

Jingtao Hou

List of Publications by Year in descending order

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Version: 2024-02-01

22
papers

1,110
citations

623188

14
h-index

713013

21
g-index

22
all docs

22
docs citations

22
times ranked

1201
citing authors

#	ARTICLE	IF	CITATIONS
1	Effect of giant oxygen vacancy defects on the catalytic oxidation of OMS-2 nanorods. <i>Journal of Materials Chemistry A</i> , 2013, 1, 6736.	5.2	256
2	Tuning the K ⁺ Concentration in the Tunnel of OMS-2 Nanorods Leads to a Significant Enhancement of the Catalytic Activity for Benzene Oxidation. <i>Environmental Science & Technology</i> , 2013, 47, 13730-13736.	4.6	198
3	Tremendous Effect of the Morphology of Birnessite-Type Manganese Oxide Nanostructures on Catalytic Activity. <i>ACS Applied Materials & Interfaces</i> , 2014, 6, 14981-14987.	4.0	175
4	As(III) adsorption on Fe-Mn binary oxides: Are Fe and Mn oxides synergistic or antagonistic for arsenic removal?. <i>Chemical Engineering Journal</i> , 2020, 389, 124470.	6.6	98
5	The effect of Ce ion substituted OMS-2 nanostructure in catalytic activity for benzene oxidation. <i>Nanoscale</i> , 2014, 6, 15048-15058.	2.8	62
6	The remarkable effect of the coexisting arsenite and arsenate species ratios on arsenic removal by manganese oxide. <i>Chemical Engineering Journal</i> , 2017, 315, 159-166.	6.6	58
7	Tremendous effect of oxygen vacancy defects on the oxidation of arsenite to arsenate on cryptomelane-type manganese oxide. <i>Chemical Engineering Journal</i> , 2016, 306, 597-606.	6.6	43
8	Morphology-dependent enhancement of arsenite oxidation to arsenate on birnessite-type manganese oxide. <i>Chemical Engineering Journal</i> , 2017, 327, 235-243.	6.6	38
9	Insights into the underlying mechanisms of stability working for As(III) removal by Fe-Mn binary oxide as a highly efficient adsorbent. <i>Water Research</i> , 2021, 203, 117558.	5.3	27
10	Al-substitution-induced defect sites enhance adsorption of Pb ²⁺ on hematite. <i>Environmental Science: Nano</i> , 2019, 6, 1323-1331.	2.2	26
11	Phosphate speciation on Al-substituted goethite: ATR-FTIR/2D-COS and CD-MUSIC modeling. <i>Environmental Science: Nano</i> , 2019, 6, 3625-3637.	2.2	25
12	The remarkable effect of alkali earth metal ion on the catalytic activity of OMS-2 for benzene oxidation. <i>Chemosphere</i> , 2020, 250, 126211.	4.2	19
13	Ce ion substitution position effect on catalytic activity of OMS-2 for benzene oxidation. <i>Materials Research Bulletin</i> , 2019, 118, 110497.	2.7	17
14	Enhanced catalytic activity of OMS-2 for carcinogenic benzene elimination by tuning Sr ²⁺ contents in the tunnels. <i>Journal of Hazardous Materials</i> , 2020, 398, 122958.	6.5	15
15	Enhanced oxidation of arsenite to arsenate using tunable K ⁺ concentration in the OMS-2 tunnel. <i>Environmental Pollution</i> , 2018, 238, 524-531.	3.7	11
16	Insights into the improving mechanism of defect-mediated As(V) adsorption on hematite nanoplates. <i>Chemosphere</i> , 2021, 280, 130597.	4.2	11
17	Formation and Morphology Evolution from Ferrihydrite to Hematite in the Presence of Tartaric Acid. <i>ACS Earth and Space Chemistry</i> , 2019, 3, 562-570.	1.2	9
18	Simultaneous introduction of K ⁺ and Rb ⁺ into OMS-2 tunnels as an available strategy for substantially increasing the catalytic activity for benzene elimination. <i>Environmental Research</i> , 2020, 191, 110146.	3.7	7

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19	Facet-dependent surface charge and Pb ²⁺ adsorption characteristics of hematite nanoparticles: CD-MUSIC-eSGC modeling. <i>Environmental Research</i> , 2021, 196, 110383.	3.7	6
20	Complexation mechanism of Pb ²⁺ at the ferrihydrite-water interface: The role of Al-substitution. <i>Chemosphere</i> , 2022, 307, 135627.	4.2	6
21	Insights into a "seesaw effect" between reducibility and hydrophobicity induced by cobalt doping: influence on OMS-2 nanomaterials for catalytic degradation of carcinogenic benzene. <i>Environmental Science: Nano</i> , 2021, 8, 3376-3386.	2.2	3
22	Peroxymonosulfate Improves the Activity and Stability of Manganese Oxide for Oxidation of Arsenite to Arsenate. <i>Clean - Soil, Air, Water</i> , 2020, 48, 1900195.	0.7	0