

Xin Zhong

List of Publications by Year in descending order

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43
papers

4,639
citations

147566

31
h-index

264894

42
g-index

43
all docs

43
docs citations

43
times ranked

2729
citing authors

#	ARTICLE	IF	CITATIONS
1	Application of carbon dots and their composite materials for the detection and removal of radioactive ions: A review. <i>Chemosphere</i> , 2022, 287, 132313.	4.2	82
2	Insight into the performance and mechanism of persimmon tannin functionalized waste paper for U(VI) and Cr(VI) removal. <i>Chemosphere</i> , 2022, 287, 132199.	4.2	92
3	Application of aluminosilicate clay mineral-based composites in photocatalysis. <i>Journal of Environmental Sciences</i> , 2022, 115, 190-214.	3.2	74
4	Effect of <i>Shewanella oneidensis</i> MR-1 on U(VI) sequestration by montmorillonite. <i>Journal of Environmental Radioactivity</i> , 2022, 242, 106798.	0.9	44
5	Removal of U(VI) from aqueous solutions by an effective bio-adsorbent from walnut shell and cellulose composite-stabilized iron sulfide nanoparticles. <i>RSC Advances</i> , 2022, 12, 2675-2683.	1.7	9
6	Synthesis of carbon-based nanomaterials and their application in pollution management. <i>Nanoscale Advances</i> , 2022, 4, 1246-1262.	2.2	30
7	Effect of Bi ₂ WO ₆ nanoflowers on the U(VI) removal from water: Roles of adsorption and photoreduction. <i>Journal of Environmental Chemical Engineering</i> , 2022, 10, 107170.	3.3	17
8	Efficient Selective Removal of Radionuclides by Sorption and Catalytic Reduction Using Nanomaterials. <i>Nanomaterials</i> , 2022, 12, 1443.	1.9	7
9	High efficient photoreduction of U(VI) by a new synergistic photocatalyst of Fe ₃ O ₄ nanoparticle on GO/g-C ₃ N ₄ composites. <i>Journal of Materials Research and Technology</i> , 2022, 18, 4248-4255.	2.6	8
10	Adsorption-photocatalysis processes: The performance and mechanism of a bifunctional covalent organic framework for removing uranium ions from water. <i>Applied Surface Science</i> , 2022, 597, 153621.	3.1	22
11	Modified biochar: synthesis and mechanism for removal of environmental heavy metals. , 2022, 1, .		165
12	High-speed and efficient removal of uranium (VI) from aqueous solution by hydroxyapatite-modified ordered mesoporous carbon (CMK-3). <i>Environmental Science and Pollution Research</i> , 2022, 29, 78989-79001.	2.7	5
13	Recent developments of doped g-C ₃ N ₄ photocatalysts for the degradation of organic pollutants. <i>Critical Reviews in Environmental Science and Technology</i> , 2021, 51, 751-790.	6.6	346
14	Recent advances on preparation and environmental applications of MOF-derived carbons in catalysis. <i>Science of the Total Environment</i> , 2021, 760, 143333.	3.9	342
15	Aluminum-based metal-organic frameworks (CAU-1) highly efficient UO ₂ ²⁺ and TcO ₄ ⁻ ions immobilization from aqueous solution. <i>Journal of Hazardous Materials</i> , 2021, 407, 124729.	6.5	86
16	The photocatalytic reduction of U(VI) into U(IV) by ZIF-8/g-C ₃ N ₄ composites at visible light. <i>Environmental Research</i> , 2021, 196, 110349.	3.7	131
17	High effective enrichment of U(VI) from aqueous solutions on versatile crystalline carbohydrate polymer-functionalized graphene oxide. <i>Dalton Transactions</i> , 2021, 50, 14009-14017.	1.6	6
18	Construction of Core-Shell MOFs@COF Hybrids as a Platform for the Removal of UO ₂ ²⁺ and Eu ³⁺ Ions from Solution. <i>ACS Applied Materials & Interfaces</i> , 2021, 13, 13883-13895.	4.0	71

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19	Challenges of organic pollutant photocatalysis by biochar-based catalysts. <i>Biochar</i> , 2021, 3, 117-123.	6.2	174
20	Reductive and adsorptive elimination of U(VI) ions in aqueous solution by SFeS@Biochar composites. <i>Environmental Science and Pollution Research</i> , 2021, 28, 55176-55185.	2.7	49
21	Highly efficient U(VI) capture by amidoxime/carbon nitride composites: Evidence of EXAFS and modeling. <i>Chemosphere</i> , 2021, 274, 129743.	4.2	130
22	Synthesis and application of perovskite-based photocatalysts in environmental remediation: A review. <i>Journal of Molecular Liquids</i> , 2021, 334, 116029.	2.3	52
23	The study of MnO ₂ with different crystalline structures for U(VI) elimination from aqueous solution. <i>Journal of Molecular Liquids</i> , 2021, 335, 116296.	2.3	8
24	Adsorption and reduction of Cr(VI) from aqueous solution using cost-effective caffeic acid functionalized corn starch. <i>Chemosphere</i> , 2021, 279, 130539.	4.2	139
25	Extremely stable amidoxime functionalized covalent organic frameworks for uranium extraction from seawater with high efficiency and selectivity. <i>Science Bulletin</i> , 2021, 66, 1994-2001.	4.3	172
26	Removal of organic compounds by nanoscale zero-valent iron and its composites. <i>Science of the Total Environment</i> , 2021, 792, 148546.	3.9	242
27	Constructing new Fe ₃ O ₄ @MnO with 3D hollow structure for efficient recovery of uranium from simulated seawater. <i>Chemosphere</i> , 2021, 283, 131241.	4.2	60
28	Functionalized mesoporous carbon nanospheres for efficient uranium extraction from aqueous solutions. <i>Environmental Nanotechnology, Monitoring and Management</i> , 2021, 16, 100510.	1.7	1
29	Recent advances in metal-organic framework membranes for water treatment: A review. <i>Science of the Total Environment</i> , 2021, 800, 149662.	3.9	450
30	In-situ growth of COF on BiOBr 2D material with excellent visible-light-responsive activity for U(VI) photocatalytic reduction. <i>Separation and Purification Technology</i> , 2021, 279, 119627.	3.9	52
31	High Sorption and Selective Extraction of Actinides from Aqueous Solutions. <i>Molecules</i> , 2021, 26, 7101.	1.7	2
32	Highly efficient enrichment mechanism of U(VI) and Eu(III) by covalent organic frameworks with intramolecular hydrogen-bonding from solutions. <i>Applied Surface Science</i> , 2020, 504, 144403.	3.1	112
33	The fabrication of 3D hierarchical flower-like $\hat{\Gamma}$ -MnO ₂ @COF nanocomposites for the efficient and ultra-fast removal of UO ₂ ²⁺ ions from aqueous solution. <i>Environmental Science: Nano</i> , 2020, 7, 3303-3317.	2.2	93
34	The magnetic covalent organic framework as a platform for high-performance extraction of Cr(VI) and bisphenol a from aqueous solution. <i>Journal of Hazardous Materials</i> , 2020, 393, 122353.	6.5	220
35	Efficient removal of U(VI) from aqueous solutions using the magnetic biochar derived from the biomass of a bloom-forming cyanobacterium (<i>Microcystis aeruginosa</i>). <i>Chemosphere</i> , 2020, 254, 126898.	4.2	55
36	Preparation of core-shell structure Fe ₃ O ₄ @C@MnO ₂ nanoparticles for efficient elimination of U(VI) and Eu(III) ions. <i>Science of the Total Environment</i> , 2019, 685, 986-996.	3.9	101

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37	Enhanced Photoreduction of U(VI) on C ₃ N ₄ by Cr(VI) and Bisphenol A: ESR, XPS, and EXAFS Investigation. Environmental Science & Technology, 2019, 53, 6454-6461.	4.6	269
38	Biochar-based materials and their applications in removal of organic contaminants from wastewater: state-of-the-art review. Biochar, 2019, 1, 45-73.	6.2	255
39	Adsorptive and reductive removal of U(VI) by Dictyophora indusiata-derived biochar supported sulfide NZVI from wastewater. Chemical Engineering Journal, 2019, 366, 368-377.	6.6	200
40	Plasma-enhanced amidoxime/magnetic graphene oxide for efficient enrichment of U(VI) investigated by EXAFS and modeling techniques. Chemical Engineering Journal, 2019, 357, 66-74.	6.6	53
41	XANES and EXAFS investigation of uranium incorporation on nZVI in the presence of phosphate. Chemosphere, 2018, 201, 764-771.	4.2	67
42	Synthesis of magnetic Fe ₃ O ₄ /CFA composites for the efficient removal of U(VI) from wastewater. Chemical Engineering Journal, 2017, 320, 448-457.	6.6	108
43	The adsorption of U(VI) on carbonaceous nanofibers: A combined batch, EXAFS and modeling techniques. Separation and Purification Technology, 2017, 175, 140-146.	3.9	38