

Fuminori Ito

List of Publications by Year in descending order

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30
papers

819
citations

687363

13
h-index

501196

28
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30
all docs

30
docs citations

30
times ranked

965
citing authors

#	ARTICLE	IF	CITATIONS
1	Optimum conditions for efficient phagocytosis of rifampicin-loaded PLGA microspheres by alveolar macrophages. <i>Journal of Controlled Release</i> , 2007, 119, 69-76.	9.9	151
2	Efficient intracellular delivery of rifampicin to alveolar macrophages using rifampicin-loaded PLGA microspheres: effects of molecular weight and composition of PLGA on release of rifampicin. <i>Colloids and Surfaces B: Biointerfaces</i> , 2004, 36, 35-42.	5.0	121
3	Preparation and properties of monodispersed rifampicin-loaded poly(lactide-co-glycolide) microspheres. <i>Colloids and Surfaces B: Biointerfaces</i> , 2004, 39, 17-21.	5.0	78
4	Incorporation of water-soluble drugs in PLGA microspheres. <i>Colloids and Surfaces B: Biointerfaces</i> , 2007, 54, 173-178.	5.0	52
5	Selective delivery of rifampicin incorporated into poly(dl-lactic-co-glycolic) acid microspheres after phagocytotic uptake by alveolar macrophages, and the killing effect against intracellular <i>Mycobacterium bovis</i> Calmette-Guérin. <i>Microbes and Infection</i> , 2006, 8, 2484-2491.	1.9	51
6	Study of types and mixture ratio of organic solvent used to dissolve polymers for preparation of drug-containing PLGA microspheres. <i>European Polymer Journal</i> , 2009, 45, 658-667.	5.4	49
7	Effect of polyethylene glycol on preparation of rifampicin-loaded PLGA microspheres with membrane emulsification technique. <i>Colloids and Surfaces B: Biointerfaces</i> , 2008, 66, 65-70.	5.0	48
8	Factors affecting the loading efficiency of water-soluble drugs in PLGA microspheres. <i>Colloids and Surfaces B: Biointerfaces</i> , 2008, 61, 25-29.	5.0	44
9	Preparation and properties of PLGA microspheres containing hydrophilic drugs by the SPG (shirasu) Tj ETQq1 1 0.784314 rgBT /Overlo 20-25.	5.0	37
10	Control of drug loading efficiency and drug release behavior in preparation of hydrophilic-drug-containing monodisperse PLGA microspheres. <i>Journal of Materials Science: Materials in Medicine</i> , 2010, 21, 1563-1571.	3.6	20
11	Facile technique for preparing organic-inorganic composite particles: Monodisperse poly(lactide-co-glycolide) (PLGA) particles having silica nanoparticles on the surface. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2010, 361, 109-117.	4.7	20
12	Development of high-performance polymer membranes for CO ₂ separation by combining functionalities of polyvinyl alcohol (PVA) and sodium polyacrylate (PAANa). <i>Journal of Polymer Research</i> , 2019, 26, 1.	2.4	20
13	Possibility for the development of cosmetics with PLGA nanospheres. <i>Drug Development and Industrial Pharmacy</i> , 2013, 39, 752-761.	2.0	15
14	Development of CO ₂ Molecular Gate Membranes for IGCC Process with CO ₂ Capture. <i>Energy Procedia</i> , 2017, 114, 613-620.	1.8	15
15	Rapid preparation of monodisperse biodegradable polymer nanospheres using a membrane emulsification technique under low gas pressure. <i>Journal of Polymer Research</i> , 2011, 18, 2077-2085.	2.4	12
16	Technique to encapsulate a low molecular weight hydrophilic drug in biodegradable polymer particles in a liquid-liquid system. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2011, 384, 368-373.	4.7	12
17	Facile technique for the preparation of monodispersed biodegradable polymer nanospheres using a solvent evaporation method. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2015, 482, 734-739.	4.7	10
18	Examination of Selection and Combination of Water-Absorbing Agent to Blend with Polyvinyl Alcohol (PVA) in Preparing CO ₂ -Separation Membrane with High-Performance. <i>Macromolecular Research</i> , 2020, 28, 365-372.	2.4	10

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19	Optimized preparation of biodegradable polymer particles encapsulating low-molecular-weight hydrophilic drugs. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2012, 402, 29-36.	4.7	7
20	Proposition of CO ₂ Removable Technology Using Membrane for Hydrogen Station. <i>ECS Transactions</i> , 2013, 51, 259-264.	0.5	7
21	Dermal administration of manganese porphyrin by iontophoresis. <i>Materials Science and Engineering C</i> , 2014, 41, 349-353.	7.3	7
22	Effects of the polymer composite composition and amine-based additives on the performance of a polymer composite CO ₂ separation membrane. <i>Polymer Bulletin</i> , 2021, 78, 513-528.	3.3	7
23	CO ₂ facilitated transport membranes prepared by blending polyvinyl alcohol and various water-absorbing agents. <i>Journal of Applied Polymer Science</i> , 2021, 138, 50191.	2.6	6
24	Preparation of (hydrophilic) INZ/PLGA particles (microcapsules) employing a unique frozen water phase investigation of optimal formulation. <i>Colloids and Surfaces A: Physicochemical and Engineering Aspects</i> , 2014, 443, 356-362.	4.7	4
25	Optimization of a simple technique for preparation of monodisperse poly(lactide-co-glycolide) nanospheres. <i>Journal of Nanoparticle Research</i> , 2016, 18, 1.	1.9	4
26	Physical properties of microspheres prepared by blending poly(lactide-co-glycolide) and poly lactide. <i>Bulletin of Materials Science</i> , 2021, 44, 1.	1.7	4
27	Factors for improving the performance of the separation membranes prepared by the blending of polyvinyl alcohol and a water absorbing agent. <i>Polymer-Plastics Technology and Materials</i> , 2021, 60, 659-669.	1.3	3
28	Preparation of Biodegradable Polymer Nanospheres Containing Manganese Porphyrin (Mn-Porphyrin). <i>Journal of Inorganic and Organometallic Polymers and Materials</i> , 2019, 29, 1010-1018.	3.7	2
29	High performance CO ₂ -facilitated transport membrane fabricated by compounding amine-terminated dendrimer in composite of polyvinyl alcohol and water-absorbing agent. <i>Journal of Macromolecular Science - Pure and Applied Chemistry</i> , 2021, 58, 849-859.	2.2	2
30	Development of a polyvinyl alcohol/sodium polyacrylate composite polymer membrane with cesium carbonate as a mobile carrier for high-performance CO ₂ capture. <i>Polymers for Advanced Technologies</i> , 2022, 33, 1677-1684.	3.2	1