts. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2020 , 117, 12572-12575	11.5	43
44	Theoretical Investigations of the Electrochemical Reduction of CO on Single Metal Atoms Embedded in Graphene. <i>ACS Central Science</i> , 2017 , 3, 1286-1293	16.8	41
43	Solvent-Adsorbate Interactions and Adsorbate-Specific Solvent Structure in Carbon Dioxide Reduction on a Stepped Cu Surface. <i>Journal of Physical Chemistry C</i> , 2019 , 123, 5999-6009	3.8	39
42	Understanding the Influence of [EMIM]Cl on the Suppression of the Hydrogen Evolution Reaction on Transition Metal Electrodes. <i>Langmuir</i> , 2017 , 33, 9464-9471	4	36
41	Self-Selective Catalyst Synthesis for CO ₂ Reduction. <i>Joule</i> , 2019 , 3, 1927-1936	27.8	35
40	Theoretical Study of EMIM ⁺ Adsorption on Silver Electrode Surfaces. <i>Journal of Physical Chemistry C</i> , 2015 , 119, 20023-20029	3.8	33
39	Understanding the apparent fractional charge of protons in the aqueous electrochemical double layer. <i>Nature Communications</i> , 2018 , 9, 3202	17.4	31

38	Scaling Relations for Adsorption Energies on Doped Molybdenum Phosphide Surfaces. <i>ACS Catalysis</i> , 2017 , 7, 2528-2534	13.1	30
37	A few basic concepts in electrochemical carbon dioxide reduction. <i>Nature Communications</i> , 2020 , 11, 5954	17.4	30
36	Unified mechanistic understanding of CO ₂ reduction to CO on transition metal and single atom catalysts. <i>Nature Catalysis</i> ,	36.5	29
35	Scaling Relations on Basal Plane Vacancies of Transition Metal Dichalcogenides for CO ₂ Reduction. <i>Journal of Physical Chemistry C</i> , 2019 , 123, 4256-4261	3.8	28
34	A Two-Dimensional MoS Catalysis Transistor by Solid-State Ion Gating Manipulation and Adjustment (SIGMA). <i>Nano Letters</i> , 2019 , 19, 7293-7300	11.5	24
33	Modeling Hydrogen Evolution Reaction Kinetics through Explicit Water/Metal Interfaces. <i>Journal of Physical Chemistry C</i> , 2020 , 124, 28083-28092	3.8	24
32	Fingerprint Voltammograms of Copper Single Crystals under Alkaline Conditions: A Fundamental Mechanistic Analysis. <i>Journal of Physical Chemistry Letters</i> , 2020 , 11, 1450-1455	6.4	23
31	Facile Electron Transfer to CO ₂ during Adsorption at the Metal/Solution Interface. <i>Journal of Physical Chemistry C</i> , 2019 , 123, 29278-29283	3.8	23
30	Practical Considerations for Continuum Models Applied to Surface Electrochemistry. <i>ChemPhysChem</i> , 2019 , 20, 3074-3080	3.2	21
29	Electrolyte Effects on the Stability of Ni-Mo Cathodes for the Hydrogen Evolution Reaction. <i>ChemSusChem</i> , 2019 , 12, 3491-3500	8.3	20
28	Preparation of open-through anodized aluminium oxide films with a clean method. <i>Nanotechnology</i> , 2007 , 18, 245304	3.4	20
27	Energy-entropy competition in cation-hydroxyl interactions at the liquid water-Pt(111) interface. <i>Chemical Communications</i> , 2020 , 56, 427-430	5.8	17
26	Rational design of stable sulfur vacancies in molybdenum disulfide for hydrogen evolution. <i>Journal of Catalysis</i> , 2020 , 382, 320-328	7.3	15
25	Ultrastable molybdenum disulfide-based electrocatalyst for hydrogen evolution in acidic media. <i>Journal of Power Sources</i> , 2020 , 456, 227998	8.9	13
24	Water balance model for polymer electrolyte fuel cells with ultrathin catalyst layers. <i>Physical Chemistry Chemical Physics</i> , 2014 , 16, 2106-17	3.6	13
23	Atomistic Insight into Cation Effects on Binding Energies in Cu-Catalyzed Carbon Dioxide Reduction. <i>Journal of Physical Chemistry C</i> , 2020 , 124, 24765-24775	3.8	12
22	Spin Uncoupling in Chemisorbed OCCO and CO ₂ : Two High-Energy Intermediates in Catalytic CO ₂ Reduction. <i>Journal of Physical Chemistry C</i> , 2018 , 122, 12251-12258	3.8	12
21	Impedance Model of Oxygen Reduction in Water-Flooded Pores of Ionomer-Free PEFC Catalyst Layers. <i>Journal of the Electrochemical Society</i> , 2011 , 159, B155-B164	3.9	12

20	Theoretical Investigations of Transition Metal Surface Energies under Lattice Strain and CO Environment. <i>Journal of Physical Chemistry C</i> , 2018 , 122, 14481-14487	3.8	12
19	Improving the intrinsic activity of electrocatalysts for sustainable energy conversion: where are we and where can we go?. <i>Chemical Science</i> , 2021 , 13, 14-26	9.4	11
18	Generalizable Trends in Electrochemical Protonation Barriers. <i>Journal of Physical Chemistry Letters</i> , 2021 , 12, 5193-5200	6.4	11
17	The determination of the HOR/HER reaction mechanism from experimental kinetic data. <i>Physical Chemistry Chemical Physics</i> , 2021 ,	3.6	6
16	Electrochemical Oxidation of CO on Cu Single Crystals under Alkaline Conditions. <i>ACS Energy Letters</i> , 2020 , 5, 3437-3442	20.1	6
15	Insights into the Hydrogen Evolution Reaction on 2D Transition-Metal Dichalcogenides. <i>Journal of Physical Chemistry C</i> , 2022 , 126, 5151-5158	3.8	6
14	Using pH Dependence to Understand Mechanisms in Electrochemical CO Reduction. <i>ACS Catalysis</i> , 2021 , 11, 4344-4357	4.3	6
13	Implications of the fractional charge of hydroxide at the electrochemical interface. <i>Physical Chemistry Chemical Physics</i> , 2020 , 22, 6964-6969	3.6	5
12	How pH affects electrochemical processes.. <i>Science</i> , 2022 , 375, 379-380	33.3	5
11	Computational Screening of Single and Di-Atom Catalysts for Electrochemical CO ₂ Reduction. <i>ACS Catalysis</i> , 2021 , 11, 4818-4824	13.1	5
10	Understanding the complementarities of surface-enhanced infrared and Raman spectroscopies in CO adsorption and electrochemical reduction.. <i>Nature Communications</i> , 2022 , 13, 2656	17.4	5
9	Transition Metal Arsenide Catalysts for the Hydrogen Evolution Reaction. <i>Journal of Physical Chemistry C</i> , 2019 , 123, 24007-24012	3.8	4
8	Towards constant potential modeling of CO-CO coupling at liquid water-Cu(1 0 0) interfaces. <i>Journal of Catalysis</i> , 2021 , 396, 251-260	7.3	4
7	Mechanism for acetate formation in CO(2) reduction on Cu: Selectivity trends with pH and nanostructuring derive from mass transport		3
6	Interaction of CO with Gold in an Electrochemical Environment. <i>Journal of Physical Chemistry C</i> , 2021 , 125, 17684-17689	3.8	3
5	Transients in Electrochemical CO Reduction Explained by Mass Transport of Buffers. <i>ACS Catalysis</i> , 2021 , 11, 5155-5161	5.6	3
4	Understanding the reaction mechanism of Kolbe electrolysis on Pt anodes. <i>Chem Catalysis</i> , 2022 ,		2
3	How to extract adsorption energies, adsorbate-adsorbate interaction parameters and saturation coverages from temperature programmed desorption experiments. <i>Physical Chemistry Chemical Physics</i> , 2021 , 23, 24396-24402	3.6	1

- 2 OH Binding Energy as a Universal Descriptor of the Potential of Zero Charge on Transition Metal Surfaces. *Journal of Physical Chemistry C*, **2022**, 126, 5521-5528 3.8 1
- 1 Nanoscale Phenomena in Catalyst Layers for PEM Fuel Cells: From Fundamental Physics to Benign Design **2010**, 317-369