

Damien Daval

List of Publications by Year in descending order

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Version: 2024-02-01

58
papers

2,372
citations

201674

27
h-index

206112

48
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all docs

58
docs citations

58
times ranked

2227
citing authors

#	ARTICLE	IF	CITATIONS
1	Influence of amorphous silica layer formation on the dissolution rate of olivine at 90°C and elevated pCO ₂ . <i>Chemical Geology</i> , 2011, 284, 193-209.	3.3	251
2	Unifying natural and laboratory chemical weathering with interfacial dissolution-precipitation: A study based on the nanometer-scale chemistry of fluid-silicate interfaces. <i>Chemical Geology</i> , 2012, 294-295, 203-216.	3.3	234
3	Carbonation of Ca-bearing silicates, the case of wollastonite: Experimental investigations and kinetic modeling. <i>Chemical Geology</i> , 2009, 265, 63-78.	3.3	225
4	Mechanism of wollastonite carbonation deduced from micro- to nanometer length scale observations. <i>American Mineralogist</i> , 2009, 94, 1707-1726.	1.9	117
5	Linking nm-scale measurements of the anisotropy of silicate surface reactivity to macroscopic dissolution rate laws: New insights based on diopside. <i>Geochimica Et Cosmochimica Acta</i> , 2013, 107, 121-134.	3.9	90
6	Early entombment within silica minimizes the molecular degradation of microorganisms during advanced diagenesis. <i>Chemical Geology</i> , 2016, 437, 98-108.	3.3	75
7	Enhanced Olivine Carbonation within a Basalt as Compared to Single-Phase Experiments: Reevaluating the Potential of CO ₂ Mineral Sequestration. <i>Environmental Science & Technology</i> , 2014, 48, 5512-5519.	10.0	70
8	Pore-Scale Geochemical Reactivity Associated with CO ₂ Storage: New Frontiers at the Fluid-Solid Interface. <i>Accounts of Chemical Research</i> , 2017, 50, 759-768.	15.6	70
9	The role of Fe and redox conditions in olivine carbonation rates: An experimental study of the rate limiting reactions at 90 and 150 °C in open and closed systems. <i>Geochimica Et Cosmochimica Acta</i> , 2013, 118, 157-183.	3.9	68
10	Lizardite serpentine dissolution kinetics as a function of pH and temperature, including effects of elevated pCO ₂ . <i>Chemical Geology</i> , 2013, 351, 245-256.	3.3	66
11	Geothermal implications for fracture-filling hydrothermal precipitation. <i>Geothermics</i> , 2016, 64, 235-245.	3.4	58
12	pH-dependent control of feldspar dissolution rate by altered surface layers. <i>Chemical Geology</i> , 2016, 442, 148-159.	3.3	53
13	CO ₂ geological storage: The environmental mineralogy perspective. <i>Comptes Rendus - Geoscience</i> , 2011, 343, 246-259.	1.2	52
14	Dissolution kinetics of diopside as a function of solution saturation state: Macroscopic measurements and implications for modeling of geological storage of CO ₂ . <i>Geochimica Et Cosmochimica Acta</i> , 2010, 74, 2615-2633.	3.9	48
15	Does crystallographic anisotropy prevent the conventional treatment of aqueous mineral reactivity? A case study based on K-feldspar dissolution kinetics. <i>Geochimica Et Cosmochimica Acta</i> , 2016, 190, 294-308.	3.9	48
16	Organic molecular heterogeneities can withstand diagenesis. <i>Scientific Reports</i> , 2017, 7, 1508.	3.3	48
17	Burial-induced oxygen-isotope re-equilibration of fossil foraminifera explains ocean paleotemperature paradoxes. <i>Nature Communications</i> , 2017, 8, 1134.	12.8	48
18	The deleterious effect of secondary phases on olivine carbonation yield: Insight from time-resolved aqueous-fluid sampling and FIB-TEM characterization. <i>Chemical Geology</i> , 2013, 357, 186-202.	3.3	47

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19	Growth of Nanosized Calcite through Gas-Solid Carbonation of Nanosized Portlandite under Anisobaric Conditions. <i>Crystal Growth and Design</i> , 2010, 10, 4823-4830.	3.0	43
20	The dependence of albite feldspar dissolution kinetics on fluid saturation state at acid and basic pH: Progress towards a universal relation. <i>Comptes Rendus - Geoscience</i> , 2010, 342, 676-684.	1.2	43
21	Fayalite (Fe ₂ SiO ₄) dissolution kinetics determined by X-ray absorption spectroscopy. <i>Chemical Geology</i> , 2010, 275, 161-175.	3.3	40
22	Influence of etch pit development on the surface area and dissolution kinetics of the orthoclase (001) surface. <i>Chemical Geology</i> , 2016, 447, 79-92.	3.3	34
23	Examination of crystal dissolution in 3D: A way to reconcile dissolution rates in the laboratory?. <i>Geochimica Et Cosmochimica Acta</i> , 2020, 273, 1-25.	3.9	33
24	Carbon dioxide sequestration through silicate degradation and carbon mineralisation: promises and uncertainties. <i>Npj Materials Degradation</i> , 2018, 2, .	5.8	31
25	From mixed flow reactor to column experiments and modeling: Upscaling of calcite dissolution rate. <i>Chemical Geology</i> , 2018, 487, 63-75.	3.3	31
26	Mineralogical evolution of Fe-Si-rich layers at the olivine-water interface during carbonation reactions. <i>American Mineralogist</i> , 2015, 100, 2655-2669.	1.9	30
27	Time-dependent feldspar dissolution rates resulting from surface passivation: Experimental evidence and geochemical implications. <i>Earth and Planetary Science Letters</i> , 2018, 498, 226-236.	4.4	30
28	Integrative analysis of the mineralogical and chemical composition of modern microbialites from ten Mexican lakes: What do we learn about their formation?. <i>Geochimica Et Cosmochimica Acta</i> , 2021, 305, 148-184.	3.9	28
29	Dynamics of altered surface layer formation on dissolving silicates. <i>Geochimica Et Cosmochimica Acta</i> , 2017, 209, 51-69.	3.9	27
30	Geochemical Conditions Allowing the Formation of Modern Lacustrine Microbialites. <i>Procedia Earth and Planetary Science</i> , 2017, 17, 380-383.	0.6	27
31	Comparing the reactivity of glasses with their crystalline equivalents: The case study of plagioclase feldspar. <i>Geochimica Et Cosmochimica Acta</i> , 2019, 254, 122-141.	3.9	27
32	Experimental study and numerical simulation of the dissolution anisotropy of tricalcium silicate. <i>Chemical Geology</i> , 2018, 497, 64-73.	3.3	26
33	Effect of pH on the stability of passivating gel layers formed on International Simple Glass. <i>Journal of Nuclear Materials</i> , 2019, 524, 21-38.	2.7	25
34	In-situ dissolution rates of silicate minerals and associated bacterial communities in the critical zone (Strengbach catchment, France). <i>Geochimica Et Cosmochimica Acta</i> , 2019, 249, 95-120.	3.9	24
35	Early stages of bacterial community adaptation to silicate aging. <i>Geology</i> , 2018, 46, 555-558.	4.4	15
36	Direct measurement of fungal contribution to silicate weathering rates in soil. <i>Geology</i> , 2021, 49, 1055-1058.	4.4	15

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37	Formation of Amorphous Silica Surface Layers by Dissolution-Reprecipitation During Chemical Weathering: Implications for CO ₂ Uptake. <i>Procedia Earth and Planetary Science</i> , 2013, 7, 346-349.	0.6	14
38	Rate-limiting reaction of C ₃ S hydration - A reply to the discussion "A new view on the kinetics of tricalcium silicate hydration" by E. Gartner. <i>Cement and Concrete Research</i> , 2018, 104, 118-122.	11.0	14
39	Control of the mobilization of arsenic and other natural pollutants in groundwater by calcium carbonate concretions in the Pampean Aquifer, southeast of the Buenos Aires province, Argentina. <i>Science of the Total Environment</i> , 2019, 674, 532-543.	8.0	13
40	Physical Properties of Interfacial Layers Developed on Weathered Silicates: A Case Study Based on Labradorite Feldspar. <i>Journal of Physical Chemistry C</i> , 2019, 123, 24520-24532.	3.1	12
41	Linear growth rate of nanosized calcite synthesized via gas-solid carbonation of Ca(OH) ₂ particles in a static bed reactor. <i>Chemical Engineering Journal</i> , 2012, 180, 237-244.	12.7	10
42	Importance of accessory minerals for the control of water chemistry of the Pampean aquifer, province of Buenos Aires, Argentina. <i>Catena</i> , 2018, 160, 112-123.	5.0	10
43	Generalized Sub-Gaussian Processes: Theory and Application to Hydrogeological and Geochemical Data. <i>Water Resources Research</i> , 2020, 56, e2020WR027436.	4.2	10
44	The Dissolution Anisotropy of Pyroxenes: Experimental Validation of a Stochastic Dissolution Model Based on Enstatite Dissolution. <i>Journal of Physical Chemistry C</i> , 2020, 124, 3122-3140.	3.1	10
45	Theoretical Considerations on the Characteristic Timescales of Hydrogen Generation by Serpentinization Reactions on Enceladus. <i>Journal of Geophysical Research E: Planets</i> , 2022, 127, .	3.6	10
46	Barite Growth Rates as a Function of Crystallographic Orientation, Temperature, And Solution Saturation State. <i>Crystal Growth and Design</i> , 2020, 20, 3663-3672.	3.0	9
47	Fast and pervasive diagenetic isotope exchange in foraminifera tests is species-dependent. <i>Nature Communications</i> , 2022, 13, 113.	12.8	9
48	Experimental Study of Dissolution Kinetics of K-feldspar as a Function of Crystal Structure Anisotropy under Hydrothermal Conditions. <i>Procedia Earth and Planetary Science</i> , 2017, 17, 165-168.	0.6	7
49	Dissolution Anisotropy of Pyroxenes: A Surrogate Model for Steady-State Enstatite Dissolution Resulting from Stochastic Simulations of the Hydrolysis Process. <i>Journal of Physical Chemistry C</i> , 2020, 124, 13113-13126.	3.1	7
50	Dissolution Anisotropy of Pyroxenes: Role of Edges and Corners Inferred from Stochastic Simulations of Enstatite Dissolution. <i>Journal of Physical Chemistry C</i> , 2021, 125, 7658-7674.	3.1	7
51	A comparative study of the dissolution mechanisms of amorphous and crystalline feldspars at acidic pH conditions. <i>Npj Materials Degradation</i> , 2022, 6, .	5.8	7
52	GaMin TM 11 " an International Inter-laboratory Comparison for Geochemical CO ₂ - Saline Fluid - Mineral Interaction Experiments. <i>Energy Procedia</i> , 2014, 63, 5538-5543.	1.8	6
53	Estimating the activation energy of bond hydrolysis by time-resolved weighing of dissolving crystals. <i>Npj Materials Degradation</i> , 2021, 5, .	5.8	6
54	Symbiotic cooperation between freshwater rock-boring bivalves and microorganisms promotes silicate bioerosion. <i>Scientific Reports</i> , 2020, 10, 13385.	3.3	5

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55	Statistical Description of Calcite Surface Roughness Resulting from Dissolution at Close-to-Equilibrium Conditions. ACS Earth and Space Chemistry, 2021, 5, 3115-3129.	2.7	4
56	Experimental determination of the reactivity of the Frio Sandstone, Texas, and the fate of heavy metals resulting from carbon dioxide sequestration. Environmental Earth Sciences, 2015, 74, 5501-5516.	2.7	3
57	Reply to 'No substantial long-term bias in the Cenozoic benthic foraminifera oxygen-isotope record'. Nature Communications, 2018, 9, 2874.	12.8	1
58	Species-specific foraminiferal ultrastructures modulate surfaces available for diagenesis. Microscopy and Microanalysis, 2021, 27, 274-275.	0.4	1