

# Oriana Brea

## List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/5479068/publications.pdf>

Version: 2024-02-01

24  
papers

288  
citations

933447

10  
h-index

940533

16  
g-index

27  
all docs

27  
docs citations

27  
times ranked

265  
citing authors

#	ARTICLE	IF	CITATIONS
1	Binding and Assembly of a Benzotriazole Cavitand in Water. <i>Angewandte Chemie - International Edition</i> , 2022, 61, .	13.8	5
2	Mechanisms of Formation and Rearrangement of Benziodoxole-Based CF <sub>3</sub> and SCF <sub>3</sub> Transfer Reagents. <i>Journal of Organic Chemistry</i> , 2020, 85, 15577-15585.	3.2	4
3	Modulating the intrinsic reactivity of molecules through non-covalent interactions. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 2222-2233.	2.8	13
4	Modeling Decomposition of <i>N</i> -Nitrosoamides in a Self-Assembled Capsule. <i>Journal of Organic Chemistry</i> , 2019, 84, 7354-7361.	3.2	5
5	Gas-phase reactivity tuned through the interaction with alkaline-earth derivatives. <i>Theoretical Chemistry Accounts</i> , 2019, 138, 1.	1.4	5
6	Mechanism(s) of thermal decomposition of <i>N</i> -Nitrosoamides: A density functional theory study. <i>Tetrahedron</i> , 2019, 75, 929-935.	1.9	10
7	Super Strong Be–Be Bonds: Theoretical Insight into the Electronic Structure of Be–Be Complexes with Radical Ligands. <i>Journal of Physical Chemistry A</i> , 2018, 122, 2258-2265.	2.5	23
8	Are beryllium-containing biphenyl derivatives efficient anion sponges?. <i>Journal of Molecular Modeling</i> , 2018, 24, 16.	1.8	7
9	Molecular Modelling of the H <sub>2</sub> Adsorptive Properties of Tetrazolate-Based Metal–Organic Frameworks: From the Cluster Approach to Periodic Simulations. <i>ChemPhysChem</i> , 2018, 19, 1349-1357.	2.1	6
10	Exergonic and Spontaneous Production of Radicals through Beryllium Bonds. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 8736-8739.	13.8	22
11	On the existence of intramolecular one-electron Be–Be bonds. <i>Chemical Communications</i> , 2016, 52, 9656-9659.	4.1	28
12	Exergonic and Spontaneous Production of Radicals through Beryllium Bonds. <i>Angewandte Chemie</i> , 2016, 128, 8878-8881.	2.0	9
13	The Spin-Partitioned Total-Position Spread Tensor: An Application To Diatomic Molecules. <i>Journal of Physical Chemistry A</i> , 2016, 120, 5230-5238.	2.5	7
14	Beryllium-Based Anion Sponges: Close Relatives of Proton Sponges. <i>Chemistry - A European Journal</i> , 2016, 22, 18322-18325.	3.3	24
15	The total position-spread tensor: Spin partition. <i>Journal of Chemical Physics</i> , 2015, 142, 094113.	3.0	16
16	Ga <sup>+</sup> Basicity and Affinity Scales Based on High-Level Ab Initio Calculations. <i>ChemPhysChem</i> , 2015, 16, 3206-3213.	2.1	0
17	Creating If Holes through the Formation of Beryllium Bonds. <i>Chemistry - A European Journal</i> , 2015, 21, 12676-12682.	3.3	38
18	Why Is the Spontaneous Deprotonation of [Cu(uracil) <sub>2</sub> ] <sup>2+</sup> Complexes Accompanied by Enolization of the System?. <i>ChemPhysChem</i> , 2015, 16, 2375-2382.	2.1	2

#	ARTICLE	IF	CITATIONS
19	Spin delocalization in hydrogen chains described with the spin-partitioned total position-spread tensor. <i>Theoretical Chemistry Accounts</i> , 2015, 134, 1.	1.4	10
20	On the stability of [(uracil) <sub>2</sub> -Cu] <sup>2+</sup> complexes in the gas phase. Different pathways for the formation of [(uracil-H)(uracil)-Cu] <sup>+</sup> monocations. <i>Organic and Biomolecular Chemistry</i> , 2013, 11, 3862.	2.8	14
21	Behavior of the Position-Spread Tensor in Diatomic Systems. <i>Journal of Chemical Theory and Computation</i> , 2013, 9, 5286-5295.	5.3	19
22	Quantum chemical calculations of the thermal isomerization of 2-methyl-4,5-dihydrofuran. <i>Computational and Theoretical Chemistry</i> , 2012, 993, 53-59.	2.5	1
23	Theoretical study of methoxy group influence in the gas-phase elimination kinetics of methoxyalkyl chlorides. <i>International Journal of Quantum Chemistry</i> , 2012, 112, 2504-2514.	2.0	15
24	Binding and Assembly of a Benzotriazole Cavitand in Water. <i>Angewandte Chemie</i> , 0, , .	2.0	4