

NaÃ-r RodrÃ-guez-Hornedo

List of Publications by Year in descending order

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59
papers

6,658
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126907

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60
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docs citations

60
times ranked

3667
citing authors

#	ARTICLE	IF	CITATIONS
1	Polymorphs, Salts, and Cocrystals: What's in a Name?. <i>Crystal Growth and Design</i> , 2012, 12, 2147-2152.	3.0	767
2	Solubility Advantage of Pharmaceutical Cocrystals. <i>Crystal Growth and Design</i> , 2009, 9, 2252-2264.	3.0	709
3	Pharmaceutical cocrystals and poorly soluble drugs. <i>International Journal of Pharmaceutics</i> , 2013, 453, 101-125.	5.2	501
4	Crystal Engineering of the Composition of Pharmaceutical Phases: Multiple-Component Crystalline Solids Involving Carbamazepine. <i>Crystal Growth and Design</i> , 2003, 3, 909-919.	3.0	493
5	General principles of pharmaceutical solid polymorphism A supramolecular perspective. <i>Advanced Drug Delivery Reviews</i> , 2004, 56, 241-274.	13.7	373
6	Screening strategies based on solubility and solution composition generate pharmaceutically acceptable cocrystals of carbamazepine. <i>CrystEngComm</i> , 2008, 10, 856.	2.6	325
7	Phase Solubility Diagrams of Cocrystals Are Explained by Solubility Product and Solution Complexation. <i>Crystal Growth and Design</i> , 2006, 6, 592-600.	3.0	278
8	Reaction Crystallization of Pharmaceutical Molecular Complexes. <i>Molecular Pharmaceutics</i> , 2006, 3, 362-367.	4.6	263
9	A rapid thermal method for cocrystal screening. <i>CrystEngComm</i> , 2008, 10, 665.	2.6	259
10	Cocrystal Formation during Cogrounding and Storage is Mediated by Amorphous Phase. <i>Pharmaceutical Research</i> , 2006, 23, 2381-2392.	3.5	215
11	Cocrystals and Salts of Gabapentin: pH Dependent Cocrystal Stability and Solubility. <i>Crystal Growth and Design</i> , 2009, 9, 378-385.	3.0	164
12	Analysis of 50 Crystal Structures Containing Carbamazepine Using the <i>Materials</i> Module of <i>Mercury CSD</i> . <i>Crystal Growth and Design</i> , 2009, 9, 1869-1888.	3.0	161
13	Cocrystals to facilitate delivery of poorly soluble compounds beyond-rule-of-5. <i>Advanced Drug Delivery Reviews</i> , 2016, 101, 143-166.	13.7	160
14	Role of Cocrystal and Solution Chemistry on the Formation and Stability of Cocrystals with Different Stoichiometry. <i>Crystal Growth and Design</i> , 2009, 9, 889-897.	3.0	148
15	Understanding and Predicting the Effect of Cocrystal Components and pH on Cocrystal Solubility. <i>Crystal Growth and Design</i> , 2009, 9, 3976-3988.	3.0	147
16	Effect of initial buffer composition on pH changes during far-from-equilibrium freezing of sodium phosphate buffer solutions. <i>Pharmaceutical Research</i> , 2001, 18, 90-97.	3.5	143
17	Cocrystal Eutectic Constants and Prediction of Solubility Behavior. <i>Crystal Growth and Design</i> , 2010, 10, 1028-1032.	3.0	117
18	Mechanisms by Which Moisture Generates Cocrystals. <i>Molecular Pharmaceutics</i> , 2007, 4, 360-372.	4.6	115

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19	Phase transition and heterogeneous/epitaxial nucleation of hydrated and anhydrous theophylline crystals. <i>International Journal of Pharmaceutics</i> , 1992, 85, 149-162.	5.2	109
20	pH-Dependent Solubility of Indomethacin-Saccharin and Carbamazepine-Saccharin Cocrystals in Aqueous Media. <i>Molecular Pharmaceutics</i> , 2012, 9, 2605-2612.	4.6	97
21	Tailoring aqueous solubility of a highly soluble compound via cocrystallization: effect of coformer ionization, pH _{max} and solute-solvent interactions. <i>CrystEngComm</i> , 2012, 14, 4801.	2.6	71
22	Factors that influence the spontaneous formation of pharmaceutical cocrystals by simply mixing solid reactants. <i>CrystEngComm</i> , 2009, 11, 493-500.	2.6	70
23	Cocrystals: Molecular Design of Pharmaceutical Materials. <i>Molecular Pharmaceutics</i> , 2007, 4, 299-300.	4.6	61
24	Multidrug Cocrystal of Anticonvulsants: Influence of Strong Intermolecular Interactions on Physiochemical Properties. <i>Crystal Growth and Design</i> , 2017, 17, 5012-5016.	3.0	58
25	Effect of Micellar Solubilization on Cocrystal Solubility and Stability. <i>Crystal Growth and Design</i> , 2010, 10, 2050-2053.	3.0	52
26	Cocrystal Transition Points: Role of Cocrystal Solubility, Drug Solubility, and Solubilizing Agents. <i>Molecular Pharmaceutics</i> , 2015, 12, 3535-3546.	4.6	47
27	Posaconazole Cocrystal with Superior Solubility and Dissolution Behavior. <i>Crystal Growth and Design</i> , 2019, 19, 6592-6602.	3.0	47
28	Evaluation and optimized selection of supersaturating drug delivery systems of posaconazole (BCS) Tj ETQq0 0 0 rgBT /Overlock 10 Tf 5 <i>Journal of Pharmaceutical Sciences</i> , 2018, 115, 258-269.	4.0	43
29	Engineering cocrystal solubility, stability, and pH _{max} by micellar solubilization. <i>Journal of Pharmaceutical Sciences</i> , 2011, 100, 5219-5234.	3.3	42
30	Cocrystals Mitigate Negative Effects of High pH on Solubility and Dissolution of a Basic Drug. <i>Crystal Growth and Design</i> , 2018, 18, 1358-1366.	3.0	42
31	Solvent Effects on the Crystallization and Preferential Nucleation of Carbamazepine Anhydrous Polymorphs: A Molecular Recognition Perspective. <i>Organic Process Research and Development</i> , 2009, 13, 1291-1300.	2.7	41
32	Transformation Pathways of Cocrystal Hydrates When Coformer Modulates Water Activity. <i>Journal of Pharmaceutical Sciences</i> , 2010, 99, 3977-3985.	3.3	37
33	Mechanistic Analysis of Cocrystal Dissolution as a Function of pH and Micellar Solubilization. <i>Molecular Pharmaceutics</i> , 2016, 13, 1030-1046.	4.6	36
34	Cocrystal Solubility Advantage Diagrams as a Means to Control Dissolution, Supersaturation, and Precipitation. <i>Molecular Pharmaceutics</i> , 2019, 16, 3887-3895.	4.6	35
35	Understanding the Differences Between Cocrystal and Salt Aqueous Solubilities. <i>Journal of Pharmaceutical Sciences</i> , 2018, 107, 113-120.	3.3	34
36	Engineering cocrystal thermodynamic stability and eutectic points by micellar solubilization and ionization. <i>CrystEngComm</i> , 2011, 13, 5409.	2.6	32

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37	Tadalafilâ€“Malonic Acid Cocrystal: Physicochemical Characterization, pH-Solubility, and Supersaturation Studies. <i>Crystal Growth and Design</i> , 2018, 18, 4378-4387.	3.0	31
38	The role of pH and dose/solubility ratio on cocrystal dissolution, drug supersaturation and precipitation. <i>European Journal of Pharmaceutical Sciences</i> , 2020, 152, 105422.	4.0	30
39	Dependence of cocrystal formation and thermodynamic stability on moisture sorption by amorphous polymer. <i>CrystEngComm</i> , 2011, 13, 1181-1189.	2.6	29
40	Fourier transform infrared spectroscopy for the analysis of neutralizer-carbomer and surfactant-carbomer interactions in aqueous, hydroalcoholic, and anhydrous gel formulations. <i>AAPS Journal</i> , 2004, 6, 61-67.	4.4	26
41	Growth and morphology of L-alanine crystals: influence of additive adsorption. <i>Pharmaceutical Research</i> , 1993, 10, 1008-1014.	3.5	24
42	Crystal growth kinetics of theophylline monohydrate. <i>Pharmaceutical Research</i> , 1991, 08, 643-648.	3.5	23
43	Cocrystal Solubilization in Biorelevant Media and its Prediction from Drug Solubilization. <i>Journal of Pharmaceutical Sciences</i> , 2015, 104, 4153-4163.	3.3	21
44	Linking the Gastrointestinal Behavior of Ibuprofen with the Systemic Exposure between and within Humansâ€“Part 1: Fasted State Conditions. <i>Molecular Pharmaceutics</i> , 2018, 15, 5454-5467.	4.6	21
45	Turning Liquid Propofol into Solid (without Freezing It): Thermodynamic Characterization of Pharmaceutical Cocrystals Built with a Liquid Drug. <i>Crystal Growth and Design</i> , 2016, 16, 6547-6555.	3.0	20
46	Cocrystal Solubility Advantage and Dose/Solubility Ratio Diagrams: A Mechanistic Approach To Selecting Additives and Controlling Dissolutionâ€“Supersaturationâ€“Precipitation Behavior. <i>Molecular Pharmaceutics</i> , 2020, 17, 4286-4301.	4.6	19
47	Correction for Polymorphs, Salts and Cocrystals: Whatâ€™s in a Name?. <i>Crystal Growth and Design</i> , 2012, 12, 4290-4291.	3.0	17
48	Mechanistic Basis of Cocrystal Dissolution Advantage. <i>Journal of Pharmaceutical Sciences</i> , 2018, 107, 380-389.	3.3	17
49	Exploring Bioequivalence of Dexketoprofen Trometamol Drug Products with the Gastrointestinal Simulator (GIS) and Precipitation Pathways Analyses. <i>Pharmaceutics</i> , 2019, 11, 122.	4.5	17
50	pH-Induced Nanosegregation of Ritonavir to Lyotropic Liquid Crystal of Higher Solubility than Crystalline Polymorphs. <i>Molecular Pharmaceutics</i> , 2008, 5, 956-967.	4.6	13
51	Solvent Systems for Crystallization and Polymorph Selection. , 2007, , 53-109.		12
52	Mechanistic Analysis of Cocrystal Dissolution, Surface pH, and Dissolution Advantage as a Guide for Rational Selection. <i>Journal of Pharmaceutical Sciences</i> , 2019, 108, 243-251.	3.3	12
53	Stability of Pharmaceutical Co-Crystals at Humid Conditions Can Be Predicted. <i>Pharmaceutics</i> , 2021, 13, 433.	4.5	11
54	An Expandable Mechanopharmaceutical Device (1): Measuring the Cargo Capacity of Macrophages in a Living Organism. <i>Pharmaceutical Research</i> , 2019, 36, 12.	3.5	8

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55	Synchronization of Cocrystal Dissolution and Drug Precipitation to Sustain Drug Supersaturation. <i>Molecular Pharmaceutics</i> , 2022, 19, 2765-2775.	4.6	7
56	Co-Crystal Screening by Vapor Sorption of Organic Solvents. <i>Crystal Growth and Design</i> , 2021, 21, 4445-4455.	3.0	4
57	Gas-Assisted Cocrystal Desublimation. <i>Crystal Growth and Design</i> , 2022, 22, 1528-1532.	3.0	1
58	NUCLEATION AND CRYSTAL GROWTH EFFECTS ON PARTICLE CHARACTERISTICS. <i>Particulate Science and Technology</i> , 1992, 10, 33-35.	2.1	0
59	General Principles of Pharmaceutical Solid Polymorphism. A Supramolecular Perspective. <i>ChemInform</i> , 2004, 35, no.	0.0	0