

Olga Cheremisina

List of Publications by Citations

Source: <https://exaly.com/author-pdf/3853864/olga-cheremisina-publications-by-citations.pdf>

Version: 2024-04-25

This document has been generated based on the publications and citations recorded by exaly.com. For the latest version of this publication list, visit the link given above.

The third column is the impact factor (IF) of the journal, and the fourth column is the number of citations of the article.

46
papers

93
citations

5
h-index

6
g-index

56
ext. papers

124
ext. citations

1.2
avg, IF

2.69
L-index

#	Paper	IF	Citations
46	Specific features of solvent extraction of REM from phosphoric acid solutions with DEHPA. <i>Mineral Processing and Extractive Metallurgy: Transactions of the Institute of Mining and Metallurgy</i> , 2019 , 1-7	0.8	6
45	Sorption of rare earth coordination compounds. <i>Journal of Mining Institute</i> , 244 , 474-481	3	6
44	Solvent sublation and ion flotation in aqueous salt solutions containing Ce(III) and Y(III) in the presence of a surfactant. <i>Russian Journal of Applied Chemistry</i> , 2014 , <i>87</i> , 1863-1867	0.8	5
43	Kinetics of oxidation of phenol with manganese dioxide. <i>Russian Journal of General Chemistry</i> , 2011 , <i>81</i> , 704-709	0.7	5
42	Thermodynamic Model of Ion-Exchange Process as Exemplified by Cerium Sorption from Multisalt Solutions. <i>Journal of Mining Institute</i> , 2019 , <i>237</i> , 307-316	3	5
41	Kinetics Study of Solvent and Solid-Phase Extraction of Rare Earth Metals with Di-2-Ethylhexylphosphoric Acid. <i>Metals</i> , 2020 , <i>10</i> , 687	2.3	4
40	Isotherm of strontium and sodium cation exchange on iron-manganese nodules. <i>Russian Journal of Applied Chemistry</i> , 2006 , <i>79</i> , 367-371	0.8	4
39	Determination of the Surface Area of Minerals by Sorption of Methylene Blue and Thermal Desorption of Argon. <i>Russian Journal of Applied Chemistry</i> , 2003 , <i>76</i> , 663-665	0.8	4
38	Thermodynamic characteristics of sorption extraction and chromatographic separation of anionic complexes of erbium and cerium with Trilon B on weakly basic anionite. <i>Russian Journal of Physical Chemistry A</i> , 2016 , <i>90</i> , 664-670	0.7	3
37	Process of Extraction of Gallium from Technological Solutions with the Use of Ion Exchange Resins. <i>Metallurgist</i> , 2019 , <i>63</i> , 206-214	0.8	3
36	Thermodynamic investigation into extraction of cerium(III) by tributyl phosphate from phosphoric acid solutions. <i>Russian Journal of Non-Ferrous Metals</i> , 2015 , <i>56</i> , 615-621	0.8	3
35	Sorption of aluminate from alkaline solutions on D-403 anion exchanger. <i>Russian Journal of Physical Chemistry A</i> , 2011 , <i>85</i> , 1995-1999	0.7	3
34	Isotherm of Strontium Sorption on Clay. <i>Russian Journal of Applied Chemistry</i> , 2003 , <i>76</i> , 727-730	0.8	3
33	Influence of anion nature on acid leaching of silicate minerals and solvent extraction of rare and rare-earth elements. <i>Chemie Der Erde</i> , 2020 , <i>80</i> , 125507	4.3	3
32	Kinetic Features of the Hydrogen Sulfide Sorption on the Ferro-Manganese Material. <i>Metals</i> , 2021 , <i>11</i> , 90	2.3	3
31	Sorption recovery of gallium and aluminum from alkaline solutions on an AN-31 anion exchanger. <i>Russian Journal of Non-Ferrous Metals</i> , 2017 , <i>58</i> , 365-372	0.8	2
30	Sorptive separation of yttrium and cerium on a weakly basic anionite. <i>Russian Journal of Physical Chemistry A</i> , 2015 , <i>89</i> , 119-124	0.7	2

29	Rare Earth Metal Extraction from Apatite Ores. <i>Metallurgist</i> , 2019 , 63, 300-307	0.8	2
28	Sorption of gallium from the alkali solutions based on anionites. <i>Russian Journal of Non-Ferrous Metals</i> , 2013 , 54, 201-208	0.8	2
27	Thermodynamic study of cerium sorption onto anionite from sulfate media. <i>Russian Journal of Physical Chemistry A</i> , 2013 , 87, 288-295	0.7	2
26	Thermodynamics of the sorption of cerium complex compounds on anionite. <i>Russian Journal of Physical Chemistry A</i> , 2013 , 87, 1562-1569	0.7	2
25	Sorption of Iron(II) on Ferromanganese Nodules. <i>Russian Journal of Applied Chemistry</i> , 2005 , 78, 592-598	0.8	2
24	PROBLEMS OF PROTECTION OF URBAN AREAS FROM RADIONUCLIDES STRONTIUM-90 AND CAESIUM-137 AFTER TECHNOLOGICAL DISASTERS. <i>Journal of Ecological Engineering</i> , 2017 , 18, 97-103	2	2
23	COMPARISON OF EXTRACTION METHODS FOR EXTRACTION OF IRON, ALUMINUM, MANGANESE AND TITANIUM USING CARBOXYLIC ACIDS AND NATURAL VEGETABLE OILS FROM WATER-SALT SYSTEMS 2017 ,		2
22	THE USEGE OF A MULTIFUNCTIONAL SORBENT BASED ON FERROMANGANESE NODULES FOR NEUTRALIZING WASTEWATER FROM OIL REFINERIES 2017 ,		2
21	Extraction of Rare Earth Metals by Solid-Phase Extractants from Phosphoric Acid Solution. <i>Metals</i> , 2021 , 11, 991	2.3	2
20	Complex processing technology of gold-bearing concentrates: Autoclave leaching with subsequent roasting. <i>Russian Journal of Non-Ferrous Metals</i> , 2015 , 56, 404-408	0.8	1
19	Sorption thermodynamics of cobalt(II) cations on iron-manganese concretions. <i>Russian Journal of Applied Chemistry</i> , 2011 , 84, 588-591	0.8	1
18	Temperature effects on the thermodynamic parameters of sorption of germanium on an anionite. <i>Russian Journal of Physical Chemistry A</i> , 2008 , 82, 2147-2151	0.7	1
17	The hydrometallurgical method of obtaining of pure zinc and germanium oxides from the slag of copper-lead production. <i>Russian Journal of Non-Ferrous Metals</i> , 2008 , 49, 356-362	0.8	1
16	Application of the Organic Waste-Based Sorbent for the Purification of Aqueous Solutions. <i>Water (Switzerland)</i> , 2021 , 13, 3101	3	1
15	Determination of the mutual entrainment of the extractant and the aqueous phase in the extraction of rare-earth elements from the technological phosphoric acid solution. <i>Journal of Physics: Conference Series</i> , 2019 , 1399, 055025	0.3	1
14	Concentration and Separation of Heavy Rare-Earth Metals at Stripping Stage. <i>Metals</i> , 2019 , 9, 1317	2.3	1
13	Quantitative x-ray spectral determination of rare-earth metals in products of metallurgy. <i>Journal of Physics: Conference Series</i> , 2018 , 1118, 012012	0.3	1
12	Thermodynamic Characteristics of the Hydrogen Sulfide Sorption Process by Ferromanganese Materials.. <i>ACS Omega</i> , 2022 , 7, 3007-3015	3.9	0

11	Interaction Features of Sodium Oleate and Oxyethylated Phosphoric Acid Esters with the Apatite Surface.. <i>ACS Omega</i> , 2022 , 7, 3016-3023	3.9	o
10	Increasing the efficiency of rare earth metal recovery from technological solutions during processing of apatite raw materials. <i>Journal of Mining Institute</i> , 252, 1-10	3	o
9	Kinetics of phenol oxidation with iron-manganese concretions. <i>Russian Journal of General Chemistry</i> , 2012 , 82, 685-692	0.7	
8	Kinetics of the oxidation of hexacyanoferrate(III) with pyrolusite. <i>Russian Journal of Physical Chemistry A</i> , 2013 , 87, 915-918	0.7	
7	Improving performance characteristics of semiconductor sensors based on adsorption SnO ₂ using photons stimulation. <i>Russian Journal of Physical Chemistry B</i> , 2012 , 6, 637-642	1.2	
6	Isotherm of exchange of sodium and copper cations on ferrimanganese concretions. <i>Russian Journal of Applied Chemistry</i> , 2009 , 82, 231-235	0.8	
5	Kinetics of the ion exchange of lead and sodium cations on the surface of iron-manganese concretions. <i>Russian Journal of Applied Chemistry</i> , 2010 , 83, 1540-1543	0.8	
4	Isotherm of Pb-Na cation exchange on iron-manganese concretions. <i>Russian Journal of Applied Chemistry</i> , 2010 , 83, 1762-1766	0.8	
3	Isotherm of exchange of nickel and sodium cations on iron-manganese nodules. <i>Russian Journal of Applied Chemistry</i> , 2006 , 79, 1091-1095	0.8	
2	Exchange Isotherm of Strontium(II) and Iron(III) Ions on Clay. <i>Russian Journal of Applied Chemistry</i> , 2004 , 77, 576-578	0.8	
1	Thermodynamic Study of Iron(III) Sorption on Clay. <i>Russian Journal of Applied Chemistry</i> , 2003 , 76, 892-895	0.8	