Fien Degryse

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88 3,255 34 55 h-index g-index citations papers 92 5.1 3,732 5.39 L-index avg, IF ext. citations ext. papers

#	Paper	IF	Citations
88	Partitioning of metals (Cd, Co, Cu, Ni, Pb, Zn) in soils: concepts, methodologies, prediction and applications (a review. <i>European Journal of Soil Science</i> , 2009 , 60, 590-612	3.4	258
87	Predicting availability of mineral elements to plants with the DGT technique: a review of experimental data and interpretation by modelling. <i>Environmental Chemistry</i> , 2009 , 6, 198	3.2	185
86	Fate and effect of zinc from tire debris in soil. Environmental Science & Envi	10 0.3	158
85	Metal complexation properties of freshwater dissolved organic matter are explained by its aromaticity and by anthropogenic ligands. <i>Environmental Science & Environmental Sci</i>	10.3	140
84	Labile Cd complexes increase Cd availability to plants. <i>Environmental Science & Environmental Science</i>	10.3	138
83	Copper speciation and isotopic fractionation in plants: uptake and translocation mechanisms. <i>New Phytologist</i> , 2013 , 199, 367-378	9.8	110
82	Solubility and toxicity of antimony trioxide (Sb2O3) in soil. <i>Environmental Science & Environmental &</i>	10.3	103
81	Soil solution concentration of Cd and Zn canbe predicted with a CaCl2 soil extract. <i>European Journal of Soil Science</i> , 2003 , 54, 149-158	3.4	86
80	Dissolution rate and agronomic effectiveness of struvite fertilizers Leffect of soil pH, granulation and base excess. <i>Plant and Soil</i> , 2017 , 410, 139-152	4.2	83
79	Metal complexes increase uptake of Zn and Cu by plants: implications for uptake and deficiency studies in chelator-buffered solutions. <i>Plant and Soil</i> , 2006 , 289, 171-185	4.2	83
78	The copper-mobilizing-potential of dissolved organic matter in soils varies 10-fold depending on soil incubation and extraction procedures. <i>Environmental Science & Environmental Science & Environme</i>	10.3	81
77	Efficacy of Hydroxyapatite Nanoparticles as Phosphorus Fertilizer in Andisols and Oxisols. <i>Soil Science Society of America Journal</i> , 2015 , 79, 551-558	2.5	79
76	The UV-absorbance of dissolved organic matter predicts the fivefold variation in its affinity for mobilizing Cu in an agricultural soil horizon. <i>European Journal of Soil Science</i> , 2008 , 59, 1087-1095	3.4	70
75	Graphene Oxide: A New Carrier for Slow Release of Plant Micronutrients. <i>ACS Applied Materials & Amp; Interfaces</i> , 2017 , 9, 43325-43335	9.5	66
74	Zinc toxicity to nitrification in soil and soilless culture can be predicted with the same biotic ligand model. <i>Environmental Science & Environmental Science & Environmental</i>	10.3	64
73	Radio-labile cadmium and zinc in soils as affected by pH and source of contamination. <i>European Journal of Soil Science</i> , 2004 , 55, 113-122	3.4	64
72	Diffusion limitations in root uptake of cadmium and zinc, but not nickel, and resulting bias in the Michaelis constant. <i>Plant Physiology</i> , 2012 , 160, 1097-109	6.6	57

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71	Copper isotope fractionation during equilibration with natural and synthetic ligands. <i>Environmental Science & Environmental &</i>	10.3	54	
70	Effect of organic P forms and P present in inorganic colloids on the determination of dissolved P in environmental samples by the diffusive gradient in thin films technique, ion chromatography, and colorimetry. <i>Analytical Chemistry</i> , 2011 , 83, 5317-23	7.8	54	
69	Mobilization of Cu and Zn by root exudates of dicotyledonous plants in resin-buffered solutions and in soil. <i>Plant and Soil</i> , 2008 , 306, 69-84	4.2	54	
68	The performance of DGT versus conventional soil phosphorus tests in tropical soils - An isotope dilution study. <i>Plant and Soil</i> , 2012 , 359, 267-279	4.2	53	
67	Relating soil solution Zn concentration to diffusive gradients in thin films measurements in contaminated soils. <i>Environmental Science & Environmental Science & Environmenta</i>	10.3	52	
66	Agronomic Effectiveness of Zinc Sources as Micronutrient Fertilizer. <i>Advances in Agronomy</i> , 2016 , 139, 215-267	7.7	49	
65	Modelling the effects of ageing on Cd, Zn, Ni and Cu solubility in soils using an assemblage model. <i>European Journal of Soil Science</i> , 2008 , 59, 1160-1170	3.4	47	
64	Speciation of nickel in surface waters measured with the Donnan membrane technique. <i>Analytica Chimica Acta</i> , 2006 , 578, 195-202	6.6	47	
63	Aluminum-Activated Malate Transporters Can Facilitate GABA Transport. Plant Cell, 2018, 30, 1147-116	5411.6	45	
62	Labile lead in polluted soils measured by stable isotope dilution. <i>European Journal of Soil Science</i> , 2007 , 58, 1-7	3.4	42	
61	Zinc speciation in mining and smelter contaminated overbank sediments by EXAFS spectroscopy. <i>Geochimica Et Cosmochimica Acta</i> , 2010 , 74, 3707-3720	5.5	41	
60	Characterization of zinc in contaminated soils: complementary insights from isotopic exchange, batch extractions and XAFS spectroscopy. <i>European Journal of Soil Science</i> , 2011 , 62, 318-330	3.4	38	
59	Agronomic Effectiveness of Granulated and Powdered P-Exchanged Mg-Al LDH Relative to Struvite and MAP. <i>Journal of Agricultural and Food Chemistry</i> , 2017 , 65, 6736-6744	5.7	36	
58	Mobility of Cd and Zn in polluted and unpolluted Spodosols. <i>European Journal of Soil Science</i> , 2006 , 57, 122-133	3.4	35	
57	Natural colloidal P and its contribution to plant P uptake. <i>Environmental Science & Environmental Sci</i>	10.3	34	
56	Isotopic fractionation of Zn in tomato plants suggests the role of root exudates on Zn uptake. <i>Plant and Soil</i> , 2013 , 370, 605-613	4.2	34	
55	Critical loads of metals and other trace elements to terrestrial environments. <i>Environmental Science & Environmental </i>	10.3	34	
54	Elemental Sulfur Oxidation in Australian Cropping Soils. <i>Soil Science Society of America Journal</i> , 2015 , 79, 89-96	2.5	31	

53	Uptake of Metals from Soil into Vegetables 2011 , 325-367		31
52	First observation of diffusion-limited plant root phosphorus uptake from nutrient solution. <i>Plant, Cell and Environment,</i> 2012 , 35, 1558-66	8.4	29
51	Phosphorus Diffusion from Fertilizer: Visualization, Chemical Measurements, and Modeling. <i>Soil Science Society of America Journal</i> , 2014 , 78, 832-842	2.5	28
50	DGT-measured fluxes explain the chloride-enhanced cadmium uptake by plants at low but not at high Cd supply. <i>Plant and Soil</i> , 2009 , 318, 127-135	4.2	28
49	Improving the efficacy of selenium fertilizers for wheat biofortification. Scientific Reports, 2019, 9, 195	20 _{4.9}	27
48	An anion resin membrane technique to overcome detection limits of isotopically exchanged P in P-sorbing soils. <i>European Journal of Soil Science</i> , 2004 , 55, 63-69	3.4	26
47	Mechanisms of enhanced mobilisation of trace metals by anionic surfactants in soil. <i>Environmental Pollution</i> , 2011 , 159, 809-16	9.3	25
46	Oxidation of Elemental Sulfur in Granular Fertilizers Depends on the Soil-Exposed Surface Area. <i>Soil Science Society of America Journal</i> , 2016 , 80, 294-305	2.5	24
45	Manganese Toxicity in Barley is Controlled by Solution Manganese and Soil Manganese Speciation. <i>Soil Science Society of America Journal</i> , 2012 , 76, 399-407	2.5	24
44	Enhanced sorption and fixation of radiocaesium in soils amended with K-bentonites, submitted to wetting drying cycles. <i>European Journal of Soil Science</i> , 2004 , 55, 513-522	3.4	24
43	Limited Dissolved Phosphorus Runoff Losses from Layered Double Hydroxide and Struvite Fertilizers in a Rainfall Simulation Study. <i>Journal of Environmental Quality</i> , 2018 , 47, 371-377	3.4	22
42	Sequestration of Phosphorus-Binding Cations by Complexing Compounds is not a Viable Mechanism to Increase Phosphorus Efficiency. <i>Soil Science Society of America Journal</i> , 2013 , 77, 2050-20	05 ² 9 ⁵	22
41	Availability of fertiliser sulphate and elemental sulphur to canola in two consecutive crops. <i>Plant and Soil</i> , 2016 , 398, 313-325	4.2	20
40	Fluid Fertilizers Improve Phosphorus Diffusion but not Lability in Andisols and Oxisols. <i>Soil Science Society of America Journal</i> , 2014 , 78, 214-224	2.5	20
39	Cadmium and nickel uptake by tomato and spinach seedlings: plant or transport control?. <i>Environmental Chemistry</i> , 2012 , 9, 48	3.2	20
38	The dissociation kinetics of Cu-dissolved organic matter complexes from soil and soil amendments. <i>Analytica Chimica Acta</i> , 2010 , 670, 24-32	6.6	20
37	Abundance and diversity of sulphur-oxidising bacteria and their role in oxidising elemental sulphur in cropping soils. <i>Biology and Fertility of Soils</i> , 2017 , 53, 159-169	6.1	19
36	An Agar Gel Technique Demonstrates Diffusion Limitations to Cadmium Uptake by Higher Plants. <i>Environmental Chemistry</i> , 2006 , 3, 419	3.2	18

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35	Formulation, synthesis and characterization of boron phosphate (BPO4) compounds as raw materials to develop slow-release boron fertilizers. <i>Journal of Plant Nutrition and Soil Science</i> , 2014 , 177, 860-868	2.3	16	
34	Mobilization of Zn upon waterlogging riparian Spodosols is related to reductive dissolution of Fe minerals. <i>European Journal of Soil Science</i> , 2010 , 61, 1014-1024	3.4	16	
33	Uptake of elemental or sulfate-S from fall- or spring-applied co-granulated fertilizer by corn stable isotope and modeling study. <i>Field Crops Research</i> , 2018 , 221, 322-332	5.5	15	
32	Sulfur and Zinc Availability from Co-granulated Zn-Enriched Elemental Sulfur Fertilizers. <i>Journal of Agricultural and Food Chemistry</i> , 2017 , 65, 1108-1115	5.7	13	
31	Mobilization of Cd upon acidification of agricultural soils: column study and field modelling. <i>European Journal of Soil Science</i> , 2007 , 58, 152-165	3.4	13	
30	Model studies of corrosion-induced copper runoff fate in soil. <i>Environmental Toxicology and Chemistry</i> , 2006 , 25, 683-91	3.8	12	
29	Diffusion and solubility control of fertilizer-applied zinc: chemical assessment and visualization. <i>Plant and Soil</i> , 2015 , 386, 195-204	4.2	11	
28	Labile complexes facilitate cadmium uptake by Caco-2 cells. <i>Science of the Total Environment</i> , 2012 , 426, 90-9	10.2	10	
27	Agronomic Effectiveness of Granular and Fluid Phosphorus Fertilizers in Andisols and Oxisols. <i>Soil Science Society of America Journal</i> , 2015 , 79, 577-584	2.5	10	
26	Slow and Fast-Release Boron Sources in Potash Fertilizers: Spatial Variability, Nutrient Dissolution and Plant Uptake. <i>Soil Science Society of America Journal</i> , 2018 , 82, 1437-1448	2.5	10	
25	Effects of pH and ionic strength on elemental sulphur oxidation in soil. <i>Biology and Fertility of Soils</i> , 2017 , 53, 247-256	6.1	9	
24	Responses of Canola to the Application of Slow-Release Boron Fertilizers and Their Residual Effect. <i>Soil Science Society of America Journal</i> , 2015 , 79, 97-103	2.5	9	
23	Boron phosphates (BPO4) as a seedling-safe boron fertilizer source. <i>Plant and Soil</i> , 2015 , 391, 153-160	4.2	8	
22	Slow-release boron fertilisers: co-granulation of boron sources with mono-ammonium phosphate (MAP). <i>Soil Research</i> , 2015 , 53, 505	1.8	8	
21	A stable-isotope methodology for measurement of soil-applied zinc-fertilizer recovery in durum wheat (Triticum durum). <i>Journal of Plant Nutrition and Soil Science</i> , 2013 , 176, 756-763	2.3	8	
20	Engineered Phosphate Fertilizers with Dual-Release Properties. <i>Industrial & Engineering Chemistry Research</i> , 2020 , 59, 5512-5524	3.9	7	
19	Model-based rationalization of sulphur mineralization in soils using 35S isotope dilution. <i>Soil Biology and Biochemistry</i> , 2018 , 120, 1-11	7.5	7	
18	A column perfusion test to assess the kinetics of nutrient release by soluble, sparingly soluble and coated granular fertilizers. <i>Journal of Plant Nutrition and Soil Science</i> , 2019 , 182, 763-771	2.3	7	

17	Rapid and Low-Cost Method for Evaluation of Nutrient Release from Controlled-Release Fertilizers Using Electrical Conductivity. <i>Journal of Polymers and the Environment</i> , 2018 , 26, 4388-4395	4.5	7
16	Effect of Cogranulation on Oxidation of Elemental Sulfur: Theoretical Model and Experimental Validation. <i>Soil Science Society of America Journal</i> , 2016 , 80, 1244-1253	2.5	6
15	Efficiency of soil-applied 67Zn-enriched fertiliser across three consecutive crops. <i>Pedosphere</i> , 2021 , 31, 531-537	5	4
14	Sulfur Uptake from Fertilizer Fortified with Sulfate and Elemental S in Three Contrasting Climatic Zones. <i>Agronomy</i> , 2020 , 10, 1035	3.6	3
13	DGT and Bioavailability216-262		3
12	Effect of soil properties on time-dependent fixation (ageing) of selenate. <i>Geoderma</i> , 2021 , 383, 114741	6.7	3
11	Low Effective Surface Area Explains Slow Oxidation of Co-Granulated Elemental Sulfur. <i>Soil Science Society of America Journal</i> , 2016 , 80, 911-918	2.5	2
10	Comparison and modelling of extraction methods to assess agronomic effectiveness of fertilizer zinc. <i>Journal of Plant Nutrition and Soil Science</i> , 2020 , 183, 248-259	2.3	1
9	Development and Testing of Improved Efficiency Boron-Enriched Diammonium Phosphate Fertilizers. <i>Journal of Soil Science and Plant Nutrition</i> , 2021 , 21, 1134-1143	3.2	1
8	Mechanochemical Synthesis of Zinc Borate for Use as a Dual-Release B Fertilizer. <i>ACS Sustainable Chemistry and Engineering</i> , 2021 , 9, 15995-16004	8.3	О
7	Layered Double Hydroxides as Slow-Release Fertilizer Compounds for the Micronutrient Molybdenum. <i>Journal of Agricultural and Food Chemistry</i> , 2021 , 69, 14501-14511	5.7	O
6	Long-term fate of fertilizer sulfate- and elemental S in co-granulated fertilizers. <i>Nutrient Cycling in Agroecosystems</i> , 2021 , 120, 31-48	3.3	Ο
5	Application method influences the oxidation rate of biologically and chemically produced elemental sulfur fertilizers. <i>Soil Science Society of America Journal</i> , 2021 , 85, 746-759	2.5	О
4	Isotopic signatures reveal zinc cycling in the natural habitat of hyperaccumulator Dichapetalum gelonioides subspecies from Malaysian Borneo. <i>BMC Plant Biology</i> , 2021 , 21, 437	5.3	Ο
3	Using Se-Labelled Foliar Fertilisers to Determine How Se Transfers Within Wheat Over Time. <i>Frontiers in Nutrition</i> , 2021 , 8, 732409	6.2	
2	Fixation of Cadmium and Zinc in Soils 2006 , 157-172		
1	Screening fertilizers for their phosphorus runoff risk using laboratory methods. <i>Journal of Environmental Quality</i> , 2021 , 50, 955-966	3.4	