

Tatiana G Levitskaia

List of Publications by Year in descending order

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#	ARTICLE	IF	CITATIONS
1	Iodine immobilization by materials through sorption and redox-driven processes: A literature review. <i>Science of the Total Environment</i> , 2020, 716, 132820.	3.9	59
2	Nanostructured MgFe and CoCr layered double hydroxides for removal and sequestration of iodine anions. <i>Chemical Engineering Journal</i> , 2020, 380, 122408.	6.6	47
3	Highly Selective Colorimetric and Luminescence Response of a Square-Planar Platinum(II) Terpyridyl Complex to Aqueous TcO_4^- . <i>Inorganic Chemistry</i> , 2015, 54, 9914-9923.	1.9	39
4	Mechanisms of neptunium redox reactions in nitric acid solutions. <i>Inorganic Chemistry Frontiers</i> , 2017, 4, 581-594.	3.0	39
5	Development of Online Spectroscopic pH Monitoring for Nuclear Fuel Reprocessing Plants: Weak Acid Schemes. <i>Analytical Chemistry</i> , 2015, 87, 5139-5147.	3.2	31
6	Nitric Acid and Water Extraction by T2EHDGA in <i>n</i> -Dodecane. <i>Solvent Extraction and Ion Exchange</i> , 2017, 35, 586-603.	0.8	31
7	An Advanced TALSPEAK Concept for Separating Minor Actinides. Part 1. Process Optimization and Flowsheet Development. <i>Solvent Extraction and Ion Exchange</i> , 2017, 35, 377-395.	0.8	26
8	Comparative analysis of ZVI materials for reductive separation of $^{99}Tc(VII)$ from aqueous waste streams. <i>Journal of Hazardous Materials</i> , 2019, 380, 120836.	6.5	25
9	Inorganic tin aluminophosphate nanocomposite for reductive separation of pertechnetate. <i>Environmental Science: Nano</i> , 2016, 3, 1003-1013.	2.2	24
10	RedOx-controlled sorption of iodine anions by hydrotalcite composites. <i>RSC Advances</i> , 2016, 6, 76042-76055.	1.7	23
11	Extraction Behavior of Ln(III) Ions by T2EHDGA/ <i>n</i> -Dodecane from Nitric Acid and Sodium Nitrate Solutions. <i>Solvent Extraction and Ion Exchange</i> , 2018, 36, 331-346.	0.8	21
12	Aqueous Binary Lanthanide(III) Nitrate $Ln(NO_3)_3$ Electrolytes Revisited: Extended Pitzer and Bromley Treatments. <i>Journal of Chemical & Engineering Data</i> , 2015, 60, 2974-2988.	1.0	20
13	Technetium immobilization by materials through sorption and redox-driven processes: A literature review. <i>Science of the Total Environment</i> , 2020, 716, 132849.	3.9	19
14	Hybrid Sorbents for I_2 Capture from Contaminated Groundwater. <i>ACS Applied Materials & Interfaces</i> , 2020, 12, 26113-26126.	4.0	19
15	Evolution of Acid-Dependent Am^{3+} and Eu^{3+} Organic Coordination Environment: Effects on the Extraction Efficiency. <i>Inorganic Chemistry</i> , 2020, 59, 4453-4467.	1.9	19
16	Spectroelectrochemistry of $EuCl_3$ in Four Molten Salt Eutectics; $3\lambda\text{-LiCl}\sim\text{NaCl}$, $3\lambda\text{-LiCl}\sim 2\lambda\text{-KCl}$, $LiCl\sim\text{RbCl}$, and $3\lambda\text{-LiCl}\sim 2\lambda\text{-CsCl}$; at 873 K. <i>Electroanalysis</i> , 2016, 28, 2158-2165.	1.5	16
17	Evaluation of materials for iodine and technetium immobilization through sorption and redox-driven processes. <i>Science of the Total Environment</i> , 2020, 716, 136167.	3.9	16
18	Surprising formation of quasi-stable $Tc(VI)$ in high ionic strength alkaline media. <i>Inorganic Chemistry Frontiers</i> , 2018, 5, 2081-2091.	3.0	15

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19	Identification and Quantification of Technetium Species in Hanford Waste Tank AN-102. Analytical Chemistry, 2020, 92, 13961-13970.	3.2	14
20	Overcoming Oxidation State-Dependent Spectral Interferences: Online Monitoring of U(VI) Reduction to U(IV) via Raman and UV-Vis Spectroscopy. Industrial & Engineering Chemistry Research, 2020, 59, 8894-8901.	1.8	13
21	Simultaneous removal of cesium and iodate using prussian blue functionalized CoCr layered double hydroxide (PB-LDH). Journal of Environmental Chemical Engineering, 2022, 10, 107477.	3.3	12
22	Impact of zero valent iron aging on reductive removal of technetium-99. Journal of Environmental Chemical Engineering, 2020, 8, 103767.	3.3	11
23	Theoretical Modeling of ⁹⁹ Tc NMR Chemical Shifts. Inorganic Chemistry, 2016, 55, 8341-8347.	1.9	10
24	Spectroscopic Characterization of Aqua [fac-Tc(CO) ₃] ⁺ Complexes at High Ionic Strength. Inorganic Chemistry, 2018, 57, 6903-6912.	1.9	10
25	Effect of HEH[EHP] impurities on the ALSEP solvent extraction process. Solvent Extraction and Ion Exchange, 2018, 36, 22-40.	0.8	9
26	Removal of iodine (I ⁻ and IO ₃ ⁻) from aqueous solutions using CoAl and NiAl layered double hydroxides. Chemical Engineering Journal, 2022, 430, 132788.	6.6	9
27	The abiotic reductive removal and subsequent incorporation of Tc(^{IV}) into iron oxides: a frontier review. Environmental Science: Nano, 2019, 6, 3492-3500.	2.2	8
28	Characterization of spent Purolite A530E resin with implications for long-term radioactive contaminant removal. Journal of Environmental Chemical Engineering, 2020, 8, 104155.	3.3	8
29	Spontaneous redox continuum reveals sequestered technetium clusters and retarded mineral transformation of iron. Communications Chemistry, 2020, 3, .	2.0	8
30	Elemental characterization of crystalline silicotitanate following Hanford tank waste processing. Separation Science and Technology, 2021, 56, 1457-1465.	1.3	8
31	Mechanisms of Plutonium Redox Reactions in Nitric Acid Solutions. Inorganic Chemistry, 2020, 59, 6826-6838.	1.9	7
32	Redox and volatility of Tc(CO) ₃ ⁺ compounds in waste glass melting. Journal of Nuclear Materials, 2019, 515, 199-205.	1.3	6
33	A Review of Bismuth(III)-Based Materials for Remediation of Contaminated Sites. ACS Earth and Space Chemistry, 2022, 6, 883-908.	1.2	6
34	Inorganic Sn nanocomposite materials for sulfate sequestration from complex aqueous solutions. Environmental Science: Nano, 2018, 5, 890-903.	2.2	5
35	Time dependent zero valent iron oxidation and the reductive removal of pertechnetate at variable pH. Journal of Hazardous Materials, 2022, 424, 127400.	6.5	5
36	Simultaneous immobilization of aqueous co-contaminants using a bismuth layered material. Journal of Environmental Radioactivity, 2021, 237, 106711.	0.9	5

#	ARTICLE	IF	CITATIONS
37	Elemental iron: reduction of pertechnetate in the presence of silica and periodicity of precipitated nano-structures. <i>Environmental Science: Nano</i> , 2021, 8, 97-109.	2.2	2
38	Kinetics and mechanism of rhenium-ethylenediaminetetraacetic acid (Re(IV)-EDTA) complex degradation; For 99Tc-EDTA degradation in the natural environment. <i>Environmental Technology and Innovation</i> , 2022, 27, 102492.	3.0	0
39	Removable coatings: Thermal stability and decontamination of steel surfaces from 241Am. <i>Chemosphere</i> , 2022, 301, 134680.	4.2	0