

Paul Jerabek

List of Publications by Year in descending order

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Version: 2024-02-01

49
papers

1,650
citations

304743

22
h-index

289244

40
g-index

57
all docs

57
docs citations

57
times ranked

1550
citing authors

#	ARTICLE	IF	CITATIONS
1	A Crystalline Singlet Phosphinonitrene: A Nitrogen Atomâ€“Transfer Agent. <i>Science</i> , 2012, 337, 1526-1528.	12.6	148
2	Isolation of Neutral Monoâ€“and Dinuclear Gold Complexes of Cyclic (Alkyl)(amino)carbenes. <i>Angewandte Chemie - International Edition</i> , 2013, 52, 8964-8967.	13.8	119
3	Experimental Charge Density Study of a Silylone. <i>Angewandte Chemie - International Edition</i> , 2014, 53, 2766-2770.	13.8	115
4	Reductive elimination: a pathway to low-valent aluminium species. <i>Chemical Communications</i> , 2013, 49, 2858.	4.1	94
5	Coinage Metals Binding as Main Group Elements: Structure and Bonding of the Carbene Complexes [TM(cAAC) ₂] and [TM(cAAC) ₂] ⁺ (TM = Cu, Ag, Au). <i>Journal of the American Chemical Society</i> , 2014, 136, 17123-17135.	13.7	84
6	Electron and Nucleon Localization Functions of Oganesson: Approaching the Thomas-Fermi Limit. <i>Physical Review Letters</i> , 2018, 120, 053001.	7.8	79
7	Dative and electronâ€“sharing bonding in transition metal compounds. <i>Journal of Computational Chemistry</i> , 2019, 40, 247-264.	3.3	74
8	Isolation of Neutral Mononuclear Copper Complexes Stabilized by Two Cyclic (Alkyl)(amino)carbenes. <i>Journal of the American Chemical Society</i> , 2014, 136, 6235-6238.	13.7	73
9	The Îƒâ€“Aromatic Clusters [Zn ₃] ⁺ and [Zn ₂ Cu]: Embryonic Brass. <i>Angewandte Chemie - International Edition</i> , 2015, 54, 4370-4374.	13.8	72
10	Ligandâ€“Stabilized [P ₄] ²⁺ Cations. <i>Angewandte Chemie - International Edition</i> , 2012, 51, 2964-2967.	13.8	67
11	[3+2] Fragmentation of an [RP ₅ Cl] ⁺ Cage Cation Induced by an Nâ€“Heterocyclic Carbene. <i>Angewandte Chemie - International Edition</i> , 2013, 52, 11078-11082.	13.8	49
12	A Novel Concept for the Synthesis of Multiply Doped Gold Clusters [(M@Au _n)M ² L _k] ^{q+} . <i>Angewandte Chemie - International Edition</i> , 2014, 53, 4327-4331.	13.8	34
13	A relativistic coupled-cluster interaction potential and rovibrational constants for the xenon dimer. <i>Molecular Physics</i> , 2018, 116, 1-8.	1.7	31
14	Znâ€“Zn interactions at nickel and palladium centers. <i>Chemical Science</i> , 2016, 7, 6413-6421.	7.4	30
15	Magnesium- and intermetallic alloys-based hydrides for energy storage: modelling, synthesis and properties. <i>Progress in Energy</i> , 2022, 4, 032007.	10.9	29
16	Comparative bonding analysis of N ₂ and P ₂ versus tetrahedral N ₄ and P ₄ . <i>Theoretical Chemistry Accounts</i> , 2014, 133, 1.	1.4	27
17	The Rich Chemistry of [Zn ₂ Cp* ₂]: Trapping Three Different Types of Zinc Ligands in the PdZn ₇ Complex [Pd(ZnCp*) ₄ (ZnMe) ₂ {Zn(tmeda)}]. <i>Inorganic Chemistry</i> , 2011, 50, 10486-10492.	4.0	26
18	Nearly Degenerate Isomers of C(BH) ₂ : Cumulene, Carbene, or Carbone?. <i>Chemistry - A European Journal</i> , 2013, 19, 15941-15954.	3.3	25

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19	Catalytic Asymmetric Fluorination of Copper Carbene Complexes: Preparative Advances and a Mechanistic Rationale. <i>Chemistry - A European Journal</i> , 2020, 26, 2509-2515.	3.3	24
20	Solid Oganesson via a Many-Body Interaction Expansion Based on Relativistic Coupled-Cluster Theory and from Plane-Wave Relativistic Density Functional Theory. <i>Journal of Physical Chemistry A</i> , 2019, 123, 4201-4211.	2.5	23
21	Oganesson Is a Semiconductor: On the Relativistic Band ϵ Gap Narrowing in the Heaviest Noble ϵ Gas Solids. <i>Angewandte Chemie - International Edition</i> , 2019, 58, 14260-14264.	13.8	22
22	Dizinc Cation [Zn ₂] ²⁺ Trapped In A Homoleptic Metalloid Coordination Environment Stabilized by Dispersion Forces: [Zn ₂ (GaCp*) ₆][BAR ₄ F] ₂ . <i>Inorganic Chemistry</i> , 2015, 54, 352-358.	4.0	21
23	Influence of Relativistic Effects on Bonding Modes in M(II) Dinuclear Complexes (M = Au, Ag, and Cu). <i>Inorganic Chemistry</i> , 2017, 56, 14624-14631.	4.0	21
24	Oganesson: A Noble Gas Element That Is Neither Noble Nor a Gas. <i>Angewandte Chemie - International Edition</i> , 2020, 59, 23636-23640.	13.8	21
25	Oligonuclear Molecular Models of Intermetallic Phases: A Case Study on [Pd ₂ Zn ₆ Ga ₂ (Cp*) ₅ (CH ₃) ₃]. <i>Chemistry - A European Journal</i> , 2012, 18, 4909-4915.	3.3	16
26	Hume ϵ Rothery Phase-Inspired Metal-Rich Molecules: Cluster Expansion of [Ni(ZnMe) ₆ (ZnCp*) ₂] by Face Capping with Ni ⁰ (I ⁶ -toluene) and Ni ^I (I ⁵ -Cp*). <i>Inorganic Chemistry</i> , 2014, 53, 10403-10411.	4.0	15
27	Understanding the F 1s NEXAFS Dichroism in Fluorinated Organic Semiconductors. <i>Journal of Physical Chemistry C</i> , 2016, 120, 12693-12705.	3.1	15
28	Suppressed Phosphine Dissociation by Polarization Effects on the Donor ϵ Acceptor Bonds in [Ni(PEt ₃) ₄] ⁿ⁺ (ECp*) ⁿ⁻ (E = Al, Ga). <i>Inorganic Chemistry</i> , 2018, 57, 12657-12664.	4.0	15
29	First-principles melting of krypton and xenon based on many-body relativistic coupled-cluster interaction potentials. <i>Physical Review B</i> , 2020, 101, .	3.2	15
30	A Hundred ϵ Year ϵ Old Experiment Re ϵ evaluated: Accurate Ab ϵ ...Initio Monte ϵ ...Carlo Simulations of the Melting of Radon. <i>Angewandte Chemie - International Edition</i> , 2018, 57, 9961-9964.	13.8	14
31	The Oxygen ϵ Rich Beryllium Oxides BeO ₄ and BeO ₆ . <i>Angewandte Chemie - International Edition</i> , 2016, 55, 10863-10867.	13.8	13
32	Tipping the Balance between Ligand and Metal Protonation due to Relativistic Effects: Unusually High Proton Affinity in Gold(I) Pincer Complexes. <i>Chemistry - A European Journal</i> , 2018, 24, 6047-6051.	3.3	12
33	The Organozinc Rich Compounds [Cp*M(ZnR) ₅] (M = Fe, Ru; R = Cp*, Me, Cl, Br). <i>Inorganic Chemistry</i> , 2013, 52, 7152-7160.	4.0	11
34	Experimental FTIR and theoretical investigation of the molecular structure and vibrational spectra of acetanilide using DFT and dispersion correction to DFT. <i>Journal of Theoretical and Computational Chemistry</i> , 2019, 18, 1950009.	1.8	9
35	Quantitative and Chemically Intuitive Evaluation of the Nature of M ϵ L Bonds in Paramagnetic Compounds: Application of EDA ϵ NOCV Theory to Spin Crossover Complexes. <i>Chemistry - A European Journal</i> , 2020, 26, 13677-13685.	3.3	9
36	Stepwise Synthesis of Siloxane-Substituted Oligoarsanes and Structural Investigation of Alkaline Earth Metal Derivatives. <i>European Journal of Inorganic Chemistry</i> , 2015, 2015, 3264-3273.	2.0	8

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37	Oganesson ist ein Halbleiter: Äber die relativistische Bandlückenkontraktion in den schwersten Edelgasen. Angewandte Chemie, 2019, 131, 14398-14402.	2.0	8
38	Hunting dimers. Theoretical Chemistry Accounts, 2012, 131, 1.	1.4	6
39	Static dipole polarizability of palladium from relativistic coupled-cluster theory. Physical Review A, 2018, 98, .	2.5	6
40	Ligands Based on Phosphine- σ -Stabilized Aluminum(I), Boron(I), and Carbon(0). Chemistry - A European Journal, 2019, 25, 3068-3076.	3.3	6
41	Oganesson: Ein Edelgas, das weder edel noch ein Gas ist. Angewandte Chemie, 2020, 132, 23843-23848.	2.0	5
42	Ein hundert Jahre altes Experiment auf dem Prüfstand: hochgenaue Ab-initio-Monte-Carlo-Schmelzsimulationen von Radon. Angewandte Chemie, 2018, 130, 10109-10113.	2.0	4
43	Hydrogenation via a low energy mechanochemical approach: the MgB_2 case. JPhys Energy, 2021, 3, 044001.	5.3	4
44	Quantitative Assessment of Ligand Substituent Effects on σ - and π -Contributions to Fe-N Bonds in Spin Crossover Fe^{II} Complexes. Chemistry - A European Journal, 2022, 28, .	3.3	4
45	Bonding analysis of telluroketones $H_2A=Te$ (A = C, Si, Ge). Journal of Molecular Modeling, 2014, 20, 2433.	1.8	3
46	Modeling the thermodynamics of the FeTi hydrogenation under para-equilibrium: An ab-initio and experimental study. Calphad: Computer Coupling of Phase Diagrams and Thermochemistry, 2022, 77, 102426.	1.6	3
47	Structures and energies of C_4S_4q ($q=+1, 0, -1$) isomers. A theoretical study. International Journal of Mass Spectrometry, 2013, 354-355, 342-345.	1.5	0
48	The Light-Driven Isomerization of Aqueous Nitrate: A Theoretical Perspective. ChemPhotoChem, 2018, 2, 725-733.	3.0	0
49	The Light-Driven Isomerization of Aqueous Nitrate: A Theoretical Perspective. ChemPhotoChem, 2018, 2, 702-702.	3.0	0