

Associaç prof Ludo B F Juurlink

List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/4968705/publications.pdf>

Version: 2024-02-01

55
papers

1,080
citations

394421

19
h-index

454955

30
g-index

56
all docs

56
docs citations

56
times ranked

1194
citing authors

#	ARTICLE	IF	CITATIONS
1	Density functional theory study of adsorption of H ₂ O, H, O, and OH on stepped platinum surfaces. Journal of Chemical Physics, 2014, 140, 134708.	3.0	83
2	Site-specific reactivity of molecules with surface defects—the case of H ₂ dissociation on Pt. Science, 2019, 363, 155-157.	12.6	72
3	Hydrogen adsorption and desorption at the Pt(110)-(1 \times 2) surface: experimental and theoretical study. Physical Chemistry Chemical Physics, 2013, 15, 6323.	2.8	67
4	The influence of step geometry on the desorption characteristics of O ₂ , D ₂ , and H ₂ O from stepped Pt surfaces. Journal of Chemical Physics, 2010, 132, 174705.	3.0	59
5	Co-adsorption of O and H ₂ O on Nanostructured Platinum Surfaces: Does OH Form at Steps?. Angewandte Chemie - International Edition, 2010, 49, 6572-6575.	13.8	50
6	Double-Stranded Water on Stepped Platinum Surfaces. Physical Review Letters, 2016, 116, 136101.	7.8	45
7	Subsurface Oxygen on Pt(111) and Its Reactivity for CO Oxidation. Catalysis Letters, 2012, 142, 1-6.	2.6	38
8	The Energy Dependence of the Ratio of Step and Terrace Reactivity for H ₂ Dissociation on Stepped Platinum. Angewandte Chemie - International Edition, 2011, 50, 5174-5177.	13.8	33
9	Initial stages of water solvation of stepped platinum surfaces. Physical Chemistry Chemical Physics, 2016, 18, 3416-3422.	2.8	32
10	Long-range influence of steps on water adsorption on clean and D-covered Pt surfaces. Physical Chemistry Chemical Physics, 2015, 17, 8530-8537.	2.8	27
11	Anomalous Dependence of the Reactivity on the Presence of Steps: Dissociation of D ₂ on Cu(211). Journal of Physical Chemistry Letters, 2018, 9, 170-175.	4.6	27
12	Hydrophobic interactions between water and pre-adsorbed D on the stepped Pt(533) surface. Physical Chemistry Chemical Physics, 2008, 10, 7169.	2.8	26
13	Hydrogen adsorption and desorption from Cu(111) and Cu(211). Physical Chemistry Chemical Physics, 2018, 20, 22477-22488.	2.8	26
14	A detailed TPD study of H ₂ O and pre-adsorbed O on the stepped Pt(553) surface. Physical Chemistry Chemical Physics, 2011, 13, 1629-1638.	2.8	25
15	Hydrophilic Interaction Between Low-Coordinated Au and Water: H ₂ O/Au(310) Studied with TPD and XPS. Journal of Physical Chemistry C, 2016, 120, 8693-8703.	3.1	23
16	CO Blocking of D ₂ Dissociative Adsorption on Ru(0001). ChemPhysChem, 2008, 9, 2372-2378.	2.1	21
17	The molecular dynamics of adsorption and dissociation of O ₂ on Pt(553). Journal of Chemical Physics, 2015, 143, 014703.	3.0	21
18	Steps on Pt stereodynamically filter sticking of O ₂ . Proceedings of the National Academy of Sciences of the United States of America, 2019, 116, 13862-13866.	7.1	21

#	ARTICLE	IF	CITATIONS
19	The interaction of water with Ni(111) and H/Ni(111) studied by TPD and HREELS. Physical Chemistry Chemical Physics, 2008, 10, 2227.	2.8	20
20	Interaction between H ₂ O and Preadsorbed D on the Stepped Pt(553) Surface. Journal of Physical Chemistry C, 2012, 116, 18706-18712.	3.1	20
21	Desorption of Water from Distinct Step Types on a Curved Silver Crystal. Molecules, 2014, 19, 10845-10862.	3.8	19
22	Transferability of the Specific Reaction Parameter Density Functional for H ₂ + Pt(111) to H ₂ + Pt(211). Journal of Physical Chemistry C, 2019, 123, 2973-2986.	3.1	18
23	The Interaction between H ₂ O and Preadsorbed O on the Stepped Pt(533) Surface. Journal of Physical Chemistry C, 2010, 114, 18953-18960.	3.1	17
24	Step-Type Selective Oxidation of Platinum Surfaces. Journal of Physical Chemistry C, 2016, 120, 22927-22935.	3.1	17
25	Step-type and step-density influences on CO adsorption probed by reflection absorption infrared spectroscopy using a curved Pt(1 1 1) surface. Journal of Vacuum Science and Technology A: Vacuum, Surfaces and Films, 2017, 35, .	2.1	16
26	Employing a cylindrical single crystal in gas-surface dynamics. Journal of Chemical Physics, 2012, 136, 114201.	3.0	15
27	Macroscopic and Microscopic Wettability of Graphene. Langmuir, 2021, 37, 4049-4055.	3.5	15
28	Tuning Hydrophobicity of Platinum by Small Changes in Surface Morphology. Physical Review Letters, 2011, 107, 146103.	7.8	14
29	Co-adsorption of water and hydrogen on Ni(111). Physical Chemistry Chemical Physics, 2008, 10, 4994.	2.8	13
30	Identification of Hydroxyl on Ni(111). ChemPhysChem, 2009, 10, 270-275.	2.1	13
31	Surface Structure Dependence in Desorption and Crystallization of Thin Interfacial Water Films on Platinum. Journal of Physical Chemistry Letters, 2016, 7, 1682-1685.	4.6	13
32	Reaction dynamics of initial O ₂ sticking on Pd(100). Journal of Chemical Physics, 2015, 142, 214708.	3.0	12
33	Recent advances in the use of curved single crystal surfaces. Progress in Surface Science, 2021, 96, 100627.	8.3	12
34	CO and H ₂ O adsorption and reaction on Au(310). Surface Science, 2011, 605, 1726-1731.	1.9	11
35	A molecular beam study of D ₂ dissociation on Pt(1̄1̄1̄): Testing SRP-DFT calculations. Chemical Physics Letters, 2018, 706, 680-683.	2.6	11
36	Scaling Platinum-Catalyzed Hydrogen Dissociation on Corrugated Surfaces. Angewandte Chemie - International Edition, 2020, 59, 20973-20979.	13.8	11

#	ARTICLE	IF	CITATIONS
37	Heterogeneous Catalytic Oxidation of Simple Alcohols by Transition Metals. Journal of Chemical Education, 2017, 94, 1285-1287.	2.3	10
38	Elucidation of temperature-programmed desorption of high-coverage hydrogen on Pt(211), Pt(221), Pt(533) and Pt(553) based on density functional theory calculations. Physical Chemistry Chemical Physics, 2019, 21, 17142-17151.	2.8	10
39	Exposure of Pt(5% ₃) and Rh(1% ₁) to atomic and molecular oxygen: do defects enhance subsurface oxygen formation?. Journal of Physics Condensed Matter, 2017, 29, 164002.	1.8	9
40	An Inexpensive 3D Printed Periscope-Type Smartphone-Based Spectrophotometer for Emission, Absorption, and Fluorescence Spectrometry. Journal of Chemical Education, 2022, 99, 2168-2174.	2.3	9
41	A Comparison of CO Oxidation by Hydroxyl and Atomic Oxygen from Water on Low-Coordinated Au Atoms. ACS Catalysis, 2016, 6, 7051-7058.	11.2	8
42	Coverage-dependent adsorption and desorption of oxygen on Pd(100). Journal of Chemical Physics, 2016, 144, 244706.	3.0	8
43	IR spectroscopic characterization of the co-adsorption of CO ₂ and H ₂ onto cationic Cu _n ⁺ clusters. Physical Chemistry Chemical Physics, 2021, 23, 26661-26673.	2.8	8
44	It's not just the defects – a curved crystal study of H ₂ O desorption from Ag. Physical Chemistry Chemical Physics, 2019, 21, 15422-15430.	2.8	7
45	Oxygen-induced surface reconstructions on curved Ag(111). Journal of Vacuum Science and Technology A: Vacuum, Surfaces and Films, 2021, 39, .	2.1	6
46	Hydrogen Dissociation on Stepped Pt Surfaces. Springer Series in Surface Sciences, 2013, , 101-129.	0.3	6
47	Heterogeneous Catalytic Oxidation of Ammonia by Various Transition Metals. Journal of Chemical Education, 2019, 96, 2266-2270.	2.3	5
48	Absolute dissociation cross sections for D ₂ dissociation on Pt steps. Chemical Physics Letters, 2021, 776, 138679.	2.6	5
49	Evidence of stable high-temperature Dx-CO intermediates on the Ru(0001) surface. Journal of Chemical Physics, 2012, 136, 114710.	3.0	4
50	Misconceptions in the Exploding Flask Demonstration Resolved through Students'™ Critical Thinking. Journal of Chemical Education, 2017, 94, 1209-1216.	2.3	4
51	Structural Inhibition of Silver Surface Oxidation. Journal of Physical Chemistry C, 2021, 125, 14702-14708.	3.1	4
52	Stepped surfaces. Journal of Physics Condensed Matter, 2018, 30, 090301.	1.8	3
53	Chiral Surface Characterisation and Reactivity Toward H ² D Exchange of a Curved Platinum Crystal. Topics in Catalysis, 2020, 63, 1558-1568.	2.8	3
54	Two Design Principles for the Design of Demonstrations to Enhance Structure-Property Reasoning. Education Sciences, 2021, 11, 504.	2.6	2

#	ARTICLE	IF	CITATIONS
55	Scaling Platinum-Catalyzed Hydrogen Dissociation on Corrugated Surfaces. <i>Angewandte Chemie</i> , 2020, 132, 21159-21165.	2.0	1