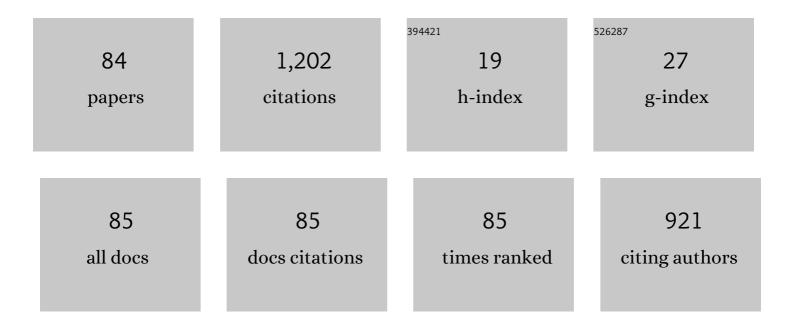
## List of Publications by Year in descending order

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SENLIAO

#	Article	IF	CITATIONS
1	Thermal degradation kinetics study of curcumin with nonlinear methods. Food Chemistry, 2014, 155, 81-86.	8.2	90
2	Self-Assembly of Luminescent Hexanuclear Lanthanide Salen Complexes. Crystal Growth and Design, 2012, 12, 970-974.	3.0	71
3	Near-UV light excited Eu3+, Tb3+, Bi3+ co-doped LaPO4 phosphors: Synthesis and enhancement of red emission for WLEDs. Ceramics International, 2015, 41, 5525-5530.	4.8	36
4	Nonisothermal Kinetic Study: IV. Comparative Methods To Evaluate <i>E</i> <sub>a</sub> for Thermal Decomposition of KZn <sub>2</sub> (PO <sub>4</sub> )(HPO <sub>4</sub> ) Synthesized by a Simple Route. Industrial & Engineering Chemistry Research, 2012, 51, 8985-8991.	3.7	32
5	Grinding kinetics of waste glass powder and its composite effect as pozzolanic admixture in cement concrete. Construction and Building Materials, 2020, 239, 117876.	7.2	32
6	Application of isoconversional calculation procedure to non-isothermal kinetic study: III. Thermal decomposition of ammonium cobalt phosphate hydrate. Thermochimica Acta, 2012, 543, 205-210.	2.7	28
7	High luminescent thermal stability and water resistance of K2SiF6:Mn4+@CaF2 red emitting phosphor. Ceramics International, 2021, 47, 33172-33179.	4.8	26
8	Nonisothermal Kinetics Study with Isoconversional Procedure and DAEM: LiCoPO <sub>4</sub> Synthesized from Thermal Decomposition of the Precursor. Industrial & Engineering Chemistry Research, 2013, 52, 1870-1876.	3.7	25
9	Na+ induced electric-dipole dominated transition (5D0→7F2) of Eu3+ emission in AMgPO4:Eu3+ (A=Li+,) Tj E	[Qq110.7	84314 rgBT
10	Preparation of nanocrystalline BiFeO3 via a simple and novel method and its kinetics of crystallization. Journal of Thermal Analysis and Calorimetry, 2012, 107, 625-632.	3.6	24
11	Novel emission bands of Na2TiF6:Mn4+ phosphors induced by the cation exchange method. Ceramics International, 2019, 45, 6243-6249.	4.8	24
12	Concentration and separation of vanadium from alkaline media by strong alkaline anion-exchange resin 717. Rare Metals, 2010, 29, 439-443.	7.1	23
13	Products and non-isothermal kinetics of thermal decomposition of MgFe2(C2O4)3·6H2O. Journal of Thermal Analysis and Calorimetry, 2012, 110, 781-787.	3.6	23
14	Facile synthesis of hydrotalcite and its thermal decomposition kinetics mechanism study with masterplots method. Thermochimica Acta, 2014, 579, 50-55.	2.7	22
15	Preparation of new sunscreen materials Ce1â^'x Zn x O2â^'x via solid-state reaction at room temperature and study on their properties. Rare Metals, 2010, 29, 149-153.	7.1	21
16	Kinetics and thermodynamics of thermal decomposition of NH4NiPO4·6H2O. Journal of Thermal Analysis and Calorimetry, 2011, 103, 805-812.	3.6	21
17	Preparation of LiZnPO4·H2O via a novel modified method and its non-isothermal kinetics and thermodynamics of thermal decomposition. Journal of Thermal Analysis and Calorimetry, 2012, 108, 1235-1242.	3.6	21
18	K2SiF6:Mn4+@K2SiF6 phosphor with remarkable negative thermal quenching and high water resistance for warm white LEDs. Journal of Luminescence, 2021, 234, 117968.	3.1	21

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19	The dual charge compensation effect of Na+ ions on the luminescence behavior of red phosphor NaMgPO4:Eu3+. Materials Letters, 2015, 160, 436-439.	2.6	20
20	Nanocrystalline Zn0.5Ni0.5Fe2O4. Journal of Thermal Analysis and Calorimetry, 2012, 110, 1143-1151.	3.6	19
21	Preparation of nanocrystalline LiMnPO4 via a simple and novel method and its isothermal kinetics of crystallization. Journal of Materials Science, 2011, 46, 2474-2478.	3.7	18
22	Preparation of LiZn0.9PO4:Mn0.1·H2O via a simple and novel method and its non-isothermal kinetics using iso-conversional calculation procedure. Thermochimica Acta, 2012, 533, 74-80.	2.7	18
23	LaPO4: Ce, Tb, Yb phosphor—synthesis and kinetics study for thermal process of precursor by Vyazovkin, OFW, KAS, Starink, and Mastplosts methods. Journal of Thermal Analysis and Calorimetry, 2015, 120, 1635-1643.	3.6	18
24	Novel luminescence enhancement and splitting of excitation and emission bands of Na2SiF6:Mn4+,Li+ phosphors induced by Li+ co-doping. Journal of Luminescence, 2020, 217, 116770.	3.1	18
25	Preparation of nano-sized cerium and titanium pyrophosphates via solid-state reaction at room temperature. Rare Metals, 2009, 28, 33-38.	7.1	17
26	Novel Method for Preparing NH <sub>4</sub> NiPO <sub>4</sub> ·6H <sub>2</sub> O: Hydrogen Bonding Coacervate Selective Selfâ€assembly. Chinese Journal of Chemistry, 2010, 28, 2389-2393.	4.9	17
27	Application of isoconversional calculation procedure to non-isothermal kinetics study. Journal of Thermal Analysis and Calorimetry, 2013, 111, 313-321.	3.6	17
28	A novel orange emissive phosphor LaPO4:Bi, Sm with sharp and splitting emission peaks of Sm3+. Materials Letters, 2014, 123, 112-115.	2.6	17
29	Cation exchange synthesis and cations doped effects of red-emitting phosphors K2TiF6:Mn4+, M2+ (M = Mg, Ca, Sr, Ba, and Zn). Journal of Materials Science: Materials in Electronics, 2017, 28, 11878-1185	8 <del>5</del> :2	17
30	Improvement of the luminescent thermal stability and water resistance of K2SiF6:Mn4+ by surface passivation. Ceramics International, 2022, 48, 17253-17260.	4.8	17
31	Selective self-assembly synthesis of MnV2O6·4H2O with controlled morphologies and study on its thermal decomposition. Journal of Thermal Analysis and Calorimetry, 2012, 109, 163-169.	3.6	16
32	Enhanced photoluminescence and energy transfer in the novel red emitting phosphors SrZn2(PO4)2:Eu3+, Tb3+, Li+. Journal of Materials Science: Materials in Electronics, 2017, 28, 657-660.	2.2	15
33	H2O2-free preparation of K2SiF6:Mn4+ and remarkable high luminescent thermal stability induced by coating with graphene quantum dots. Materials Chemistry and Physics, 2021, 260, 124149.	4.0	15
34	Preparation of Magnetic Cu0.5Mg0.5Fe2O4 Nanoparticles and Kinetics of Thermal Process of Precursor. Journal of Superconductivity and Novel Magnetism, 2012, 25, 1971-1977.	1.8	14
35	Preparation of magnetic nanocrystalline Mn0.5Mg0.5Fe2O4 and kinetics of thermal decomposition of precursor. Journal of Thermal Analysis and Calorimetry, 2013, 114, 205-212.	3.6	14
36	NaF induced enhancement of luminous efficiency in narrow-band red-emitting K2TiF6:Mn4+@NaF phosphors. Journal of Materials Science: Materials in Electronics, 2018, 29, 12536-12542.	2.2	14

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37	High Water Resistance and Luminescent Thermal Stability of Li <sub><i>y</i></sub> Na <sub>(2–<i>y</i>)</sub> SiF <sub>6</sub> : Mn <sup>4+</sup> Red-Emitting Phosphor Induced by Codoping of Li <sup>+</sup> . Inorganic Chemistry, 2022, 61, 5484-5494.	4.0	14
38	A Simple and Novel Route for The Preparation of Chiral Sodium Zincophosphate. Chinese Journal of Chemistry, 2008, 26, 281-285.	4.9	13
39	Synthesis and Regulation between NaH(ZnPO <sub>4</sub> ) <sub>2</sub> and <i>α</i> â€Hopeite via a Solid State Reaction at Lowâ€heating Temperatures. Chinese Journal of Chemistry, 2008, 26, 1837-1842.	4.9	13
40	Strain-Induced Enhancement of Eu3+ Emission in Red Phosphor NaMgPO4:Eu3+, Al3+. Journal of Electronic Materials, 2017, 46, 911-916.	2.2	13
41	Significant enhancement and broadening of excitation bands of K2SiF6:Mn4+, NH4+ phosphors induced by NH4+. Materials Research Bulletin, 2020, 121, 110622.	5.2	13
42	Effects of graphene quantum dots coating on the luminescence properties of K2SiF6:Mn4+ red-emitting phosphors. Journal of Materials Science: Materials in Electronics, 2020, 31, 444-456.	2.2	13
43	Enhancement of the luminescent thermal stability and water resistance of K2SiF6:Mn4+,Na+ by double coating of GQDs and K2SiF6. Journal of Alloys and Compounds, 2022, 898, 162819.	5.5	13
44	An organic–inorganic hybrid K <sub>2</sub> TiF <sub>6</sub> : Mn <sup>4+</sup> red-emitting phosp with remarkable improvement of emission and luminescent thermal stability. RSC Advances, 2022, 12, 3788-3795.	ohor 3.6	13
45	Magnetic Properties of Cu0.48Ni0.52Fe2O4 and Thermal Process of Precursor. Journal of Superconductivity and Novel Magnetism, 2013, 26, 2153-2158.	1.8	12
46	Synthesis, luminescence properties and nephelauxetic effect of nano stick phosphors K3AlF6:Mn4+ for warm white LED. Journal of Materials Science: Materials in Electronics, 2019, 30, 1870-1877.	2.2	12
47	Enhancement of the luminescent thermal stability and water resistance of K2TiF6:Mn4+ by organic inorganic hybrid matrix and surface passivation. Journal of Luminescence, 2022, 247, 118885.	3.1	12
48	Synthesis of Layered Sodium Manganese Phosphate via Lowâ€heating Solid State Reaction and Its Properties. Chinese Journal of Chemistry, 2010, 28, 2394-2398.	4.9	11
49	A facile surface passivation strategy for Na2SiF6:Mn4+,Li+ phosphors to achieve high moisture resistance and luminescent thermal stability. Journal of Luminescence, 2022, 243, 118643.	3.1	11
50	Magnetic properties and crystallization kinetics of Zn0.5Ni0.5Fe2O4. Rare Metals, 2011, 30, 621-626.	7.1	10
51	Improvement in luminescent properties and thermo-optical conversion mechanism of Na <sub>2</sub> SiF <sub>6</sub> :Mn <sup>4+</sup> ,K <sup>+</sup> @GQDs. RSC Advances, 2021, 11, 23023-23035.	3.6	10
52	The formation of KF induced red-emitting phosphors K 2 TiF 6 ·BaF(HF 2 ):Mn 4+ by cation exchange. Journal of Luminescence, 2017, 188, 307-312.	3.1	9
53	Non-isothermal kinetics of thermal decomposition of NH4ZrH(PO4)2·H2O. Journal of Thermal Analysis and Calorimetry, 2011, 104, 685-691.	3.6	8
54	Non-isothermal kinetics study with isoconversional procedure andÂDAEM: Thermal decomposition of LaPO4:Ce,Tb·0.5H2O. Materials Chemistry and Physics, 2013, 142, 453-458.	4.0	8

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55	Coating synthesis and enhanced photoluminescence of NaCl@LiLa(MoO4)2:Eu3+. Materials Letters, 2017, 208, 77-81.	2.6	8
56	Water resistance, thermal stability, luminescence enhancement of core-double shell structure K2TiF6:Mn4+ phosphor. Journal of Luminescence, 2022, 244, 118728.	3.1	8
57	Synthesis and regulation of α-LiZnPO4·H2O via a solid-state reaction at low-heating temperatures. Materials Research Bulletin, 2009, 44, 1428-1431.	5.2	7
58	Nanocrystalline LiMn2O4 preparation and kinetics of thermal process of precursor. Journal of Thermal Analysis and Calorimetry, 2013, 112, 1391-1399.	3.6	7
59	Magnetic Nanocrystalline Mg0.5Zn0.5Fe2O4: Preparation, Morphology Evolution, and Kinetics of Thermal Decomposition of Precursor. Journal of Superconductivity and Novel Magnetism, 2014, 27, 511-518.	1.8	7
60	Simple coating synthesis and enhanced luminescence behaviour of LiLa(MoO4)2:Eu3+@NaF. Materials Research Bulletin, 2018, 103, 181-185.	5.2	7
61	Conversion of thermal energy to light energy and energy transfer in KGdF4: Eu3+,Tb3+ phosphors. Inorganic Chemistry Communication, 2021, 127, 108549.	3.9	7
62	Selective Synthesis of a Hexagonal Co(II)‣ubstituted Sodium Zincophosphate via a Simple and Novel Route. Chinese Journal of Chemistry, 2010, 28, 50-54.	4.9	6
63	Synthesis of Spinel MnCo2O4 by Thermal Decomposition of Carbonates and Kinetics of Thermal Decomposition of Precursor. Journal of Superconductivity and Novel Magnetism, 2014, 27, 1249-1256.	1.8	6
64	Nonisothermal kinetics study with advanced isoconversional procedure and DAEM. Journal of Thermal Analysis and Calorimetry, 2014, 115, 237-245.	3.6	6
65	Thermochemical Study on the Chiral Sodium Zincophosphate Nanocrystalline. Chinese Journal of Chemistry, 2006, 24, 453-456.	4.9	5
66	Preparation of Ammonium Cerium Phosphate via Lowâ€heating Solid State Reaction and Its Catalysis for Benzyl Acetate Synthesis. Chinese Journal of Chemistry, 2010, 28, 378-382.	4.9	5
67	Nanocrystalline ZrO2 preparation and kinetics research of phase transition. Rare Metals, 2012, 31, 51-57.	7.1	5
68	Application of simplified version of advanced isoconversional procedure in non-isothermal kinetic study. Journal of Thermal Analysis and Calorimetry, 2013, 113, 649-657.	3.6	5
69	Synthesis of CeO2 by thermal decomposition of oxalate and kinetics of thermal decomposition of precursor. Journal of Thermal Analysis and Calorimetry, 2014, 117, 499-506.	3.6	5
70	Synthesis of a new phosphor (LaPO4:Ce, Li, Mn) and kinetics study for thermal process of its precursor. Advanced Powder Technology, 2015, 26, 861-867.	4.1	5
71	Enhancement of zero phonon line for Na2TiF6:Mn4+, Li+ phosphors induced by Li+. Journal of Materials Science: Materials in Electronics, 2019, 30, 14646-14656.	2.2	5
72	Synthesis of (La0.8Y0.2)PO4: Sm3+, Eu3+, Na+ and kinetics mechanism study with Z(α) master plots method for thermal process of its precursor. Journal of Thermal Analysis and Calorimetry, 2019, 136, 2487-2494.	3.6	5

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73	Eu <sup>3</sup> <sup>+</sup> , Sm <sup>3</sup> <sup>+</sup> Co-Doped La <sub>0.8</sub> Y <sub>0.2</sub> PO <sub>4</sub> : A Novel and Potential Red-Emitting Phosphor for UV-Based White Light-Emitting Diodes. Science of Advanced Materials, 2016, 8, 1093-1100.	0.7	4
74	Synthesis of Nano-Lamellar KZnPO <sub>4</sub> via Solid-State Reaction and its Data Mining Technology. Integrated Ferroelectrics, 2013, 147, 78-84.	0.7	3
75	Synthesis of Perovskite Pr1.1MnO3.15 and Phase Evolution and Magnetic Properties. Journal of Superconductivity and Novel Magnetism, 2014, 27, 2751-2756.	1.8	2
76	A novel energy transfer inducing strong enhancement of electric dipole transition in Na <sub>3</sub> Mo <sub>12</sub> PO <sub>40</sub> : <i>x</i> Eu <sup>3+</sup> phosphors. Materials Research Express, 2017, 4, 086305.	1.6	2
77	Synthesis of Gd(III)-MOF: Dy3+ phosphor and kinetics study of its thermal decomposition. Journal of Thermal Analysis and Calorimetry, 0, , 1.	3.6	2
78	Negative thermal quenching of K3AlF6:Mn4+@GQDs phosphors caused by enhancement of the conversion of heat energy into light energy. Journal of Materials Science: Materials in Electronics, 2021, 32, 26384-26396.	2.2	2
79	A new type of zero thermal quenching red emitting phosphor β-NaYF4:Eu3+ for NUV LEDs. Journal of Solid State Chemistry, 2022, , 123099.	2.9	2
80	Novel splitting of excitation and emission spectra of K2TiF6:Mn4+ phosphors induced by graphene quantum dots. Journal of Materials Science: Materials in Electronics, 2021, 32, 485-495.	2.2	1
81	Formation and enhancement of negative thermal quenching in emission of KGdF4:Eu3+, Yb3+@GQDs. RSC Advances, 2021, 11, 36222-36229.	3.6	1
82	Preparation and ultraviolet–visible ray transmission property of nanocrystalline InPO4. Journal of Thermal Analysis and Calorimetry, 2014, 115, 1705-1709.	3.6	0
83	Standard Molar Formation Enthalpy of NH4Fe(HPO4)2. Integrated Ferroelectrics, 2014, 154, 89-96.	0.7	0
84	Kinetics Study with Rigorous Nonlinear Methods for Thermal Decomposition of Polysaccharide Iron Complex. Food Biophysics, 2014, 9, 277-284.	3.0	0