

# Andreas Luttge

## List of Publications by Year in descending order

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95  
papers

6,956  
citations

71061

41  
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58549

82  
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101  
all docs

101  
docs citations

101  
times ranked

6925  
citing authors

#	ARTICLE	IF	CITATIONS
1	A Kinetic Monte Carlo Approach to Model Barite Dissolution: The Role of Reactive Site Geometry. Minerals (Basel, Switzerland), 2022, 12, 639.	0.8	3
2	Influence of chemical zoning on sandstone calcite cement dissolution: The case of manganese and iron. Chemical Geology, 2021, 559, 119952.	1.4	7
3	Gypsum Precipitation under Saline Conditions: Thermodynamics, Kinetics, Morphology, and Size Distribution. Minerals (Basel, Switzerland), 2021, 11, 141.	0.8	35
4	Influence of Muscovite (001) Surface Nanotopography on Radionuclide Adsorption Studied by Kinetic Monte Carlo Simulations. Minerals (Basel, Switzerland), 2021, 11, 468.	0.8	4
5	Multiscale investigation of olivine (0 1 0) face dissolution from a surface control perspective. Applied Surface Science, 2021, 549, 149317.	3.1	9
6	Mineral Dissolution Kinetics: Pathways to Equilibrium. ACS Earth and Space Chemistry, 2021, 5, 1657-1673.	1.2	8
7	Influence of Processing Route on the Surface Reactivity of Cu <sub>47</sub> Ti <sub>33</sub> Zr <sub>11</sub> Ni <sub>6</sub> Sn <sub>2</sub> Si <sub>1</sub> Metallic Glass. Metals, 2021, 11, 1173.	1.0	5
8	A modeling approach for unveiling adsorption of toxic ions on iron oxide nanocrystals. Journal of Hazardous Materials, 2021, 417, 126005.	6.5	7
9	The role of crystal heterogeneity in alkali feldspar dissolution kinetics. Geochimica Et Cosmochimica Acta, 2021, 309, 329-351.	1.6	17
10	Discrimination of Ceramic Surface Finishing by Vertical Scanning Interferometry. Archaeometry, 2019, 61, 31-42.	0.6	6
11	A Statistical Approach for Analysis of Dissolution Rates Including Surface Morphology. Minerals (Basel, Switzerland), 2019, 9, 458.	0.8	23
12	Kinetic concepts for quantitative prediction of fluid-solid interactions. Chemical Geology, 2019, 504, 216-235.	1.4	42
13	Inherited control of crystal surface reactivity. Applied Geochemistry, 2018, 91, 140-148.	1.4	36
14	The evaluation of arsenic contamination potential, speciation and hydrogeochemical behaviour in aquifers of Punjab, Pakistan. Chemosphere, 2018, 199, 737-746.	4.2	119
15	Pulsating dissolution of crystalline matter. Proceedings of the National Academy of Sciences of the United States of America, 2018, 115, 897-902.	3.3	37
16	Arsenic removal by Japanese oak wood biochar in aqueous solutions and well water: Investigating arsenic fate using integrated spectroscopic and microscopic techniques. Science of the Total Environment, 2018, 621, 1642-1651.	3.9	175
17	Arsenic removal by perilla leaf biochar in aqueous solutions and groundwater: An integrated spectroscopic and microscopic examination. Environmental Pollution, 2018, 232, 31-41.	3.7	297
18	Crystal Dissolution Kinetics Studied by a Combination of Monte Carlo and Voronoi Methods. Minerals (Basel, Switzerland), 2018, 8, 133.	0.8	14

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19	Kinetics of pipeline steel corrosion studied by Raman spectroscopy-coupled vertical scanning interferometry. <i>Npj Materials Degradation</i> , 2018, 2, .	2.6	11
20	Temporal Evolution of Calcite Surface Dissolution Kinetics. <i>Minerals (Basel, Switzerland)</i> , 2018, 8, 256.	0.8	31
21	The effect of crystal size variation on the rate of dissolution – A kinetic Monte Carlo study. <i>Geochimica Et Cosmochimica Acta</i> , 2017, 212, 167-175.	1.6	53
22	Variability of Zinc Oxide Dissolution Rates. <i>Environmental Science &amp; Technology</i> , 2017, 51, 4297-4305.	4.6	37
23	Beyond the conventional understanding of water–rock reactivity. <i>Earth and Planetary Science Letters</i> , 2017, 457, 100-105.	1.8	57
24	Kinetic Monte Carlo Approach To Study Carbonate Dissolution. <i>Journal of Physical Chemistry C</i> , 2016, 120, 6482-6492.	1.5	70
25	Direct Measurement of Surface Dissolution Rates in Potential Nuclear Waste Forms: The Example of Pyrochlore. <i>ACS Applied Materials &amp; Interfaces</i> , 2015, 7, 17857-17865.	4.0	48
26	Calcite Dissolution Kinetics. <i>Aquatic Geochemistry</i> , 2015, 21, 415-422.	1.5	12
27	Variability of crystal surface reactivity: What do we know?. <i>Applied Geochemistry</i> , 2014, 43, 132-157.	1.4	125
28	Lateral Resolution Enhancement of Vertical Scanning Interferometry by Sub-Pixel Sampling. <i>Microscopy and Microanalysis</i> , 2014, 20, 90-98.	0.2	15
29	A Stochastic Treatment of Crystal Dissolution Kinetics. <i>Elements</i> , 2013, 9, 183-188.	0.5	131
30	Fundamental Controls of Dissolution Rate Spectra: Comparisons of Model and Experimental Results. <i>Procedia Earth and Planetary Science</i> , 2013, 7, 537-540.	0.6	16
31	Plasmonic Nature of the Terahertz Conductivity Peak in Single-Wall Carbon Nanotubes. <i>Nano Letters</i> , 2013, 13, 5991-5996.	4.5	143
32	Kinetic Monte Carlo Simulations of Silicate Dissolution: Model Complexity and Parametrization. <i>Journal of Physical Chemistry C</i> , 2013, 117, 24894-24906.	1.5	38
33	A comprehensive stochastic model of phyllosilicate dissolution: Structure and kinematics of etch pits formed on muscovite basal face. <i>Geochimica Et Cosmochimica Acta</i> , 2013, 120, 545-560.	1.6	47
34	Midinfrared third-harmonic generation from macroscopically aligned ultralong single-wall carbon nanotubes. <i>Physical Review B</i> , 2013, 87, .	1.1	6
35	Does the stepwave model predict mica dissolution kinetics?. <i>Geochimica Et Cosmochimica Acta</i> , 2012, 97, 120-130.	1.6	32
36	How predictable are dissolution rates of crystalline material?. <i>Geochimica Et Cosmochimica Acta</i> , 2012, 98, 177-185.	1.6	169

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37	Modeling and simulation of cement hydration kinetics and microstructure development. <i>Cement and Concrete Research</i> , 2011, 41, 1257-1278.	4.6	328
38	Experimental techniques for cement hydration studies. <i>Studia Universitatis Babes-Bolyai, Geologia</i> , 2011, 56, 3-15.	1.0	5
39	Reduction of Graphene Oxide <i>via</i> Bacterial Respiration. <i>ACS Nano</i> , 2010, 4, 4852-4856.	7.3	539
40	Reactions at Surfaces: A New Approach Integrating Interferometry and Kinetic Simulations. <i>Journal of the American Ceramic Society</i> , 2010, 93, 3519-3530.	1.9	43
41	Retention of Latex Colloids on Calcite as a Function of Surface Roughness and Topography. <i>Langmuir</i> , 2010, 26, 4743-4752.	1.6	48
42	Fluorite dissolution at acidic pH: In situ AFM and ex situ VSI experiments and Monte Carlo simulations. <i>Geochimica Et Cosmochimica Acta</i> , 2010, 74, 4298-4311.	1.6	33
43	Mineral dissolution kinetics as a function of distance from equilibrium – New experimental results. <i>Chemical Geology</i> , 2010, 269, 79-88.	1.4	103
44	Simultaneous Interferometric Measurement of Corrosive or Demineralizing Bacteria and Their Mineral Interfaces. <i>Applied and Environmental Microbiology</i> , 2009, 75, 1445-1449.	1.4	8
45	Correlation between sub-micron surface roughness of iron oxide encrustations and trace element concentrations. <i>Science of the Total Environment</i> , 2009, 407, 4703-4710.	3.9	11
46	Theoretical approach to evaluating plagioclase dissolution mechanisms. <i>Geochimica Et Cosmochimica Acta</i> , 2009, 73, 2832-2849.	1.6	54
47	Morphological evolution of dissolving feldspar particles with anisotropic surface kinetics and implications for dissolution rate normalization and grain size dependence: A kinetic modeling study. <i>Geochimica Et Cosmochimica Acta</i> , 2009, 73, 6757-6770.	1.6	34
48	Shape control of new Fe <sub>3</sub> O <sub>4</sub> and Fe <sub>1-x</sub> Mn <sub>x</sub> O <sub>4</sub> nanostructures. <i>Advanced Functional Materials</i> , 2008, 18, 1661-1667.	1.4	47
49	Interferometric observations and kinetic modeling of the chemical cleaning of humic materials deposited on membranes. <i>Journal of Membrane Science</i> , 2008, 313, 127-134.	4.1	13
50	Microorganisms, mineral surfaces, and aquatic environments: Learning from the past for future progress. <i>Geobiology</i> , 2008, 6, 201-213.	1.1	12
51	In search of the microbe/mineral interface: quantitative analysis of bacteria on metal surfaces using vertical scanning interferometry. <i>Geobiology</i> , 2008, 6, 254-262.	1.1	9
52	Relationship between Micrometer to Submicrometer Surface Roughness and Topography Variations of Natural Iron Oxides and Trace Element Concentrations. <i>Langmuir</i> , 2008, 24, 3250-3266.	1.6	16
53	Aluminosilicate Dissolution Kinetics: A General Stochastic Model. <i>Journal of Physical Chemistry B</i> , 2008, 112, 1736-1742.	1.2	41
54	Interferometric study of pyrite surface reactivity in acidic conditions. <i>American Mineralogist</i> , 2008, 93, 508-519.	0.9	23

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55	Etch pit formation on iron silicate surfaces during siderophore-promoted dissolution. <i>Chemical Geology</i> , 2007, 240, 326-342.	1.4	98
56	Al,Si order in albite and its effect on albite dissolution processes: A Monte Carlo study. <i>American Mineralogist</i> , 2007, 92, 1316-1324.	0.9	27
57	Calcium Carbonate Formation and Dissolution. <i>Chemical Reviews</i> , 2007, 107, 342-381.	23.0	862
58	Nanoparticle Shape Conservation in the Conversion of MnO Nanocrosses into Mn <sub>3</sub> O <sub>4</sub> . <i>Chemistry of Materials</i> , 2007, 19, 1369-1375.	3.2	64
59	Converged surface roughness parameters A new tool to quantify rock surface morphology and reactivity alteration. <i>Numerische Mathematik</i> , 2007, 307, 955-973.	0.7	48
60	Kinetic inhibition of calcite (104) dissolution by aqueous manganese(II). <i>Journal of Crystal Growth</i> , 2007, 307, 116-125.	0.7	44
61	Calcite and dolomite dissolution rates in the context of microbe?mineral surface interactions. <i>Geobiology</i> , 2007, 5, 191-205.	1.1	47
62	Iron Phosphide Nanostructures Produced from a Single-Source Organometallic Precursor:â€‰ Nanorods, Bundles, Crosses, and Spherulites. <i>Nano Letters</i> , 2007, 7, 2920-2925.	4.5	87
63	Manganese(II) Oxide Nanohexapods:â€‰ Insight into Controlling the Form of Nanocrystals. <i>Chemistry of Materials</i> , 2006, 18, 1821-1829.	3.2	88
64	Magnesium inhibition of calcite dissolution kinetics. <i>Geochimica Et Cosmochimica Acta</i> , 2006, 70, 583-594.	1.6	98
65	Albite dissolution kinetics as a function of distance from equilibrium: Implications for natural feldspar weathering. <i>Geochimica Et Cosmochimica Acta</i> , 2006, 70, 1402-1420.	1.6	122
66	Characterization of fibroblast morphology on bioactive surfaces using vertical scanning interferometry. <i>Matrix Biology</i> , 2006, 25, 523-533.	1.5	9
67	Crystal dissolution kinetics and Gibbs free energy. <i>Journal of Electron Spectroscopy and Related Phenomena</i> , 2006, 150, 248-259.	0.8	139
68	A comparison of vertical scanning interferometry (VSI) and atomic force microscopy (AFM) for characterizing membrane surface topography. <i>Journal of Membrane Science</i> , 2006, 278, 410-417.	4.1	70
69	Incongruent dissolution of wollastonite measured with vertical scanning interferometry. <i>American Mineralogist</i> , 2006, 91, 430-434.	0.9	31
70	Inhibitive Properties and Surface Morphology of a Group of Heterocyclic Diazoles as Inhibitors for Acidic Iron Corrosion. <i>Langmuir</i> , 2005, 21, 12187-12196.	1.6	184
71	Development and validation of vertical scanning interferometry as a novel method for acquiring chondrocyte geometry. <i>Journal of Biomedical Materials Research Part B</i> , 2005, 72A, 83-90.	3.0	17
72	Mineral surfaces and their implications for microbial attachment: Results from Monte Carlo simulations and direct surface observations. <i>Numerische Mathematik</i> , 2005, 305, 766-790.	0.7	29

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73	Quantifying the relationship between microbial attachment and mineral surface dynamics using vertical scanning interferometry (VSI). <i>Numerische Mathematik</i> , 2005, 305, 727-751.	0.7	18
74	Multiple length-scale kinetics: an integrated study of calcite dissolution rates and strontium inhibition. <i>Numerische Mathematik</i> , 2005, 305, 119-146.	0.7	57
75	Etch pit coalescence, surface area, and overall mineral dissolution rates. <i>American Mineralogist</i> , 2005, 90, 1776-1783.	0.9	75
76	Phosphonate mediated surface reaction and reorganization: implications for the mechanism controlling cement hydration inhibition. <i>Chemical Communications</i> , 2005, , 2354.	2.2	18
77	Kinetic Justification of the Solubility Product: Application of a General Kinetic Dissolution Model. <i>Journal of Physical Chemistry B</i> , 2005, 109, 1635-1642.	1.2	34
78	Mineralogical approaches to fundamental crystal dissolution kinetics - Dissolution of an A3B structure. <i>European Journal of Mineralogy</i> , 2004, 16, 713-729.	0.4	29
79	Towards a consistent rate law: glass corrosion kinetics near saturation. <i>Geological Society Special Publication</i> , 2004, 236, 579-594.	0.8	11
80	Mineralogical approaches to fundamental crystal dissolution kinetics. <i>American Mineralogist</i> , 2004, 89, 527-540.	0.9	84
81	Direct Observation of Microbial Inhibition of Calcite Dissolution. <i>Applied and Environmental Microbiology</i> , 2004, 70, 1627-1632.	1.4	73
82	A kinetic model of metamorphism: an application to siliceous dolomites. <i>Contributions To Mineralogy and Petrology</i> , 2004, 146, 546-565.	1.2	21
83	Single-crystal plagioclase feldspar dissolution rates measured by vertical scanning interferometry. <i>American Mineralogist</i> , 2004, 89, 51-56.	0.9	49
84	Crystal dissolution kinetics studied by vertical scanning interferometry and Monte Carlo simulations: A brief review and outlook. <i>Nanostructure Science and Technology</i> , 2004, , 209-247.	0.1	5
85	Interferometric study of the dolomite dissolution: a new conceptual model for mineral dissolution. <i>Geochimica Et Cosmochimica Acta</i> , 2003, 67, 1099-1116.	1.6	140
86	Variation in calcite dissolution rates:. <i>Geochimica Et Cosmochimica Acta</i> , 2003, 67, 1623-1634.	1.6	317
87	Vertical Scanning Interferometry: Super-Resolution and in Situ Capabilities for the Studies of Gashydrates. <i>Energy Exploration and Exploitation</i> , 2003, 21, 329-332.	1.1	6
88	A model for crystal dissolution. <i>European Journal of Mineralogy</i> , 2003, 15, 603-615.	0.4	130
89	Calculation of fluid fluxes in Earth's crust. <i>Geochimica Et Cosmochimica Acta</i> , 2001, 65, 1161-1185.	1.6	16
90	Variation of Crystal Dissolution Rate Based on a Dissolution Stepwave Model. <i>Science</i> , 2001, 291, 2400-2404.	6.0	425

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91	Melting experiments in the systems CaO-MgO-Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> and MgO-SiO <sub>2</sub> at 3 to 15 GPa. American Mineralogist, 1998, 83, 491-500.	0.9	20
92	CO <sub>2</sub> -H <sub>2</sub> O fluid inclusions in forsterite: An experimental study. European Journal of Mineralogy, 1996, 8, 997-1014.	0.4	4
93	An experimental calibration of the temperature dependence of oxygen isotope fractionation between apatite and calcite at high temperatures (350–800°C). Chemical Geology, 1995, 125, 281-290.	1.4	23
94	Oxygen isotope fractionation between fluorphlogopite and calcite: an experimental investigation of temperature dependence and F-/OH- effects. European Journal of Mineralogy, 1994, 6, 53-66.	0.4	25
95	Mechanism and kinetics of the reaction: 1 dolomite + 2 quartz = 1 diopside + 2 CO <sub>2</sub> : a comparison of rock-sample and of powder experiments. Contributions To Mineralogy and Petrology, 1993, 115, 155-164.	1.2	25