## Daniel E Giammar

# List of Publications by Year in Descending Order

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The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

62 119 4,471 40 h-index g-index citations papers 8.3 130 5,200 5.91 avg, IF L-index ext. citations ext. papers

#	Paper	IF	Citations
119	Influence of point-of-use filters and stagnation on water quality at a preschool and under laboratory conditions <i>Water Research</i> , <b>2022</b> , 211, 118034	12.5	O
118	Impact of dissolved oxygen and pH on the removal of selenium from water by iron electrocoagulation <i>Water Research</i> , <b>2022</b> , 213, 118159	12.5	1
117	Lead phosphate deposition in porous media and implications for lead remediation <i>Water Research</i> , <b>2022</b> , 214, 118200	12.5	1
116	Cost and Energy Metrics for Municipal Water Reuse. ACS ES&T Engineering, 2022, 2, 489-507		1
115	Ligand-Induced U Mobilization from Chemogenic Uraninite and Biogenic Noncrystalline U(IV) under Anoxic Conditions <i>Environmental Science &amp; Environmental Science &amp; Environme</i>	10.3	O
114	Copper availability governs nitrous oxide accumulation in wetland soils and stream sediments. <i>Geochimica Et Cosmochimica Acta</i> , <b>2022</b> , 327, 96-115	5.5	0
113	Consistent Controls on Trace Metal Micronutrient Speciation in Wetland Soils and Stream Sediments. <i>Geochimica Et Cosmochimica Acta</i> , <b>2021</b> ,	5.5	1
112	National Alliance for Water Innovation (NAWI) Municipal Sector Technology Roadmap 2021 <b>2021</b> ,		2
111	Pilot-scale comparison of sodium silicates, orthophosphate and pH adjustment to reduce lead release from lead service lines. <i>Water Research</i> , <b>2021</b> , 195, 116955	12.5	4
110	Intercomparison and Refinement of Surface Complexation Models for U(VI) Adsorption onto Goethite Based on a Metadata Analysis. <i>Environmental Science &amp; Environmental Science </i>	10.3	3
109	Effect of sodium silicate on lead release from lead service lines. Water Research, 2021, 188, 116485	12.5	8
108	Lead Phosphate Particles in Tap Water: Challenges for Point-of-Use Filters. <i>Environmental Science and Technology Letters</i> , <b>2021</b> , 8, 244-249	11	7
107	Effects of Cu(II) and Zn(II) on PbO Reductive Dissolution under Drinking Water Conditions: Short-term Inhibition and Long-term Enhancement. <i>Environmental Science &amp; amp; Technology</i> , <b>2021</b> , 55, 14397-14406	10.3	1
106	Worth a Closer Look: Raman Spectra of Lead-Pipe Scale. <i>Minerals (Basel, Switzerland)</i> , <b>2021</b> , 11, 1047	2.4	2
105	Surface functionalized nanoscale metal oxides for arsenic(V), chromium(VI), and uranium(VI) sorption: considering single- and multi-sorbate dynamics. <i>Environmental Science: Nano</i> , <b>2020</b> , 7, 3805-3	873 <sup>1</sup>	2
104	Interplay of transport processes and interfacial chemistry affecting chromium reduction and reoxidation with iron and manganese. <i>Frontiers of Environmental Science and Engineering</i> , <b>2020</b> , 14, 1	5.8	5
103	Modeling performance of rhamnolipid-coated engineered magnetite nanoparticles for U(VI) sorption and separation. <i>Environmental Science: Nano</i> , <b>2020</b> , 7, 2010-2020	7.1	6

### (2018-2020)

102	Formation and Transport of Cr(III)-NOM-Fe Colloids upon Reaction of Cr(VI) with NOM-Fe(II) Colloids at Anoxic-Oxic Interfaces. <i>Environmental Science &amp; Environmental Science </i>	10.3	27
101	Effect of Cu(II) on Mn(II) Oxidation by Free Chlorine To Form Mn Oxides at Drinking Water Conditions. <i>Environmental Science &amp; Environmental Science &amp;</i>	10.3	8
100	Effect of Aluminum on Lead Release to Drinking Water from Scales of Corrosion Products. <i>Environmental Science &amp; Environmental Science &amp; Environmental</i>	10.3	7
99	Impact of orthophosphate on lead release from pipe scale in high pH, low alkalinity water. <i>Water Research</i> , <b>2020</b> , 177, 115764	12.5	13
98	The Ability of Phosphate To Prevent Lead Release from Pipe Scale When Switching from Free Chlorine to Monochloramine. <i>Environmental Science &amp; Environmental Science &amp; Environ</i>	10.3	17
97	Accumulation on and extraction of lead from point-of-use filters for evaluating lead exposure from drinking water. <i>Environmental Science: Water Research and Technology</i> , <b>2020</b> , 6, 2734-2741	4.2	5
96	Cr(VI) Formation from CrxFe1II(OH)3 Induced by Mn(II) Oxidation on the Surface of CrxFe1II(OH)3. <i>ACS Earth and Space Chemistry</i> , <b>2020</b> , 4, 1558-1564	3.2	6
95	Impact of iron-rich scale in service lines on lead release to water. AWWA Water Science, 2020, 2, e1188	1.6	2
94	Cr(VI) Adsorption on Engineered Iron Oxide Nanoparticles: Exploring Complexation Processes and Water Chemistry. <i>Environmental Science &amp; Environmental Science &amp; Environmental</i>	10.3	37
93	Understanding the Roles of Dissolution and Diffusion in Cr(OH)3 Oxidation by EMnO2. <i>ACS Earth and Space Chemistry</i> , <b>2019</b> , 3, 357-365	3.2	13
92	Role of Manganese in Accelerating the Oxidation of Pb(II) Carbonate Solids to Pb(IV) Oxide at Drinking Water Conditions. <i>Environmental Science &amp; Environmental Science &amp; Envi</i>	10.3	24
91	Tackling Deficiencies in the Presentation and Interpretation of Adsorption Results for New Materials. <i>Environmental Science &amp; Environmental Science &amp;</i>	10.3	17
90	Geochemical Stability of Dissolved Mn(III) in the Presence of Pyrophosphate as a Model Ligand: Complexation and Disproportionation. <i>Environmental Science &amp; Environmental Sci</i>	10.3	33
89	Impact of Cu(II) and Zn(II) on the Reductive Dissolution of Pb(IV) Oxide. <i>Environmental Science and Technology Letters</i> , <b>2019</b> , 6, 745-751	11	3
88	CO2 Mineral Sequestration in Naturally Porous Basalt. <i>Environmental Science and Technology Letters</i> , <b>2018</b> , 5, 142-147	11	28
87	Enhanced Uranium Immobilization by Phosphate Amendment under Variable Geochemical and Flow Conditions: Insights from Reactive Transport Modeling. <i>Environmental Science &amp; Environmental Science &amp; Env</i>	10.3	19
86	Engineered superparamagnetic nanomaterials for arsenic(V) and chromium(VI) sorption and separation: quantifying the role of organic surface coatings. <i>Environmental Science: Nano</i> , <b>2018</b> , 5, 556-5	563 <sup>1</sup>	19
85	Heterogeneous Lead Phosphate Nucleation at Organic Water Interfaces: Implications for Lead Immobilization. ACS Earth and Space Chemistry, 2018, 2, 869-877	3.2	10

84	Permanent CO Trapping through Localized and Chemical Gradient-Driven Basalt Carbonation. <i>Environmental Science &amp; Environmental Science &amp; Environmenta</i>	10.3	15
83	Water metal contaminants in a potentially mineral-deficient population of Haiti. <i>International Journal of Environmental Health Research</i> , <b>2018</b> , 28, 626-634	3.6	4
82	Formation and Aggregation of Lead Phosphate Particles: Implications for Lead Immobilization in Water Supply Systems. <i>Environmental Science &amp; Environmental Science &amp; Environm</i>	10.3	40
81	Carbon Sequestration in Olivine and Basalt Powder Packed Beds. <i>Environmental Science &amp; Emp; Technology</i> , <b>2017</b> , 51, 2105-2112	10.3	15
80	Spatially-variable carbonation reactions in polycrystalline olivine. <i>Geochimica Et Cosmochimica Acta</i> , <b>2017</b> , 204, 252-266	5.5	5
79	Effect of Humic Acid on the Removal of Chromium(VI) and the Production of Solids in Iron Electrocoagulation. <i>Environmental Science &amp; Electrocoagulation</i> (2017), 51, 6308-6318	10.3	58
78	Formation, Aggregation, and Deposition Dynamics of NOM-Iron Colloids at Anoxic-Oxic Interfaces. <i>Environmental Science &amp; Environmental Science &amp; Envir</i>	10.3	58
77	Rates of Cr(VI) Generation from CrFe(OH) Solids upon Reaction with Manganese Oxide. <i>Environmental Science &amp; Environmental Sci</i>	10.3	49
76	CO2 mineral trapping in fractured basalt. International Journal of Greenhouse Gas Control, 2017, 66, 204	I-2.1 <sub>2</sub> 7	23
75	Roles of Transport Limitations and Mineral Heterogeneity in Carbonation of Fractured Basalts. <i>Environmental Science &amp; Environmental &amp;</i>	10.3	15
74	Effect of transport limitations and fluid properties on reaction products in fractures of unaltered and serpentinized basalt exposed to high PCO2 fluids. <i>International Journal of Greenhouse Gas Control</i> , <b>2017</b> , 63, 310-320	4.2	19
73	Measurement and Surface Complexation Modeling of U(VI) Adsorption to Engineered Iron Oxide Nanoparticles. <i>Environmental Science &amp; Environmental Scien</i>	10.3	36
72	Dissolution and surface roughening of Columbia River flood basalt at geologic carbon sequestration conditions. <i>Chemical Geology</i> , <b>2017</b> , 467, 100-109	4.2	8
71	Evidence from 29Si Solid-State Nuclear Magnetic Resonance of Dissolution Reactions of Forsterite. <i>Environmental Engineering Science</i> , <b>2016</b> , 33, 799-805	2	8
70	Synergistic Effects between Biogenic Ligands and a Reductant in Fe Acquisition from Calcareous Soil. <i>Environmental Science &amp; Environmental Science &amp; </i>	10.3	18
69	Effect of Reaction Pathway on the Extent and Mechanism of Uranium(VI) Immobilization with Calcium and Phosphate. <i>Environmental Science &amp; Environmental Science &amp; Environmenta</i>	10.3	33
68	Element mobilization from Bakken shales as a function of water chemistry. <i>Chemosphere</i> , <b>2016</b> , 149, 286-93	8.4	29
67	Phosphate-Induced Immobilization of Uranium in Hanford Sediments. <i>Environmental Science &amp; Environmental Science &amp; Technology</i> , <b>2016</b> , 50, 13486-13494	10.3	22

### (2014-2016)

66	Dynamics of Chromium(VI) Removal from Drinking Water by Iron Electrocoagulation. <i>Environmental Science &amp; Environmental Scienc</i>	10.3	78
65	Effect of phosphate on U(VI) sorption to montmorillonite: Ternary complexation and precipitation barriers. <i>Geochimica Et Cosmochimica Acta</i> , <b>2016</b> , 175, 86-99	5.5	52
64	MINFIT: A Spreadsheet-Based Tool for Parameter Estimation in an Equilibrium Speciation Software Program. <i>Environmental Science &amp; Environmental Scienc</i>	10.3	17
63	Effects of pH, dissolved oxygen, and aqueous ferrous iron on the adsorption of arsenic to lepidocrocite. <i>Journal of Colloid and Interface Science</i> , <b>2015</b> , 448, 331-8	9.3	73
62	Measurement and Modeling of U(IV) Adsorption to Metal Oxide Minerals. <i>Environmental Science and Technology Letters</i> , <b>2015</b> , 2, 227-232	11	31
61	Long-term in situ oxidation of biogenic uraninite in an alluvial aquifer: impact of dissolved oxygen and calcium. <i>Environmental Science &amp; Enp.; Technology</i> , <b>2015</b> , 49, 7340-7	10.3	21
60	Impact of Water Chemistry on Element Mobilization from Eagle Ford Shale. <i>Environmental Engineering Science</i> , <b>2015</b> , 32, 310-320	2	37
59	Interaction of Fe(II) with phosphate and sulfate on iron oxide surfaces. <i>Geochimica Et Cosmochimica Acta</i> , <b>2015</b> , 158, 130-146	5.5	63
58	Synergistic effect of reductive and ligand-promoted dissolution of goethite. <i>Environmental Science &amp; Environmental Science</i> & Environmental Science & Environmental &	10.3	54
57	Transport of U(VI) through sediments amended with phosphate to induce in situ uranium immobilization. <i>Water Research</i> , <b>2015</b> , 69, 307-317	12.5	33
56	Metal Contaminant Oxidation Mediated by Manganese Redox Cycling in Subsurface Environment. <i>ACS Symposium Series</i> , <b>2015</b> , 29-50	0.4	15
55	Determining pH at elevated pressure and temperature using in situ IIC NMR. <i>Environmental Science &amp; Environmental Science</i>	10.3	5
54	Effect of co-solutes on the products and solubility of uranium(VI) precipitated with phosphate. <i>Chemical Geology</i> , <b>2014</b> , 364, 66-75	4.2	57
53	Speciation and reactivity of uranium products formed during in situ bioremediation in a shallow alluvial aquifer. <i>Environmental Science &amp; Environmental Science &amp; Environment</i>	10.3	42
52	Effects of Mn(II) on UO2 dissolution under anoxic and oxic conditions. <i>Environmental Science &amp; Environmental Science &amp; Technology</i> , <b>2014</b> , 48, 5546-54	10.3	29
51	Forsterite Carbonation in Zones with Transport Limited by Diffusion. <i>Environmental Science and Technology Letters</i> , <b>2014</b> , 1, 333-338	11	18
50	Effect of water chemistry on the dissolution rate of the lead corrosion product hydrocerussite. <i>Water Research</i> , <b>2014</b> , 54, 237-46	12.5	42
49	Oxidative UO2 dissolution induced by soluble Mn(III). <i>Environmental Science &amp; amp; Technology</i> , <b>2014</b> , 48, 289-98	10.3	69

48	Impacts of diffusive transport on carbonate mineral formation from magnesium silicate-CO2-water reactions. <i>Environmental Science &amp; Environmental Scie</i>	10.3	19
47	Impacts of geochemical reactions on geologic carbon sequestration. <i>Environmental Science &amp; Environmental Science &amp; Technology</i> , <b>2013</b> , 47, 3-8	10.3	107
46	Mass action expressions for bidentate adsorption in surface complexation modeling: theory and practice. <i>Environmental Science &amp; Environmental Science</i>	10.3	68
45	Uraninite oxidation and dissolution induced by manganese oxide: A redox reaction between two insoluble minerals. <i>Geochimica Et Cosmochimica Acta</i> , <b>2013</b> , 100, 24-40	5.5	76
44	Kinetics of lead(IV) oxide (PbO2) reductive dissolution: role of lead(II) adsorption and surface speciation. <i>Journal of Colloid and Interface Science</i> , <b>2013</b> , 389, 236-43	9.3	26
43	Forsterite dissolution in saline water at elevated temperature and high CO2 pressure. <i>Environmental Science &amp; Environmental &amp;</i>	10.3	44
42	Adsorption of uranium(VI) to manganese oxides: X-ray absorption spectroscopy and surface complexation modeling. <i>Environmental Science &amp; Environmental Science &amp; Environmental</i>	10.3	160
41	Relative reactivity of biogenic and chemogenic uraninite and biogenic noncrystalline U(IV). <i>Environmental Science &amp; Diagram (Note: Technology, 2013)</i> , 47, 9756-63	10.3	69
40	Effect of connection methods on lead release from galvanic corrosion. <i>Journal - American Water Works Association</i> , <b>2013</b> , 105, E337-E351	0.5	25
39	Impact of galvanic corrosion on lead release from aged lead service lines. Water Research, 2012, 46, 50	49 <u>-6</u> .g	53
38	Effect of diffusive transport limitations on UO2 dissolution. Water Research, 2012, 46, 6023-32	12.5	14
37	Kinetics of the reductive dissolution of lead(IV) oxide by iodide. <i>Environmental Science &amp; Environmental Science &amp; Technology</i> , <b>2012</b> , 46, 5859-66	10.3	14
36	Metal release and speciation changes during wet aging of coal fly ashes. <i>Environmental Science &amp; Environmental &amp; Envi</i>	10.3	37
35	Effect of Ca2+ and Zn2+ on UO2 dissolution rates. <i>Environmental Science &amp; Environmental Science &amp; Env</i>	10.3	29
34	Molecular-scale structure of uranium(VI) immobilized with goethite and phosphate. <i>Environmental Science &amp; Environmental</i> (Science & Environmental	10.3	75
33	Fate of Metals in Fly Ash During Aging in Laboratory-Scale Ash Impoundments. <i>Environmental Engineering Science</i> , <b>2012</b> , 29, 1085-1091	2	4
32	Precipitation of Magnesium Carbonates as a Function of Temperature, Solution Composition, and Presence of a Silicate Mineral Substrate. <i>Environmental Engineering Science</i> , <b>2011</b> , 28, 881-889	2	26
31	Uranium speciation and stability after reductive immobilization in aquifer sediments. <i>Geochimica Et Cosmochimica Acta</i> , <b>2011</b> , 75, 6497-6510	5.5	95

### (2008-2011)

30	Effects of water chemistry on arsenic removal from drinking water by electrocoagulation. <i>Water Research</i> , <b>2011</b> , 45, 384-92	12.5	164
29	Effects of flow and water chemistry on lead release rates from pipe scales. <i>Water Research</i> , <b>2011</b> , 45, 6525-34	12.5	78
28	U(VI) reduction by Fe(II) on hematite nanoparticles. <i>Journal of Nanoparticle Research</i> , <b>2011</b> , 13, 3741-37	<b>5<u>4</u>3</b>	19
27	Speciation of Selenium, Arsenic, and Zinc in Class C Fly Ash. <i>Energy &amp; Description</i> 25, 2980-2987	4.1	50
26	Oxidative Dissolution of Biogenic Uraninite in Groundwater at Old Rifle, CO. <i>Environmental Science</i> & Samp; Technology, <b>2011</b> , 45, 8748-54	10.3	63
25	Speciation-Dependent Kinetics of Uranium(VI) Bioreduction. <i>Geomicrobiology Journal</i> , <b>2011</b> , 28, 396-40	92.5	25
24	Formation of lead(IV) oxides from lead(II) compounds. <i>Environmental Science &amp; Environmental Science &amp;</i>	10.3	45
23	Impact of chlorine disinfectants on dissolution of the lead corrosion product PbO2. <i>Environmental Science &amp; Environmental Sci</i>	10.3	35
22	Impact of phosphate on U(VI) immobilization in the presence of goethite. <i>Geochimica Et Cosmochimica Acta</i> , <b>2010</b> , 74, 6324-6343	5.5	77
21	Effects of pH and carbonate concentration on dissolution rates of the lead corrosion product PbO(2). <i>Environmental Science &amp; Environmental Science &amp; </i>	10.3	40
20	Effect of Mn(II) on the structure and reactivity of biogenic uraninite. <i>Environmental Science &amp; Environmental Science &amp; Technology</i> , <b>2009</b> , 43, 6541-7	10.3	30
19	Comparative dissolution kinetics of biogenic and chemogenic uraninite under oxidizing conditions in the presence of carbonate. <i>Geochimica Et Cosmochimica Acta</i> , <b>2009</b> , 73, 6065-6083	5.5	86
18	Nanoscale size effects on uranium(VI) adsorption to hematite. <i>Environmental Science &amp; Environmental &amp;</i>	10.3	124
17	Indirect UO2 oxidation by Mn(II)-oxidizing spores of Bacillus sp. strain SG-1 and the effect of U and Mn concentrations. <i>Environmental Science &amp; Environmental Science &amp; Envi</i>	10.3	42
16	Individual and competitive adsorption of arsenate and phosphate to a high-surface-area iron oxide-based sorbent. <i>Environmental Science &amp; Environmental Science &amp; Environmenta</i>	10.3	157
15	Report from the third workshop on future directions of solid-state chemistry: The status of solid-state chemistry and its impact in the physical sciences. <i>Progress in Solid State Chemistry</i> , <b>2008</b> , 36, 1-133	8	51
14	Effects of water chemistry and flow rate on arsenate removal by adsorption to an iron oxide-based sorbent. <i>Water Research</i> , <b>2008</b> , 42, 4629-36	12.5	54
13	Microbial reduction of Fe(III) in hematite nanoparticles by Geobacter sulfurreducens. <i>Environmental Science &amp; Environmental &amp;</i>	10.3	55

12	Dissolution of biogenic and synthetic UO2 under varied reducing conditions. <i>Environmental Science &amp; Environmental Science &amp; Environmental Science</i>	10.3	83
11	Evaluation of Nanostructured Sorbents in Differential Bed Reactors for Elemental Mercury Capture. <i>Environmental Engineering Science</i> , <b>2008</b> , 25, 1061-1070	2	13
10	Immobilization of Lead with Nanocrystalline Carbonated Apatite Present in Fish Bone. <i>Environmental Engineering Science</i> , <b>2008</b> , 25, 725-736	2	31
9	Equilibrium solubility and dissolution rate of the lead phosphate chloropyromorphite. <i>Environmental Science &amp; Environmental S</i>	10.3	50
8	Evaluation of a sequential extraction process used for determining mercury binding mechanisms to coal combustion byproducts. <i>Journal of the Air and Waste Management Association</i> , <b>2007</b> , 57, 856-67	2.4	27
7	Evaluation of chemical indicators for tracking and apportionment of phosphorus sources to Table Rock Lake in Southwest Missouri, USA. <i>Water Research</i> , <b>2007</b> , 41, 1525-33	12.5	1
6	Effects of Particle Size and Crystalline Phase on Lead Adsorption to Titanium Dioxide Nanoparticles. <i>Environmental Engineering Science</i> , <b>2007</b> , 24, 85-95	2	133
5	Forsterite dissolution and magnesite precipitation at conditions relevant for deep saline aquifer storage and sequestration of carbon dioxide. <i>Chemical Geology</i> , <b>2005</b> , 217, 257-276	4.2	284
4	Influence of dissolved sodium and cesium on uranyl oxide hydrate solubility. <i>Environmental Science &amp; Environmental Science</i>	10.3	35
3	Equilibrium and kinetic aspects of soddyite dissolution and secondary phase precipitation in aqueous suspension. <i>Geochimica Et Cosmochimica Acta</i> , <b>2002</b> , 66, 3235-3245	5.5	24
2	Time scales for sorption-desorption and surface precipitation of uranyl on goethite. <i>Environmental Science &amp; Environmental Sc</i>	10.3	116
1	Copper Complexation with the Mellitic Acid Series. <i>Journal of Solution Chemistry</i> , <b>1998</b> , 27, 89-105	1.8	12