

Danna E Freedman

List of Publications by Year in descending order

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71
papers

4,451
citations

117625

34
h-index

102487

66
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76
all docs

76
docs citations

76
times ranked

4341
citing authors

#	ARTICLE	IF	CITATIONS
1	Controlled nâ€Doping of Naphthaleneâ€Diimideâ€Based 2D Polymers. <i>Advanced Materials</i> , 2022, 34, e2101932.	21.0	13
2	Chemical control of spinâ€lattice relaxation to discover a room temperature molecular qubit. <i>Chemical Science</i> , 2022, 13, 7034-7045.	7.4	16
3	Synthesis of the Candidate Topological Compound Ni ₃ Pb ₂ . <i>Journal of the American Chemical Society</i> , 2022, 144, 11943-11948.	13.7	1
4	Dynamic Nuclear Polarization with Vanadium(IV) Metal Centers. <i>CheM</i> , 2021, 7, 421-435.	11.7	20
5	A Molecular Approach to Quantum Sensing. <i>ACS Central Science</i> , 2021, 7, 712-723.	11.3	52
6	Spectral Addressability in a Modular Two Qubit System. <i>Journal of the American Chemical Society</i> , 2021, 143, 8069-8077.	13.7	39
7	Computationally Directed Discovery of MoBi ₂ . <i>Journal of the American Chemical Society</i> , 2021, 143, 214-222.	13.7	17
8	Strong Magnetocrystalline Anisotropy Arising from Metalâ€Ligand Covalency in a Metalâ€Organic Candidate for 2D Magnetic Order. <i>Chemistry of Materials</i> , 2021, 33, 8712-8721.	6.7	8
9	Tunable Cr ⁴⁺ Molecular Color Centers. <i>Journal of the American Chemical Society</i> , 2021, 143, 21350-21363.	13.7	29
10	Taking titanium for a spin. <i>Nature Chemistry</i> , 2020, 12, 670-671.	13.6	1
11	Trigonal Bipyramidal V ³⁺ Complex as an Optically Addressable Molecular Qubit Candidate. <i>Journal of the American Chemical Society</i> , 2020, 142, 20400-20408.	13.7	46
12	Spin and Phonon Design in Modular Arrays of Molecular Qubits. <i>Chemistry of Materials</i> , 2020, 32, 10200-10206.	6.7	37
13	Optically addressable molecular spins for quantum information processing. <i>Science</i> , 2020, 370, 1309-1312.	12.6	148
14	Nickel(II) Metal Complexes as Optically Addressable Qubit Candidates. <i>Journal of the American Chemical Society</i> , 2020, 142, 14826-14830.	13.7	46
15	Pressure-Induced Collapse of Magnetic Order in Jarosite. <i>Physical Review Letters</i> , 2020, 125, 077202.	7.8	3
16	Orbital energy mismatch engenders high-spin ground states in heterobimetallic complexes. <i>Chemical Science</i> , 2020, 11, 9971-9977.	7.4	4
17	Fast and programmable locomotion of hydrogel-metal hybrids under light and magnetic fields. <i>Science Robotics</i> , 2020, 5, .	17.6	163
18	Synthetic investigation of competing magnetic interactions in 2D metalâ€chloranilate radical frameworks. <i>Chemical Science</i> , 2020, 11, 5922-5928.	7.4	13

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19	Control of the Porosity in Manganese Trimer-Based Metal-Organic Frameworks by Linker Functionalization. <i>Inorganic Chemistry</i> , 2020, 59, 8444-8450.	4.0	11
20	Exploiting chemistry and molecular systems for quantum information science. <i>Nature Reviews Chemistry</i> , 2020, 4, 490-504.	30.2	247
21	Magnetic Anisotropy in Heterobimetallic Complexes. <i>Inorganic Chemistry</i> , 2019, 58, 11893-11902.	4.0	19
22	A concentrated array of copper porphyrin candidate qubits. <i>Chemical Science</i> , 2019, 10, 1702-1708.	7.4	58
23	Metal-ligand covalency enables room temperature molecular qubit candidates. <i>Chemical Science</i> , 2019, 10, 6707-6714.	7.4	50
24	MnBi ₂ : A Metastable High-Pressure Phase in the Mn-Bi System. <i>Chemistry of Materials</i> , 2019, 31, 3083-3088.	6.7	6
25	Charting a course for chemistry. <i>Nature Chemistry</i> , 2019, 11, 286-294.	13.6	18
26	A chemical path to quantum information. <i>Science</i> , 2019, 366, 1070-1071.	12.6	20
27	Controlling Dimensionality in the Ni-Bi System with Pressure. <i>Chemistry of Materials</i> , 2019, 31, 955-959.	6.7	8
28	High-pressure synthesis of the perovskite. <i>Physical Review Materials</i> , 2019, 3, 041101.	2.4	7
29	High-pressure synthesis of the perovskite. <i>Physical Review Materials</i> , 2019, 3, 041101.	2.4	12
30	Progress towards creating optically addressable molecular qubits. <i>Chemical Communications</i> , 2018, 54, 13773-13781.	4.1	34
31	Size Determines Efficacy of Nanoparticle Magnetoresistance. <i>ACS Central Science</i> , 2018, 4, 1092-1094.	11.3	1
32	Discovery of Cu ₃ Pb. <i>Angewandte Chemie</i> , 2018, 130, 12991-12995.	2.0	3
33	Discovery of Cu ₃ Pb. <i>Angewandte Chemie - International Edition</i> , 2018, 57, 12809-12813.	13.8	7
34	High-Pressure Synthesis: A New Frontier in the Search for Next-Generation Intermetallic Compounds. <i>Accounts of Chemical Research</i> , 2018, 51, 1315-1323.	15.6	32
35	Impact of Pressure on Magnetic Order in Jarosite. <i>Journal of the American Chemical Society</i> , 2018, 140, 12001-12009.	13.7	9
36	Octacyanometallate qubit candidates. <i>Dalton Transactions</i> , 2018, 47, 11744-11748.	3.3	8

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37	Synthetic Approach To Determine the Effect of Nuclear Spin Distance on Electronic Spin Decoherence. <i>Journal of the American Chemical Society</i> , 2017, 139, 3196-3201.	13.7	72
38	Forging Solid-State Qubit Design Principles in a Molecular Furnace. <i>Chemistry of Materials</i> , 2017, 29, 1885-1897.	6.7	94
39	A Porous Array of Clock Qubits. <i>Journal of the American Chemical Society</i> , 2017, 139, 7089-7094.	13.7	86
40	Creating Binary Cu-Bi Compounds via High-Pressure Synthesis: A Combined Experimental and Theoretical Study. <i>Chemistry of Materials</i> , 2017, 29, 5276-5285.	6.7	39
41	High-pressure discovery of $\text{I}^2\text{-NiBi}$. <i>Chemical Communications</i> , 2017, 53, 11241-11244.	4.1	11
42	Probing Nuclear Spin Effects on Electronic Spin Coherence via EPR Measurements of Vanadium(IV) Complexes. <i>Inorganic Chemistry</i> , 2017, 56, 8106-8113.	4.0	37
43	Magnetic Anisotropy from Main-Group Elements: Halides versus Group 14 Elements. <i>Inorganic Chemistry</i> , 2017, 56, 8195-8202.	4.0	19
44	Discovery of a Superconducting Cu-Bi Intermetallic Compound by High-Pressure Synthesis. <i>Angewandte Chemie - International Edition</i> , 2016, 55, 13446-13449.	13.8	46
45	Transformation of the coordination complex $[\text{Co}(\text{C}_3\text{S}_5)_2]^{2+}$ from a molecular magnet to a potential qubit. <i>Chemical Science</i> , 2016, 7, 6160-6166.	7.4	40
46	Unexpected suppression of spin-lattice relaxation via high magnetic field in a high-spin iron(III) complex. <i>Chemical Communications</i> , 2016, 52, 10175-10178.	4.1	14
47	Enhancement of magnetic anisotropy in a Mn-Bi heterobimetallic complex. <i>Chemical Communications</i> , 2016, 52, 11394-11397.	4.1	13
48	Magnetic transitions in the topological magnon insulator $\text{Cu}(1,3\text{-bdc})$. <i>Physical Review B</i> , 2016, 93, .	3.2	25
49	Discovery of a Superconducting Cu-Bi Intermetallic Compound by High-Pressure Synthesis. <i>Angewandte Chemie</i> , 2016, 128, 13644-13647.	2.0	14
50	Discovery of FeBi_2 . <i>ACS Central Science</i> , 2016, 2, 867-871.	11.3	35
51	Long Coherence Times in Nuclear Spin-Free Vanadyl Qubits. <i>Journal of the American Chemical Society</i> , 2016, 138, 14678-14685.	13.7	118
52	A flexible iron(II) complex in which zero-field splitting is resistant to structural variation. <i>Chemical Science</i> , 2016, 7, 416-423.	7.4	28
53	Employing Forbidden Transitions as Qubits in a Nuclear Spin-Free Chromium Complex. <i>Journal of the American Chemical Society</i> , 2016, 138, 1344-1348.	13.7	74
54	Topological Magnon Bands in a Kagome Lattice Ferromagnet. <i>Physical Review Letters</i> , 2015, 115, 147201.	7.8	289

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55	Qubit Control Limited by Spin-Orbit Lattice Relaxation in a Nuclear Spin-Free Iron(III) Complex. <i>Inorganic Chemistry</i> , 2015, 54, 12027-12031.	4.0	38
56	Millisecond Coherence Time in a Tunable Molecular Electronic Spin Qubit. <i>ACS Central Science</i> , 2015, 1, 488-492.	11.3	296
57	(BiSe) _{1.23} CrSe ₂ and (BiSe) _{1.22} (Cr _{1.2} Se ₂) ₂ : Magnetic Anisotropy in the First Structurally Characterized Bi-Se-Cr Ternary Compounds. <i>Inorganic Chemistry</i> , 2015, 54, 2765-2771.	4.0	14
58	Influence of Electronic Spin and Spin-Orbit Coupling on Decoherence in Mononuclear Transition Metal Complexes. <i>Journal of the American Chemical Society</i> , 2014, 136, 7623-7626.	13.7	120
59	Multiple Quantum Coherences from Hyperfine Transitions in a Vanadium(IV) Complex. <i>Journal of the American Chemical Society</i> , 2014, 136, 15841-15844.	13.7	81
60	A Mononuclear Transition Metal Single-Molecule Magnet in a Nuclear Spin-Free Ligand Environment. <i>Inorganic Chemistry</i> , 2014, 53, 10716-10721.	4.0	132
61	Frustrated magnetism in the S = 1 kagom� lattice BaNi ₃ (OH) ₂ (VO ₄) ₂ . <i>Chemical Communications</i> , 2012, 48, 64-66.	4.1	53
62	Dinitrogen binding at vanadium in a tris(alkoxide) ligand environment. <i>Chemical Communications</i> , 2011, 47, 10242.	4.1	38
63	Slow Magnetic Relaxation and Charge-Transfer in Cyano-Bridged Coordination Clusters Incorporating [Re(CN) ₇] ³⁻ . <i>Inorganic Chemistry</i> , 2010, 49, 8886-8896.	4.0	72
64	Slow Magnetic Relaxation in a Family of Trigonal Pyramidal Iron(II) Pyrrolide Complexes. <i>Journal of the American Chemical Society</i> , 2010, 132, 18115-18126.	13.7	317
65	Site Specific X-ray Anomalous Dispersion of the Geometrically Frustrated Kagom� Magnet, Herbertsmithite, ZnCu ₃ (OH) ₆ Cl ₂ . <i>Journal of the American Chemical Society</i> , 2010, 132, 16185-16190.	13.7	166
66	A Cu ₂ (S = 1/2) Kagom� Antiferromagnet: MgxCu ₄ x(OH) ₆ Cl ₂ . <i>Journal of the American Chemical Society</i> , 2010, 132, 5570-5571.	13.7	36
67	Slow Magnetic Relaxation in a High-Spin Iron(II) Complex. <i>Journal of the American Chemical Society</i> , 2010, 132, 1224-1225.	13.7	457
68	Strong magnetic exchange coupling in the cyano-bridged coordination clusters [(PY ₅ Me ₂) ₄ V ₄ M(CN) ₆] ₅₊ (M = Cr, Mo). <i>Chemical Communications</i> , 2009, , 4829.	4.1	44
69	A Redox-Switchable Single-Molecule Magnet Incorporating [Re(CN) ₇] ³⁻ . <i>Journal of the American Chemical Society</i> , 2008, 130, 2884-2885.	13.7	235
70	Symmetry-breaking substitutions of [Re(CN) ₈] ³⁻ into the centered, face-capped octahedral clusters (CH ₃ OH) ₂₄ M ₉ M ₆ (CN) ₄₈ (M = Mn, Co; M ²⁺ = Mo, W). <i>Dalton Transactions</i> , 2006, , 2829-2834.	3.3	91
71	2,3-Dihydroxy-N-methylbenzamide monohydrate. <i>Acta Crystallographica Section E: Structure Reports Online</i> , 2004, 60, o1296-o1298.	0.2	1