

Beatriz Valle

List of Publications by Year in descending order

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45
papers

2,989
citations

159358

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h-index

233125

45
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47
all docs

47
docs citations

47
times ranked

2224
citing authors

#	ARTICLE	IF	CITATIONS
1	Role of zeolite properties in bio-oil deoxygenation and hydrocarbons production by catalytic cracking. Fuel Processing Technology, 2022, 227, 107130.	3.7	36
2	Unveiling the deactivation by coke of NiAl ₂ O ₄ spinel derived catalysts in the bio-oil steam reforming: Role of individual oxygenates. Fuel, 2022, 321, 124009.	3.4	17
3	Combined effect of bio-oil composition and temperature on the stability of Ni spinel derived catalyst for hydrogen production by steam reforming. Fuel, 2022, 326, 124966.	3.4	16
4	Feasibility of online pre-reforming step with dolomite for improving Ni spinel catalyst stability in the steam reforming of raw bio-oil. Fuel Processing Technology, 2021, 215, 106769.	3.7	20
5	Effect of reaction conditions on the deactivation by coke of a NiAl ₂ O ₄ spinel derived catalyst in the steam reforming of bio-oil. Applied Catalysis B: Environmental, 2021, 297, 120445.	10.8	44
6	Dual catalyst-sorbent role of dolomite in the steam reforming of raw bio-oil for producing H ₂ -rich syngas. Fuel Processing Technology, 2020, 200, 106316.	3.7	28
7	Deactivation of Ni spinel derived catalyst during the oxidative steam reforming of raw bio-oil. Fuel, 2020, 276, 117995.	3.4	26
8	Recent research progress on bio-oil conversion into bio-fuels and raw chemicals: a review. Journal of Chemical Technology and Biotechnology, 2019, 94, 670-689.	1.6	124
9	Origin and Nature of Coke in Ethanol Steam Reforming and Its Role in Deactivation of Ni/La ₂ O ₃ -Al ₂ O ₃ Catalyst. Industrial & Engineering Chemistry Research, 2019, 58, 14736-14751.	1.8	70
10	Cost-effective upgrading of biomass pyrolysis oil using activated dolomite as a basic catalyst. Fuel Processing Technology, 2019, 195, 106142.	3.7	43
11	Effect of phenols extraction on the behavior of Ni-spinel derived catalyst for raw bio-oil steam reforming. International Journal of Hydrogen Energy, 2019, 44, 12593-12603.	3.8	35
12	Temperature Programmed Oxidation Coupled with In-situ Techniques Reveal the Nature and Location of Coke Deposited on a Ni/La ₂ O ₃ -Al ₂ O ₃ Catalyst in the Steam Reforming of Bio-oil. ChemCatChem, 2018, 10, 2311-2321.	1.8	44
13	Kinetic Model for the Conversion of Chloromethane into Hydrocarbons over a HZSM-5 Zeolite Catalyst. Industrial & Engineering Chemistry Research, 2018, 57, 908-919.	1.8	11
14	Steam reforming of raw bio-oil over Ni/La ₂ O ₃ -Al ₂ O ₃ : Influence of temperature on product yields and catalyst deactivation. Fuel, 2018, 216, 463-474.	3.4	89
15	Biomass to hydrogen-rich gas via steam reforming of raw bio-oil over Ni/La ₂ O ₃ -Al ₂ O ₃ catalyst: Effect of space-time and steam-to-carbon ratio. Fuel, 2018, 216, 445-455.	3.4	79
16	Reaction network of the chloromethane conversion into light olefins using a HZSM-5 zeolite catalyst. Journal of Industrial and Engineering Chemistry, 2018, 61, 427-436.	2.9	10
17	Kinetic model considering catalyst deactivation for the steam reforming of bio-oil over Ni/La ₂ O ₃ -Al ₂ O ₃ . Chemical Engineering Journal, 2018, 332, 192-204.	6.6	36
18	Role of oxygenates and effect of operating conditions in the deactivation of a Ni supported catalyst during the steam reforming of bio-oil. Green Chemistry, 2017, 19, 4315-4333.	4.6	97

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19	Strategies for maximizing the bio-oil valorization by catalytic transformation. <i>Journal of Cleaner Production</i> , 2015, 88, 345-348.	4.6	11
20	Hydrogen production by steam reforming of bio-oil/bio-ethanol mixtures in a continuous thermal-catalytic process. <i>International Journal of Hydrogen Energy</i> , 2014, 39, 6889-6898.	3.8	31
21	Compositional Insights and Valorization Pathways for Carbonaceous Material Deposited During Bio-oil Thermal Treatment. <i>ChemSusChem</i> , 2014, 7, 2597-2608.	3.6	41
22	Upgrading of Bio-Oil in a Continuous Process with Dolomite Catalyst. <i>Energy & Fuels</i> , 2014, 28, 6419-6428.	2.5	42
23	Effect of calcination/reduction conditions of Ni/La ₂ O ₃ -Al ₂ O ₃ catalyst on its activity and stability for hydrogen production by steam reforming of raw bio-oil/ethanol. <i>Applied Catalysis B: Environmental</i> , 2014, 147, 402-410.	10.8	111
24	Operating conditions for attenuating Ni/La ₂ O ₃ -Al ₂ O ₃ catalyst deactivation in the steam reforming of bio-oil aqueous fraction. <i>Fuel Processing Technology</i> , 2013, 115, 222-232.	3.7	122
25	Steam Reforming of Raw Bio-oil in a Fluidized Bed Reactor with Prior Separation of Pyrolytic Lignin. <i>Energy & Fuels</i> , 2013, 27, 7549-7559.	2.5	71
26	Catalysts of Ni/Al ₂ O ₃ and Ni/La ₂ O ₃ -Al ₂ O ₃ for hydrogen production by steam reforming of bio-oil aqueous fraction with pyrolytic lignin retention. <i>International Journal of Hydrogen Energy</i> , 2013, 38, 1307-1318.	3.8	111
27	Steam Reforming of the Bio-Oil Aqueous Fraction in a Fluidized Bed Reactor with in Situ CO ₂ Capture. <i>Industrial & Engineering Chemistry Research</i> , 2013, 52, 17087-17098.	1.8	40
28	Effect of operating conditions on the coke nature and HZSM-5 catalysts deactivation in the transformation of crude bio-oil into hydrocarbons. <i>Catalysis Today</i> , 2012, 195, 106-113.	2.2	101
29	Deactivating species in the transformation of crude bio-oil with methanol into hydrocarbons on a HZSM-5 catalyst. <i>Journal of Catalysis</i> , 2012, 285, 304-314.	3.1	175
30	Deactivation kinetics of a HZSM-5 zeolite catalyst treated with alkali for the transformation of bio-ethanol into hydrocarbons. <i>AIChE Journal</i> , 2012, 58, 526-537.	1.8	27
31	Kinetic modelling for the transformation of bioethanol into olefins on a hydrothermally stable Ni-HZSM-5 catalyst considering the deactivation by coke. <i>Chemical Engineering Journal</i> , 2011, 167, 262-277.	6.6	73
32	Pyrolytic lignin removal for the valorization of biomass pyrolysis crude bio-oil by catalytic transformation. <i>Journal of Chemical Technology and Biotechnology</i> , 2010, 85, 132-144.	1.6	159
33	Hydrothermally stable HZSM-5 zeolite catalysts for the transformation of crude bio-oil into hydrocarbons. <i>Applied Catalysis B: Environmental</i> , 2010, 100, 318-327.	10.8	124
34	Hydrothermal stability of HZSM-5 catalysts modified with Ni for the transformation of bioethanol into hydrocarbons. <i>Fuel</i> , 2010, 89, 3365-3372.	3.4	96
35	Selective production of olefins from bioethanol on HZSM-5 zeolite catalysts treated with NaOH. <i>Applied Catalysis B: Environmental</i> , 2010, 97, 299-306.	10.8	135
36	Selective Production of Aromatics by Crude Bio-oil Valorization with a Nickel-Modified HZSM-5 Zeolite Catalyst. <i>Energy & Fuels</i> , 2010, 24, 2060-2070.	2.5	164

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37	Olefin Production by Catalytic Transformation of Crude Bio-Oil in a Two-Step Process. <i>Industrial & Engineering Chemistry Research</i> , 2010, 49, 123-131.	1.8	119
38	Kinetic Model for the Transformation of Bioethanol into Olefins over a HZSM-5 Zeolite Treated with Alkali. <i>Industrial & Engineering Chemistry Research</i> , 2010, 49, 10836-10844.	1.8	52
39	Attenuation of Catalyst Deactivation by Cofeeding Methanol for Enhancing the Valorisation of Crude Bio-oil. <i>Energy & Fuels</i> , 2009, 23, 4129-4136.	2.5	88
40	Integration of Thermal Treatment and Catalytic Transformation for Upgrading Biomass Pyrolysis Oil. <i>International Journal of Chemical Reactor Engineering</i> , 2007, 5, .	0.6	21
41	Development of Alternative Catalysts Based on HZSM-5 Zeolite for the BTO Process. <i>International Journal of Chemical Reactor Engineering</i> , 2007, 5, .	0.6	3
42	Regeneration of a HZSM-5 zeolite catalyst deactivated in the transformation of aqueous ethanol into hydrocarbons. <i>Catalysis Today</i> , 2005, 107-108, 410-416.	2.2	29
43	Undesired components in the transformation of biomass pyrolysis oil into hydrocarbons on an HZSM-5 zeolite catalyst. <i>Journal of Chemical Technology and Biotechnology</i> , 2005, 80, 1244-1251.	1.6	135
44	Effect of nickel incorporation on the acidity and stability of HZSM-5 zeolite in the MTO process. <i>Catalysis Today</i> , 2005, 106, 118-122.	2.2	62
45	Kinetic Behavior of the SAPO-18 Catalyst in the Transformation of Methanol into Olefins. <i>Industrial & Engineering Chemistry Research</i> , 2005, 44, 6605-6614.	1.8	17