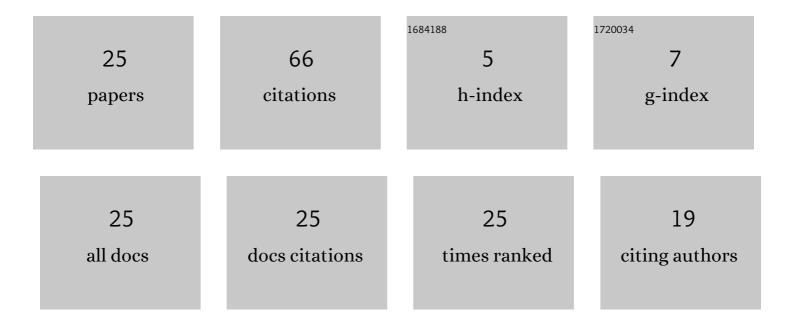
## Rafail A Yusupov

List of Publications by Year in descending order

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Version: 2024-02-01



#	Article	IF	CITATIONS
1	PH shifts in the Pb(II)-H2O-OHâ^' system. AIP Conference Proceedings, 2021, , .	0.4	Ο
2	Modeling a Heterogeneous Cu(II)–H2O–OH––NH3 System with Data from Measuring pH. Russian Journal of Physical Chemistry A, 2020, 94, 2459-2466.	0.6	1
3	Study of Equilibria in the "CuCl2–H2O–NaOH―System by Potentiometric Titration. Russian Journal of Physical Chemistry A, 2019, 93, 970-975.	0.6	4
4	Model and Algorithm for Calculating the Formation of Phases in Aqueous Solutions of Metal Salts. Russian Journal of Physical Chemistry A, 2019, 93, 845-850.	0.6	2
5	Evaluation of analytical capabilities of total reflection X-ray fluorescence spectrometry for the analysis of drinks with sucrose matrix. Analitika I Kontrol, 2019, 23, 483-493.	0.2	2
6	Calculating Equilibrium Constants in the SnCl2–H2O–NaOH System According to Potentiometric Titration Data. Russian Journal of Physical Chemistry A, 2018, 92, 1025-1031.	0.6	2
7	Equilibria in a ZnCl2–H2O–NaOH System, According to Data from Potentiometric Titration, and Selecting Conditions for the Hydrochemical Synthesis of ZnS and ZnSe Films. Russian Journal of Physical Chemistry A, 2018, 92, 2575-2582.	0.6	3
8	Effect of the morphology of cadmium sulfide films on the process of ion-exchange substitution at the interface with a lead salt solution. Russian Journal of Physical Chemistry A, 2017, 91, 1539-1547.	0.6	1
9	Simulating equilibrium processes in the Ga(NO3)3–H2O–NaOH system. Russian Journal of Physical Chemistry A, 2016, 90, 1274-1279.	0.6	4
10	Calculating the equilibrium constants of a Sn(II)-H2O-OHâ^' system with allowance for precipitation. Russian Journal of Physical Chemistry A, 2014, 88, 927-931.	0.6	6
11	Diagrams of the formation of In2S3 and In2Se3 films on vitroceramic upon precipitation, according to potentiometric titration. Russian Journal of Physical Chemistry A, 2013, 87, 1771-1777.	0.6	10
12	Calculation of sediment existence regions in metal ion-H2O-complex forming agent systems taking intermediate solubilities into account. Russian Journal of Physical Chemistry A, 2010, 84, 1263-1265.	0.6	6
13	Heteronuclear compounds formed in the systems based on Fe(II), Fe(III), Al(III), SO 4 2â^' , Clâ^-H2O-OHâ^', and NH3. Russian Journal of Applied Chemistry, 2010, 83, 593-597.	0.5	1
14	X-ray irradiation-assisted low-temperature synthesis of yellow lead(II) oxide (massicot) in aqueous solution. High Energy Chemistry, 2010, 44, 178-180.	0.9	0
15	Calculation of the regions of solid phase precipitations in the metal ion-water-complexing agent systems. Russian Journal of Physical Chemistry A, 2009, 83, 2188-2190.	0.6	9
16	The interaction of disodium salt of 4,6-dinitro-1-oxobenzo-[6,5-c]-2,1,3-oxadiazolediol-5,7 with NaOH and HCl. Russian Journal of Coordination Chemistry/Koordinatsionnaya Khimiya, 2008, 34, 864-868.	1.0	0
17	Coagulation in lipid- and protein-containing systems. Russian Journal of Applied Chemistry, 2006, 79, 1935-1939.	0.5	1
18	Agar-immobilized Metal Sulfides and Their Sorption Power with Respect to Ag(I) Ions. Russian Journal of General Chemistry, 2004, 74, 791-792.	0.8	0

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#	Article	IF	CITATIONS
19	Reactions of PtCl4and Na2PtCl6with 18-Crown-6 and Dibenzo-18-crown-6 in Various Solvents. Russian Journal of General Chemistry, 2004, 74, 1126-1127.	0.8	0
20	Reaction of H2PtCl6 with 18-crown-6 and dibenzo-18-crown-6 in 1,2-dichloroethane and acetonitrile. Russian Journal of General Chemistry, 2004, 74, 1478-1484.	0.8	0
21	Electrophilic Substitution Ni(II)→M(II) in Ni2[Fe(CN)6] Gelatin-immobilized Matrix Materials. Russian Journal of General Chemistry, 2003, 73, 847-854.	0.8	5
22	Ion Exchange of Silver(I) on Thin PbS Films as Influenced by Diffusion. Russian Journal of General Chemistry, 2003, 73, 999-1004.	0.8	1
23	Lead(II) Sulfide. Synthesis in Lead(II) Tetraoxophosphate(V) Gelatin-immobilized Matrix Implantates and Sorption Activity Toward Silver(I) Ions. Russian Journal of General Chemistry, 2003, 73, 1183-1187.	0.8	3
24	Title is missing!. Russian Journal of General Chemistry, 2001, 71, 217-221.	0.8	0
25	Overcoming absorption effects in the determination of light elements in beverages by totalâ€reflection Xâ€ray spectrometry. X-Ray Spectrometry, 0, , .	1.4	5