

# Jane Hvolbæk Nielsen

## List of Publications by Year in descending order

Source: <https://exaly.com/author-pdf/4890869/publications.pdf>

Version: 2024-02-01

41  
papers

11,573  
citations

257450

24  
h-index

289244

40  
g-index

42  
all docs

42  
docs citations

42  
times ranked

14562  
citing authors

#	ARTICLE	IF	CITATIONS
1	Identification of Active Edge Sites for Electrochemical H <sub>2</sub> Evolution from MoS <sub>2</sub> Nanocatalysts. <i>Science</i> , 2007, 317, 100-102.	12.6	5,149
2	Biomimetic Hydrogen Evolution: MoS <sub>2</sub> Nanoparticles as Catalyst for Hydrogen Evolution. <i>Journal of the American Chemical Society</i> , 2005, 127, 5308-5309.	13.7	3,497
3	Role of Steps in N <sub>2</sub> Activation on Ru(0001). <i>Physical Review Letters</i> , 1999, 83, 1814-1817.	7.8	706
4	Structure sensitivity of the methanation reaction: H <sub>2</sub> -induced CO dissociation on nickel surfaces. <i>Journal of Catalysis</i> , 2008, 255, 6-19.	6.2	411
5	The Effect of Size on the Oxygen Electroreduction Activity of Mass-Selected Platinum Nanoparticles. <i>Angewandte Chemie - International Edition</i> , 2012, 51, 4641-4643.	13.8	319
6	Mass-selected nanoparticles of Pt <sub>x</sub> Y as model catalysts for oxygen electroreduction. <i>Nature Chemistry</i> , 2014, 6, 732-738.	13.6	298
7	From fundamental studies of reactivity on single crystals to the design of catalysts. <i>Surface Science Reports</i> , 1999, 35, 163-222.	7.2	209
8	Modification of Ni(111) reactivity toward CH <sub>4</sub> , CO, and D <sub>2</sub> by two-dimensional alloying. <i>Journal of Chemical Physics</i> , 1996, 104, 7289-7295.	3.0	107
9	Designing surface alloys with specific active sites. <i>Catalysis Letters</i> , 1996, 40, 131-135.	2.6	77
10	CO dissociation on Ni: The effect of steps and of nickel carbonyl. <i>Surface Science</i> , 2008, 602, 733-743.	1.9	72
11	N <sub>2</sub> dissociation on Fe(110) and Fe/Ru(0001): what is the role of steps?. <i>Surface Science</i> , 2001, 491, 183-194.	1.9	67
12	The morphology of mass selected ruthenium nanoparticles from a magnetron-sputter gas-aggregation source. <i>Journal of Nanoparticle Research</i> , 2010, 12, 1249-1262.	1.9	53
13	Increased dissociation probability of CH <sub>4</sub> on Co/Cu(111). <i>Surface Science</i> , 1998, 405, 62-73.	1.9	50
14	Dynamic Behavior of CuZn Nanoparticles under Oxidizing and Reducing Conditions. <i>Journal of Physical Chemistry C</i> , 2015, 119, 2804-2812.	3.1	49
15	Dissociative sticking of CH <sub>4</sub> on Ru(0001). <i>Journal of Chemical Physics</i> , 1999, 110, 2637-2642.	3.0	46
16	Self Blocking of CO Dissociation on a Stepped Ruthenium Surface. <i>Topics in Catalysis</i> , 2010, 53, 357-364.	2.8	44
17	Molecular beam study of N <sub>2</sub> dissociation on Ru(0001). <i>Physical Chemistry Chemical Physics</i> , 2001, 3, 2007-2011.	2.8	34
18	Combined spectroscopy and microscopy of supported MoS <sub>2</sub> nanoparticles. <i>Surface Science</i> , 2009, 603, 1182-1189.	1.9	30

#	ARTICLE	IF	CITATIONS
19	Scanning Tunneling Microscopy Evidence for the Dissociation of Carbon Monoxide on Ruthenium Steps. <i>Journal of Physical Chemistry C</i> , 2012, 116, 14350-14359.	3.1	30
20	Methanol Decomposition on Pt/ZnO(0001)âˆ™Zn Model Catalysts. <i>Journal of Physical Chemistry B</i> , 2001, 105, 9273-9279.	2.6	26
21	Methanol Synthesis on Potassium-Modified Cu(100) from CO + H <sub>2</sub> and CO + CO <sub>2</sub> + H <sub>2</sub> . <i>Topics in Catalysis</i> , 2003, 22, 151-160.	2.8	26
22	A comparative STM study of Ru nanoparticles deposited on HOPG by mass-selected gas aggregation versus thermal evaporation. <i>Surface Science</i> , 2009, 603, 3420-3430.	1.9	25
23	Probing the active sites for CO dissociation on ruthenium nanoparticles. <i>Physical Chemistry Chemical Physics</i> , 2012, 14, 8005.	2.8	25
24	Is the methanation reaction over Ru single crystals structure dependent?. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 4486.	2.8	21
25	Methanation on mass-selected Ru nanoparticles on a planar SiO <sub>2</sub> model support: The importance of under-coordinated sites. <i>Journal of Catalysis</i> , 2013, 308, 282-290.	6.2	20
26	Structural Modification of Platinum Model Systems under High Pressure CO Annealing. <i>Journal of Physical Chemistry C</i> , 2012, 116, 15353-15360.	3.1	19
27	Enthalpies of adsorption of metal atoms on single-crystalline surfaces by microcalorimetry. <i>Journal of Chemical Thermodynamics</i> , 2001, 33, 333-345.	2.0	18
28	Growth and decomposition of lithium and lithium hydride on nickel. <i>Surface Science</i> , 2006, 600, 1468-1474.	1.9	18
29	Catalytic oxidation of graphite by mass-selected ruthenium nanoparticles. <i>Carbon</i> , 2011, 49, 376-385.	10.3	14
30	Biomimetic Hydrogen Evolution: MoS <sub>2</sub> Nanoparticles as Catalyst for Hydrogen Evolution. <i>ChemInform</i> , 2005, 36, no.	0.0	12
31	Reduction of a Ni/Spinel Catalyst for Methane Reforming. <i>Journal of Physical Chemistry C</i> , 2015, 119, 1424-1432.	3.1	12
32	Probing the crossover in CO desorption from single crystal to nanoparticulate Ru model catalysts. <i>Physical Chemistry Chemical Physics</i> , 2011, 13, 10333.	2.8	11
33	Catalyst dynamics: consequences for classical kinetic descriptions of reactors. <i>Chemical Engineering Journal</i> , 2001, 82, 219-230.	12.7	9
34	Decomposition of lithium amide and imide films on nickel. <i>Surface Science</i> , 2007, 601, 830-836.	1.9	7
35	Shape-Selection of Thermodynamically Stabilized Colloidal Pd and Pt Nanoparticles Controlled via Support Effects. <i>Journal of Physical Chemistry C</i> , 2015, 119, 29178-29185.	3.1	7
36	Enhanced reactivity of pseudomorphic Co on Cu(111). <i>Catalysis Letters</i> , 1998, 52, 1-5.	2.6	3

#	ARTICLE	IF	CITATIONS
37	PtRu Colloid Nanoparticles for CO Oxidation in Microfabricated Reactors. <i>Catalysis Letters</i> , 2006, 109, 7-12.	2.6	3
38	An Open-Source Data Storage and Visualization Back End for Experimental Data. <i>Journal of the Association for Laboratory Automation</i> , 2014, 19, 183-190.	2.8	3
39	Novel micro-reactor flow cell for investigation of model catalysts using <i>in situ</i> grazing-incidence X-ray scattering. <i>Journal of Synchrotron Radiation</i> , 2016, 23, 455-463.	2.4	2
40	Batch chemical microreactors: Reversible, <i>in situ</i> UHV sealing of a microcavity. <i>Microelectronic Engineering</i> , 2009, 86, 1389-1392.	2.4	0
41	Morphology of Ruthenium Particles for Methanation under Reactive Conditions. <i>Microscopy and Microanalysis</i> , 2014, 20, 416-417.	0.4	0