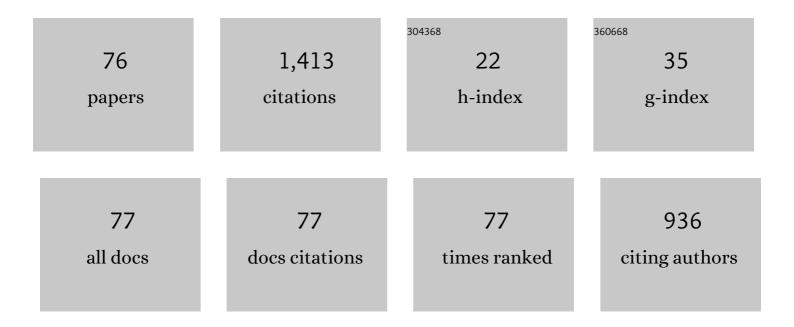
## Aleksandr Glotov

List of Publications by Year in descending order

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| #  | Article   | IF   | CITATIONS |
|----|---|------|-----------|
| 1  | Mesoporous Metal Catalysts Templated on Clay Nanotubes. Bulletin of the Chemical Society of Japan, 2019, 92, 61-69.   | 2.0  | 89        |
| 2  | Core/Shell Ruthenium–Halloysite Nanocatalysts for Hydrogenation of Phenol. Industrial &<br>Engineering Chemistry Research, 2017, 56, 14043-14052.   | 1.8  | 83        |
| 3  | Interfacial Self-Assembly in Halloysite Nanotube Composites. Langmuir, 2019, 35, 8646-8657.   | 1.6  | 82        |
| 4  | Clay nanotube-metal core/shell catalysts for hydroprocesses. Chemical Society Reviews, 2021, 50,<br>9240-9277.  | 18.7 | 73        |
| 5  | Oxidative Desulfurization of Fuels Using Heterogeneous Catalysts Based on MCM-41. Energy &<br>Fuels, 2018, 32, 10898-10903.   | 2.5  | 58        |
| 6  | Nanoparticles Formed onto/into Halloysite Clay Tubules: Architectural Synthesis and Applications.<br>Chemical Record, 2018, 18, 858-867.  | 2.9  | 56        |
| 7  | Halloysite nanotube-based cobalt mesocatalysts for hydrogen production from sodium borohydride.<br>Journal of Solid State Chemistry, 2018, 268, 182-189.  | 1.4  | 54        |
| 8  | Templated self-assembly of ordered mesoporous silica on clay nanotubes. Chemical Communications, 2019, 55, 5507-5510.   | 2.2  | 50        |
| 9  | Ru/CdS Quantum Dots Templated on Clay Nanotubes as Visibleâ€Lightâ€Active Photocatalysts: Optimization of S/Cd Ratio and Ru Content. Chemistry - A European Journal, 2020, 26, 13085-13092.               | 1.7  | 48        |
| 10 | Catalytic cracking additives based on mesoporous MCM-41 for sulfur removal. Fuel Processing<br>Technology, 2016, 153, 50-57.  | 3.7  | 39        |
| 11 | Methane Hydrate Formation in Halloysite Clay Nanotubes. ACS Sustainable Chemistry and Engineering, 2020, 8, 7860-7868.  | 3.2  | 37        |
| 12 | Mesoporous Al-HMS and Al-MCM-41 supported Ni-Mo sulfide catalysts for HYD and HDS via in situ<br>hydrogen generation through a WGSR. Catalysis Today, 2019, 329, 156-166.                                 | 2.2  | 36        |
| 13 | Ruthenium Catalysts Templated on Mesoporous MCM-41 Type Silica and Natural Clay Nanotubes for<br>Hydrogenation of Benzene to Cyclohexane. Catalysts, 2020, 10, 537.                                       | 1.6  | 33        |
| 14 | Deep Oxidative Desulfurization of Fuels in the Presence of Brönsted Acidic Polyoxometalate-Based<br>Ionic Liquids. Molecules, 2020, 25, 536.  | 1.7  | 30        |
| 15 | Ruthenium-Loaded Halloysite Nanotubes as Mesocatalysts for Fischer–Tropsch Synthesis. Molecules,<br>2020, 25, 1764.   | 1.7  | 29        |
| 16 | Dispersed Ni-Mo sulfide catalysts from water-soluble precursors for HDS of BT and DBT via in situ<br>produced H2 under Water gas shift conditions. Applied Catalysis B: Environmental, 2021, 282, 119616. | 10.8 | 29        |
| 17 | Micro-mesoporous MCM-41/ZSM-5 supported Pt and Pd catalysts for hydroisomerization of C-8 aromatic fraction. Applied Catalysis A: General, 2020, 603, 117764.   | 2.2  | 28        |
| 18 | Core-shell nanoarchitecture: Schiff-base assisted synthesis of ruthenium in clay nanotubes. Pure and<br>Applied Chemistry, 2018, 90, 825-832.   | 0.9  | 26        |

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|----|---|------|-----------|
| 19 | Aluminosilicates supported La-containing sulfur reduction additives for FCC catalyst: Correlation between activity, support structure and acidity. Catalysis Today, 2019, 329, 135-141.   | 2.2  | 26        |
| 20 | Nanoreactors based on hydrophobized tubular aluminosilicates decorated with ruthenium: Highly active and stable catalysts for aromatics hydrogenation. Catalysis Today, 2021, 378, 33-42.   | 2.2  | 26        |
| 21 | Enhanced HDS and HYD activity of sulfide Co-PMo catalyst supported on alumina and structured mesoporous silica composite. Catalysis Today, 2021, 377, 82-91.  | 2.2  | 25        |
| 22 | Nanostructured Ruthenium Catalysts in Hydrogenation of Aromatic Compounds. Petroleum Chemistry, 2018, 58, 1221-1226.  | 0.4  | 24        |
| 23 | Hydrodeoxygenation of guaiacol via in situ H2 generated through a water gas shift reaction over<br>dispersed NiMoS catalysts from oil-soluble precursors: Tuning the selectivity towards cyclohexene.<br>Applied Catalysis B: Environmental, 2022, 312, 121403. | 10.8 | 24        |
| 24 | Selective hydrogenation of terminal alkynes over palladium nanoparticles within the pores of amino-modified porous aromatic frameworks. Catalysis Today, 2020, 357, 176-184.  | 2.2  | 22        |
| 25 | Architectural design of core–shell nanotube systems based on aluminosilicate clay. Nanoscale<br>Advances, 2022, 4, 2823-2835.   | 2.2  | 22        |
| 26 | Manganese and Cobalt Doped Hierarchical Mesoporous Halloysite-Based Catalysts for Selective<br>Oxidation of p-Xylene to Terephthalic Acid. Catalysts, 2020, 10, 7.  | 1.6  | 21        |
| 27 | Transition Metal Sulfides- and Noble Metal-Based Catalysts for N-Hexadecane Hydroisomerization: A<br>Study of Poisons Tolerance. Catalysts, 2020, 10, 594.  | 1.6  | 21        |
| 28 | Isomerization of Xylenes in the Presence of Pt-Containing Catalysts Based on Halloysite<br>Aluminosilicate Nanotubes. Russian Journal of Applied Chemistry, 2018, 91, 1353-1362.  | 0.1  | 18        |
| 29 | A Study of Platinum Catalysts Based on Ordered Alâ€"ĐœĐ¡Đœ-41 Aluminosilicate and Natural Halloysite<br>Nanotubes in Xylene Isomerization. Petroleum Chemistry, 2019, 59, 1226-1234.  | 0.4  | 17        |
| 30 | CdS Quantum Dots in Hierarchical Mesoporous Silica Templated on Clay Nanotubes: Implications for<br>Photocatalytic Hydrogen Production. ACS Applied Nano Materials, 2022, 5, 605-614.   | 2.4  | 16        |
| 31 | Ruthenium Catalysts on ZSM-5/MCM-41 Micro-Mesoporous Support for Hydrodeoxygenation of Guaiacol in the Presence of Water. Russian Journal of Applied Chemistry, 2019, 92, 1170-1178.  | 0.1  | 14        |
| 32 | Effect of the ruthenium deposition method on the nanostructured catalyst activity in the deep hydrogenation of benzene. Russian Chemical Bulletin, 2020, 69, 260-264.   | 0.4  | 14        |
| 33 | Ni–Mo sulfide nanosized catalysts from water-soluble precursors for hydrogenation of aromatics<br>under water gas shift conditions. Pure and Applied Chemistry, 2020, 92, 949-966.  | 0.9  | 14        |
| 34 | Selective Hydrogenation of Acetylene over Pd-Mn/Al2O3 Catalysts. Catalysts, 2020, 10, 624.  | 1.6  | 13        |
| 35 | Halloysite as a Zeolite Catalyst Component for Converting Dimethyl Ether Into Hydrocarbons.<br>Chemistry and Technology of Fuels and Oils, 2020, 55, 682-688.   | 0.2  | 13        |
| 36 | Nanoarchitectural approach for synthesis of highly crystalline zeolites with a low Si/Al ratio from natural clay nanotubes. Microporous and Mesoporous Materials, 2022, 330, 111622.  | 2.2  | 13        |

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|----|--|-----|-----------|
| 37 | Carbon Dioxide Reforming of Methane. Russian Journal of Applied Chemistry, 2020, 93, 765-787.  | 0.1 | 12        |
| 38 | Alkali Earth Catalysts Based on Mesoporous MCM-41 and Al-SBA-15 for Sulfone Removal from Middle<br>Distillates. ACS Omega, 2019, 4, 12736-12744.   | 1.6 | 11        |
| 39 | Hydroconversion of Aromatic Hydrocarbons over Bimetallic Catalysts. Catalysts, 2019, 9, 384.   | 1.6 | 11        |
| 40 | Hydrogenation of Aromatic Substrates over Dispersed Ni–Mo Sulfide Catalysts in System H2O/CO.<br>Petroleum Chemistry, 2018, 58, 528-534.   | 0.4 | 9         |
| 41 | Bizeolite Pt/ZSM-5:ZSM-12/Al2O3 catalyst for hydroisomerization of C-8 fraction with various ethylbenzene content. Catalysis Today, 2021, 378, 83-95.  | 2.2 | 9         |
| 42 | Hydroconversion of Thiophene Derivatives over Dispersed Ni–Mo Sulfide Catalysts. Petroleum<br>Chemistry, 2018, 58, 1227-1232.  | 0.4 | 8         |
| 43 | Natural clay nanotube supported Mo and W catalysts for exhaustive oxidative desulfurization of model fuels. Pure and Applied Chemistry, 2021, 93, 231-241.   | 0.9 | 8         |
| 44 | CuO-In2O3 Catalysts Supported on Halloysite Nanotubes for CO2 Hydrogenation to Dimethyl Ether.<br>Catalysts, 2021, 11, 1151.   | 1.6 | 8         |
| 45 | Isomerization of Xylenes (a Review). Petroleum Chemistry, 2021, 61, 1158-1177.   | 0.4 | 8         |
| 46 | Core-shell catalysts with CoMoS phase embedded in clay nanotubes for dibenzothiophene<br>hydrodesulfurization. Catalysis Today, 2022, 397-399, 121-128.  | 2.2 | 8         |
| 47 | Bimetallic Sulfur-Reducing Additives Based on Al–MCM-41 Structured Aluminosilicate for Cracking<br>Catalysts. Petroleum Chemistry, 2018, 58, 214-219.  | 0.4 | 7         |
| 48 | Halloysite Based Core-Shell Nanosystems: Synthesis and Application. , 2019, , 203-256.   |     | 7         |
| 49 | Formation of ruthenium nanoparticles inside aluminosilicate nanotubes and their catalytic activity in aromatics hydrogenation: the impact of complexing agents and reduction procedure. Pure and Applied Chemistry, 2020, 92, 909-918. | 0.9 | 6         |
| 50 | CO <sub>2</sub> hydrogenation to dimethyl ether over In <sub>2</sub> O <sub>3</sub> catalysts supported on aluminosilicate halloysite nanotubes. Green Processing and Synthesis, 2021, 10, 594-605.                                    | 1.3 | 6         |
| 51 | Use of ionic liquids in cyclohexene epoxidation with hydrogen peroxide. Petroleum Chemistry, 2013, 53, 110-116.  | 0.4 | 5         |
| 52 | Micro-Mesoporous Catalyst Based on Dealuminated Halloysite Nanotubes for Isomerization of C-8<br>Aromatic Fraction. Petroleum Chemistry, 2021, 61, 1085-1095.  | 0.4 | 5         |
| 53 | Structured composite catalyst Pd/Ce0.75Zr0.25O2-x/Î,-Al2O3/FeCrAlloy for complete oxidation of methane. Materials Letters, 2022, 310, 131481.  | 1.3 | 5         |
| 54 | Natural aluminosilicate nanotubes loaded with RuCo as nanoreactors for Fischer-Tropsch synthesis.<br>Science and Technology of Advanced Materials, 2022, 23, 17-30.  | 2.8 | 5         |

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|----|--|-----|-----------|
| 55 | Catalytic Cracking of Petroleum Feedstock in the Presence of Additives Derived from Cross–Linked<br>Mesoporous Oxides for Reduction of the Sulfur Content in Liquid Products. Chemistry and<br>Technology of Fuels and Oils, 2016, 52, 171-174.  | 0.2 | 4         |
| 56 | Hydroconversion of Oxidation Products of Sulfur-Containing Aromatic Compounds. Russian Journal of Applied Chemistry, 2018, 91, 981-989.  | 0.1 | 4         |
| 57 | Hydrocracking of Vacuum Gas Oil on Bimetallic Ni-Mo Sulfide Catalysts Based on Mesoporous<br>Aluminosilicate Al-HMS. Chemistry and Technology of Fuels and Oils, 2016, 52, 515-526.  | 0.2 | 3         |
| 58 | Sulfur-Reducing Additives Based on Aluminosilicates Al-SBA-15 and Al-SBA-16 for Cracking catalysts.<br>Chemistry and Technology of Fuels and Oils, 2018, 54, 15-23.  | 0.2 | 3         |
| 59 | Bimetallic Sulfur Reduction Additives Based on Alumosilicate of Al-MCM-41 Type For Cracking<br>Catalysts: Desulfurazing Activity vs. Ratio of Components in a Support. Russian Journal of Applied<br>Chemistry, 2019, 92, 562-568.               | 0.1 | 3         |
| 60 | Synthesis and studies of structured support Ce0.75Zr0.25O2/Î,-Al2O3/FeCrAl. Materials Letters, 2021, 283, 128855.  | 1.3 | 3         |
| 61 | Structured catalysts for the conversion of liquefied petroleum gas to hydrogen-rich gas and for<br>anode off-gas afterburning. International Journal of Hydrogen Energy, 2021, 46, 35853-35865.  | 3.8 | 3         |
| 62 | Heterogeneous Catalysts for Petrochemical Synthesis and Oil Refining. Catalysts, 2021, 11, 602.  | 1.6 | 3         |
| 63 | Synthesis, Physicochemical Properties, and Strength Profile of Hydroprocessing Catalyst Supports<br>Based on Aluminosilicate Halloysite Nanotubes. Chemistry and Technology of Fuels and Oils, 2021, 57,<br>250.                                 | 0.2 | 3         |
| 64 | The mesoporous silicate-alumina composites application as supports for bifunctional sulfide catalysts for n-hexadecane hydroconversion. Journal of Porous Materials, 2021, 28, 1449-1458.  | 1.3 | 3         |
| 65 | Mathematical Modeling of the Catalytic Cracking of Oil Sludge that has Been Subjected to Electromagnetic Activation. Chemistry and Technology of Fuels and Oils, 2016, 51, 663-672.  | 0.2 | 2         |
| 66 | Sulfur-reduction additives based on ordered hexagonal mesoporous silica in the catalytic cracking of vacuum gas oil. Theoretical Foundations of Chemical Engineering, 2017, 51, 825-829.   | 0.2 | 2         |
| 67 | Study of the Oxidation Products of Light Oil Aromatic Compounds Using Ultrahigh Resolution Mass Spectrometry. Chemistry and Technology of Fuels and Oils, 2018, 53, 891-896.   | 0.2 | 2         |
| 68 | Hydroconversion of 2-methylnaphtalene and dibenzothiophene over sulfide catalysts in the presence of water under CO pressure. Russian Chemical Bulletin, 2020, 69, 280-288.  | 0.4 | 2         |
| 69 | Structured catalytic burner for deep oxidation of hydrocarbons. Catalysis Communications, 2021, 149, 106198.   | 1.6 | 2         |
| 70 | Influence of the Procedure for Preparing Ruthenium Nanoparticles on the Internal Surface of<br>Aluminosilicate Nanotubes on Their Catalytic Properties in Benzene Hydrogenation in the Presence of<br>Water. Petroleum Chemistry, 2021, 61, 676. | 0.4 | 2         |
| 71 | Micro-Mesoporous Catalyst Based on Natural Aluminosilicate Nanotubes and ZSM-5 Zeolite for<br>Methanol Conversion to Hydrocarbons. Petroleum Chemistry, 2021, 61, 773-780.   | 0.4 | 2         |
| 72 | Hydroconversion of n-Hexadecane on Zeolite-Containing Sulfide-Based Catalysts: Influence of<br>Nitrogen Impurity in the Feedstock on the Hydroisomerization Selectivity. Petroleum Chemistry, 2021,<br>61, 739-747.                              | 0.4 | 2         |

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|----|---|-----|-----------|
| 73 | Ruthenium-Containing Catalysts Based on Halloysite Aluminosilicate Nanotubes of Different Origin in<br>Benzene Hydrogenation. Petroleum Chemistry, 2021, 61, 1104-1110. | 0.4 | 2         |
| 74 | Two-stage oxidative desulfurization of material containing oil sludge. Theoretical Foundations of Chemical Engineering, 2017, 51, 830-834.                              | 0.2 | 1         |
| 75 | Biofuels energetics: Measurements and evaluation of calorific values of triglycerides. Fuel, 2022, 326, 125101.   | 3.4 | 1         |

The 18<sup>th</sup> IUPAC International Symposium Macromolecular-Metal Complexes ( $10\hat{a}\in 13$  June,) Tj ETQq0.0.0 rgBT /Overlock  $\frac{10}{0.9}$