

# Andrew Kerridge

## List of Publications by Year in descending order

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53  
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#	ARTICLE	IF	CITATIONS
1	Does Reduction-Induced Isomerization of a Uranium(III) Aryl Complex Proceed via C-H Oxidative Addition and Reductive Elimination across the Uranium(II/IV) Redox Couple?. <i>Inorganic Chemistry</i> , 2022, 61, 8955-8965.	4.0	7
2	Structural, spectroscopic, and computational evaluations of cation-cation and halogen bonding interactions in heterometallic uranyl hybrid materials. <i>Inorganic Chemistry Frontiers</i> , 2021, 8, 1128-1141.	6.0	7
3	Systematic Investigation of the Molecular and Electronic Structure of Thorium and Uranium Phosphorus and Arsenic Complexes. <i>Inorganic Chemistry</i> , 2021, 60, 10614-10630.	4.0	15
4	Photoluminescence of Pentavalent Uranyl Amide Complexes. <i>Journal of the American Chemical Society</i> , 2021, 143, 13184-13194.	13.7	5
5	<i>Ab initio</i> molecular dynamics studies of hydroxide coordination of alkaline earth metals and uranyl. <i>Physical Chemistry Chemical Physics</i> , 2019, 21, 13809-13820.	2.8	6
6	Factors affecting the nucleus-independent chemical shift in NMR studies of microporous carbon electrode materials. <i>Energy Storage Materials</i> , 2019, 21, 335-346.	18.0	18
7	Oxygen Vacancy Formation and Water Adsorption on Reduced $\text{AnO}_2$ {111}, {110}, and {100} Surfaces (An = U, Pu): A Computational Study. <i>Journal of Physical Chemistry C</i> , 2018, 122, 7149-7165.	3.1	45
8	Why Do Some Molecules Form Hydrates or Solvates?. <i>Crystal Growth and Design</i> , 2018, 18, 1903-1908.	3.0	47
9	How to Bend the Uranyl Cation via Crystal Engineering. <i>Inorganic Chemistry</i> , 2018, 57, 2714-2723.	4.0	17
10	Energy-Degeneracy-Driven Covalency in Actinide Bonding. <i>Journal of the American Chemical Society</i> , 2018, 140, 17977-17984.	13.7	108
11	Decomposition of d- and f-Shell Contributions to Uranium Bonding from the Quantum Theory of Atoms in Molecules: Application to Uranium and Uranyl Halides. <i>Inorganics</i> , 2018, 6, 88.	2.7	19
12	Probing hydrogen and halogen-oxo interactions in uranyl coordination polymers: a combined crystallographic and computational study. <i>CrystEngComm</i> , 2018, 20, 4916-4925.	2.6	23
13	Solubility prediction from first principles: a density of states approach. <i>Physical Chemistry Chemical Physics</i> , 2018, 20, 20981-20987.	2.8	26
14	Coordination Chemistry and QTAIM Analysis of Homoleptic Dithiocarbamate Complexes, $\text{M}(\text{S}_2\text{CN})_4$ (M = Ti, Zr, Hf, Th, U, Np). <i>Inorganic Chemistry</i> , 2018, 57, 10518-10524.	4.0	24
15	Elucidation of the inverse <i>trans</i> influence in uranyl and its imido and carbene analogues <i>via</i> quantum chemical simulation. <i>Chemical Communications</i> , 2018, 54, 9761-9764.	4.1	16
16	Ligand size dependence of U-N and U-O bond character in a series of uranyl hexaphyrin complexes: quantum chemical simulation and density based analysis. <i>Physical Chemistry Chemical Physics</i> , 2017, 19, 7546-7559.	2.8	10
17	The inverse-trans-influence in tetravalent lanthanide and actinide bis(carbene) complexes. <i>Nature Communications</i> , 2017, 8, 14137.	12.8	128
18	Water Adsorption on $\text{AnO}_2$ {111}, {110}, and {100} Surfaces (An = U and Pu): A Density Functional Theory + <i>U</i> Study. <i>Journal of Physical Chemistry C</i> , 2017, 121, 1675-1682.	3.1	48

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19	Quantification of f-element covalency through analysis of the electron density: insights from simulation. <i>Chemical Communications</i> , 2017, 53, 6685-6695.	4.1	102
20	Double Reduction of 4,4'-Bipyridine and Reductive Coupling of Pyridine by Two Thorium(III) Single-Electron Transfers. <i>Chemistry - A European Journal</i> , 2017, 23, 2290-2293.	3.3	26
21	Actinide covalency measured by pulsed electron paramagnetic resonance spectroscopy. <i>Nature Chemistry</i> , 2017, 9, 578-583.	13.6	102
22	Unravelling the electronic structure and dynamics of an isolated molecular rotary motor in the gas-phase. <i>Chemical Science</i> , 2017, 8, 6141-6148.	7.4	13
23	Assessing covalency in equatorial U <sup>IV</sup> -N bonds: density based measures of bonding in BTP and isoamethyrin complexes of uranyl. <i>Physical Chemistry Chemical Physics</i> , 2016, 18, 16830-16839.	2.8	24
24	Ionic adsorption on the brucite (0001) surface: A periodic electrostatic embedded cluster method study. <i>Journal of Chemical Physics</i> , 2016, 145, 204708.	3.0	7
25	Electronic structure of bulk AnO <sub>2</sub> (An = U, Np, Pu) and water adsorption on the (111) and (110) surfaces of UO <sub>2</sub> and PuO <sub>2</sub> from hybrid density functional theory within the periodic electrostatic embedded cluster method. <i>Journal of Nuclear Materials</i> , 2016, 482, 124-134.	2.7	30
26	Concomitant Carboxylate and Oxalate Formation From the Activation of CO <sub>2</sub> by a Thorium(III) Complex. <i>Chemistry - A European Journal</i> , 2016, 22, 17976-17979.	3.3	39
27	Topological Study of Bonding in Aquo and Bis(triazinyl)pyridine Complexes of Trivalent Lanthanides and Actinides: Does Covalency Imply Stability?. <i>Inorganic Chemistry</i> , 2016, 55, 10034-10042.	4.0	41
28	Emergence of comparable covalency in isostructural cerium(IV) and uranium(IV) carbon multiple bonds. <i>Chemical Science</i> , 2016, 7, 3286-3297.	7.4	90
29	Should environmental effects be included when performing QAIM calculations on actinide systems? A comparison of QAIM metrics for Cs <sub>2</sub> UO <sub>2</sub> Cl <sub>4</sub> , U(Se <sub>2</sub> PPh <sub>2</sub> ) <sub>4</sub> and Np(Se <sub>2</sub> PPh <sub>2</sub> ) <sub>4</sub> in gas phase, COSMO and PEECM. <i>Polyhedron</i> , 2016, 116, 57-63.	2.2	14
30	U <sup>IV</sup> -O <sub>yl</sub> Stretching Vibrations as a Quantitative Measure of the Equatorial Bond Covalency in Uranyl Complexes: A Quantum-Chemical Investigation. <i>Inorganic Chemistry</i> , 2016, 55, 573-583.	4.0	53
31	White phosphorus activation by a Th(III) complex. <i>Dalton Transactions</i> , 2016, 45, 2390-2393.	3.3	30
32	Assessing Covalency in Cerium and Uranium Hexachlorides: A Correlated Wavefunction and Density Functional Theory Study. <i>Inorganics</i> , 2015, 3, 482-499.	2.7	33
33	The importance of second shell effects in the simulation of hydrated Sr <sup>2+</sup> hydroxide complexes. <i>Dalton Transactions</i> , 2015, 44, 11572-11581.	3.3	9
34	Dithio- and Diselenophosphinate Thorium(IV) and Uranium(IV) Complexes: Molecular and Electronic Structures, Spectroscopy, and Transmetalation Reactivity. <i>Inorganic Chemistry</i> , 2015, 54, 11625-11636.	4.0	35
35	Yttrium Complexes of Arsine, Arsenide, and Arsinidene Ligands. <i>Angewandte Chemie - International Edition</i> , 2015, 54, 4255-4258.	13.8	28
36	Neptunyl(VI) centred visible LMCT emission directly observable in the presence of uranyl(VI). <i>Chemical Communications</i> , 2015, 51, 5402-5405.	4.1	18

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37	Optical excitation of MgO nanoparticles; a computational perspective. <i>Physical Chemistry Chemical Physics</i> , 2014, 16, 22052-22061.	2.8	33
38	f-Orbital covalency in the actinocenes (An = Th–Cm): multiconfigurational studies and topological analysis. <i>RSC Advances</i> , 2014, 4, 12078-12086.	3.6	78
39	Emission spectroscopy of uranium(iv) compounds: a combined synthetic, spectroscopic and computational study. <i>RSC Advances</i> , 2013, 3, 4350.	3.6	57
40	Oxidation state and covalency in f-element metallocenes (M = Ce, Th, Pu): a combined CASSCF and topological study. <i>Dalton Transactions</i> , 2013, 42, 16428.	3.3	90
41	A RASSCF study of free base, magnesium and zinc porphyrins: accuracy versus efficiency. <i>Physical Chemistry Chemical Physics</i> , 2013, 15, 2197.	2.8	17
42	The coordination of Sr <sup>2+</sup> by hydroxide: a density functional theoretical study. <i>Dalton Transactions</i> , 2011, 40, 11258.	3.3	8
43	Quantum Chemical Studies of the Hydration of Sr <sup>2+</sup> in Vacuum and Aqueous Solution. <i>Chemistry - A European Journal</i> , 2011, 17, 5060-5067.	3.3	17
44	All-electron CASPT2 study of Ce(III)C <sub>8</sub> H <sub>6</sub> . <i>Comptes Rendus Chimie</i> , 2010, 13, 853-859.	0.5	23
45	Are the Ground States of the Later Actinocenes Multiconfigurational? All-Electron Spin-Orbit Coupled CASPT2 Calculations on An <sup>2+</sup> (An = Th, Pa, U, Np, Pu, Am, Cm). <i>Journal of Physical Chemistry A</i> , 2010, 114, 10784-10791.	0.784314	14
46	Is Cerocene Really a Ce(III) Compound? All-Electron Spin-Orbit Coupled CASPT2 Calculations on M <sup>2+</sup> (M = Th, Pa, Ce). <i>Journal of Physical Chemistry A</i> , 2009, 113, 2896-2905.	2.5	72
47	A mystery solved? Photoelectron spectroscopic and quantum chemical studies of the ion states of CeCp <sub>3</sub> <sup>+</sup> . <i>Dalton Transactions</i> , 2009, , 5943.	3.3	21
48	Structure-dependent exchange in the organic magnets Cu(II)Pc and Mn(II)Pc. <i>Physical Review B</i> , 2008, 77, .	3.2	38
49	Electron dynamics in quantum gate operation. <i>Journal of Physics Condensed Matter</i> , 2007, 19, 282201.	1.8	7
50	Molecular Thin Films: A New Type of Magnetic Switch. <i>Advanced Materials</i> , 2007, 19, 3618-3622.	21.0	133
51	Time dependent quantum simulations of two-qubit gates based on donor states in silicon. <i>Journal of Physics Condensed Matter</i> , 2006, 18, S767-S776.	1.8	5
52	Importance of Quantum Tunneling in Vacancy-Hydrogen Complexes in Diamond. <i>Physical Review Letters</i> , 2005, 95, 105502.	7.8	42
53	Quantum behaviour of hydrogen and muonium in vacancy-containing complexes in diamond. <i>Journal of Physics Condensed Matter</i> , 2004, 16, 8743-8751.	1.8	19