## Cem Kahruman

## List of Publications by Year

 in descending orderSource: https:/|exaly.com/author-pdf/4029936/publications.pdf
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1. Kinetics of celestite conversion to acidic strontium oxalate hydrate in aqueous solution of oxalic acid. Transactions of Nonferrous Metals Society of China, 2019, 29, 1332-1345.

Reaction mechanism of strontium cobaltite synthesis from equimolar mixture of $\mathrm{Sr}(\mathrm{NO} 3) 2$ and $\mathrm{Co}(\mathrm{NO} 3) 2 \hat{a}^{\wedge}{ }^{\mathrm{TM}} 6 \mathrm{H} 2 \mathrm{O}$ under air atmosphere. Thermochimica Acta, 2019, 676, 52-63.

Determination of conversion reaction mechanism of celestite to acidic strontium oxalate hydrate in aqueous solution of H 2 C 2 O 4. Hydrometallurgy, 2017, 171, 53-60.
4.3

Conversion kinetics of SrSO4 to SrCO3 in solutions obtained by dissolving/hydrolyzing of equimolar

Kinetics of conversion of celestite to strontium carbonate in solutions containing carbonate,
5 bicarbonate and ammonium ions and dissolved ammonia. Journal of the Serbian Chemical Society, 2014,
$0.8 \quad 12$ 79, 345-359.
$6 \quad$ Retention and surface changes of zirconia primary crowns with secondary crowns of different materials. Clinical Oral Investigations, 2014, 18, 2023-2035.
3.0

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3 Determination of conversion reaction mechanism of celestite to acidic strontium oxalate hydrate in

$7 \quad$ Correlation between Bifilm Index and Toughness of Aluminum Alloys., 2014, , 171-176.

The Use of Oxalic Acid as a Chelating Agent in the Dissolution Reaction of Calcium Molybdate.
8 Metallurgical and Materials Transactions B: Process Metallurgy and Materials Processing Science, 2013, 44, 495-505.

9 Thermal decomposition of ammonium paratungstate hydrate in air and inert gas atmospheres.
International Journal of Refractory Metals and Hard Materials, 2013, 37, 106-116.

The investigation of dissolution behavior of gangue materials during the dissolution of scheelite concentrate in oxalic acid solution. Hydrometallurgy, 2013, 136, 15-26.
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\text { Kinetic investigation of dissolution of } \mathrm{CaWO} \text { <sub>4</sub>in oxalic acid solution. Canadian }
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Metallurgical Quarterly, 2013, 52, 348-357.

The investigation of thermal decomposition mechanism of ammonium phosphotungstate hydrate in inert gas atmosphere. Thermochimica Acta, 2012, 546, 1-7.
2.7

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Thermal decomposition mechanism of ammonium phosphotungstate hydrate in air atmosphere. International Journal of Refractory Metals and Hard Materials, 2012, 31, 14-20.

Characterization of the thermal decomposition products of ammonium phosphomolybdate hydrate. Journal of Analytical and Applied Pyrolysis, 2007, 78, 363-370.

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[^0]:    Leaching kinetics of synthetic CaWO4 in HCl solutions containing H3PO4 as chelating agent.
    Hydrometallurgy, 2006, 81, 182-189.

