Alain Tressaud

List of Publications by Year in descending order

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54 1,731 21 39 g-index

87 87 87 87 2022

87 87 87 2022 all docs docs citations times ranked citing authors

#	Article	IF	CITATIONS
1	Access to Heteroleptic Fluorido yanido Complexes with a Large Magnetic Anisotropy by Fluoride Abstraction. Angewandte Chemie, 2020, 132, 10392-10396.	2.0	2
2	Access to Heteroleptic Fluorido yanido Complexes with a Large Magnetic Anisotropy by Fluoride Abstraction. Angewandte Chemie - International Edition, 2020, 59, 10306-10310.	13.8	6
3	Fluorinated carbonaceous nanoparticles as active material in primary lithium battery. Journal of Fluorine Chemistry, 2019, 219, 1-9.	1.7	9
4	Use of inorganic fluorinated materials in lithium batteries and in energy conversion systems. Chemical Communications, 2018, 54, 11375-11382.	4.1	28
5	[OsF ₆] ^{<i>x</i>23, 11244-11248.}	3.3	18
6	Iridates from the molecular side. Nature Communications, 2016, 7, 12195.	12.8	41
7	Long-term stability of fluorine–graphite intercalation compound prepared under high pressure of fluorine. Carbon, 2015, 82, 176-183.	10.3	8
8	Crystal Chemistry and Selected Physical Properties of Inorganic Fluorides and Oxide-Fluorides. Chemical Reviews, 2015, 115, 1191-1254.	47.7	135
9	Dual Lithium Insertion and Conversion Mechanisms in a Titanium-Based Mixed-Anion Nanocomposite. Journal of the American Chemical Society, 2011, 133, 13240-13243.	13.7	34
10	Structural architecture and physical properties of some inorganic fluoride series: a review. Journal of Fluorine Chemistry, 2011, 132, 651-659.	1.7	10
11	Surface modification of phyllosilicate minerals by fluorination methods. Journal of Vacuum Science and Technology A: Vacuum, Surfaces and Films, 2010, 28, 373-381.	2.1	9
12	Fluorination of alumino-silicate minerals: The example of lepidolite. Journal of Fluorine Chemistry, 2009, 130, 799-805.	1.7	13
13	Highly Fluorinated Silica Obtained by Direct F ₂ -Gas Fluorination: Stability and Unprecedented Fluorosilicate Species Revealed by Solid State NMR Investigations. Journal of Physical Chemistry C, 2009, 113, 18652-18660.	3.1	23
14	The Single Crystal Structures of the Copper Usovites Ba2CaCuV2F14and Ba2CaCuCr2F14. Zeitschrift Fur Anorganische Und Allgemeine Chemie, 2008, 634, 2737-2739.	1.2	5
15	Coupling Sol–Gel Synthesis and Microwaveâ€Assisted Techniques: A New Route from Amorphous to Crystalline High‧urfaceâ€Area Aluminium Fluoride. Chemistry - A European Journal, 2008, 14, 6205-6212.	3.3	34
16	Nanostructured Aluminium Hydroxyfluorides Derived from β-AlF ₃ . Chemistry of Materials, 2008, 20, 1459-1469.	6.7	64
17	Microwave Synthesis of an Aluminum Fluoride Hydrate with Cationic Vacancies: Structure, Thermal Stability, and Acidic Properties. Chemistry of Materials, 2008, 20, 7095-7106.	6.7	22
18	Nanostructured Al-based Fluoride-Oxide Materials with a Core-Shell Morphology. Journal of Physical Chemistry C, 2008, 112, 12374-12380.	3.1	19

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19	Optical Properties of the (CrF ₆) ^{3â^'} Complex in A ₂ BMF ₆ :Cr ³⁺ Elpasolite Crystals: Variation with Mâ^'F Bond Distance and Hydrostatic Pressure. Inorganic Chemistry, 2008, 47, 10288-10298.	4.0	36
20	Probing the Local Environments of Fluorine in Ce-Based Fluorite-Type Oxyfluorides with ¹⁹ F MAS NMR Spectroscopy. Journal of Physical Chemistry C, 2008, 112, 860-866.	3.1	16
21	Surface structure and electrochemical characteristics of plasma-fluorinated petroleum cokes for lithium ion battery. Journal of Power Sources, 2007, 168, 265-271.	7.8	20
22	Charge/discharge behavior of plasma-fluorinated natural graphites in propylene carbonate-containing solvent. Journal of Power Sources, 2007, 171, 932-937.	7.8	12
23	Rare Earth Oxyfluorosulfides:  A New Class of Compounds with Modulated Structure. Chemistry of Materials, 2006, 18, 6121-6131.	6.7	7
24	The low-temperature form of Rb2KCrF6 and Rb2KGaF6: The first example of an elpasolite-derived structure with pentagonal bipyramid in the B-sublattice. Journal of Solid State Chemistry, 2006, 179, 3607-3614.	2.9	29
25	Henri Moissan: Winner of the Nobel Prize for Chemistry 1906. Angewandte Chemie - International Edition, 2006, 45, 6792-6796.	13.8	19
26	Surface Modification of Carbonaceous Materials by Fluorinated Gases and Use as Anode in Lithium-Ion Battery. ECS Transactions, 2006, 3, 167-176.	0.5	2
27	Analysis of the Refractive Indices of TiO2, TiOF2, and TiF4: Concept of Optical Channel as a Guide to Understand and Design Optical Materials ChemInform, 2005, 36, no.	0.0	0
28	Evolution of the optical band gap in Titanium-based Oxy-(Hydroxy)-Fluorides series. Materials Research Society Symposia Proceedings, 2005, 891, 1.	0.1	2
29	Analysis of the Refractive Indices of TiO2, TiOF2, and TiF4:  Concept of Optical Channel as a Guide To Understand and Design Optical Materials. Inorganic Chemistry, 2005, 44, 3589-3593.	4.0	38
30	Palladium Chemistry in Anhydrous HF/AsF5 Superacid Medium. European Journal of Inorganic Chemistry, 2004, 2004, 1827-1834.	2.0	15
31	Palladium Chemistry in Anhydrous HF/AsF5 Superacid Medium ChemInform, 2004, 35, no.	0.0	0
32	Compressed Octahedral Coordination in Chain Compounds Containing Divalent Copper: Structure and Magnetic Properties of CuFAsF6 and CsCuAlF6. Chemistry - A European Journal, 2004, 10, 5052-5058.	3.3	27
33	Surface modification of several carbon-based materials: comparison between CF4 rf plasma and direct F2-gas fluorination routes. Journal of Fluorine Chemistry, 2004, 125, 1639-1648.	1.7	102
34	The dielectric function of LnSF rare-earth fluorosulfides (Ln=La, Ce): experiment and theory. Journal of Solid State Chemistry, 2004, 177, 2833-2840.	2.9	20
35	Infrared Study of the Surface Properties of HTB-Type Alâ^', Crâ^', Feâ^'Hydroxyfluorides. Journal of Physical Chemistry B, 2004, 108, 3246-3255.	2.6	44
36	Synthesis and characterization of Al3+, Cr3+, Fe3+ and Ga3+ hydroxyfluorides: correlations between structural features, thermal stability and acidic properties. Journal of Materials Chemistry, 2003, 13, 2330.	6.7	40

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37	Surface Structure And Electrochemical Characteristics Of Graphite Fluorinated By Elemental Fluorine And Plasma Treatment Using Cf 4. Molecular Crystals and Liquid Crystals, 2002, 388, 103-108.	0.9	15
38	Heat capacity and T–p phase diagram of Cs2NH4GaF6 elpasolite. Solid State Sciences, 2002, 4, 15-18.	3.2	8
39	Structural features of new rare earth-based mixed anions (O, S, F) compounds: relationships between optical absorption and rare earth environment. Solid State Sciences, 2002, 4, 1471-1479.	3.2	21
40	Electrochemical behavior of plasma-fluorinated graphite for lithium ion batteries. Journal of Power Sources, 2002, 104, 108-114.	7.8	85
41	Electrochemical properties and structures of surface-fluorinated graphite for the lithium ion secondary battery. Journal of Fluorine Chemistry, 2002, 114, 209-214.	1.7	43
42	Preparation, Magnetic Properties, and Pressure-Induced Transitions of Some MIIMIVF6 (MII=Ni, Pd, Cu;) Tj ETQq(0 0 <u>0 ,</u> gBT	/Oyerlock 10
43	Synthesis and characterization of boron-substituted carbons. Carbon, 2000, 38, 1461-1467.	10.3	212
44	Surface Modification of Inorganic Materials by Fluorination Treatments., 2000,, 437-492.		5
45	Supercritical fluid processing: a new route for materials synthesis. Journal of Materials Chemistry, 1999, 9, 67-75.	6.7	112
46	Synthesis, structure and polarized optical spectroscopy of two new fluoromanganese(III) complexes â€. Journal of the Chemical Society Dalton Transactions, 1997, , 4335-4340.	1.1	21
47	Cation Ordering in the Sillén X1-Type Oxychloride, BaBiO2Cl. Journal of Solid State Chemistry, 1995, 117, 201-205.	2.9	20
48	Aspects of Intercalation of Fluorine and Metal Fluorides into Graphite. Molecular Crystals and Liquid Crystals, 1994, 244, 13-28.	0.3	15
49	Influence of fluorine-gas treatment on the structural and physical properties of La2CuO4. Journal of the Less Common Metals, 1990, 164-165, 832-839.	0.8	19
50	New heteronuclear trimers in fluorides with usovite-type structure. Materials Research Bulletin, 1988, 23, 637-645.	5. 2	7
51	Crystal Chemistry of Fluorides. , 1985, , 77-203.		66
52	Ferro- and Ferrimagnetism in Fluorides. , 1985, , 371-394.		11
53	The tetrafluorides of iridium, rhodium and palladium. Journal of Inorganic and Nuclear Chemistry, 1976, 28, 23-28.	0.5	33
54	Crystal structure of rhodium pentafluoride. Inorganic Chemistry, 1973, 12, 2640-2644.	4.0	52