## Asghar Azizi

## List of Publications by Citations

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| #  | Paper  | IF               | Citations |
|----|--|------------------|-----------|
| 35 | Fabrication and investigation of MnFe2O4/MWCNTs nanocomposite by hydrothermal technique and adsorption of cationic and anionic dyes. <i>Applied Surface Science</i> , <b>2017</b> , 419, 70-83   | 6.7              | 67        |
| 34 | Adsorption of gold from cyanide leaching solution onto activated carbon originating from coconut shell ptimization, kinetics and equilibrium studies. <i>Journal of Industrial and Engineering Chemistry</i> , <b>2017</b> , 54, 464-471 | 6.3              | 49        |
| 33 | Alkaline leaching of lead and zinc by sodium hydroxide: kinetics modeling. <i>Journal of Materials Research and Technology</i> , <b>2018</b> , 7, 118-125  | 5.5              | 44        |
| 32 | Investigating the first-order flotation kinetics models for Sarcheshmeh copper sulfide ore. <i>International Journal of Mining Science and Technology</i> , <b>2015</b> , 25, 849-854  | 7.1              | 36        |
| 31 | Modeling and optimization of Direct Red 16 adsorption from aqueous solutions using nanocomposite of MnFe2O4/MWCNTs: RSM-CCRD model. <i>Journal of Molecular Liquids</i> , <b>2017</b> , 233, 370-3                                       | <del>6</del> 77  | 34        |
| 30 | Adsorption of lead(II) and chromium(VI) from aqueous environment onto metal-organic framework MIL-100(Fe): Synthesis, kinetics, equilibrium and thermodynamics. <i>Journal of Solid State Chemistry</i> , <b>2020</b> , 291, 121636      | 3.3              | 32        |
| 29 | Synthesis and characterization of manganese ferrite nanostructure by co-precipitation, sol-gel, and hydrothermal methods. <i>Particulate Science and Technology</i> , <b>2019</b> , 37, 904-910  | 2                | 29        |
| 28 | Leaching of zinc from a lead-zinc flotation tailing sample using ferric sulphate and sulfuric acid media. <i>Journal of Environmental Chemical Engineering</i> , <b>2017</b> , 5, 4769-4775  | 6.8              | 25        |
| 27 | Estimation of flotation rate constant and particle-bubble interactions considering key hydrodynamic parameters and their interrelations. <i>Minerals Engineering</i> , <b>2019</b> , 141, 105836   | 4.9              | 24        |
| 26 | Investigating the best mixture extraction systems in the separation of rare earth elements from nitric acid solution using Cyanex272, D2EHPA, and 8-Hydroxyquinoline. <i>Geosystem Engineering</i> , <b>2016</b> , 19, 32-38             | 1.2              | 15        |
| 25 | Solvent extraction of zinc from sulphate leaching solution of a sulphide-oxide sample using D2EHPA and Cyanex 272. <i>Journal of Dispersion Science and Technology</i> , <b>2018</b> , 39, 1328-1334                                     | 1.5              | 14        |
| 24 | Galvanic Interaction between Chalcopyrite and Pyrite with Low Alloy and High Carbon Chromium Steel Ball. <i>Journal of Chemistry</i> , <b>2013</b> , 2013, 1-9   | 2.3              | 10        |
| 23 | Parametric Optimization in Rougher Flotation Performance of a Sulfidized Mixed Copper Ore. <i>Minerals (Basel, Switzerland)</i> , <b>2020</b> , 10, 660  | 2.4              | 10        |
| 22 | A study on the modified flotation parameters and selectivity index in copper flotation. <i>Particulate Science and Technology</i> , <b>2017</b> , 35, 38-44  | 2                | 9         |
| 21 | Solvent Extraction of Copper and Zinc from Sulfate Leach Solution Derived from a Porcelain Stone Tailings Sample with Chemorex CP-150 and D2EHPA. <i>Journal of Sustainable Metallurgy</i> , <b>2020</b> , 6, 250-258                    | 3 <sup>2.7</sup> | 8         |
| 20 | The effect of pH, solid content, water chemistry and ore mineralogy on the galvanic interactions between chalcopyrite and pyrite and steel balls. <i>Frontiers of Chemical Science and Engineering</i> , <b>2013</b> , 7, 464-471        | 4.5              | 8         |
| 19 | A comparative analysis of the dissolution kinetics of lead from low grade oxide ores in HCl, H2SO4, HNO3 and citric acid solutions. <i>Metallurgical Research and Technology</i> , <b>2017</b> , 114, 406                                | 0.9              | 8         |

## (2021-2015)

| 18 | Optimization of Rougher Flotation Parameters of the Sarcheshmeh Copper Ore Using a Statistical Technique. <i>Journal of Dispersion Science and Technology</i> , <b>2015</b> , 36, 1066-1072   | 1.5                            | 8 |  |
|----|---|--------------------------------|---|--|
| 17 | Optimizing and evaluating the operational factors affecting the cyanide leaching circuit of the Aghdareh gold processing plant using a CCD model. <i>Proceedings of the Royal Society A:</i> Mathematical, Physical and Engineering Sciences, 2015, 471, 20150681 | 2.4                            | 7 |  |
| 16 | Influence of collector dosage and pulp chemistry on copper flotation. <i>Geosystem Engineering</i> , <b>2014</b> , 17, 311-316  | 1.2                            | 6 |  |
| 15 | Mechanochemical sulfidization of a mixed oxide-sulphide copper ore by co-grinding with sulfur and its effect on the flotation efficiency. <i>Chinese Journal of Chemical Engineering</i> , <b>2020</b> , 28, 743-748  | 3.2                            | 6 |  |
| 14 | Relative floatability as a criterion for evaluating the separation performance of phosphate from iron. <i>International Journal of Mining Science and Technology</i> , <b>2017</b> , 27, 451-458  | 7.1                            | 5 |  |
| 13 | Investigating the controllable factors influencing the weight loss of grinding ball using SEM/EDX analysis and RSM model <b>2015</b> , 18, 278-285  |                                | 5 |  |
| 12 | A comprehensive study of the leaching behavior and dissolution kinetics of copper oxide ore in sulfuric acid lixiviant. <i>Scientia Iranica</i> , <b>2018</b> , 0-0   | 1.5                            | 5 |  |
| 11 | Leaching of copper and zinc from the tailings sample obtained from a porcelain stone mine: feasibility, modeling, and optimization. <i>Environmental Science and Pollution Research</i> , <b>2020</b> , 27, 6239-62   | .5 <sup>5</sup> 2 <sup>1</sup> | 4 |  |
| 10 | Experimental and Kinetic Modeling Investigation of Copper Dissolution Process from an Iranian Mixed Oxide/Sulfide Copper Ore. <i>Journal of Sustainable Metallurgy</i> , <b>2020</b> , 6, 437-450   | 2.7                            | 3 |  |
| 9  | Modeling and prediction of wear rate of grinding media in mineral processing industry using multiple kernel support vector machine. <i>SN Applied Sciences</i> , <b>2020</b> , 2, 1   | 1.8                            | 3 |  |
| 8  | A Study on the Corrosive and Abrasive Wear of Grinding Media in Grinding of Minerals Using Fuzzy Analytical Hierarchy Delphi Method. <i>Arabian Journal for Science and Engineering</i> , <b>2014</b> , 39, 3373-3382   |                                | 2 |  |
| 7  | An Investigation into the Extraction Behavior of Copper from Sulfate Leach Liquor Using Acorga M5640 Extractant: Mechanism, Equilibrium, and Thermodynamics. <i>Mining, Metallurgy and Exploration</i> , <b>2020</b> , 37, 1673-1680                              | 1.1                            | 2 |  |
| 6  | Modelling and simulation of the cyanidation process of Aghdareh gold ore using artificial neural network and multiple linear regression. <i>International Journal of Mining and Mineral Engineering</i> , <b>2016</b> , 7, 139                                    | 0.7                            | 2 |  |
| 5  | Optimizing the alkaline oxidation pretreatment of a refractory gold ore using taguchi orthogonal array method. <i>Materials Research Express</i> , <b>2018</b> , 5, 126516  | 1.7                            | 2 |  |
| 4  | An investigation into the recovery of oxide copper from a complex copper ore using sulphidisation technique and hydroxamate and potassium amyl xanthate collectors. <i>Geosystem Engineering</i> , <b>2020</b> , 23, 43-50  | 1.2                            | 1 |  |
| 3  | Solvent extraction and kinetic studies of copper from a heap leach liquor using CuPRO MEX-3302. <i>Separation Science and Technology</i> ,1-18  | 2.5                            | 1 |  |
| 2  | Kinetic Investigation on Leaching of Copper from a Low-Grade Copper Oxide Deposit in Sulfuric Acid Solution: A Case Study of the Crushing Circuit Reject of a Copper Heap Leaching Plant. <i>Journal of Sustainable Metallurgy</i> , <b>2021</b> , 7, 1154-1168   | 2.7                            | 1 |  |
| 1  | Recycling lead from a zinc plant residue (ZPR) using brine leaching and cementation with aluminum powder. <i>Environmental Science and Pollution Research</i> , <b>2021</b> , 28, 42121-42134   | 5.1                            | О |  |