

Richard G Finke

List of Publications by Year in descending order

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times ranked

10408
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#	ARTICLE	IF	CITATIONS
1	A review of the problem of distinguishing true homogeneous catalysis from soluble or other metal-particle heterogeneous catalysis under reducing conditions. <i>Journal of Molecular Catalysis A</i> , 2003, 198, 317-341.	4.8	1,134
2	A review of modern transition-metal nanoclusters: their synthesis, characterization, and applications in catalysis. <i>Journal of Molecular Catalysis A</i> , 1999, 145, 1-44.	4.8	852
3	Transition Metal Nanocluster Formation Kinetic and Mechanistic Studies. A New Mechanism When Hydrogen Is the Reductant: A Slow, Continuous Nucleation and Fast Autocatalytic Surface Growth. <i>Journal of the American Chemical Society</i> , 1997, 119, 10382-10400.	6.6	834
4	Protein aggregation kinetics, mechanism, and curve-fitting: A review of the literature. <i>Biochimica Et Biophysica Acta - Proteins and Proteomics</i> , 2009, 1794, 375-397.	1.1	572
5	Transition-metal nanocluster stabilization for catalysis: A critical review of ranking methods and putative stabilizers. <i>Coordination Chemistry Reviews</i> , 2007, 251, 1075-1100.	9.5	418
6	Electrocatalytic Water Oxidation Beginning with the Cobalt Polyoxometalate [Co ₄ (H ₂ O) ₂ (PW ₉ O ₃₄) ₂] ¹⁰⁻ : Identification of Heterogeneous CoO _x as the Dominant Catalyst. <i>Journal of the American Chemical Society</i> , 2011, 133, 14872-14875.	6.6	394
7	Trivalent heteropolytungstate derivatives. 3. Rational syntheses, characterization, two-dimensional tungsten-183 NMR, and properties of tungstometallophosphates P ₂ W ₁₈ M ₄ (H ₂ O) ₂ O ₆₈ 10- and P ₄ W ₃₀ M ₄ (H ₂ O) ₂ O ₁₁₂ 16- (M = cobalt, copper, zinc). <i>Inorganic Chemistry</i> , 1987, 26, 3886-3896.	1.9	393
8	Nanocluster nucleation and growth kinetic and mechanistic studies: A review emphasizing transition-metal nanoclusters. <i>Journal of Colloid and Interface Science</i> , 2008, 317, 351-374.	5.0	329
9	Highly oxidation resistant inorganic-porphyrin analog polyoxometalate oxidation catalysts. 1. The synthesis and characterization of aqueous-soluble potassium salts of .alpha.2-P ₂ W ₁₇ O ₆₁ (Mn+.cntdot.OH ₂)(n-10) and organic solvent soluble tetra-n-butylammonium salts of .alpha.2-P ₂ W ₁₇ O ₆₁ (Mn+.cntdot.Br)(n-11) (M = Mn ³⁺ , Fe ³⁺ , Co ²⁺ , Ni ²⁺ , Cu ²⁺). <i>Journal of the American Chemical Society</i> , 1991, 113, 7309-7321.	6.6	325
10	A More General Approach to Distinguishing "Homogeneous" from "Heterogeneous" Catalysis: Discovery of Polyoxoanion- and Bu ₄ N ⁺ -Stabilized, Isolable and Redissolvable, High-Reactivity Ir.apprx.190-450 Nanocluster Catalysts. <i>Inorganic Chemistry</i> , 1994, 33, 4891-4910.	1.9	299
11	Nanocluster Formation and Stabilization Fundamental Studies: A Ranking Commonly Employed Anionic Stabilizers via the Development, Then Application, of Five Comparative Criteria. <i>Journal of the American Chemical Society</i> , 2002, 124, 5796-5810.	6.6	283
12	Fitting Neurological Protein Aggregation Kinetic Data via a 2-Step, Minimal/Ockham's Razor Model: The Finke-Watzky Mechanism of Nucleation Followed by Autocatalytic Surface Growth. <i>Biochemistry</i> , 2008, 47, 2413-2427.	1.2	265
13	Thermolysis of the cobalt-carbon bond of adenosylcobalamin. 2. Products, kinetics, and cobalt-carbon bond dissociation energy in aqueous solution. <i>Journal of the American Chemical Society</i> , 1986, 108, 4820-4829.	6.6	242
14	Is It Homogeneous or Heterogeneous Catalysis? Identification of Bulk Ruthenium Metal as the True Catalyst in Benzene Hydrogenations Starting with the Monometallic Precursor, Ru(II)(1-6-C ₆ Me ₆)(OAc) ₂ , Plus Kinetic Characterization of the Heterogeneous Nucleation, Then Autocatalytic Surface-Growth Mechanism of Metal Film Formation. <i>Journal of the American Chemical Society</i> , 2003, 125, 10301-10310.	6.6	236
15	Novel Polyoxoanion- and Bu ₄ N ⁺ -Stabilized, Isolable, and Redissolvable, 20-30-ANG. Ir ³⁰⁰⁻⁹⁰⁰ Nanoclusters: The Kinetically Controlled Synthesis, Characterization, and Mechanism of Formation of Organic Solvent-Soluble, Reproducible Size, and Reproducible Catalytic Activity Metal Nanoclusters. <i>Journal of the American Chemical Society</i> , 1994, 116, 8335-8353.	6.6	233
16	Nanocluster Size-Control and "Magic Number" Investigations. Experimental Tests of the "Living-Metal Polymer" Concept and of Mechanism-Based Size-Control Predictions Leading to the Syntheses of Iridium(0) Nanoclusters Centering about Four Sequential Magic Numbers. <i>Chemistry of Materials</i> , 1997, 9, 3083-3095.	3.2	210
17	A Mechanism for Transition-Metal Nanoparticle Self-Assembly. <i>Journal of the American Chemical Society</i> , 2005, 127, 8179-8184.	6.6	202
18	Distinguishing Homogeneous from Heterogeneous Water Oxidation Catalysis when Beginning with Polyoxometalates. <i>ACS Catalysis</i> , 2014, 4, 909-933.	5.5	195

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19	Rh(0) Nanoclusters in Benzene Hydrogenation Catalysis: Kinetic and Mechanistic Evidence that a Putative [(C ₈ H ₁₇) ₃ NCH ₃] ⁺ [RhCl ₄] ⁻ Ion-Pair Catalyst Is Actually a Distribution of Cl ⁻ and [(C ₈ H ₁₇) ₃ NCH ₃] ⁺ Stabilized Rh(0) Nanoclusters. <i>Journal of the American Chemical Society</i> , 1998, 120, 5653-5666.	6.6	188
20	A perspective on nanocluster catalysis: polyoxoanion and (n-C ₄ H ₉) ₄ N ⁺ stabilized Ir(0) ^{1/3} 300 nanocluster soluble heterogeneous catalysts™. <i>Journal of Molecular Catalysis A</i> , 1996, 114, 29-51.	4.8	157
21	Nanocluster Nucleation, Growth, and Then Agglomeration Kinetic and Mechanistic Studies: A More General, Four-Step Mechanism Involving Double Autocatalysis. <i>Chemistry of Materials</i> , 2005, 17, 4925-4938.	3.2	154
22	Transition-Metal Nanocluster Size vs Formation Time and the Catalytically Effective Nucleus Number: A Mechanism-Based Treatment. <i>Journal of the American Chemical Society</i> , 2008, 130, 11959-11969.	6.6	153
23	Is It Homogeneous or Heterogeneous Catalysis Derived from [RhCp*Cl ₂] ₂ ? In Operando XAFS, Kinetic, and Crucial Kinetic Poisoning Evidence for Subnanometer Rh ₄ Cluster-Based Benzene Hydrogenation Catalysis. <i>Journal of the American Chemical Society</i> , 2011, 133, 18889-18902.	6.6	147
24	A review of the kinetics and mechanisms of formation of supported-nanoparticle heterogeneous catalysts. <i>Journal of Molecular Catalysis A</i> , 2012, 355, 1-38.	4.8	144
25	Additional Investigations of a New Kinetic Method To Follow Transition-Metal Nanocluster Formation, Including the Discovery of Heterolytic Hydrogen Activation in Nanocluster Nucleation Reactions. <i>Chemistry of Materials</i> , 2001, 13, 312-324.	3.2	138
26	Trivacant heteropolytungstate derivatives: the rational synthesis, characterization, and tungsten-183 NMR spectra of P ₂ W ₁₈ M ₄ (H ₂ O) ₂ O ₆₈ 10 ⁻ (M = cobalt, copper, zinc). <i>Journal of the American Chemical Society</i> , 1981, 103, 1587-1589.	6.6	132
27	Nanocluster Formation Synthetic, Kinetic, and Mechanistic Studies. The Detection of, and Then Methods To Avoid, Hydrogen Mass-Transfer Limitations in the Synthesis of Polyoxoanion- and Tetrabutylammonium-Stabilized, Near-Monodisperse 40 Å ± 6 Å... Rh(0) Nanoclusters. <i>Journal of the American Chemical Society</i> , 1998, 120, 9545-9554.	6.6	127
28	Water Oxidation Catalysis Beginning with 2.5 × 10 ⁻⁴ M [Co ₄ (H ₂ O) ₂ (PW ₉ O ₃₄) ₂] ¹⁰⁻ : Investigation of the True Electrochemically Driven Catalyst at a ¥600 mV Overpotential at a Glassy Carbon Electrode. <i>ACS Catalysis</i> , 2013, 3, 1209-1219.	5.5	124
29	An All-Inorganic, Polyoxometalate-Based Catechol Dioxygenase That Exhibits >100,000 Catalytic Turnovers. <i>Journal of the American Chemical Society</i> , 1999, 121, 9831-9842.	6.6	123
30	Is It Homogeneous or Heterogeneous Catalysis? Compelling Evidence for Both Types of Catalysts Derived from [Rh(<i>i</i> -5-C ₅ Me ₅)Cl ₂] ₂ as a Function of Temperature and Hydrogen Pressure. <i>Journal of the American Chemical Society</i> , 2005, 127, 4423-4432.	6.6	123
31	Is There a Minimal Chemical Mechanism Underlying Classical Avrami-Erofeev Treatments of Phase-Transformation Kinetic Data?. <i>Chemistry of Materials</i> , 2009, 21, 4692-4705.	3.2	122
32	Supramolecular Triruthenium Cluster-Based Benzene Hydrogenation Catalysis: A Fact or Fiction?. <i>Organometallics</i> , 2005, 24, 1819-1831.	1.1	117
33	LaMer's 1950 Model for Particle Formation of Instantaneous Nucleation and Diffusion-Controlled Growth: A Historical Look at the Model's Origins, Assumptions, Equations, and Underlying Sulfur Sol Formation Kinetics Data. <i>Chemistry of Materials</i> , 2019, 31, 7116-7132.	3.2	111
34	Trisubstituted heteropolytungstates as soluble metal oxide analogues. 4. The synthesis and characterization of organic solvent-soluble (Bu ₄ N) ₁₂ H ₄ P ₄ W ₃₀ Nb ₆ O ₁₂₃ and (Bu ₄ N) ₉ P ₂ W ₁₅ Nb ₃ O ₆₂ and solution spectroscopic and other evidence for the supported organometallic derivatives (Bu ₄ N) ₇ [(C ₅ Me ₅)Rh.cntdot.P ₂ W ₁₅ Nb ₃ O ₆₂] and (Bu ₄ N) ₇ [(C ₆ H ₆)Ru.cntdot.P ₂ W ₁₅ Nb ₃ O ₆₂]. <i>Organometallics</i> , 1988, 7, 1692-1704.	1.1	110
35	Visible-Light-Assisted Photoelectrochemical Water Oxidation by Thin Films of a Phosphonate-Functionalized Perylene Diimide Plus CoO Cocatalyst. <i>ACS Applied Materials & Interfaces</i> , 2014, 6, 13367-13377.	4.0	108
36	Polyoxoanion-Supported Catalyst Precursors. Synthesis and Characterization of the Iridium(I) and Rhodium(I) Precatalysts [(n-C ₄ H ₉) ₄ N] ₅ Na ₃ [(1,5-COD)M.cntdot.P ₂ W ₁₅ Nb ₃ O ₆₂] (M = Ir, Rh). <i>Inorganic Chemistry</i> , 1995, 34, 1413-1429.	1.9	107

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37	Sigmoidal Nucleation and Growth Curves Across Nature Fit by the Finke-Watzky Model of Slow Continuous Nucleation and Autocatalytic Growth: Explicit Formulas for the Lag and Growth Times Plus Other Key Insights. <i>Journal of Physical Chemistry C</i> , 2017, 121, 5302-5312.	1.5	105
38	Nanoclusters in Catalysis: A Comparison of CS ₂ Catalyst Poisoning of Polyoxoanion- and Tetrabutylammonium-Stabilized 40 Å ± 6 Å... Rh(0) Nanoclusters to 5 Rh/Al ₂ O ₃ , Including an Analysis of the Literature Related to the CS ₂ to Metal Stoichiometry Issue. <i>Inorganic Chemistry</i> , 2002, 41, 1625-1638.	1.9	104
39	Synthetic and mechanistic studies of the reduction of .alpha.,.beta.-unsaturated carbonyl compounds by the binuclear cluster, sodium hydrogen octacarbonyldiferrate. <i>Journal of the American Chemical Society</i> , 1978, 100, 1119-1140.	6.6	103
40	Effects of paramagnetic and diamagnetic transition-metal monosubstitutions on tungsten-183 and phosphorus-31 NMR spectra for Keggin and Wells-Dawson heteropolytungstate derivatives. Correlations and corrections. Tungsten-183 NMR two-dimensional INADEQUATE studies of .alpha.-[(D ₂ O)ZnO ₄ X _n +W ₁₁ O ₃₄](10-n)- wherein X _n + = Si ⁴⁺ and P ⁵⁺ . <i>Journal of the American Chemical Society</i> , 1987, 109, 7402-7408.	6.6	103
41	Polyoxometalate Catalyst Precursors. Improved Synthesis, H ⁺ -Titration Procedure, and Evidence for ³¹ P NMR as a Highly Sensitive Support-Site Indicator for the Prototype Polyoxoanion-Organometallic-Support System [(n-C ₄ H ₉) ₄ N] ₉ P ₂ W ₁₅ Nb ₃ O ₆₂ . <i>Inorganic Chemistry</i> , 1996, 35, 7905-7913.	1.9	88
42	Nanocluster Formation and Stabilization Fundamental Studies: Investigating "Solvent-Only" Stabilization En Route to Discovering Stabilization by the Traditionally Weakly Coordinating Anion BF ₄ ⁻ Plus High Dielectric Constant Solvents. <i>Inorganic Chemistry</i> , 2006, 45, 8382-8393.	1.9	88
43	Molecular insights for how preferred oxoanions bind to and stabilize transition-metal nanoclusters: a tridentate, C ₃ symmetry, lattice size-matching binding model. <i>Coordination Chemistry Reviews</i> , 2004, 248, 135-146.	9.5	87
44	Water-oxidation photoanodes using organic light-harvesting materials: a review. <i>Journal of Materials Chemistry A</i> , 2017, 5, 19560-19592.	5.2	87
45	The Four-Step, Double-Autocatalytic Mechanism for Transition-Metal Nanocluster Nucleation, Growth, and Then Agglomeration: Metal, Ligand, Concentration, Temperature, and Solvent Dependency Studies. <i>Chemistry of Materials</i> , 2008, 20, 1956-1970.	3.2	85
46	Transition-Metal Nanocluster Kinetic and Mechanistic Studies Emphasizing Nanocluster Agglomeration: Demonstration of a Kinetic Method That Allows Monitoring of All Three Phases of Nanocluster Formation and Aging. <i>Chemistry of Materials</i> , 2004, 16, 139-150.	3.2	83
47	Polyoxoanion- and Tetrabutylammonium-Stabilized, Near-Monodisperse, 40 Å ± 6 Å... Rh(0) ^{1/4} 1500 to Rh(0) ^{1/4} 3700 Nanoclusters: Synthesis, Characterization, and Hydrogenation Catalysis. <i>Chemistry of Materials</i> , 1999, 11, 1035-1047.	3.2	82
48	Iridium(0) Nanocluster, Acid-Assisted Catalysis of Neat Acetone Hydrogenation at Room Temperature: A Exceptional Activity, Catalyst Lifetime, and Selectivity at Complete Conversion. <i>Journal of the American Chemical Society</i> , 2005, 127, 4800-4808.	6.6	79
49	Fitting Yeast and Mammalian Prion Aggregation Kinetic Data with the Finke-Watzky Two-Step Model of Nucleation and Autocatalytic Growth. <i>Biochemistry</i> , 2008, 47, 10790-10800.	1.2	79
50	Water Oxidation Catalysis Beginning with Co ₄ (H ₂ O) ₂ (PW ₉ O ₃₄) ₂ × 10H ₂ O When Driven by the Chemical Oxidant Ruthenium(III)tris(2,2'-bipyridine): Stoichiometry, Kinetic, and Mechanistic Studies en Route to Identifying the True Catalyst. <i>ACS Catalysis</i> , 2014, 4, 79-89.	5.5	74
51	Transition-Metal Nanocluster Stabilization Fundamental Studies: Hydrogen Phosphate as a Simple, Effective, Readily Available, Robust, and Previously Unappreciated Stabilizer for Well-Formed, Isolable, and Redissolvable Ir(0) and Other Transition-Metal Nanoclusters. <i>Langmuir</i> , 2003, 19, 6247-6260.	1.6	72
52	I [±] -Synuclein aggregation variable temperature and variable pH kinetic data: A re-analysis using the Finke-Watzky 2-step model of nucleation and autocatalytic growth. <i>Biophysical Chemistry</i> , 2009, 140, 9-15.	1.5	70
53	Agglomerative Sintering of an Atomically Dispersed Ir ₁ /Zeolite Y Catalyst: Compelling Evidence Against Ostwald Ripening but for Bimolecular and Autocatalytic Agglomeration Catalyst Sintering Steps. <i>ACS Catalysis</i> , 2015, 5, 3514-3527.	5.5	66
54	Electrochemically Driven Water-Oxidation Catalysis Beginning with Six Exemplary Cobalt Polyoxometalates: Is It Molecular, Homogeneous Catalysis or Electrode-Bound, Heterogeneous Co ₀ Catalysis?. <i>Journal of the American Chemical Society</i> , 2018, 140, 12040-12055.	6.6	63

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55	Nucleation is Second Order: An Apparent Kinetically Effective Nucleus of Two for Ir(0) _n Nanoparticle Formation from [(1,5-COD)Ir ₃ O ₆₂] ⁶⁻ Plus Hydrogen. Journal of the American Chemical Society, 2014, 136, 17601-17615.	6.6	62
56	In Situ Formed Weakly Ligated/Labile Ligand Iridium(0) Nanoparticles and Aggregates as Catalysts for the Complete Hydrogenation of Neat Benzene at Room Temperature and Mild Pressures. Langmuir, 2010, 26, 12455-12464.	1.6	61
57	Nanocluster Formation and Stabilization Fundamental Studies. 2. Proton Sponge as an Effective H ⁺ -Scavenger and Expansion of the Anion Stabilization Ability Series. Langmuir, 2002, 18, 7653-7662.	1.6	58
58	Monitoring Supported-Nanocluster Heterogeneous Catalyst Formation: Product and Kinetic Evidence for a 2-Step, Nucleation and Autocatalytic Growth Mechanism of Pt(0) _n Formation from H ₂ /PtCl ₆ on Al ₂ O ₃ or TiO ₂ . Journal of the American Chemical Society, 2009, 131, 6389-6396.	6.6	58
59	LaMer's 1950 model of particle formation: a review and critical analysis of its classical nucleation and fluctuation theory basis, of competing models and mechanisms for phase-changes and particle formation, and then of its application to silver halide, semiconductor, metal, and metal-oxide nanoparticles. Materials Advances, 2021, 2, 186-235.	2.6	58
60	Fitting and Interpreting Transition-Metal Nanocluster Formation and Other Sigmoidal-Appearing Kinetic Data: A More Thorough Testing of Dispersive Kinetic vs Chemical-Mechanism-Based Equations and Treatments for 4-Step Type Kinetic Data. Chemistry of Materials, 2009, 21, 4468-4479.	3.2	56
61	Development Plus Kinetic and Mechanistic Studies of a Prototype Supported-Nanoparticle Heterogeneous Catalyst Formation System in Contact with Solution: Ir(1,5-COD)Cl/γ-Al ₂ O ₃ and Its Reduction by H ₂ to Ir(0)/γ-Al ₂ O ₃ . Journal of the American Chemical Society, 2010, 132, 9701-9714.	6.6	54
62	Electrochemical Water Oxidation Catalysis Beginning with Co(II) Polyoxometalates: The Case of the Precatalyst Co ₄ V ₂ W ₁₈ O ₆₈ . ACS Catalysis, 2017, 7, 7-16.	5.5	54
63	Quantitative 1,10-Phenanthroline Catalyst-Poisoning Kinetic Studies of Rh(0) Nanoparticle and Rh ₄ Cluster Benzene Hydrogenation Catalysts: Estimates of the Poison <i>K</i> association Binding Constants, of the Equivalents of Poison Bound and of the Number of Catalytically Active Sites for Each Catalyst. ACS Catalysis, 2012, 2, 1967-1975.	5.5	53
64	A Four-Step Mechanism for the Formation of Supported-Nanoparticle Heterogeneous Catalysts in Contact with Solution: The Conversion of Ir(1,5-COD)Cl/γ-Al ₂ O ₃ to Ir(0) _n /γ-Al ₂ O ₃ . Journal of the American Chemical Society, 2014, 136, 1930-1941.	6.6	48
65	Mechanism-Enabled Population Balance Modeling of Particle Formation en Route to Particle Average Size and Size Distribution Understanding and Control. Journal of the American Chemical Society, 2019, 141, 15827-15839.	6.6	48
66	Oxygenation Catalysis by All-Inorganic, Oxidation-Resistant, Dawson-Type Polyoxoanion-Supported Transition Metal Precatalysts, [(CH ₃ CN) _x M] ⁿ⁺ Plus P ₂ W ₁₅ Nb ₃ O ₆₂ - (M = MnII, FeII, CoII, NiII, CuI, CuII,) Tj ETQq0100 rgBT / Overlock 1	10.0	10
67	Mononuclear Zeolite-Supported Iridium: Kinetic, Spectroscopic, Electron Microscopic, and Size-Selective Poisoning Evidence for an Atomically Dispersed True Catalyst at 22 Å°C. ACS Catalysis, 2012, 2, 1947-1957.	5.5	47
68	Nanoparticle Nucleation Is Termolecular in Metal and Involves Hydrogen: Evidence for a Kinetically Effective Nucleus of Three {Ir ₃ H ₂ Å-P ₂ W ₁₅ Nb ₃ O ₆₂ } ⁶⁻ in Ir(0) _n Nanoparticle Formation From [(1,5-COD)IrÅ-P ₂ W ₁₅ Nb ₃ O ₆₂] ⁸⁻ Plus Dihydrogen. Journal of the American Chemical Society, 2017, 139, 5444-5457.	6.6	46
69	The solid-state rearrangement of the Wells-Dawson K ₆ P ₂ W ₁₈ O ₆₂ ·1/2 10H ₂ O to a stable Keggin-type heteropolyanion phase: a catalyst for the selective oxidation of isobutane to isobutene. Catalysis Letters, 1996, 36, 75-79.	1.4	43
70	Kinetic and Mechanistic Studies of Vanadium-Based, Extended Catalytic Lifetime Catechol Dioxygenases. Journal of the American Chemical Society, 2005, 127, 13988-13996.	6.6	42
71	Gold Nanocluster Agglomeration Kinetic Studies: Evidence for Parallel Bimolecular Plus Autocatalytic Agglomeration Pathways as a Mechanism-Based Alternative to an Avrami-Based Analysis. Chemistry of Materials, 2012, 24, 1718-1725.	3.2	42
72	Sensitization of Nanocrystalline Metal Oxides with a Phosphonate-Functionalized Perylene Diimide for Photoelectrochemical Water Oxidation with a CoO _x Catalyst. ACS Applied Materials & Interfaces, 2017, 9, 27625-27637.	4.0	40

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73	A Test of the Transition-Metal Nanocluster Formation and Stabilization Ability of the Most Common Polymeric Stabilizer, Poly(vinylpyrrolidone), as Well as Four Other Polymeric Protectants. <i>Langmuir</i> , 2006, 22, 9357-9367.	1.6	39
74	Platinum-Catalyzed Phenyl and Methyl Group Transfer from Tin to Iridium: Evidence for an Autocatalytic Reaction Pathway with an Unusual Preference for Methyl Transfer. <i>Journal of the American Chemical Society</i> , 2008, 130, 1839-1841.	6.6	36
75	Transition-Metal Nanocluster Catalysts: Scaled-up Synthesis, Characterization, Storage Conditions, Stability, and Catalytic Activity before and after Storage of Polyoxoanion- and Tetrabutylammonium-Stabilized Ir(0) Nanoclusters. <i>Chemistry of Materials</i> , 2003, 15, 899-909.	3.2	33
76	Supported-Nanoparticle Heterogeneous Catalyst Formation in Contact with Solution: Kinetics and Proposed Mechanism for the Conversion of Ir(1,5-COD)Cl/Al ₂ O ₃ to Ir(0)/Al ₂ O ₃ . <i>Journal of the American Chemical Society</i> , 2011, 133, 7744-7756.	6.6	32
77	Palladium(0) Nanoparticle Formation, Stabilization, and Mechanistic Studies: Pd(acac) ₂ as a Preferred Precursor, [Bu ₄ N] ₂ HPO ₄ Stabilizer, plus the Stoichiometry, Kinetics, and Minimal, Four-Step Mechanism of the Palladium Nanoparticle Formation and Subsequent Agglomeration Reactions. <i>Langmuir</i> , 2016, 32, 3699-3716.	1.6	32
78	Transition-Metal Nanocluster Stabilization versus Agglomeration Fundamental Studies: Measurement of the Two Types of Rate Constants for Agglomeration Plus Their Activation Parameters under Catalytic Conditions. <i>Chemistry of Materials</i> , 2008, 20, 2592-2601.	3.2	31
79	Particle Size Distributions via Mechanism-Enabled Population Balance Modeling. <i>Journal of Physical Chemistry C</i> , 2020, 124, 4852-4880.	1.5	30
80	Silver Nanoparticles Synthesized by Microwave Heating: A Kinetic and Mechanistic Re-Analysis and Re-Interpretation. <i>Journal of Physical Chemistry C</i> , 2017, 121, 27643-27654.	1.5	29
81	Determination of the Dominant Catalyst Derived from the Classic [RhCp*Cl] ₂ Precatalyst System: Is it Single-Metal Rh ₁ Cp*-Based, Subnanometer Rh ₄ Cluster-Based, or Rh(O) _n Nanoparticle-Based Cyclohexene Hydrogenation Catalysis at Room Temperature and Mild Pressures?. <i>ACS Catalysis</i> , 2015, 5, 2876-2886.	5.5	28
82	Unintuitive Inverse Dependence of the Apparent Turnover Frequency on Precatalyst Concentration: A Quantitative Explanation in the Case of Ziegler-Type Nanoparticle Catalysts Made from [(1,5-COD)Ir(1/4-O) ₂ C ₈ H ₁₅] ₂ and AlEt ₃ . <i>ACS Catalysis</i> , 2015, 5, 3342-3353.	5.5	27
83	Autoxidation-Product-Initiated Dioxygenases: Vanadium-Based, Record Catalytic Lifetime Catechol Dioxygenase Catalysis. <i>Inorganic Chemistry</i> , 2005, 44, 8521-8530.	1.9	26
84	Dust Effects on Nucleation Kinetics and Nanoparticle Product Size Distributions: Illustrative Case Study of a Prototype Ir(0) _n Transition-Metal Nanoparticle Formation System. <i>Langmuir</i> , 2017, 33, 6550-6562.	1.6	24
85	Gold Nanoparticle Formation Kinetics and Mechanism: A Critical Analysis of the Redox Crystallization Mechanism. <i>ACS Omega</i> , 2018, 3, 1555-1563.	1.6	23
86	Stereospecific Polymerization of Chiral Oxazolidinone-Functionalized Alkenes. <i>Macromolecules</i> , 2010, 43, 7504-7514.	2.2	22
87	The Second Isolable B12-Thiolate Complex, (Pentafluorophenylthiolato)cobalamin: Synthesis and Characterization. <i>Inorganic Chemistry</i> , 1998, 37, 5109-5116.	1.9	21
88	Metal Complexes of the Lacunary Heteropolytungstates [B ₁₂ -PW ₉ O ₃₄] ⁹⁻ and [B ₁₂ -P ₂ W ₁₅ O ₅₆] ¹²⁻ . <i>Inorganic Syntheses</i> , 2007, , 167-185.	0.3	20
89	Cobalt Polyoxometalate Co ₄ V ₂ W ₁₈ O ₆₈ ¹⁰⁺ : A Critical Investigation of Its Synthesis, Purity, and Observed ⁵¹ V Quadrupolar NMR. <i>Inorganic Chemistry</i> , 2016, 55, 5343-5355.	1.9	19
90	Nucleation Kinetics and Molecular Mechanism in Transition-Metal Nanoparticle Formation: The Intriguing, Informative Case of a Bimetallic Precursor, [(1,5-COD)Ir ⁺] ₂ HPO ₄ ²⁻ . <i>Chemistry of Materials</i> , 2019, 31, 2848-2862.	3.2	19

#	ARTICLE	IF	CITATIONS
91	Polyoxoanion-Supported, Atomically Dispersed Iridium(I) and Rhodium(I): Na ₃ [(C ₄ H ₉) ₄ N] ₅ [Ir [±] -Nb ₃ P ₂ W ₁₅ O ₆₂]{ [±] 4-C ₈ H ₁₂ }] and Na ₃ [(C ₄ H ₉) ₄ N] ₅ [Rh [±] -Nb ₃ P ₂ W ₁₅ O ₆₂]{ [±] 4-C ₈ H ₁₂ }}. <i>Inorganic Syntheses</i> 2007, , 186-201.		18
92	Copper ion vs copper metalâ€‘organic framework catalyzed NO release from bioavailable S-Nitrosoglutathione en route to biomedical applications: Direct 1H NMR monitoring in water allowing identification of the distinct, true reaction stoichiometries and thiol dependencies. <i>Journal of Inorganic Biochemistry</i> , 2019, 199, 110760.	1.5	18
93	Particle formation mechanisms supported by <i>in situ</i> synchrotron XAFS and SAXS studies: a review of metal, metal-oxide, semiconductor and selected other nanoparticle formation reactions. <i>Materials Advances</i> , 2021, 2, 6532-6568.	2.6	18
94	Synthesis and Characterization of [Ir(1,5-Cyclooctadiene)(^{1/4} -H)] ₄ : A Tetrametallic Ir ₄ H ₄ -Core, Coordinatively Unsaturated Cluster. <i>Inorganic Chemistry</i> , 2012, 51, 3186-3193.	1.9	17
95	Triniobium, Wellsâ€‘Dawson-Type Polyoxoanion, [(<i>n</i> -C ₄ H ₉) ₄ N] ₉ P ₂ W ₁₅ Nb ₃ O ₆₅ Improvements in the Synthesis, Its Reliability, the Purity of the Product, and the Detailed Synthetic Procedure. <i>Inorganic Chemistry</i> , 2014, 53, 2666-2676.	1.9	17
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97	Supersensitivity of Transition-Metal Nanoparticle Formation to Initial Precursor Concentration and Reaction Temperature: Understanding Its Origins. <i>Journal of Nanoscience and Nanotechnology</i> , 2008, 8, 1551-1556.	0.9	16
98	Kinetic Evidence for Bimolecular Nucleation in Supported-Transition-Metal-Nanoparticle Catalyst Formation in Contact with Solution: The Prototype Ir(1,5-COD)Cl/ ³ -Al ₂ O ₃ to Ir(O) ^{1/4} 900/ ³ -Al ₂ O ₃ System. <i>ACS Catalysis</i> , 2012, 2, 298-305.	5.5	16
99	Response to â€‘Particle Size Is a Primary Determinant for Sigmoidal Kinetics of Nanoparticle Formation: A Disproofâ€‘ of the Finkeâ€‘Watzky (F-W) Nanoparticle Nucleation and Growth Mechanismâ€‘. <i>Chemistry of Materials</i> , 2020, 32, 3657-3672.	3.2	16
100	â€‘Burst Nucleationâ€‘ vs Autocatalytic, â€‘Burstâ€‘ Growth in Near-Monodisperse Particle-Formation Reactions. <i>Journal of Physical Chemistry C</i> , 2020, 124, 24543-24554.	1.5	15
101	Hydrocarbon-Soluble, Isolable Ziegler-Type Ir(0) Nanoparticle Catalysts Made from [(1,5-COD)Ir(^{1/4} -O) ₂ C ₈ H ₁₅] ₂ and 2â€‘5 Equivalents of AlEt ₃ : Their High Catalytic Activity, Long Lifetime, and AlEt ₃ -Dependent, Exceptional, 200 Â°C Thermal Stability. <i>ACS Catalysis</i> , 2012, 2, 632-641.	5.5	14
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112	Pseudoelementary Steps: A Key Concept and Tool for Studying the Kinetics and Mechanisms of Complex Chemical Systems. Journal of Physical Chemistry A, 2021, 125, 10687-10705.	1.1	7
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