

Raffaella Biesuz

List of Publications by Year in descending order

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Version: 2024-02-01

81
papers

1,940
citations

236612

25
h-index

288905

40
g-index

83
all docs

83
docs citations

83
times ranked

1888
citing authors

#	ARTICLE	IF	CITATIONS
1	Towards intelligent packaging: BCP-EVOH@ optode for milk freshness measurement. <i>Talanta</i> , 2022, 241, 123230.	2.9	18
2	pH-Sensitive Sensors at Work on Poultry Meat Degradation Detection: From the Laboratory to the Supermarket Shelf. <i>AppliedChem</i> , 2022, 2, 128-141.	0.2	2
3	Screen-Printed Gold Electrode Functionalized with Deferoxamine for Iron(III) Detection. <i>Chemosensors</i> , 2022, 10, 214.	1.8	7
4	EVOH-Based pH-Sensitive Optode Array and Chemometrics: From Naked-Eye Analysis to Predictive Modeling to Detect Milk Freshness. <i>ACS Food Science & Technology</i> , 2021, 1, 819-828.	1.3	10
5	Current Trends in Polymer Based Sensors. <i>Chemosensors</i> , 2021, 9, 108.	1.8	37
6	Naked-Eye Food Freshness Detection: Innovative Polymeric Optode for High-Protein Food Spoilage Monitoring. <i>ACS Food Science & Technology</i> , 2021, 1, 165-175.	1.3	22
7	Gold and Silver Nanoparticle-Based Colorimetric Sensors: New Trends and Applications. <i>Chemosensors</i> , 2021, 9, 305.	1.8	49
8	Chelating Agents in Soil Remediation: A New Method for a Pragmatic Choice of the Right Chelator. <i>Frontiers in Chemistry</i> , 2020, 8, 597400.	1.8	21
9	Disposable and Low-Cost Colorimetric Sensors for Environmental Analysis. <i>International Journal of Environmental Research and Public Health</i> , 2020, 17, 8331.	1.2	47
10	DFO@EVOH and 3,4-HP@EVOH: Towards New Polymeric Sorbents for Iron(III). <i>Chemosensors</i> , 2020, 8, 111.	1.8	11
11	Development of a Dye-Based Device to Assess Poultry Meat Spoilage. Part I: Building and Testing the Sensitive Array. <i>Journal of Agricultural and Food Chemistry</i> , 2020, 68, 12702-12709.	2.4	16
12	Development of a Dye-Based Device to Assess the Poultry Meat Spoilage. Part II: Array on Act. <i>Journal of Agricultural and Food Chemistry</i> , 2020, 68, 12710-12718.	2.4	18
13	Colorimetric Sensor Array for Monitoring, Modelling and Comparing Spoilage Processes of Different Meat and Fish Foods. <i>Foods</i> , 2020, 9, 684.	1.9	44
14	Low-cost, disposable colourimetric sensors for metal ions detection. <i>Journal of Analytical Science and Technology</i> , 2020, 11, .	1.0	20
15	A portable, disposable, and low-cost optode for sulphide and thiol detection. <i>Analytical Methods</i> , 2019, 11, 4464-4470.	1.3	17
16	Sensing of Copper(II) by Immobilized Ligands: Comparison of Electrochemical and Surface Plasmon Resonance Transduction. <i>Proceedings (mdpi)</i> , 2019, 15, .	0.2	1
17	Inexpensive Alizarin Red S-based optical device for the simultaneous detection of Fe(III) and Al(III). <i>Microchemical Journal</i> , 2019, 149, 104036.	2.3	12
18	Simple solid-phase spectrophotometric method for free iron(III) determination. <i>Arabian Journal of Chemistry</i> , 2019, 12, 573-579.	2.3	10

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19	Salicylamide derivatives for iron and aluminium sequestration. From synthesis to complexation studies. <i>Journal of Trace Elements in Medicine and Biology</i> , 2018, 50, 580-588.	1.5	4
20	Metal complexation capacity of Antarctic lacustrine sediments. <i>Chemosphere</i> , 2018, 196, 402-408.	4.2	3
21	Unusual PLS application for Pd(II) sensing in extremely acidic solutions. <i>New Journal of Chemistry</i> , 2018, 42, 7901-7907.	1.4	5
22	Development of a sensor for trivalent iron: AHP fixed on mesoporous silica. <i>New Journal of Chemistry</i> , 2018, 42, 15237-15244.	1.4	8
23	Colorimetric detection, quantification and extraction of Fe(III) in water by acrylic polymers with pendant Kojic acid motifs. <i>Sensors and Actuators B: Chemical</i> , 2016, 233, 120-126.	4.0	17
24	Smart sensory materials for divalent cations: a dithizone immobilized membrane for optical analysis. <i>Analyst</i> , 2016, 141, 6140-6148.	1.7	14
25	A Speciation Study on the Perturbing Effects of Iron Chelators on the Homeostasis of Essential Metal Ions. <i>PLoS ONE</i> , 2015, 10, e0133050.	1.1	37
26	Deferoxamine paper for iron(III) and vanadium(V) sensing. <i>Chemical Papers</i> , 2015, 69, .	1.0	26
27	Geopolymers from low-T activated kaolin: Implications for the use of alunite-bearing raw materials. <i>Applied Clay Science</i> , 2015, 114, 530-539.	2.6	17
28	A Simple Small Size and Low Cost Sensor Based on Surface Plasmon Resonance for Selective Detection of Fe(III). <i>Sensors</i> , 2014, 14, 4657-4671.	2.1	51
29	Novel DFO-functionalized mesoporous silica for iron sensing. Part 2. Experimental detection of free iron concentration (pFe) in urine samples. <i>Analyst</i> , 2014, 139, 3940-3948.	1.7	22
30	Potentiometric Sensors Based on Molecular Imprinted Polymers. <i>Lecture Notes in Electrical Engineering</i> , 2014, , 141-144.	0.3	0
31	Determination of 10B in lymphoma human cells after boron carrier treatment: comparison of 10BPA and immuno-nanoparticles. <i>Chemical Papers</i> , 2014, 68, .	1.0	1
32	Novel DFO-SAM on mesoporous silica for iron sensing. Part I. Synthesis optimization and characterization of the material. <i>Analyst</i> , 2014, 139, 3932.	1.7	20
33	Adsorption of the Prototype Anionic Anthraquinone, Acid Blue 25, on a Modified Banana Peel: Comparison with Equilibrium and Kinetic Ligand Receptor Biochemical Data. <i>Industrial & Engineering Chemistry Research</i> , 2014, 53, 2251-2260.	1.8	15
34	Sorption of chrysoidine by row cork and cork entrapped in calcium alginate beads. <i>Arabian Journal of Chemistry</i> , 2014, 7, 133-138.	2.3	28
35	Iron(III) and aluminium(III) complexes with substituted salicyl-aldehydes and salicylic acids. <i>Journal of Inorganic Biochemistry</i> , 2013, 128, 174-182.	1.5	12
36	Biomass against emerging pollution in wastewater: Ability of cork for the removal of ofloxacin from aqueous solutions at different pH. <i>Journal of Environmental Chemical Engineering</i> , 2013, 1, 1199-1204.	3.3	35

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37	Supramolecular receptors in solid phase: developing sensors for anionic radionuclides. Dalton Transactions, 2013, 42, 6227.	1.6	17
38	Pb(II), Cu(II) and Cd(II) Removal through Untreated Rice Husk; Thermodynamics and Kinetics. Analytical Sciences, 2012, 28, 993-999.	0.8	5
39	Experimental design applied to the optimization of microwave-assisted DNA hydrolysis. Journal of Chromatography A, 2012, 1249, 8-16.	1.8	10
40	Cavity Effect on Perrhenate Recognition by Polyammonium Cages. European Journal of Inorganic Chemistry, 2012, 2012, 3410-3417.	1.0	35
41	Beyond the synthesis of novel solid phases: Review on modelling of sorption phenomena. Coordination Chemistry Reviews, 2012, 256, 28-45.	9.5	185
42	Emporeâ„¢ membrane vs. Chelex 100: Thermodynamic and kinetic studies on metals sorption. Reactive and Functional Polymers, 2011, 71, 588-598.	2.0	10
43	Nickel release from new conventional stainless steel, recycled, and nickel-free orthodontic brackets: An in vitro study. American Journal of Orthodontics and Dentofacial Orthopedics, 2010, 137, 809-815.	0.8	43
44	Optimum extraction process of polyphenols from Bridelia grandis stem bark using experimental design. Journal of Separation Science, 2010, 33, 1692-1697.	1.3	11
45	Solid phase extraction of copper(II) by fixed bed procedure on cation exchange complexing resins. Journal of Chromatography A, 2010, 1217, 1208-1218.	1.8	10
46	Chromium Release from New Stainless Steel, Recycled and Nickel-free Orthodontic Brackets. Angle Orthodontist, 2009, 79, 361-367.	1.1	46
47	Molecularly Imprinted Polymerâ€Based Sensors for Amperometric Determination of Nonelectroactive Substances. Electroanalysis, 2009, 21, 604-611.	1.5	27
48	Analytical methods for determination of free metal ion concentration, labile species fraction and metal complexation capacity of environmental waters: A review. Analytica Chimica Acta, 2009, 631, 129-141.	2.6	186
49	Usage of Emporeâ„¢ membrane in alcoholic media for copper(II) distribution studies. Talanta, 2009, 79, 603-612.	2.9	8
50	Sorption of Lead(II) on Two Chelating Resins: From the Exchange Coefficient to the Intrinsic Complexation Constant. Journal of Solution Chemistry, 2008, 37, 527-541.	0.6	16
51	Ion Exchange Complexing Resins as Sensors for the Determination of Free Metal Ion Concentration at a Low Level. Solvent Extraction and Ion Exchange, 2008, 26, 301-320.	0.8	5
52	Determination of the Total Concentration and Speciation of Metal Ions in River, Estuarine and Seawater Samples. Analytical Sciences, 2008, 24, 1605-1611.	0.8	6
53	Strong copper(II) species in estuarine and sea waters investigated by a method with high detection window. Talanta, 2007, 71, 706-714.	2.9	7
54	A comparison between the determination of free Pb(II) by two techniques: Absence of gradients and Nernstian equilibrium stripping and resin titration. Analytica Chimica Acta, 2007, 599, 41-50.	2.6	30

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55	A chelating resin as a probe for the copper(II) distribution in grape wines. <i>Reactive and Functional Polymers</i> , 2007, 67, 1083-1093.	2.0	15
56	Determination of cadmium(II), copper(II), manganese(II) and nickel(II) species in Antarctic seawater with complexing resins. <i>Marine Chemistry</i> , 2006, 101, 180-189.	0.9	25
57	Speciation of copper(II) in natural waters in the presence of ligands of high and intermediate strength. <i>Chemical Speciation and Bioavailability</i> , 2004, 16, 35-43.	2.0	8
58	Investigation of the complexation of metal-ions by strong ligands in fresh and marine water. <i>Environmental Science and Pollution Research</i> , 2003, 10, 317-320.	2.7	5
59	Characterization of the sorption of uranium(VI) on different complexing resins. <i>Analytical and Bioanalytical Chemistry</i> , 2003, 376, 1023-1029.	1.9	27
60	Separation of copper(II) and aluminium(III) from fresh waters by solid phase extraction on a complexing resin column. <i>Journal of Separation Science</i> , 2003, 26, 381-386.	1.3	17
61	Determination of the total concentration and speciation of Al(III) in tea infusions. <i>Journal of Inorganic Biochemistry</i> , 2003, 97, 79-88.	1.5	32
62	Investigation of the complexation properties of a natural water towards copper(II), manganese(II) and aluminium(III), based on sorption of metal ions on a complexing resin. <i>Polyhedron</i> , 2002, 21, 1343-1350.	1.0	15
63	Investigation on sorption equilibria of Mn(II), Cu(II) and Cd(II) on a carboxylic resin by the Gibbs-Donnan model. <i>Talanta</i> , 2001, 55, 541-550.	2.9	18
64	Estimation of Deprotonation Coefficients for Chelating Ion Exchange Resins. Comparison of Different Thermodynamic Model. <i>Journal of Physical Chemistry B</i> , 2001, 105, 4721-4726.	1.2	18
65	Evaluation of the sorption of metal ions on a complexing resin from different solutions based on the Gibbs-Donnan model. <i>Reactive and Functional Polymers</i> , 2001, 46, 233-246.	2.0	16
66	Investigation of the metal species in seawater by sorption of the metal ion on complexing resins with different sorbing properties. <i>Analytica Chimica Acta</i> , 2001, 449, 23-33.	2.6	20
67	Determination of metal ions concentration and speciation in seawater by titration with an iminodiacetic resin. <i>Analytica Chimica Acta</i> , 1999, 401, 265-276.	2.6	31
68	Investigation of the speciation of aluminium in drinking waters by sorption on a strong anionic-exchange resin AG1X8. <i>Analytica Chimica Acta</i> , 1998, 367, 215-222.	2.6	18
69	Characterization and applications of chelating resins as chemical reagents for metal ions, based on the Gibbs-Donnan model. <i>Reactive and Functional Polymers</i> , 1998, 36, 135-147.	2.0	38
70	Study of aluminium speciation in freshwaters by sorption on a chelating resin. <i>Analyst</i> , 1998, 123, 1295-1301.	1.7	26
71	Sorption of divalent metal ions on an iminodiacetic resin from artificial seawater. <i>Analytica Chimica Acta</i> , 1997, 346, 381-391.	2.6	34
72	Simultaneous Determination of Total and Free Metal Ion Concentration in Solution by Sorption on Iminodiacetate Resin. <i>Analytical Chemistry</i> , 1995, 67, 3558-3563.	3.2	41

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73	Sorption of metal ions on a weak acid cation-exchange resin containing carboxylic groups. <i>Analytica Chimica Acta</i> , 1994, 298, 225-232.	2.6	47
74	Sorption mechanism of trace amounts of divalent metal ions on a chelating resin containing iminodiacetate groups. <i>Analytical Chemistry</i> , 1993, 65, 2522-2527.	3.2	118
75	Sorption of copper, nickel and lanthanum on a strong-base anion exchanger containing Chromotrope 2B. <i>Reactive & Functional Polymers</i> , 1991, 14, 239-250.	0.8	18
76	Exchange of protons between some poly(amido-amine) resins and aqueous solutions: A thermodynamic interpretation. <i>Reactive & Functional Polymers</i> , 1989, 11, 37-45.	0.8	4
77	Determination of dissolved inorganic species of iodine by spectrophotometric titration. <i>Analyst, The</i> , 1987, 112, 1265.	1.7	2
78	Spectrophotometric determination of palladium(II) with four water-soluble heterocyclic azo dyes. <i>Analyst, The</i> , 1985, 110, 801.	1.7	6
79	Photometric titration of total iodine at trace levels in concentrated chloride solutions. <i>Analytica Chimica Acta</i> , 1984, 158, 143-146.	2.6	1
80	Role of Biogenic Amines in Protein Foods Sensing: Myths and Evidence. , 0, , .		4
81	Freshness <i>Traffic Light</i> for Fish Products: Dual-Optode Label to Monitor Fish Spoilage in Sales Packages. <i>ACS Food Science & Technology</i> , 0, , .	1.3	2