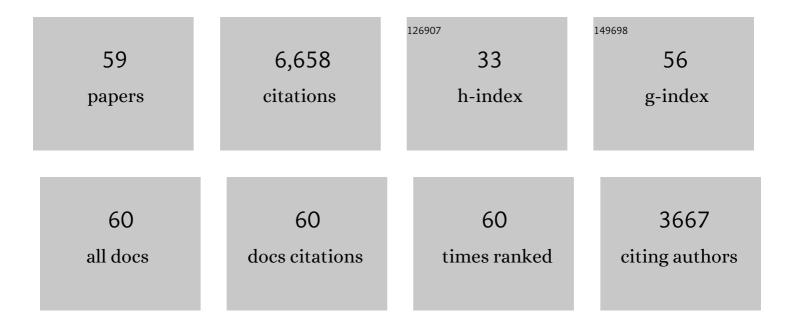
NaÃ-r RodrÃ-guez-Hornedo

List of Publications by Year in descending order

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#	Article	IF	CITATIONS
1	Gas-Assisted Cocrystal Desublimation. Crystal Growth and Design, 2022, 22, 1528-1532.	3.0	1
2	Synchronization of Cocrystal Dissolution and Drug Precipitation to Sustain Drug Supersaturation. Molecular Pharmaceutics, 2022, 19, 2765-2775.	4.6	7
3	Stability of Pharmaceutical Co-Crystals at Humid Conditions Can Be Predicted. Pharmaceutics, 2021, 13, 433.	4.5	11
4	Co-Crystal Screening by Vapor Sorption of Organic Solvents. Crystal Growth and Design, 2021, 21, 4445-4455.	3.0	4
5	Cocrystal Solubility Advantage and Dose/Solubility Ratio Diagrams: A Mechanistic Approach To Selecting Additives and Controlling Dissolution–Supersaturation–Precipitation Behavior. Molecular Pharmaceutics, 2020, 17, 4286-4301.	4.6	19
6	The role of pH and dose/solubility ratio on cocrystal dissolution, drug supersaturation and precipitation. European Journal of Pharmaceutical Sciences, 2020, 152, 105422.	4.0	30
7	Cocrystal Solubility Advantage Diagrams as a Means to Control Dissolution, Supersaturation, and Precipitation. Molecular Pharmaceutics, 2019, 16, 3887-3895.	4.6	35
8	Posaconazole Cocrystal with Superior Solubility and Dissolution Behavior. Crystal Growth and Design, 2019, 19, 6592-6602.	3.0	47
9	Exploring Bioequivalence of Dexketoprofen Trometamol Drug Products with the Gastrointestinal Simulator (GIS) and Precipitation Pathways Analyses. Pharmaceutics, 2019, 11, 122.	4.5	17
10	An Expandable Mechanopharmaceutical Device (1): Measuring the Cargo Capacity of Macrophages in a Living Organism. Pharmaceutical Research, 2019, 36, 12.	3.5	8
11	Mechanistic Analysis of Cocrystal Dissolution, Surface pH, and Dissolution Advantage as a Guide for Rational Selection. Journal of Pharmaceutical Sciences, 2019, 108, 243-251.	3.3	12
12	Cocrystals Mitigate Negative Effects of High pH on Solubility and Dissolution of a Basic Drug. Crystal Growth and Design, 2018, 18, 1358-1366.	3.0	42
13	Evaluation and optimized selection of supersaturating drug delivery systems of posaconazole (BCS) Tj ETQq1 1 (Journal of Pharmaceutical Sciences, 2018, 115, 258-269.).784314 r 4.0	gBT /Overloc 43
14	Mechanistic Basis of Cocrystal Dissolution Advantage. Journal of Pharmaceutical Sciences, 2018, 107, 380-389.	3.3	17
15	Understanding the Differences Between Cocrystal and Salt Aqueous Solubilities. Journal of Pharmaceutical Sciences, 2018, 107, 113-120.	3.3	34
16	Linking the Gastrointestinal Behavior of Ibuprofen with the Systemic Exposure between and within Humans—Part 1: Fasted State Conditions. Molecular Pharmaceutics, 2018, 15, 5454-5467.	4.6	21
17	Tadalafil–Malonic Acid Cocrystal: Physicochemical Characterization, pH-Solubility, and Supersaturation Studies. Crystal Growth and Design, 2018, 18, 4378-4387.	3.0	31
18	Multidrug Cocrystal of Anticonvulsants: Influence of Strong Intermolecular Interactions on Physiochemical Properties. Crystal Growth and Design, 2017, 17, 5012-5016.	3.0	58

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19	Cocrystals to facilitate delivery of poorly soluble compounds beyond-rule-of-5. Advanced Drug Delivery Reviews, 2016, 101, 143-166.	13.7	160
20	Turning Liquid Propofol into Solid (without Freezing It): Thermodynamic Characterization of Pharmaceutical Cocrystals Built with a Liquid Drug. Crystal Growth and Design, 2016, 16, 6547-6555.	3.0	20
21	Mechanistic Analysis of Cocrystal Dissolution as a Function of pH and Micellar Solubilization. Molecular Pharmaceutics, 2016, 13, 1030-1046.	4.6	36
22	Cocrystal Solubilization in Biorelevant Media and its Prediction from Drug Solubilization. Journal of Pharmaceutical Sciences, 2015, 104, 4153-4163.	3.3	21
23	Cocrystal Transition Points: Role of Cocrystal Solubility, Drug Solubility, and Solubilizing Agents. Molecular Pharmaceutics, 2015, 12, 3535-3546.	4.6	47
24	Pharmaceutical cocrystals and poorly soluble drugs. International Journal of Pharmaceutics, 2013, 453, 101-125.	5.2	501
25	Tailoring aqueous solubility of a highly soluble compound via cocrystallization: effect of coformer ionization, pHmax and solute–solvent interactions. CrystEngComm, 2012, 14, 4801.	2.6	71
26	pH-Dependent Solubility of Indomethacin–Saccharin and Carbamazepine–Saccharin Cocrystals in Aqueous Media. Molecular Pharmaceutics, 2012, 9, 2605-2612.	4.6	97
27	Polymorphs, Salts, and Cocrystals: What's in a Name?. Crystal Growth and Design, 2012, 12, 2147-2152.	3.0	767
28	Correction for Polymorphs, Salts and Cocrystals: What's in a Name?. Crystal Growth and Design, 2012, 12, 4290-4291.	3.0	17
29	Dependence of cocrystal formation and thermodynamic stability on moisture sorption by amorphous polymer. CrystEngComm, 2011, 13, 1181-1189.	2.6	29
30	Engineering cocrystal thermodynamic stability and eutectic points by micellar solubilization and ionization. CrystEngComm, 2011, 13, 5409.	2.6	32
31	Engineering cocrystal solubility, stability, and pHmax by micellar solubilization. Journal of Pharmaceutical Sciences, 2011, 100, 5219-5234.	3.3	42
32	Transformation Pathways of Cocrystal Hydrates When Coformer Modulates Water Activity. Journal of Pharmaceutical Sciences, 2010, 99, 3977-3985.	3.3	37
33	Cocrystal Eutectic Constants and Prediction of Solubility Behavior. Crystal Growth and Design, 2010, 10, 1028-1032.	3.0	117
34	Effect of Micellar Solubilization on Cocrystal Solubility and Stability. Crystal Growth and Design, 2010, 10, 2050-2053.	3.0	52
35	Solvent Effects on the Crystallization and Preferential Nucleation of Carbamazepine Anhydrous Polymorphs: A Molecular Recognition Perspective. Organic Process Research and Development, 2009, 13, 1291-1300.	2.7	41
36	Factors that influence the spontaneous formation of pharmaceutical cocrystals by simply mixing solid reactants. CrystEngComm, 2009, 11, 493-500.	2.6	70

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37	Role of Cocrystal and Solution Chemistry on the Formation and Stability of Cocrystals with Different Stoichiometry. Crystal Growth and Design, 2009, 9, 889-897.	3.0	148
38	Cocrystals and Salts of Gabapentin: pH Dependent Cocrystal Stability and Solubility. Crystal Growth and Design, 2009, 9, 378-385.	3.0	164
39	Analysis of 50 Crystal Structures Containing Carbamazepine Using the <i>Materials</i> Module of <i>Mercury CSD</i> . Crystal Growth and Design, 2009, 9, 1869-1888.	3.0	161
40	Solubility Advantage of Pharmaceutical Cocrystals. Crystal Growth and Design, 2009, 9, 2252-2264.	3.0	709
41	Understanding and Predicting the Effect of Cocrystal Components and pH on Cocrystal Solubility. Crystal Growth and Design, 2009, 9, 3976-3988.	3.0	147
42	A rapid thermal method for cocrystal screening. CrystEngComm, 2008, 10, 665.	2.6	259
43	Screening strategies based on solubility and solution composition generate pharmaceutically acceptable cocrystals of carbamazepine. CrystEngComm, 2008, 10, 856.	2.6	325
44	pH-Induced Nanosegregation of Ritonavir to Lyotropic Liquid Crystal of Higher Solubility than Crystalline Polymorphs. Molecular Pharmaceutics, 2008, 5, 956-967.	4.6	13
45	Solvent Systems for Crystallization and Polymorph Selection. , 2007, , 53-109.		12
46	Mechanisms by Which Moisture Generates Cocrystals. Molecular Pharmaceutics, 2007, 4, 360-372.	4.6	115
47	Cocrystals:  Molecular Design of Pharmaceutical Materials. Molecular Pharmaceutics, 2007, 4, 299-300.	4.6	61
48	Phase Solubility Diagrams of Cocrystals Are Explained by Solubility Product and Solution Complexation. Crystal Growth and Design, 2006, 6, 592-600.	3.0	278
49	Reaction Crystallization of Pharmaceutical Molecular Complexes. Molecular Pharmaceutics, 2006, 3, 362-367.	4.6	263
50	Cocrystal Formation during Cogrinding and Storage is Mediated by Amorphous Phase. Pharmaceutical Research, 2006, 23, 2381-2392.	3.5	215
51	General principles of pharmaceutical solid polymorphism A supramolecular perspective. Advanced Drug Delivery Reviews, 2004, 56, 241-274.	13.7	373
52	General Principles of Pharmaceutical Solid Polymorphism. A Supramolecular Perspective. ChemInform, 2004, 35, no.	0.0	0
53	Fourier transform infrared spectroscopy for the analysis of neutralizer-carbomer and surfactant-carbomer interactions in aqueous, hydroalcoholic, and anhydrous gel formulations. AAPS Journal, 2004, 6, 61-67.	4.4	26
54	Crystal Engineering of the Composition of Pharmaceutical Phases:  Multiple-Component Crystalline Solids Involving Carbamazepine. Crystal Growth and Design, 2003, 3, 909-919.	3.0	493

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55	Effect of initial buffer composition on pH changes during far-from-equilibrium freezing of sodium phosphate buffer solutions. Pharmaceutical Research, 2001, 18, 90-97.	3.5	143
56	Growth and morphology of L-alanine crystals: influence of additive adsorption. Pharmaceutical Research, 1993, 10, 1008-1014.	3.5	24
57	NUCLEATION AND CRYSTAL GROWTH EFFECTS ON PARTICLE CHARACTERISTICS. Particulate Science and Technology, 1992, 10, 33-35.	2.1	Ο
58	Phase transition and heterogeneous/epitaxial nucleation of hydrated and anhydrous theophylline crystals. International Journal of Pharmaceutics, 1992, 85, 149-162.	5.2	109
59	Crystal growth kinetics of theophylline monohydrate. Pharmaceutical Research, 1991, 08, 643-648.	3.5	23