## Andreas Luttge

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Calcium Carbonate Formation and Dissolution. Chemical Reviews, 2007, 107, 342-381.	23.0	862
2	Reduction of Graphene Oxide <i>via</i> Bacterial Respiration. ACS Nano, 2010, 4, 4852-4856.	7.3	539
3	Variation of Crystal Dissolution Rate Based on a Dissolution Stepwave Model. Science, 2001, 291, 2400-2404.	6.0	425
4	Modeling and simulation of cement hydration kinetics and microstructure development. Cement and Concrete Research, 2011, 41, 1257-1278.	4.6	328
5	Variation in calcite dissolution rates:. Geochimica Et Cosmochimica Acta, 2003, 67, 1623-1634.	1.6	317
6	Arsenic removal by perilla leaf biochar in aqueous solutions and groundwater: An integrated spectroscopic and microscopic examination. Environmental Pollution, 2018, 232, 31-41.	3.7	297
7	Inhibitive Properties and Surface Morphology of a Group of Heterocyclic Diazoles as Inhibitors for Acidic Iron Corrosion. Langmuir, 2005, 21, 12187-12196.	1.6	184
8	Arsenic removal by Japanese oak wood biochar in aqueous solutions and well water: Investigating arsenic fate using integrated spectroscopic and microscopic techniques. Science of the Total Environment, 2018, 621, 1642-1651.	3.9	175
9	How predictable are dissolution rates of crystalline material?. Geochimica Et Cosmochimica Acta, 2012, 98, 177-185.	1.6	169
10	Plasmonic Nature of the Terahertz Conductivity Peak in Single-Wall Carbon Nanotubes. Nano Letters, 2013, 13, 5991-5996.	4.5	143
11	Interferometric study of the dolomite dissolution: a new conceptual model for mineral dissolution. Geochimica Et Cosmochimica Acta, 2003, 67, 1099-1116.	1.6	140
12	Crystal dissolution kinetics and Gibbs free energy. Journal of Electron Spectroscopy and Related Phenomena, 2006, 150, 248-259.	0.8	139
13	A Stochastic Treatment of Crystal Dissolution Kinetics. Elements, 2013, 9, 183-188.	0.5	131
14	A model for crystal dissolution. European Journal of Mineralogy, 2003, 15, 603-615.	0.4	130
15	Variability of crystal surface reactivity: What do we know?. Applied Geochemistry, 2014, 43, 132-157.	1.4	125
16	Albite dissolution kinetics as a function of distance from equilibrium: Implications for natural feldspar weathering. Geochimica Et Cosmochimica Acta, 2006, 70, 1402-1420.	1.6	122
17	The evaluation of arsenic contamination potential, speciation and hydrogeochemical behaviour in aquifers of Punjab, Pakistan. Chemosphere, 2018, 199, 737-746.	4.2	119
18	Mineral dissolution kinetics as a function of distance from equilibrium – New experimental results. Chemical Geology, 2010, 269, 79-88.	1.4	103

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19	Magnesium inhibition of calcite dissolution kinetics. Geochimica Et Cosmochimica Acta, 2006, 70, 583-594.	1.6	98
20	Etch pit formation on iron silicate surfaces during siderophore-promoted dissolution. Chemical Geology, 2007, 240, 326-342.	1.4	98
21	Manganese(II) Oxide Nanohexapods:  Insight into Controlling the Form of Nanocrystals. Chemistry of Materials, 2006, 18, 1821-1829.	3.2	88
22	Iron Phosphide Nanostructures Produced from a Single-Source Organometallic Precursor: Nanorods, Bundles, Crosses, and Spherulites. Nano Letters, 2007, 7, 2920-2925.	4.5	87
23	Mineralogical approaches to fundamental crystal dissolution kinetics. American Mineralogist, 2004, 89, 527-540.	0.9	84
24	Etch pit coalescence, surface area, and overall mineral dissolution rates. American Mineralogist, 2005, 90, 1776-1783.	0.9	75
25	Direct Observation of Microbial Inhibition of Calcite Dissolution. Applied and Environmental Microbiology, 2004, 70, 1627-1632.	1.4	73
26	A comparison of vertical scanning interferometry (VSI) and atomic force microscopy (AFM) for characterizing membrane surface topography. Journal of Membrane Science, 2006, 278, 410-417.	4.1	70
27	Kinetic Monte Carlo Approach To Study Carbonate Dissolution. Journal of Physical Chemistry C, 2016, 120, 6482-6492.	1.5	70
28	Nanoparticle Shape Conservation in the Conversion of MnO Nanocrosses into Mn3O4. Chemistry of Materials, 2007, 19, 1369-1375.	3.2	64
29	Multiple length-scale kinetics: an integrated study of calcite dissolution rates and strontium inhibition. Numerische Mathematik, 2005, 305, 119-146.	0.7	57
30	Beyond the conventional understanding of water–rock reactivity. Earth and Planetary Science Letters, 2017, 457, 100-105.	1.8	57
31	Theoretical approach to evaluating plagioclase dissolution mechanisms. Geochimica Et Cosmochimica Acta, 2009, 73, 2832-2849.	1.6	54
32	The effect of crystal size variation on the rate of dissolution – A kinetic Monte Carlo study. Geochimica Et Cosmochimica Acta, 2017, 212, 167-175.	1.6	53
33	Single-crystal plagioclase feldspar dissolution rates measured by vertical scanning interferometry. American Mineralogist, 2004, 89, 51-56.	0.9	49
34	Converged surface roughness parameters A new tool to quantify rock surface morphology and reactivity alteration. Numerische Mathematik, 2007, 307, 955-973.	0.7	48
35	Retention of Latex Colloids on Calcite as a Function of Surface Roughness and Topography. Langmuir, 2010, 26, 4743-4752.	1.6	48
36	Direct Measurement of Surface Dissolution Rates in Potential Nuclear Waste Forms: The Example of Pyrochlore. ACS Applied Materials & Interfaces, 2015, 7, 17857-17865.	4.0	48

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37	Calcite and dolomite dissolution rates in the context of microbe?mineral surface interactions. Geobiology, 2007, 5, 191-205.	1.1	47
38	Shape control of new Fe <sub><i><b>x</b></i></sub> O–Fe <sub>3</sub> O <sub>4</sub> and Fe <sub>1–<i><b>y</b></i></sub> Mn <sub><i><b>y</b></i></sub> O–Fe <sub>3–<i><b>z</b></i>nanostructures. Advanced Functional Materials, 2008, 18, 1661-1667.</sub>	ıb>Mn <b>r</b> ∡ssub>	<i>47b&gt;z</i>
39	A comprehensive stochastic model of phyllosilicate dissolution: Structure and kinematics of etch pits formed on muscovite basal face. Geochimica Et Cosmochimica Acta, 2013, 120, 545-560.	1.6	47
40	Kinetic inhibition of calcite (104) dissolution by aqueous manganese(II). Journal of Crystal Growth, 2007, 307, 116-125.	0.7	44
41	Reactions at Surfaces: A New Approach Integrating Interferometry and Kinetic Simulations. Journal of the American Ceramic Society, 2010, 93, 3519-3530.	1.9	43
42	Kinetic concepts for quantitative prediction of fluid-solid interactions. Chemical Geology, 2019, 504, 216-235.	1.4	42
43	Aluminosilicate Dissolution Kinetics:Â A General Stochastic Model. Journal of Physical Chemistry B, 2008, 112, 1736-1742.	1.2	41
44	Kinetic Monte Carlo Simulations of Silicate Dissolution: Model Complexity and Parametrization. Journal of Physical Chemistry C, 2013, 117, 24894-24906.	1.5	38
45	Variability of Zinc Oxide Dissolution Rates. Environmental Science & Technology, 2017, 51, 4297-4305.	4.6	37
46	Pulsating dissolution of crystalline matter. Proceedings of the National Academy of Sciences of the United States of America, 2018, 115, 897-902.	3.3	37
47	Inherited control of crystal surface reactivity. Applied Geochemistry, 2018, 91, 140-148.	1.4	36
48	Gypsum Precipitation under Saline Conditions: Thermodynamics, Kinetics, Morphology, and Size Distribution. Minerals (Basel, Switzerland), 2021, 11, 141.	0.8	35
49	Kinetic Justification of the Solubility Product:Â Application of a General Kinetic Dissolution Model. Journal of Physical Chemistry B, 2005, 109, 1635-1642.	1.2	34
50	Morphological evolution of dissolving feldspar particles with anisotropic surface kinetics and implications for dissolution rate normalization and grain size dependence: A kinetic modeling study. Geochimica Et Cosmochimica Acta, 2009, 73, 6757-6770.	1.6	34
51	Fluorite dissolution at acidic pH: In situ AFM and ex situ VSI experiments and Monte Carlo simulations. Geochimica Et Cosmochimica Acta, 2010, 74, 4298-4311.	1.6	33
52	Does the stepwave model predict mica dissolution kinetics?. Geochimica Et Cosmochimica Acta, 2012, 97, 120-130.	1.6	32
53	Incongruent dissolution of wollastonite measured with vertical scanning interferometry. American Mineralogist, 2006, 91, 430-434.	0.9	31
54	Temporal Evolution of Calcite Surface Dissolution Kinetics. Minerals (Basel, Switzerland), 2018, 8, 256.	0.8	31

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55	Mineralogical approaches to fundamental crystal dissolution kinetics - Dissolution of an A3B structure. European Journal of Mineralogy, 2004, 16, 713-729.	0.4	29
56	Mineral surfaces and their implications for microbial attachment: Results from Monte Carlo simulations and direct surface observations. Numerische Mathematik, 2005, 305, 766-790.	0.7	29
57	Al,Si order in albite and its effect on albite dissolution processes: A Monte Carlo study. American Mineralogist, 2007, 92, 1316-1324.	0.9	27
58	Mechanism and kinetics of the reaction: 1 dolomite + 2 quartz = 1 diopside + 2 CO2: a comparison of rock-sample and of powder experiments. Contributions To Mineralogy and Petrology, 1993, 115, 155-164.	1.2	25
59	Oxygen isotope fractionation between fluorphlogopite and calcite: an experimental investigation of temperature dependence and F-/OH- effects. European Journal of Mineralogy, 1994, 6, 53-66.	0.4	25
60	An experimental calibration of the temperature dependence of oxygen isotope fractionation between apatite and calcite at high temperatures (350–800°C). Chemical Geology, 1995, 125, 281-290.	1.4	23
61	Interferometric study of pyrite surface reactivity in acidic conditions. American Mineralogist, 2008, 93, 508-519.	0.9	23
62	A Statistical Approach for Analysis of Dissolution Rates Including Surface Morphology. Minerals (Basel, Switzerland), 2019, 9, 458.	0.8	23
63	A kinetic model of metamorphism: an application to siliceous dolomites. Contributions To Mineralogy and Petrology, 2004, 146, 546-565.	1.2	21
64	Melting experiments in the systems CaO-MgO-Al2O3-SiO2and MgO-SiO2at 3 to 15 GPa. American Mineralogist, 1998, 83, 491-500.	0.9	20
65	Quantifying the relationship between microbial attachment and mineral surface dynamics using vertical scanning interferometry (VSI). Numerische Mathematik, 2005, 305, 727-751.	0.7	18
66	Phosphonate mediated surface reaction and reorganization: implications for the mechanism controlling cement hydration inhibition. Chemical Communications, 2005, , 2354.	2.2	18
67	Development and validation of vertical scanning interferometry as a novel method for acquiring chondrocyte geometry. Journal of Biomedical Materials Research Part B, 2005, 72A, 83-90.	3.0	17
68	The role of crystal heterogeneity in alkali feldspar dissolution kinetics. Geochimica Et Cosmochimica Acta, 2021, 309, 329-351.	1.6	17
69	Calculation of fluid fluxes in Earth's crust. Geochimica Et Cosmochimica Acta, 2001, 65, 1161-1185.	1.6	16
70	Relationship between Micrometer to Submicrometer Surface Roughness and Topography Variations of Natural Iron Oxides and Trace Element Concentrations. Langmuir, 2008, 24, 3250-3266.	1.6	16
71	Fundamental Controls of Dissolution Rate Spectra: Comparisons of Model and Experimental Results. Procedia Earth and Planetary Science, 2013, 7, 537-540.	0.6	16
72	Lateral Resolution Enhancement of Vertical Scanning Interferometry by Sub-Pixel Sampling. Microscopy and Microanalysis, 2014, 20, 90-98.	0.2	15

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73	Crystal Dissolution Kinetics Studied by a Combination of Monte Carlo and Voronoi Methods. Minerals (Basel, Switzerland), 2018, 8, 133.	0.8	14
74	Interferometric observations and kinetic modeling of the chemical cleaning of humic materials deposited on membranes. Journal of Membrane Science, 2008, 313, 127-134.	4.1	13
75	Microorganisms, mineral surfaces, and aquatic environments: Learning from the past for future progress. Geobiology, 2008, 6, 201-213.	1.1	12
76	Calcite Dissolution Kinetics. Aquatic Geochemistry, 2015, 21, 415-422.	1.5	12
77	Towards a consistent rate law: glass corrosion kinetics near saturation. Geological Society Special Publication, 2004, 236, 579-594.	0.8	11
78	Correlation between sub-micron surface roughness of iron oxide encrustations and trace element concentrations. Science of the Total Environment, 2009, 407, 4703-4710.	3.9	11
79	Kinetics of pipeline steel corrosion studied by Raman spectroscopy-coupled vertical scanning interferometry. Npj Materials Degradation, 2018, 2, .	2.6	11
80	Characterization of fibroblast morphology on bioactive surfaces using vertical scanning interferometry. Matrix Biology, 2006, 25, 523-533.	1.5	9
81	In search of the microbe/mineral interface: quantitative analysis of bacteria on metal surfaces using vertical scanning interferometry. Geobiology, 2008, 6, 254-262.	1.1	9
82	Multiscale investigation of olivine (0 1 0) face dissolution from a surface control perspective. Applied Surface Science, 2021, 549, 149317.	3.1	9
83	Simultaneous Interferometric Measurement of Corrosive or Demineralizing Bacteria and Their Mineral Interfaces. Applied and Environmental Microbiology, 2009, 75, 1445-1449.	1.4	8
84	Mineral Dissolution Kinetics: Pathways to Equilibrium. ACS Earth and Space Chemistry, 2021, 5, 1657-1673.	1.2	8
85	Influence of chemical zoning on sandstone calcite cement dissolution: The case of manganese and iron. Chemical Geology, 2021, 559, 119952.	1.4	7
86	A modeling approach for unveiling adsorption of toxic ions on iron oxide nanocrystals. Journal of Hazardous Materials, 2021, 417, 126005.	6.5	7
87	Vertical Scanning Interferometry: Super-Resolution and in Situ Capabilities for the Studies of Gashydrates. Energy Exploration and Exploitation, 2003, 21, 329-332.	1.1	6
88	Midinfrared third-harmonic generation from macroscopically aligned ultralong single-wall carbon nanotubes. Physical Review B, 2013, 87, .	1.1	6
89	Discrimination of Ceramic Surface Finishing by Vertical Scanning Interferometry. Archaeometry, 2019, 61, 31-42.	0.6	6
90	Experimental techniques for cement hydration studies. Studia Universitatis Babes-Bolyai, Geologia, 2011, 56, 3-15.	1.0	5

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91	Influence of Processing Route on the Surface Reactivity of Cu47Ti33Zr11Ni6Sn2Si1 Metallic Glass. Metals, 2021, 11, 1173.	1.0	5
92	Crystal dissolution kinetics studied by vertical scanning interferometry and Monte Carlo simulations: A brief review and outlook. Nanostructure Science and Technology, 2004, , 209-247.	0.1	5
93	Influence of Muscovite (001) Surface Nanotopography on Radionuclide Adsorption Studied by Kinetic Monte Carlo Simulations. Minerals (Basel, Switzerland), 2021, 11, 468.	0.8	4
94	CO2-H2O fluid inclusions in forsterite: An experimental study. European Journal of Mineralogy, 1996, 8, 997-1014.	0.4	4
95	A Kinetic Monte Carlo Approach to Model Barite Dissolution: The Role of Reactive Site Geometry. Minerals (Basel, Switzerland), 2022, 12, 639.	0.8	3