

Nikolla P Qafoku

List of Publications by Year in descending order

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100
papers

3,761
citations

126907

33
h-index

144013

57
g-index

113
all docs

113
docs citations

113
times ranked

3646
citing authors

#	ARTICLE	IF	CITATIONS
1	Natural, incidental, and engineered nanomaterials and their impacts on the Earth system. <i>Science</i> , 2019, 363, .	12.6	479
2	Developing a molecular picture of soil organic matter–mineral interactions by quantifying organo–mineral binding. <i>Nature Communications</i> , 2017, 8, 396.	12.8	150
3	Geochemical Implications of Gas Leakage associated with Geologic CO ₂ Storage–A Qualitative Review. <i>Environmental Science & Technology</i> , 2013, 47, 23-36.	10.0	146
4	The biogeochemistry of technetium: A review of the behavior of an artificial element in the natural environment. <i>Numerische Mathematik</i> , 2010, 310, 721-752.	1.4	142
5	Variable Charge Soils: Their Mineralogy, Chemistry and Management. <i>Advances in Agronomy</i> , 2004, 84, 159-215.	5.2	137
6	Kinetic Desorption and Sorption of U(VI) during Reactive Transport in a Contaminated Hanford Sediment. <i>Environmental Science & Technology</i> , 2005, 39, 3157-3165.	10.0	137
7	Immobilization of 99-Technetium (VII) by Fe(II)-Goethite and Limited Reoxidation. <i>Environmental Science & Technology</i> , 2011, 45, 4904-4913.	10.0	124
8	Scale–dependent desorption of uranium from contaminated subsurface sediments. <i>Water Resources Research</i> , 2008, 44, .	4.2	123
9	Geochemical, mineralogical and microbiological characteristics of sediment from a naturally reduced zone in a uranium-contaminated aquifer. <i>Applied Geochemistry</i> , 2012, 27, 1499-1511.	3.0	123
10	Abiotic Reductive Immobilization of U(VI) by Biogenic Mackinawite. <i>Environmental Science & Technology</i> , 2013, 47, 2361-2369.	10.0	100
11	Uranium in Framboidal Pyrite from a Naturally Bioreduced Alluvial Sediment. <i>Environmental Science & Technology</i> , 2009, 43, 8528-8534.	10.0	85
12	Anion Transport through Columns of Highly Weathered Acid Soil: Adsorption and Retardation. <i>Soil Science Society of America Journal</i> , 1996, 60, 132-137.	2.2	81
13	Cr(OH) ₃ (s) Oxidation Induced by Surface Catalyzed Mn(II) Oxidation. <i>Environmental Science & Technology</i> , 2014, 48, 10760-10768.	10.0	74
14	Silver-functionalized silica aerogels and their application in the removal of iodine from aqueous environments. <i>Journal of Hazardous Materials</i> , 2019, 379, 119364.	12.4	64
15	Iodine immobilization by materials through sorption and redox-driven processes: A literature review. <i>Science of the Total Environment</i> , 2020, 716, 132820.	8.0	59
16	Removal of TcO ₄ [–] from Representative Nuclear Waste Streams with Layered Potassium Metal Sulfide Materials. <i>Chemistry of Materials</i> , 2016, 28, 3976-3983.	6.7	56
17	Modeling the impact of carbon dioxide leakage into an unconfined, oxidizing carbonate aquifer. <i>International Journal of Greenhouse Gas Control</i> , 2016, 44, 290-299.	4.6	56
18	Terrestrial Nanoparticles and Their Controls on Soil-/Geo-Processes and Reactions. <i>Advances in Agronomy</i> , 2010, 107, 33-91.	5.2	55

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19	Adsorption Properties of Subtropical and Tropical Variable Charge Soils: Implications from Climate Change and Biochar Amendment. <i>Advances in Agronomy</i> , 2016, 135, 1-58.	5.2	54
20	Anion Transport in Columns of Variable Charge Subsoils: Nitrate and Chloride. <i>Journal of Environmental Quality</i> , 2000, 29, 484-493.	2.0	48
21	Arsenate Displacement from Fly Ash in Amended Soils. <i>Water, Air, and Soil Pollution</i> , 1999, 114, 185-198.	2.4	47
22	Transport-controlled kinetics of dissolution and precipitation in the sediments under alkaline and saline conditions. <i>Geochimica Et Cosmochimica Acta</i> , 2004, 68, 2981-2995.	3.9	47
23	Iron oxide waste form for stabilizing 99Tc. <i>Journal of Nuclear Materials</i> , 2012, 429, 201-209.	2.7	46
24	Aluminum Effect on Dissolution and Precipitation under Hyperalkaline Conditions. <i>Journal of Environmental Quality</i> , 2003, 32, 2364-2372.	2.0	43
25	Review of the impacts of leaking CO ₂ gas and brine on groundwater quality. <i>Earth-Science Reviews</i> , 2017, 169, 69-84.	9.1	42
26	Foam Delivery of Calcium Polysulfide to the Vadose Zone for Chromium(VI) Immobilization: A Laboratory Evaluation. <i>Vadose Zone Journal</i> , 2009, 8, 976-985.	2.2	41
27	Geochemical and mineralogical investigation of uranium in multi-element contaminated, organic-rich subsurface sediment. <i>Applied Geochemistry</i> , 2014, 42, 77-85.	3.0	40
28	Effect of Coupled Dissolution and Redox Reactions on Cr(VI) Attenuation during Transport in the Sediments under Hyperalkaline Conditions. <i>Environmental Science & Technology</i> , 2003, 37, 3640-3646.	10.0	39
29	Mineralogy and chemistry of some variable charge subsoils. <i>Communications in Soil Science and Plant Analysis</i> , 2000, 31, 1051-1070.	1.4	38
30	RETENTION AND TRANSPORT OF CALCIUM NITRATE IN VARIABLE CHARGE SUBSOILS. <i>Soil Science</i> , 2001, 166, 297-307.	0.9	38
31	Adsorption and Desorption of Indifferent Ions in Variable Charge Subsoils. <i>Soil Science Society of America Journal</i> , 2002, 66, 1231-1239.	2.2	38
32	Influence of acidic and alkaline waste solution properties on uranium migration in subsurface sediments. <i>Journal of Contaminant Hydrology</i> , 2013, 151, 155-175.	3.3	38
33	Technetium Stabilization in Low-Solubility Sulfide Phases: A Review. <i>ACS Earth and Space Chemistry</i> , 2018, 2, 532-547.	2.7	36
34	Interactions of aqueous U(VI) with soil minerals in slightly alkaline natural systems. <i>Reviews in Environmental Science and Biotechnology</i> , 2008, 7, 355-380.	8.1	34
35	Climate-Change Effects on Soils: Accelerated Weathering, Soil Carbon, and Elemental Cycling. <i>Advances in Agronomy</i> , 2015, 131, 111-172.	5.2	34
36	Coupled Geochemical Impacts of Leaking CO ₂ and Contaminants from Subsurface Storage Reservoirs on Groundwater Quality. <i>Environmental Science & Technology</i> , 2015, 49, 8202-8209.	10.0	34

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37	Evaluating impacts of CO ₂ intrusion into an unconsolidated aquifer: I. Experimental data. <i>International Journal of Greenhouse Gas Control</i> , 2016, 44, 323-333.	4.6	31
38	Proteinâ€“Mineral Interactions: Molecular Dynamics Simulations Capture Importance of Variations in Mineral Surface Composition and Structure. <i>Langmuir</i> , 2016, 32, 6194-6209.	3.5	31
39	Incorporation Modes of Iodate in Calcite. <i>Environmental Science & Technology</i> , 2018, 52, 5902-5910.	10.0	31
40	Advective Removal of Intraparticle Uranium from Contaminated Vadose Zone Sediments, Hanford, U.S.. <i>Environmental Science & Technology</i> , 2008, 42, 1565-1571.	10.0	30
41	Strong mineralogic control of soil organic matter composition in response to nutrient addition across diverse grassland sites. <i>Science of the Total Environment</i> , 2020, 736, 137839.	8.0	29
42	A review of the behavior of radioiodine in the subsurface at two DOE sites. <i>Science of the Total Environment</i> , 2019, 691, 466-475.	8.0	28
43	Relative Humidity Controls Ammonia Loss from Urea Applied to Loblolly Pine. <i>Soil Science Society of America Journal</i> , 2010, 74, 543-549.	2.2	27
44	Getters for improved technetium containment in cementitious waste forms. <i>Journal of Hazardous Materials</i> , 2018, 341, 238-247.	12.4	25
45	Aluminum Effect on Dissolution and Precipitation under Hyperalkaline Conditions. <i>Journal of Environmental Quality</i> , 2003, 32, 2354-2363.	2.0	24
46	Design and Ammoniaâ€“Recovery Evaluation of a Wind Speedâ€“Sensitive Chamber System. <i>Soil Science Society of America Journal</i> , 2001, 65, 1302-1306.	2.2	23
47	Pathways of Aqueous Cr(VI) Attenuation in a Slightly Alkaline Oxidic Subsurface. <i>Environmental Science & Technology</i> , 2009, 43, 1071-1077.	10.0	23
48	Chromium transport in an acidic waste contaminated subsurface medium: The role of reduction. <i>Chemosphere</i> , 2010, 81, 1492-1500.	8.2	23
49	Evaluating impacts of CO ₂ intrusion into an unconsolidated aquifer: II. Modeling results. <i>International Journal of Greenhouse Gas Control</i> , 2016, 44, 300-309.	4.6	23
50	Time-Dependent Iodate and Iodide Adsorption to Fe Oxides. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 2415-2420.	2.7	23
51	Mineralogy of a perudic Andosol in central Java, Indonesia. <i>Geoderma</i> , 2008, 144, 379-386.	5.1	21
52	Silver-based getters for ¹²⁹ I removal from low-activity waste. <i>Radiochimica Acta</i> , 2016, 104, 905-913.	1.2	21
53	Technetium immobilization by materials through sorption and redox-driven processes: A literature review. <i>Science of the Total Environment</i> , 2020, 716, 132849.	8.0	19
54	The function of Sn(II)-apatite as a Tc immobilizing agent. <i>Journal of Nuclear Materials</i> , 2016, 480, 393-402.	2.7	18

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55	Direct Visualization of Aggregate Morphology and Dynamics in a Model Soil Organic-Mineral System. <i>Environmental Science and Technology Letters</i> , 2017, 4, 186-191.	8.7	18
56	Efficacy of acetate-amended biostimulation for uranium sequestration: Combined analysis of sediment/groundwater geochemistry and bacterial community structure. <i>Applied Geochemistry</i> , 2017, 78, 172-185.	3.0	18
57	Technetium and iodine aqueous species immobilization and transformations in the presence of strong reductants and calcite-forming solutions: Remedial action implications. <i>Science of the Total Environment</i> , 2018, 636, 588-595.	8.0	17
58	Impact of iron and manganese nano-metal-oxides on contaminant interaction and fortification potential in agricultural systems – a review. <i>Environmental Chemistry</i> , 2019, 16, 377.	1.5	17
59	Geochemical impacts of leaking CO ₂ from subsurface storage reservoirs to an unconfined oxidizing carbonate aquifer. <i>International Journal of Greenhouse Gas Control</i> , 2016, 44, 310-322.	4.6	16
60	Element mobilization and immobilization from carbonate rocks between CO ₂ storage reservoirs and the overlying aquifers during a potential CO ₂ leakage. <i>Chemosphere</i> , 2018, 197, 399-410.	8.2	16
61	Chromate Effect on Iodate Incorporation into Calcite. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 1624-1630.	2.7	16
62	Geochemical and Geophysical Changes during Ammonia Gas Treatment of Vadose Zone Sediments for Uranium Remediation. <i>Vadose Zone Journal</i> , 2012, 11, vj2011.0158.	2.2	14
63	Mineral assemblage transformation of a metakaolin-based waste form after geopolymer encapsulation. <i>Journal of Nuclear Materials</i> , 2016, 473, 320-332.	2.7	13
64	Current Understanding of the Use of Alkaline Extractions of Soils to Investigate Soil Organic Matter and Environmental Processes. <i>Journal of Environmental Quality</i> , 2019, 48, 1561-1564.	2.0	13
65	Chemical weathering of new pyroclastic deposits from Mt. Merapi (Java), Indonesia. <i>Journal of Mountain Science</i> , 2009, 6, 240-254.	2.0	12
66	CR(VI) FATE IN MINERALOGICALLY ALTERED SEDIMENTS BY HYPERALKALINE WASTE FLUIDS. <i>Soil Science</i> , 2007, 172, 598-613.	0.9	11
67	Evidence of technetium and iodine release from a sodalite-bearing ceramic waste form. <i>Applied Geochemistry</i> , 2016, 66, 210-218.	3.0	11
68	Performance of the Fluidized Bed Steam Reforming product under hydraulically unsaturated conditions. <i>Journal of Environmental Radioactivity</i> , 2014, 131, 119-128.	1.7	10
69	Phyllosilicate mineral dissolution upon alkaline treatment under aerobic and anaerobic conditions. <i>Applied Clay Science</i> , 2020, 189, 105520.	5.2	10
70	Uranium fate in Hanford sediment altered by simulated acid waste solutions. <i>Applied Geochemistry</i> , 2015, 63, 1-9.	3.0	9
71	Fe-solid phase transformations under highly basic conditions. <i>Applied Geochemistry</i> , 2007, 22, 2054-2064.	3.0	8
72	Advective Desorption of Uranium(VI) from Contaminated Hanford Vadose Zone Sediments under Saturated and Unsaturated Conditions. <i>Vadose Zone Journal</i> , 2008, 7, 1144-1159.	2.2	8

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73	Remediation of Technetium in Vadose Zone Sediments Using Ammonia and Hydrogen Sulfide Gases. Vadose Zone Journal, 2015, 14, 1-12.	2.2	8
74	Characterizing Technetium in Subsurface Sediments for Contaminant Remediation. ACS Earth and Space Chemistry, 2018, 2, 1145-1160.	2.7	8
75	Evaluating Impacts of CO ₂ Gas Intrusion Into a Confined Sandstone aquifer: Experimental Results. Energy Procedia, 2014, 63, 3275-3284.	1.8	7
76	Glass Corrosion in the Presence of Iron-Bearing Materials and Potential Corrosion Suppressors. Materials Research Society Symposia Proceedings, 2015, 1744, 139-144.	0.1	7
77	Influence of Carbon and Microbial Community Priming on the Attenuation of Uranium in a Contaminated Floodplain Aquifer. Ground Water, 2015, 53, 600-613.	1.3	7
78	Risk of Geologic Sequestration of CO ₂ to Groundwater Aquifers: Current Knowledge and Remaining Questions. Energy Procedia, 2017, 114, 3052-3059.	1.8	7
79	Methanogenesis-induced pH-Eh shifts drives aqueous metal(loid) mobility in sulfide mineral systems under CO ₂ enriched conditions. Geochimica Et Cosmochimica Acta, 2016, 173, 232-245.	3.9	6
80	In situ precipitation of hydrous ferric oxide (HFO) for remediation of subsurface iodine contamination. Journal of Contaminant Hydrology, 2020, 235, 103705.	3.3	6
81	A Review of Bismuth(III)-Based Materials for Remediation of Contaminated Sites. ACS Earth and Space Chemistry, 2022, 6, 883-908.	2.7	6
82	Effect of extent of natural subsurface bioreduction on Fe-mineralogy of subsurface sediments. Journal of Physics: Conference Series, 2010, 217, 012047.	0.4	5
83	Microbial Methylation of Iodide in Unconfined Aquifer Sediments at the Hanford Site, USA. Frontiers in Microbiology, 2019, 10, 2460.	3.5	5
84	Charge fingerprint in relation to mineralogical composition of Quaternary volcanic ash along a climatic gradient on Java Island, Indonesia. Catena, 2019, 172, 547-557.	5.0	5
85	Iodate interactions with calcite: implications for natural attenuation. Environmental Earth Sciences, 2020, 79, 1.	2.7	5
86	Simultaneous immobilization of aqueous co-contaminants using a bismuth layered material. Journal of Environmental Radioactivity, 2021, 237, 106711.	1.7	5
87	ARSENIC, BORON, SELENIUM, AND MOLYBDENUM DISPLACEMENT AND TRANSPORT IN A FLY ASH AMENDED SOIL LEACHED WITH CALCIUM PHOSPHATE SOLUTION. Communications in Soil Science and Plant Analysis, 2001, 32, 1499-1512.	1.4	4
88	Evaluating impacts of CO ₂ and CH ₄ gas intrusion into an unconsolidated aquifer: fate of As and Cd. Frontiers in Environmental Science, 2015, 3, .	3.3	4
89	Persistence of chromate in vadose zone and aquifer sediments in Hanford, Washington. Science of the Total Environment, 2019, 676, 482-492.	8.0	4
90	Selective Interactions of Soil Organic Matter Compounds with Calcite and the Role of Aqueous Ca. ACS Earth and Space Chemistry, 0, , .	2.7	4

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91	Physical control on CCl ₄ and CHCl ₃ desorption from artificially contaminated and aged sediments with supercritical carbon dioxide. <i>Chemosphere</i> , 2009, 74, 494-500.	8.2	3
92	Fungal hyphae develop where titanomagnetite inclusions reach the surface of basalt grains. <i>Scientific Reports</i> , 2022, 12, 3407.	3.3	3
93	Silicon concentration and pH controls over competitive or simultaneous incorporation of iodate and chromate into calcium carbonate phases. <i>Applied Geochemistry</i> , 2021, 128, 104941.	3.0	2
94	Editorial: Searching for Solutions to Soil Pollution: Underlying Soil-Contaminant Interactions and Development of Innovative Land Remediation and Reclamation Techniques. <i>Frontiers in Environmental Science</i> , 2022, 9, .	3.3	2
95	Response to Comment on "Geochemical Implications of Gas Leakage associated with Geologic CO ₂ Storage" A Qualitative Review. <i>Environmental Science & Technology</i> , 2013, 47, 4951-4952.	10.0	1
96	Tchnetium Getters to Improve Cast Stone Performance. <i>Materials Research Society Symposia Proceedings</i> , 2015, 1744, 43-52.	0.1	1
97	Evaluation of gaseous substrates for microbial immobilization of contaminant mixtures in unsaturated subsurface sediments. <i>Journal of Environmental Radioactivity</i> , 2020, 214-215, 106183.	1.7	1
98	Ion Transport Dynamics in Acid Variable Charge Subsoils. <i>Soil Science and Plant Nutrition</i> , 2005, 51, 601-603.	1.9	0
99	Stability of mineral-organic matter associations under varying biogeochemical conditions. <i>Soil Science Society of America Journal</i> , 0, , .	2.2	0
100	SSSAJ 2021 Publisher's Report. <i>Soil Science Society of America Journal</i> , 2022, 86, 868-878.	2.2	0