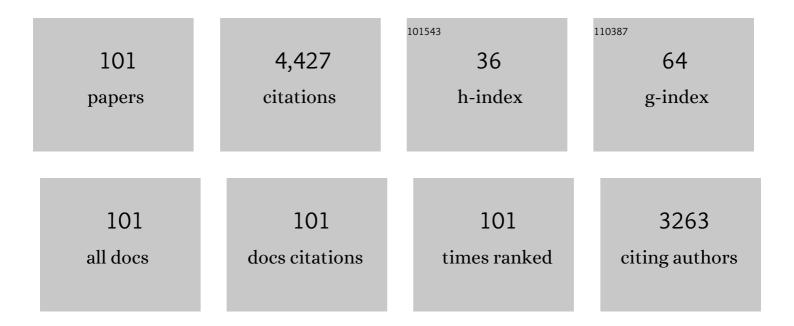
## **Tobias Reich**

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Effect of Ca(II) on U(VI) and Np(VI) retention on Ca-bentonite and clay minerals at hyperalkaline conditions - New insights from batch sorption experiments and luminescence spectroscopy. Science of the Total Environment, 2022, 842, 156837.	8.0	6
2	Development, characterization, and first application of a resonant laser secondary neutral mass spectrometry setup for the research of plutonium in the context of long-term nuclear waste storage. Analytical and Bioanalytical Chemistry, 2021, 413, 3987-3997.	3.7	6
3	Nanocomposite antimicrobials prevent bacterial growth through the enzyme-like activity of Bi-doped cerium dioxide (Ce <sub>1â^x</sub> Bi <sub>x</sub> O <sub>2â^î</sub> ). Nanoscale, 2020, 12, 21344-21358.	5.6	20
4	Solid State Fluorination on the Minute Scale: Synthesis of WO <sub>3â^'</sub> <i><sub>x</sub></i> F <i><sub>x</sub></i> with Photocatalytic Activity. Advanced Functional Materials, 2020, 30, 1909051.	14.9	15
5	Recent developments in resonance ionization mass spectrometry for ultra-trace analysis of actinide elements. Radiochimica Acta, 2019, 107, 645-652.	1.2	11
6	Modeling the sorption of Np(V) on Na-montmorillonite – effects of pH, ionic strength and CO <sub>2</sub> . Radiochimica Acta, 2019, 107, 615-622.	1.2	2
7	Investigation of the Electrophoretic Mobility of the Actinides Th, U, Np, Pu, and Am in Different Oxidation States. Analytical Chemistry, 2019, 91, 11537-11543.	6.5	8
8	Antioxidant activity of cerium dioxide nanoparticles and nanorods in scavenging hydroxyl radicals. RSC Advances, 2019, 9, 11077-11081.	3.6	48
9	Determination of the Stability Constants of the Acetate Complexes of the Actinides Am(III), Th(IV), Np(V), and U(VI) Using Capillary Electrophoresis-Inductively Coupled Plasma Mass Spectrometry. Inorganic Chemistry, 2019, 58, 4851-4858.	4.0	6
10	Spark Plasma Sintering (SPS)-Assisted Synthesis and Thermoelectric Characterization of Magnéli Phase V <sub>6</sub> O <sub>11</sub> . Inorganic Chemistry, 2018, 57, 1259-1268.	4.0	11
11	Search for an electric charge of the neutron. Physical Review D, 2018, 97, .	4.7	5
12	Determination of kinetic parameters of redox reactions using CEâ€ICPâ€MS: A case study for the reduction of Np(V) by hydroxylamine hydrochloride. Electrophoresis, 2018, 39, 3013-3021.	2.4	1
13	Ddpd as Expanded Terpyridine: Dramatic Effects of Symmetry and Electronic Properties in First Row Transition Metal Complexes. Inorganics, 2018, 6, 86.	2.7	41
14	Uptake of actinides by calcium silicate hydrate (C-S-H) phases. Applied Geochemistry, 2018, 98, 426-434.	3.0	16
15	Geochemical Interactions of Plutonium with Opalinus Clay Studied by Spatially Resolved Synchrotron Radiation Techniques. Environmental Science & Technology, 2017, 51, 7892-7902.	10.0	10
16	High-resolution in-source laser spectroscopy in perpendicular geometry. Hyperfine Interactions, 2017, 238, 1.	0.5	22
17	Application of Resonance Ionization Mass Spectrometry for Ultratrace Analysis of Technetium. Analytical Chemistry, 2017, 89, 9077-9082.	6.5	10
18	Upgrade of the ultracold neutron source at the pulsed reactor TRIGA Mainz. European Physical Journal A, 2017, 53, 1.	2.5	12

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19	Comparison of ultracold neutron sources for fundamental physics measurements. Physical Review C, 2017, 95, .	2.9	39
20	Instrumental determination of phosphorus in silicon for photovoltaics by β spectroscopy: a new approach. Journal of Radioanalytical and Nuclear Chemistry, 2017, 311, 541-548.	1.5	1
21	Distribution coefficients for the sorption of Th, U, Np, Pu, and Am on Opalinus Clay. Radiochimica Acta, 2016, 104, 33-40.	1.2	21
22	Improving material properties and performance of nuclear targets for transmutation-relevant experiments. Journal of Radioanalytical and Nuclear Chemistry, 2015, 305, 913-919.	1.5	0
23	Uranyl sorption onto birnessite: A surface complexation modeling and EXAFS study. Chemical Geology, 2014, 373, 59-70.	3.3	23
24	Performance of the solid deuterium ultra-cold neutron source at the pulsed reactor TRIGA Mainz. European Physical Journal A, 2014, 50, 1.	2.5	20
25	Smooth crack-free targets for nuclear applications produced by molecular plating. Nuclear Instruments and Methods in Physics Research, Section A: Accelerators, Spectrometers, Detectors and Associated Equipment, 2013, 714, 163-175.	1.6	22
26	The performance of thin layers produced by molecular plating as α-particle sources. Nuclear Instruments and Methods in Physics Research, Section A: Accelerators, Spectrometers, Detectors and Associated Equipment, 2013, 721, 35-44.	1.6	20
27	Influence of humic acid on neptunium(V) sorption and diffusion in Opalinus Clay. Radiochimica Acta, 2013, , 130617035320002.	1.2	6
28	Influence of temperature and background electrolyte on the sorption of neptunium(V) on Opalinus Clay. Applied Clay Science, 2012, 69, 43-49.	5.2	11
29	Sensitive redox speciation of neptunium by CE–ICP–MS. Analytical and Bioanalytical Chemistry, 2012, 404, 2143-2150.	3.7	14
30	Speciation of Np(V) uptake by Opalinus Clay using synchrotron microbeam techniques. Analytical and Bioanalytical Chemistry, 2012, 404, 2151-2162.	3.7	22
31	Determination of a three-step excitation and ionization scheme for resonance ionization and ultratrace analysis of Np-237. Spectrochimica Acta, Part B: Atomic Spectroscopy, 2011, 66, 242-247.	2.9	11
32	Sorption of neptunium(V) on Opalinus Clay under aerobic/anaerobic conditions. Radiochimica Acta, 2011, 99, 71-77.	1.2	23
33	Neptunium(V) sorption on kaolinite. Radiochimica Acta, 2011, 99, 349-357.	1.2	21
34	Study of the role of sulfur functionalities in humic acids for uranium(VI) complexation. Radiochimica Acta, 2010, 98, 467-477.	1.2	11
35	Neptunium(V) sorption onto gibbsite. Radiochimica Acta, 2009, 97, .	1.2	6
36	Emission of ThO2 valence electrons upon excitation with synchrotron radiation near the O 4,5(Th) resonance absorption threshold. Radiochemistry, 2009, 51, 560-566.	0.7	7

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37	Neptunium(V) Sorption and Diffusion in Opalinus Clay. Environmental Science & Technology, 2009, 43, 6567-6571.	10.0	55
38	Structural characterization of U(VI) surface complexes on kaolinite in the presence of humic acid using EXAFS spectroscopy. Journal of Colloid and Interface Science, 2008, 319, 40-47.	9.4	56
39	Actinide Sorption Studies Using the Isotopes <sup>237</sup> Np and <sup>239</sup> Np. Journal of Nuclear Science and Technology, 2008, 45, 133-137.	1.3	11
40	Regularization methods for the analysis of EXAFS spectra of chemical complexes. Journal of Inverse and Ill-Posed Problems, 2007, 15, .	1.0	3
41	Application of XAFS Spectroscopy to Actinide Environmental Science. AIP Conference Proceedings, 2007, , .	0.4	21
42	New Regularization Method for EXAFS Analysis. AIP Conference Proceedings, 2007, , .	0.4	2
43	Plutonium(III) complexation by humic substances studied by X-ray absorption fine structure spectroscopy. Inorganica Chimica Acta, 2006, 359, 237-242.	2.4	40
44	Spectroscopic characterization of alkaline earth uranyl carbonates. Journal of Solid State Chemistry, 2005, 178, 567-577.	2.9	54
45	Speciation analysis with synchrotron radiation. Analytical and Bioanalytical Chemistry, 2005, 383, 10-11.	3.7	3
46	EXAFS study on the neptunium(V) complexation by various humic acids under neutral pH conditions. Radiochimica Acta, 2005, 93, .	1.2	36
47	Neptunium(IV) complexation by humic substances studied by X-ray absorption fine structure spectroscopy. Radiochimica Acta, 2005, 93, 187-196.	1.2	30
48	Complexation of Uranium by Cells and S-Layer Sheets of Bacillus sphaericus JG-A12. Applied and Environmental Microbiology, 2005, 71, 5532-5543.	3.1	246
49	Spectroscopic Characterization of the Uranium Carbonate Andersonite Na2Ca[UO2(CO3)3]·6H2O. Environmental Science & Technology, 2004, 38, 6032-6036.	10.0	76
50	Complexation of uranium(VI) with protocatechuic acid?application of iterative transformation factor analysis to EXAFS spectroscopy. Analytical and Bioanalytical Chemistry, 2003, 376, 631-638.	3.7	154
51	Interaction of uranium(VI) with various modified and unmodified natural and synthetic humic substances studied by EXAFS and FTIR spectroscopy. Inorganica Chimica Acta, 2003, 351, 133-140.	2.4	103
52	A XAS study of the local environments of cations in (U, Ce)O2. Journal of Nuclear Materials, 2003, 312, 103-110.	2.7	40
53	Uranyl(VI) complexes with alpha-substituted carboxylic acids in aqueous solution. Radiochimica Acta, 2003, 91, .	1.2	54
54	Sorption of Uranium(VI) onto Ferric Oxides in Sulfate-Rich Acid Waters. Environmental Science & Technology, 2003, 37, 2898-2904.	10.0	72

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55	Uranium speciation in plants. Radiochimica Acta, 2003, 91, 319-328.	1.2	64
56	Characterization of U(VI)-Acidithiobacillus ferrooxidans complexes using EXAFS, transmission electron microscopy, and energy-dispersive X-ray analysis. Radiochimica Acta, 2003, 91, 583-592.	1.2	73
57	EXAFS and XRD investigations of zeunerite and meta-zeunerite. Zeitschrift Fur Kristallographie - Crystalline Materials, 2003, 218, 37-45.	0.8	17
58	Structure of uranium sorption complexes at montmorillonite edge sites. Radiochimica Acta, 2002, 90, 653-657.	1.2	118
59	Evidence for the existence of Tc(IV) – humic substance species by X-ray absorption near-edge spectroscopy. Radiochimica Acta, 2002, 90, 879-884.	1.2	27
60	The colloid chemistry of acid rock drainage solution from an abandoned Zn–Pb–Ag mine. Applied Geochemistry, 2002, 17, 633-648.	3.0	61
61	A theoretical study on the structures of UO2(CO3)34â^', Ca2UO2(CO3)30, and Ba2UO2(CO3)30. Chemical Physics Letters, 2002, 357, 73-77.	2.6	32
62	Uranyl(VI) carbonate complex formation: Validation of the Ca2UO2(CO3)3(aq.) species. Radiochimica Acta, 2001, 89, 511-518.	1.2	353
63	On the Structure of Np(VI) and Np(VII) Species in Alkaline Solution Studied by EXAFS and Quantum Chemical Methods. Journal of Physical Chemistry A, 2001, 105, 11441-11445.	2.5	52
64	EXAFS as a tool for bond-length determination in the environment of heavy atoms. Journal of Synchrotron Radiation, 2001, 8, 695-697.	2.4	10
65	Do Perchlorate and Triflate Anions Bind to the Uranyl Cation in an Acidic Aqueous Medium? A Combined EXAFS and Quantum Mechanical Investigation. ChemPhysChem, 2001, 2, 591-598.	2.1	76
66	A theoretical study of uranyl hydroxide monomeric and dimeric complexes. Chemical Physics Letters, 2001, 347, 127-132.	2.6	39
67	Reactivity of technetium(I) thioether carbonyl complexes towards histidine—an EXAFS study in solution. Inorganica Chimica Acta, 2001, 322, 79-86.	2.4	21
68	EXAFS investigation of uranium(VI) complexes formed at Bacillus cereus and Bacillus sphaericus surfaces. Radiochimica Acta, 2001, 89, 625-632.	1.2	77
69	Do Perchlorate and Triflate Anions Bind to the Uranyl Cation in an Acidic Aqueous Medium? A Combined EXAFS and Quantum Mechanical Investigation. ChemPhysChem, 2001, 2, 591-598.	2.1	1
70	EXAFS analyses of technetium(I) carbonyl complexes – stability studies in solutions. Radiochimica Acta, 2000, 88, 239-246.	1.2	10
71	Solution coordination chemistry of uranium in the binary UO2 2+-SO4 2- and the ternary UO2 2+-SO4 2OH- system. Radiochimica Acta, 2000, 88, 559-566.	1.2	57
72	EXAFS structural analysis of aqueous uranium(VI) complexes with lignin degradation products. Radiochimica Acta, 2000, 88, 593-598.	1.2	20

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73	The Rossendorf Beam Line ROBL – a dedicated experimental station for XAFS measurements of actinides and other radionuclides. Radiochimica Acta, 2000, 88, 633-638.	1.2	90
74	The hydrolysis of dioxouranium(VI) investigated using EXAFS and 17O-NMR. Radiochimica Acta, 2000, 88, .	1.2	67
75	The Structure of U6+ Sorption Complexes on Vermiculite and Hydrobiotite. Clays and Clay Minerals, 1999, 47, 439-457.	1.3	70
76	ROBL – a CRG beamline for radiochemistry and materials research at the ESRF. Journal of Synchrotron Radiation, 1999, 6, 1076-1085.	2.4	182
77	Phase relations in the system V/Nb/O. V. Investigation of mixed crystals V 1-x Nb x O 2. Fresenius' Journal of Analytical Chemistry, 1999, 363, 202-205.	1.5	5
78	X-ray photoelectron spectroscopy investigation of the interaction of U(VI) and Fe(III) with natural humic acid in aqueous solutions. Journal FA¼r Praktische Chemie, 1999, 341, 773-777.	0.2	6
79	Structures of Substituted-Cyclopentadienyl Uranium(III) Dimers and Related Uranium Metallocenes Deduced by EXAFS. Organometallics, 1999, 18, 1253-1258.	2.3	32
80	Modern Speciation Techniques Applied to Environmental Systems. , 1999, , 11-38.		2
81	Solution structures of rhenium (V) oxo peptide complexes of glycylglycylcysteine and cysteinylglycine as studied by capillary electrophoresis and X-ray absorption spectroscopy. Journal of Inorganic Biochemistry, 1998, 70, 99-106.	3.5	13
82	Electron spectroscopy for chemical analysis investigation of the interaction of uranyl and calcium ions with humic acids. Inorganica Chimica Acta, 1998, 273, 234-237.	2.4	11
83	An EXAFS study of uranium(VI) sorption onto silica gel and ferrihydrite. Journal of Electron Spectroscopy and Related Phenomena, 1998, 96, 237-243.	1.7	132
84	Determination of structural parameters of uranyl ions complexed with organic acids using EXAFS. Journal of Alloys and Compounds, 1998, 271-273, 123-127.	5.5	50
85	Investigation of Aquo and Chloro Complexes of UO22+, NpO2+, Np4+, and Pu3+ by X-ray Absorption Fine Structure Spectroscopy. Inorganic Chemistry, 1997, 36, 4676-4683.	4.0	349
86	Technetium coordination ability of cysteine-containing peptides: X-ray absorption spectroscopy of a 99Tc labelled endothelin derivative. Applied Radiation and Isotopes, 1997, 48, 1045-1050.	1.5	9
87	Laser and X-ray spectroscopic studies of uranium-calcite interface phenomena. Journal of Nuclear Materials, 1997, 248, 408-411.	2.7	47
88	Chemical Speciation Studies of Radionuclides by XAFS. European Physical Journal Special Topics, 1997, 7, C2-789-C2-792.	0.2	2
89	EXAFS Determinations of Uranium Structures:Â The Uranyl Ion Complexed with Tartaric, Citric, and Malic Acids. Inorganic Chemistry, 1996, 35, 784-787.	4.0	100
90	Lineshape asymmetry parameters in X-ray photoelectron spectra. Journal of Electron Spectroscopy and Related Phenomena, 1996, 77, 15-24.	1.7	7

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91	Polarized x-ray-absorption spectroscopy of the uranyl ion: Comparison of experiment and theory. Physical Review B, 1996, 54, 156-165.	3.2	155
92	Multinuclear NMR, Raman, EXAFS, and X-ray diffraction studies of uranyl carbonate complexes in near-neutral aqueous solution. X-ray structure of [C(NH2)3]6[(UO2)3(CO3)6].cntdot.6.5H2O. Inorganic Chemistry, 1995, 34, 4797-4807.	4.0	199
93	Electronic and structural investigations of technetium compounds by x-ray absorption spectroscopy. Inorganic Chemistry, 1995, 34, 193-198.	4.0	68
94	A XANES and EXAFS Investigation of the Speciation of Selenite following Bacterial Metabolization. Inorganic Chemistry, 1995, 34, 1617-1619.	4.0	21
95	Near-threshold behavior of theK-shell satellites in CO. Physical Review A, 1994, 49, 4570-4577.	2.5	32
96	Toward a soft X-ray Fourier-transform spectrometer. Nuclear Instruments and Methods in Physics Research, Section A: Accelerators, Spectrometers, Detectors and Associated Equipment, 1994, 347, 182-191.	1.6	19
97	High resolution in the soft xâ€ray range from a toroidal grating monochromator. Review of Scientific Instruments, 1993, 64, 2552-2557.	1.3	3
98	The influence of Coster—Kronig decay processes on the relative intensities of 2s and 2p photoelectron lines of Si, P, S, Cl, and Ca. Journal of Electron Spectroscopy and Related Phenomena, 1992, 58, 67-73.	1.7	6
99	Quantitative XPS surface analysis: Correction for contamination layer. Journal of Electron Spectroscopy and Related Phenomena, 1991, 56, 33-49.	1.7	12
100	Valence band offset in ZnS layers on Si(111) grown by molecular beam epitaxy. Journal of Vacuum Science & Technology an Official Journal of the American Vacuum Society B, Microelectronics Processing and Phenomena, 1991, 9, 2238.	1.6	22
101	Calculation of inelastic mean free path of photoelectrons in some solids. Journal of Electron Spectroscopy and Related Phenomena, 1988, 46, 255-267.	1.7	24