Peter Glarborg

List of Publications by Year
in descending order

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274 papers	18,364 citations	13068 68 h-index	17055 122 g-index
279 all docs	279 docs citations	279 times ranked	7201 citing authors

#	Article	IF	Citations
1	Modeling nitrogen chemistry in combustion. Progress in Energy and Combustion Science, 2018, 67, 31-68.	15.8	980
2	Oxy-fuel combustion of solid fuels. Progress in Energy and Combustion Science, 2010, 36, 581-625.	15.8	940
3	Fuel nitrogen conversion in solid fuel fired systems. Progress in Energy and Combustion Science, 2003, 29, 89-113.	15.8	764
4	Kinetic Modeling of Hydrocarbon/Nitric Oxide Interactions in a Flow Reactor. Combustion and Flame, 1998, 115, 1-27.	2.8	475
5	Review on Ammonia as a Potential Fuel: From Synthesis to Economics. Energy & Fuels, 2021, 35, 6964-7029.	2.5	403
6	Kinetic modeling and sensitivity analysis of nitrogen oxide formation in well-stirred reactors. Combustion and Flame, 1986, 65, 177-202.	2.8	398
7	An experimental and kinetic modeling study of premixed NH3/CH4/O2/Ar flames at low pressure. Combustion and Flame, 2009, 156, 1413-1426.	2.8	359
8	Chemical Effects of a High CO₂ Concentration in Oxy-Fuel Combustion of Methane. Energy & Fuels, 2008, 22, 291-296.	2.5	348
9	Ammonia chemistry in oxy-fuel combustion of methane. Combustion and Flame, 2009, 156, 1937-1949.	2.8	327
10	Release of K, Cl, and S during Pyrolysis and Combustion of High-Chlorine Biomass. Energy & Fuels, 2011, 25, 4961-4971.	2.5	312
11	The oxidation of hydrogen cyanide and related chemistry. Progress in Energy and Combustion Science, 2008, 34, 1-46.	15.8	305
12	The role of NNH in NO formation and control. Combustion and Flame, 2011, 158, 774-789.	2.8	304
13	Release to the Gas Phase of Inorganic Elements during Wood Combustion. Part 2: Influence of Fuel Composition. Energy & Fuels, 2008, 22, 1598-1609.	2.5	252
14	Ammonia chemistry below 1400ÂK under fuel-rich conditions in a flow reactor. Combustion and Flame, 2004, 136, 501-518.	2.8	228
15	Global Combustion Mechanisms for Use in CFD Modeling under Oxy-Fuel Conditions. Energy & Fuels, 2009, 23, 1379-1389.	2.5	223
16	Ammonia oxidation at high pressure and intermediate temperatures. Fuel, 2016, 181, 358-365.	3.4	223
17	Mechanism and modeling of the formation of gaseous alkali sulfates. Combustion and Flame, 2005, 141, 22-39.	2.8	203
18	Numerical modeling of straw combustion in a fixed bed. Fuel, 2005, 84, 389-403.	3.4	181

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19	Release to the Gas Phase of Inorganic Elements during Wood Combustion. Part 1:â€‰ Development and Evaluation of Quantification Methods. Energy & Fuels, 2006, 20, 964-978.	2.5	177
20	Experimental measurements and kinetic modeling of CO/H₂/O₂/NO_x conversion at high pressure. International Journal of Chemical Kinetics, 2008, 40, 454-480.	1.0	164
21	Hidden interactionsâ€”Trace species governing combustion and emissions. Proceedings of the Combustion Institute, 2007, 31, 77-98.	2.4	161
22	Formation of polycyclic aromatic hydrocarbons and soot in fuel-rich oxidation of methane in a laminar flow reactor. Combustion and Flame, 2004, 136, 91-128.	2.8	157
23	High-pressure oxidation of methane. Combustion and Flame, 2016, 172, 349-364.	2.8	157
24	Modeling the thermal DENOx process in flow reactors. Surface effects and Nitrous Oxide formation. International Journal of Chemical Kinetics, 1994, 26, 421-436.	1.0	156
25	Inhibition and sensitization of fuel oxidation by SO2. Combustion and Flame, 2001, 127, 2234-2251.	2.8	150
26	Shedding of ash deposits. Progress in Energy and Combustion Science, 2009, 35, 31-56.	15.8	148
27	Modelling and experiments of straw combustion in a grate furnace. Biomass and Bioenergy, 2000, 19, 199-208.	2.9	143
28	Impact of SO2 and NO on CO oxidation under post-flame conditions. International Journal of Chemical Kinetics, 1996, 28, 773-790.	1.0	140
29	Modeling the thermal De-NOx process: Closing in on a final solution. International Journal of Chemical Kinetics, 1999, 31, 757-765.	1.0	135
30	The thermal DeNOx process: Influence of partial pressures and temperature. Chemical Engineering Science, 1995, 50, 1455-1466.	1.9	131
31	Experimental and numerical analysis of the autoignition behavior of NH3 and NH3/H2 mixtures at high pressure. Combustion and Flame, 2020, 215, 134-144.	2.8	130
32	Release of Chlorine and Sulfur during Biomass Torrefaction and Pyrolysis. Energy & Fuels, 2014, 28, 3738-3746.	2.5	128
33	Nitrogen chemistry during burnout in fuel-staged combustion. Combustion and Flame, 1996, 107, 211-222.	2.8	123
34	Experimental and kinetic modeling study of the oxidation of benzene. International Journal of Chemical Kinetics, 2000, 32, 498-522.	1.0	121
35	Sensitizing effects of NOx on CH4 oxidation at high pressure. Combustion and Flame, 2008, 154, 529-545.	2.8	119
36	Reburning chemistry: a kinetic modeling study. Industrial & Engineering Chemistry Research, 1992, 31, 1477-1490.	1.8	118

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37	Low temperature interactions between hydrocarbons and nitric oxide: An experimental study. Combustion and Flame, 1997, 109, 25-36.	2.8	111
38	Reburn Chemistry in Oxy-fuel Combustion of Methane. Energy & Fuels, 2009, 23, 3565-3572.	2.5	111
39	The reaction of ammonia with nitrogen dioxide in a flow reactor: Implications for the NH2 + NO2 reaction. International Journal of Chemical Kinetics, 1995, 27, 1207-1220.	1.0	110
40	Influence of process parameters on nitrogen oxide formation in pulverized coal burners. Progress in Energy and Combustion Science, 1997, 23, 349-377.	15.8	110
41	Heat transfer in ash deposits: A modelling tool-box. Progress in Energy and Combustion Science, 2005, 31, 371-421.	15.8	108
42	Nitric Oxide Reduction by Non-hydrocarbon Fuels. Implications for Reburning with Gasification Gases. Energy & Fuels, 2000, 14, 828-838.	2.5	107
43	Low temperature oxidation of methane: the influence of nitrogen oxides. Combustion Science and Technology, 2000, 151, 31-71.	1.2	102
44	Ignition delay times of NH3 /DME blends at high pressure and low DME fraction: RCM experiments and simulations. Combustion and Flame, 2021, 227, 120-134.	2.8	97
45	Ammonia conversion and NOx formation in laminar coflowing nonpremixed methane-air flames. Combustion and Flame, 2002, 131, 285-298.	2.8	95
46	An exploratory study of alkali sulfate aerosol formation during biomass combustion. Fuel, 2008, 87, 1591-1600.	3.4	95
47	Influence of fast pyrolysis conditions on yield and structural transformation of biomass chars. Fuel Processing Technology, 2015, 140, 205-214.	3.7	94
48	An experimental study of biomass ignitionâˆ†. Fuel, 2003, 82, 825-833.	3.4	92
49	Screening of NiFe₂O₄ Nanoparticles as Oxygen Carrier in Chemical Looping Hydrogen Production. Energy & Fuels, 2016, 30, 4251-4262.	2.5	91
50	Experimental and kinetic modeling study of the effect of NO and SO2 on the oxidation of CO?H2 mixtures. International Journal of Chemical Kinetics, 2003, 35, 564-575.	1.0	90
51	Formation and reduction of nitric oxide in fixed-bed combustion of straw. Fuel, 2006, 85, 705-716.	3.4	90
52	Release of K, Cl, and S during combustion and co-combustion with wood of high-chlorine biomass in bench and pilot scale fuel beds. Proceedings of the Combustion Institute, 2013, 34, 2363-2372.	2.4	90
53	Autoignition studies of NH3/CH4 mixtures at high pressure. Combustion and Flame, 2020, 218, 19-26.	2.8	90
54	Kinetics of homogeneous nitrous oxide decomposition. Combustion and Flame, 1994, 99, 523-532.	2.8	89

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55	Combustion chemistry in the twenty-first century: Developing theory-informed chemical kinetics models. Progress in Energy and Combustion Science, 2021, 83, 100886.	15.8	89
56	Mechanism and modeling of hydrogen cyanide oxidation in a flow reactor. Combustion and Flame, 1994, 99, 475-483.	2.8	87
57	Hydrogen oxidation at high pressure and intermediate temperatures: Experiments and kinetic modeling. Proceedings of the Combustion Institute, 2015, 35, 553-560.	2.4	87
58	Mechanisms of radical removal by SO2. Proceedings of the Combustion Institute, 2007, 31, 339-347.	2.4	84
59	Effects of several types of biomass fuels on the yield, nanostructure and reactivity of soot from fast pyrolysis at high temperatures. Applied Energy, 2016, 171, 468-482.	5.1	82
60	A flow reactor study of HNCO oxidation chemistry. Combustion and Flame, 1994, 98, 241-258.	2.8	77
61	High-pressure oxidation of ethane. Combustion and Flame, 2017, 182, 150-166.	2.8	76
62	Oxidation of formaldehyde and its interaction with nitric oxide in a flow reactor. Combustion and Flame, 2003, 132, 629-638.	2.8	74
63	Fuel-nitrogen conversion in the combustion of small amines using dimethylamine and ethylamine as biomass-related model fuels. Combustion and Flame, 2012, 159, 2254-2279.	2.8	74
64	Nitromethane dissociation: Implications for the CH3 + NO2 reaction. International Journal of Chemical Kinetics, 1999, 31, 591-602.	1.0	73
65	Post-flame gas-phase sulfation of potassium chloride. Combustion and Flame, 2013, 160, 959-969.	2.8	72
66	Devolatilization characteristics of large particles of tyre rubber under combustion conditions. Fuel, 2006, 85, 1335-1345.	3.4	71
67	Interactions of CO, NOxand H2O Under Post-Flame Conditions. Combustion Science and Technology, 1995, 110-111, 461-485.	1.2	70
68	Release and Transformation of Inorganic Elements in Combustion of a High-Phosphorus Fuel. Energy & Fuels, 2011, 25, 2874-2886.	2.5	70
69	Ignition-promoting effect of NO2 on methane, ethane and methane/ethane mixtures in a rapid compression machine. Proceedings of the Combustion Institute, 2011, 33, 433-440.	2.4	70
70	Experimental measurements and kinetic modeling of CH₄/O₂ and CH₄/C₂H₆/O₂ conversion at high pressure. International Journal of Chemical Kinetics, 2008, 40, 778-807.	1.0	69
71	Experimental Study on Effects of Particle Shape and Operating Conditions on Combustion Characteristics of Single Biomass Particles. Energy & Fuels, 2013, 27, 507-514.	2.5	69
72	Release and transformation of chlorine and potassium during pyrolysis of KCl doped biomass. Fuel, 2017, 197, 422-432.	3.4	68

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73	Experimental and Kinetic Modeling Study of C₂H₂ Oxidation at High Pressure. International Journal of Chemical Kinetics, 2016, 48, 724-738.	1.0	67
74	Experimental and kinetic modeling study of C2H4 oxidation at high pressure. Proceedings of the Combustion Institute, 2009, 32, 367-375.	2.4	66
75	Formation of NO from N₂/O₂ Mixtures in a Flow Reactor: Toward an Accurate Prediction of Thermal NO. International Journal of Chemical Kinetics, 2015, 47, 518-532.	1.0	66
76	Co-combustion of pulverized coal and solid recovered fuel in an entrained flow reactor â€“ General combustion and ash behaviour. Fuel, 2011, 90, 1980-1991.	3.4	65
77	Impact of coal fly ash addition on ash transformation and deposition in a full-scale wood suspension-firing boiler. Fuel, 2013, 113, 632-643.	3.4	65
78	Reactions of SO3with the O/H Radical Pool under Combustion Conditionsâ€. Journal of Physical Chemistry A, 2007, 111, 3984-3991.	1.1	64
79	Characterization of free radicals by electron spin resonance spectroscopy in biochars from pyrolysis at high heating rates and at high temperatures. Biomass and Bioenergy, 2016, 94, 117-129.	2.9	64
80	Laboratory Study of the CO/NH3/NO/O2System:Â Implications for Hybrid Reburn/SNCR Strategies. Energy & Fuels, 1997, 11, 716-723.	2.5	63
81	Oxidation of Dimethyl Ether and its Interaction with Nitrogen Oxides. Israel Journal of Chemistry, 1999, 39, 73-86.	1.0	63
82	Evaluation of different oxygen carriers for biomass tar reforming (I): Carbon deposition in experiments with toluene. Fuel, 2011, 90, 1049-1060.	3.4	63
83	Homogeneous and heterogeneously catalyzed oxidation of <mml:math altimg="si19.gif" display="inline" overflow="scroll" xmlns:xocs="http://www.elsevier.com/xml/xocs/dtd" xmlns:xs="http://www.w3.org/2001/XMLSchema" xmlns:xsi="http://www.w3.org/2001/XMLSchema-instance" xmlns="http://www.elsevier.com/xml/ja/dtd" xmlns:ja="http://www.elsevier.com/xml/ja/dtd" xmlns:mml="http://www.w3.org/1998/Math/MathML" xmlns:tb="http://www.elsevier.com/xml/common/table/dtd" xmlns:sb="http://www.elsevier.com/xml/co	1.9	62
84	Effect of fast pyrolysis conditions on biomass solid residues at high temperatures. Fuel Processing Technology, 2016, 143, 118-129.	3.7	62
85	Evaluation of different oxygen carriers for biomass tar reforming (II): Carbon deposition in experiments with methane and other gases. Fuel, 2011, 90, 1370-1382.	3.4	61
86	Methanol oxidation in a flow reactor: Implications for the branching ratio of the CH₃OH+OH reaction. International Journal of Chemical Kinetics, 2008, 40, 423-441.	1.0	60
87	Dust-Firing of Straw and Additives: Ash Chemistry and Deposition Behavior. Energy & Fuels, 2011, 25, 2862-2873.	2.5	59
88	High-pressure pyrolysis and oxidation of DME and DME/CH4. Combustion and Flame, 2019, 205, 80-92.	2.8	58
89	Trace elements in co-combustion of solid recovered fuel and coal. Fuel Processing Technology, 2013, 105, 212-221.	3.7	57
90	Comparison of high temperature chars of wheat straw and rice husk with respect to chemistry, morphology and reactivity. Biomass and Bioenergy, 2016, 86, 76-87.	2.9	57

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91	Theory and modeling of relevance to prompt-NO formation at high pressure. Combustion and Flame, 2018, 195, 3-17.	2.8	57
92	An experimental and kinetic modeling study of premixed nitromethane flames at low pressure. Proceedings of the Combustion Institute, 2011, 33, 407-414.	2.4	55
93	Experimental and Kinetic Modeling Study of Methanol Ignition and Oxidation at High Pressure. International Journal of Chemical Kinetics, 2013, 45, 283-294.	1.0	55
94	Kinetic Study of NO Reduction over Biomass Char under Dynamic Conditions. Energy & Fuels, 2003, 17, 1429-1436.	2.5	52
95	A kinetic issue in reburning: the fate of HCNO. Combustion and Flame, 2003, 135, 357-362.	2.8	51
96	Branching Fraction of the NH2 + NO Reaction between 1210 and 1370 K. Journal of Physical Chemistry A, 1997, 101, 3741-3745.	1.1	49
97	High-pressure oxidation of propane. Proceedings of the Combustion Institute, 2019, 37, 461-468.	2.4	48
98	Mutually Promoted Thermal Oxidation of Nitric Oxide and Organic Compounds. Industrial & Engineering Chemistry Research, 1995, 34, 1882-1888.	1.8	47
99	Modeling Low-Temperature Gas Reburning. NOxReduction Potential and Effects of Mixing. Energy & Fuels, 1998, 12, 329-338.	2.5	47
100	Partitioning of K, Cl, S and P during combustion of poplar and brassica energy crops. Fuel, 2014, 134, 209-219.	3.4	47
101	A reduced mechanism for nitrogen chemistry in methane combustion. Proceedings of the Combustion Institute, 1992, 24, 889-898.	0.3	46
102	Modelling the Formation of N2O and NO2 in the Thermal De-NOx Process. Springer Series in Chemical Physics, 1996, , 318-333.	0.2	46
103	Ab initio and kinetic modeling studies of formic acid oxidation. Proceedings of the Combustion Institute, 2015, 35, 153-160.	2.4	45
104	Experimental investigation of no from pulverized char combustion. Proceedings of the Combustion Institute, 2000, 28, 2271-2278.	2.4	44
105	New insights in the low-temperature oxidation of acetylene. Proceedings of the Combustion Institute, 2017, 36, 355-363.	2.4	43
106	On the Rate Constant for NH₂+HO₂ and Third-Body Collision Efficiencies for NH₂+H(+M) and NH₂+NH₂(+M). Journal of Physical Chemistry A, 2021, 125, 1505-1516.	1.1	43
107	A Model for Nitrogen Chemistry in Oxy-Fuel Combustion of Pulverized Coal. Energy & Fuels, 2011, 25, 4280-4289.	2.5	42
108	The recombination of hydrogen atoms with nitric oxide at high temperatures. Proceedings of the Combustion Institute, 1998, 27, 219-226.	0.3	41

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109	High-temperature chemistry of HCl and Cl2. Combustion and Flame, 2015, 162, 2693-2704.	2.8	41
110	Theoretical kinetics predictions for NH2Â+ÂHO2. Combustion and Flame, 2022, 236, 111787.	2.8	41
111	Experimental and Modeling Study of Biomass Reburning. Energy & Fuels, 2004, 18, 1442-1450.	2.5	39
112	Thermal Dissociation of SO3at 1000â˜'1400 Kâ€. Journal of Physical Chemistry A, 2006, 110, 6654-6659.	1.1	39
113	Review: Circulation of Inorganic Elements in Combustion of Alternative Fuels in Cement Plants. Energy & Fuels, 2015, 29, 4076-4099.	2.5	39
114	High pressure oxidation of C2H4/NO mixtures. Proceedings of the Combustion Institute, 2011, 33, 449-457.	2.4	38
115	Formation of fine particles in co-combustion of coal and solid recovered fuel in a pulverized coal-fired power station. Proceedings of the Combustion Institute, 2011, 33, 2845-2852.	2.4	38
116	Oxidation of Reduced Sulfur Species: Carbonyl Sulfide. International Journal of Chemical Kinetics, 2013, 45, 429-439.	1.0	38
117	An Exploratory Flow Reactor Study of H₂S Oxidation at 30â€“100 Bar. International Journal of Chemical Kinetics, 2017, 49, 37-52.	1.0	38
118	High-pressure pyrolysis and oxidation of ethanol. Fuel, 2018, 218, 247-257.	3.4	38
119	Rate Constant and Branching Fraction for the NH₂ + NO₂ Reaction. Journal of Physical Chemistry A, 2013, 117, 9011-9022.	1.1	37
120	Oxidation of Reduced Sulfur Species: Carbon Disulfide. Journal of Physical Chemistry A, 2014, 118, 6798-6809.	1.1	37
121	Experimental and Modeling Investigation of the Effect of H₂S Addition to Methane on the Ignition and Oxidation at High Pressures. Energy & Fuels, 2017, 31, 2175-2182.	2.5	37
122	Visualization methods in analysis of detailed chemical kinetics modelling. Computers & Chemistry, 2001, 25, 161-170.	1.2	36
123	Potassium Capture by Kaolin, Part 2: K₂CO₃, KCl, and K₂SO₄. Energy & Fuels, 2018, 32, 3566-3578.	2.5	36
124	Devolatilization kinetics of woody biomass at short residence times and high heating rates and peak temperatures. Applied Energy, 2016, 162, 245-256.	5.1	35
125	Kinetic Modeling of Fuel-Nitrogen Conversion in One-Dimensional, Pulverized-Coal Flames. Combustion Science and Technology, 1991, 76, 81-109.	1.2	34
126	Potassium Capture by Kaolin, Part 1: KOH. Energy & Fuels, 2018, 32, 1851-1862.	2.5	34

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127	An experimental and modeling study on auto-ignition kinetics of ammonia/methanol mixtures at intermediate temperature and high pressure. Combustion and Flame, 2022, 242, 112160.	2.8	34
128	A study of benzene formation in a laminar flow reactor. Proceedings of the Combustion Institute, 2002, 29, 1329-1336.	2.4	33
129	Numerical simulation of nitrogen oxide formation in lean premixed turbulent H2/O2/N2 flames. Proceedings of the Combustion Institute, 2011, 33, 1591-1599.	2.4	33
130	Influence of coal quality on combustion performance. Fuel, 1998, 77, 1317-1328.	3.4	32
131	Experimental and detailed kinetic modeling study of PAH formation in laminar co-flow methane diffusion flames. Proceedings of the Combustion Institute, 2013, 34, 1811-1818.	2.4	32
132	Sulfation of Condensed Potassium Chloride by SO₂. Energy & Fuels, 2013, 27, 3283-3289.	2.5	32
133	NO Formation during Oxy-Fuel Combustion of Coal and Biomass Chars. Energy & Fuels, 2014, 28, 4684-4693.	2.5	32
134	Effects of ambient pressure on ignition and flame characteristics in diesel spray combustion. Fuel, 2019, 237, 676-685.	3.4	32
135	Inhibition of hydrogen oxidation by HBr and Br2. Combustion and Flame, 2012, 159, 528-540.	2.8	31
136	Optical investigation of gas-phase KCl/KOH sulfation in post flame conditions. Fuel, 2018, 224, 461-468.	3.4	31
137	Potassium capture by coal fly ash: K2CO3, KCl and K2SO4. Fuel Processing Technology, 2019, 194, 106115.	3.7	31
138	Thermal dissociation of nitrous oxide at medium temperatures. Proceedings of the Combustion Institute, 1992, 24, 917-923.	0.3	30
139	Parabenzoquinone pyrolysis and oxidation in a flow reactor. International Journal of Chemical Kinetics, 1998, 30, 683-697.	1.0	30
140	A Chemical Engineering Model for Predicting NO Emissions and Burnout from Pulverised Coal Flames. Combustion Science and Technology, 1998, 132, 251-314.	1.2	30
141	Propargyl recombination: estimation of the high temperature, low pressure rate constant from flame measurements. Proceedings of the Combustion Institute, 2005, 30, 1023-1031.	2.4	30
142	Experimental Investigation of Ash Deposit Shedding in a Straw-Fired Boiler. Energy & Fuels, 2006, 20, 512-519.	2.5	30
143	Direct Partial Oxidation of Natural Gas to Liquid Chemicals: Chemical Kinetic Modeling and Global Optimization. Industrial & Engineering Chemistry Research, 2008, 47, 6579-6588.	1.8	30
144	An experimental and kinetic modeling study of premixed nitroethane flames at low pressure. Proceedings of the Combustion Institute, 2013, 34, 617-624.	2.4	29

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145	Influence of Torrefaction on Single Particle Combustion of Wood. Energy & Fuels, 2016, 30, 5772-5778.	2.5	29
146	A Rhodium-Based Methane Oxidation Catalyst with High Tolerance to H₂O and SO₂. ACS Catalysis, 2020, 10, 1821-1827.	5.5	29
147	Residence time distributions in a cold, confined swirl flow. Chemical Engineering Science, 1997, 52, 2743-2756.	1.9	28
148	Effects of mixing on ammonia oxidation in combustion environments at intermediate temperatures. Proceedings of the Combustion Institute, 2005, 30, 1193-1200.	2.4	28
149	Challenges in Kinetic modeling of ammonia pyrolysis. Fuel Communications, 2022, 10, 100049.	2.0	28
150	Extension of apparent devolatilization kinetics from thermally thin to thermally thick particles in zero dimensions for woody biomass. Energy, 2016, 95, 279-290.	4.5	27
151	Density Functional Theory Study of the Role of an Carbonâ€“Oxygen Single Bond Group in the NOâ€“Char Reaction. Energy & Fuels, 2018, 32, 7734-7744.	2.5	27
152	Nitrous oxide emissions control by reburning. Combustion and Flame, 1996, 107, 453-463.	2.8	26
153	Some chemical kinetics issues in reburning: The branching fraction of the HCCO+NO reaction. Proceedings of the Combustion Institute, 1998, 27, 235-243.	0.3	26
154	Detailed modeling and laser-induced fluorescence imaging of nitric oxide in a NH3-seeded non-premixed methane/air flame. Proceedings of the Combustion Institute, 2002, 29, 2195-2202.	2.4	25
155	NO Reduction over Biomass and Coal Char during Simultaneous Combustion. Energy & Fuels, 2013, 27, 7817-7826.	2.5	25
156	Modeling the Use of Sulfate Additives for Potassium Chloride Destruction in Biomass Combustion. Energy & Fuels, 2014, 28, 199-207.	2.5	25
157	Impact of Coal Fly Ash Addition on Combustion Aerosols (PM_{2.5}) from Full-Scale Suspension-Firing of Pulverized Wood. Energy & Fuels, 2014, 28, 3217-3223.	2.5	25
158	Fly Ash Formation during Suspension Firing of Biomass: Effects of Residence Time and Fuel Type. Energy & Fuels, 2017, 31, 555-570.	2.5	25
159	Skeletal mechanisms for prediction of <mml:math xmlns:mml="http://www.w3.org/1998/Math/MathML" altimg="si1.svg"><mml:mrow><mml:msub><mml:mrow><mml:mtext>NO</mml:mtext></mml:mrow><mml:mrow><mml:mi mathvariant="normal">x</mml:mi></mml:mrow></mml:msub></mml:mrow></mml:math> emission in solid fuel combustion. Fuel, 2019, 254, 115569.	3.4	25
160	KOH capture by coal fly ash. Fuel, 2019, 242, 828-836.	3.4	25
161	Kinetic NO modelling and experimental results from single wood particle combustion. Fuel, 1997, 76, 671-682.	3.4	24
162	Post-processing of detailed chemical kinetic mechanisms onto CFD simulations. Computers and Chemical Engineering, 2004, 28, 2351-2361.	2.0	24

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163	Computer-Aided Modeling Framework for Efficient Model Development, Analysis, and Identification: Combustion and Reactor Modeling. Industrial & Engineering Chemistry Research, 2011, 50, 5253-5265.	1.8	24
164	Soot Reactivity in Conventional Combustion and Oxy-fuel Combustion Environments. Energy & Fuels, 2012, 26, 5337-5344.	2.5	24
165	Glyoxal Oxidation Mechanism: Implications for the Reactions HCO + O2and OCHCHO + HO2. Journal of Physical Chemistry A, 2015, 119, 7305-7315.	1.1	24
166	Reactivity of sewage sludge, RDF, and straw chars towards NO. Fuel, 2019, 236, 297-305.	3.4	24
167	Oxidation of methylamine. International Journal of Chemical Kinetics, 2020, 52, 893-906.	1.0	24
168	The CH3+NO rate coefficient at high temperatures: Theoretical analysis and comparison with experiment. International Journal of Chemical Kinetics, 1998, 30, 223-228.	1.0	23
169	Formation and Destruction of CH2O in the Exhaust System of a Gas Engine. Environmental Science & Technology, 2003, 37, 4512-4516.	4.6	23
170	Reactivity of coal char in reducing NO. Combustion and Flame, 2004, 136, 249-253.	2.8	23
171	Influence of potassium chloride on moist CO oxidation under reducing conditions: Experimental and kinetic modeling study. Fuel, 2006, 85, 978-988.	3.4	23
172	High-Temperature Release of SO₂ from Calcined Cement Raw Materials. Energy & Fuels, 2011, 25, 2917-2926.	2.5	23
173	Mixing large and small particles in a pilot scale rotary kiln. Powder Technology, 2011, 210, 273-280.	2.1	23
174	Deposit Probe Measurements in Large Biomass-Fired Grate Boilers and Pulverized-Fuel Boilers. Energy & Fuels, 2014, 28, 3539-3555.	2.5	23
175	Defluidization in fluidized bed gasifiers using high-alkali content fuels. Biomass and Bioenergy, 2016, 91, 160-174.	2.9	23
176	A model of the coal reburning process. Proceedings of the Combustion Institute, 1998, 27, 3027-3035.	0.3	22
177	Formation of NO from combustion of volatiles from municipal solid wastes. Combustion and Flame, 2001, 124, 195-212.	2.8	22
178	Inhibition and Promotion of Pyrolysis by Hydrogen Sulfide (H₂S) and Sulfanyl Radical (SH). Journal of Physical Chemistry A, 2016, 120, 8941-8948.	1.1	22
179	Modelling of temporal and spatial evolution of sulphur oxides and sulphuric acid under large, two-stroke marine engine-like conditions using integrated CFD-chemical kinetics. Applied Energy, 2017, 193, 60-73.	5.1	22
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