

# Valeri A Drebushchak

## List of Publications by Year in descending order

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104  
papers

1,912  
citations

279798  
23  
h-index

315739  
38  
g-index

105  
all docs

105  
docs citations

105  
times ranked

1684  
citing authors

#	ARTICLE	IF	CITATIONS
1	Polymorphism of glycine, Part I. Journal of Thermal Analysis and Calorimetry, 2003, 73, 409-418.	3.6	170
2	Polymorphism of glycine, Part II. Journal of Thermal Analysis and Calorimetry, 2003, 73, 419-428.	3.6	102
3	Calibration coefficient of a heat-flow DSC; Part II. Optimal calibration procedure. Journal of Thermal Analysis and Calorimetry, 2005, 79, 213-218.	3.6	97
4	DSC and adiabatic calorimetry study of the polymorphs of paracetamol. Journal of Thermal Analysis and Calorimetry, 2004, 77, 607-623.	3.6	91
5	The investigation of ancient pottery. Journal of Thermal Analysis and Calorimetry, 2005, 82, 617-626.	3.6	57
6	Advances in elucidating mechanochemical complexities via implementation of a simple organic system. Faraday Discussions, 2014, 170, 311-335.	3.2	47
7	Thermal expansion of solids: review on theories. Journal of Thermal Analysis and Calorimetry, 2020, 142, 1097-1113.	3.6	47
8	Low-temperature heat capacity of $\gamma$ -glycine and a phase transition at 252 K. Journal of Thermal Analysis and Calorimetry, 2005, 79, 65-70.	3.6	46
9	Low-temperature heat capacity of $\beta\pm$ and $\beta^3$ polymorphs of glycine. Journal of Thermal Analysis and Calorimetry, 2003, 74, 109-120.	3.6	44
10	Solid-state transformations in the $\beta^2$ -form of chlorpropamide on cooling to 100...K. Acta Crystallographica Section B: Structural Science, 2011, 67, 163-176.	1.8	42
11	Furosemide Solvates: Can They Serve As Precursors to Different Polymorphs of Furosemide?. Crystal Growth and Design, 2014, 14, 513-522.	3.0	38
12	Glycine phases formed from frozen aqueous solutions: Revisited. Journal of Chemical Physics, 2012, 137, 065103.	3.0	37
13	Transitions among five polymorphs of chlorpropamide near the melting point. Journal of Thermal Analysis and Calorimetry, 2008, 93, 343-351.	3.6	36
14	Thermoanalytical investigation of drugâ€“excipient interaction. Journal of Thermal Analysis and Calorimetry, 2006, 86, 303-309.	3.6	33
15	An Extended Phase Transition in Crystalline-l-Cysteine near 70 K. Journal of Physical Chemistry B, 2007, 111, 9186-9188.	2.6	33
16	The Peltier effect. Journal of Thermal Analysis and Calorimetry, 2008, 91, 311-315.	3.6	32
17	Recommendations on DSC calibration. Journal of Thermal Analysis and Calorimetry, 2016, 124, 951-958.	3.6	30
18	Synthesis and calorimetric investigation of stoichiometric Fe-spinels: MgFe <sub>2</sub> O <sub>4</sub> . Journal of Crystal Growth, 2004, 265, 165-167.	1.5	29

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19	Observation of subtle dynamic transitions by a combination of neutron scattering, X-ray diffraction and DSC: A case study of the monoclinic l-cysteine. <i>Biophysical Chemistry</i> , 2010, 148, 34-41.	2.8	29
20	Thermogravimetric investigation of ancient ceramics. <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 90, 73-79.	3.6	27
21	Low-temperature heat capacity of heulandite: comparison with clinoptilolite. <i>Thermochimica Acta</i> , 2000, 348, 33-40.	2.7	26
22	Calibration coefficient of a heat-flow DSC. Part I. Relation to the Sensitivity of a thermocouple. <i>Journal of Thermal Analysis and Calorimetry</i> , 2004, 76, 941-947.	3.6	26
23	Decreasing particle size helps to preserve metastable polymorphs. A case study of dl-cysteine. <i>CrystEngComm</i> , 2011, 13, 4417.	2.6	25
24	Low-temperature phase transition in glycine-“glutaric acid co-crystals studied by single-crystal X-ray diffraction, Raman spectroscopy and differential scanning calorimetry. <i>Acta Crystallographica Section B: Structural Science</i> , 2012, 68, 287-296.	1.8	25
25	Isoenergetic Polymorphism: The Puzzle of Tolazamide as a Case Study. <i>Chemistry - A European Journal</i> , 2015, 21, 15395-15404.	3.3	24
26	Thermal properties of the midinfrared nonlinear crystal LiInSe <sub>2</sub> . <i>Journal of Applied Physics</i> , 2004, 96, 3659-3665.	2.5	23
27	The mass-loss diagram for the ancient ceramics. <i>Journal of Thermal Analysis and Calorimetry</i> , 2011, 104, 459-466.	3.6	23
28	A new polymorph of metacetamol. <i>CrystEngComm</i> , 2015, 17, 6183-6192.	2.6	23
29	FT-IR and FT-Raman spectra of five polymorphs of chlorpropamide. Experimental study and ab initio calculations. <i>Journal of Molecular Structure</i> , 2008, 891, 75-86.	3.6	22
30	Single-crystal to single-crystal conformational polymorphic transformation in tolbutamide at 313 K. Relation to other polymorphic transformations in tolbutamide and chlorpropamide. <i>CrystEngComm</i> , 2016, 18, 5736-5743.	2.6	22
31	Heat capacity of D- and DL-serine in a temperature range of 5.5 to 300 K. <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 89, 649-654.	3.6	20
32	Thermodynamic and kinetic stability of inclusion compounds under heating. <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 90, 23-30.	3.6	20
33	Synthesis of pure pentlandite in bulk. <i>Journal of Crystal Growth</i> , 1998, 193, 728-731.	1.5	17
34	Experimental study of boron solubility and speciation in the Na <sub>2</sub> O-“B <sub>2</sub> O <sub>3</sub> -“SiO <sub>2</sub> -“H <sub>2</sub> O system. <i>Chemical Geology</i> , 2005, 223, 16-34.	3.3	17
35	Kinetic and thermodynamic stability of cluster compounds under heating. <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 88, 687-692.	3.6	17
36	Thermal and thermo-optic parameters of LiInSe <sub>2</sub> single crystals. <i>Journal of Crystal Growth</i> , 2005, 275, e1679-e1684.	1.5	16

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37	Thermoanalytical investigation of drug-excipient interaction. <i>Journal of Thermal Analysis and Calorimetry</i> , 2006, 84, 643-649.	3.6	16
38	Influence of mechanical treatment on the properties of betulin, betulin diacetate, and their mixture with water-soluble polymers. <i>Chemistry of Natural Compounds</i> , 2011, 47, 229-233.	0.8	16
39	Thermoanalytical investigations of ancient ceramics. <i>Journal of Thermal Analysis and Calorimetry</i> , 2018, 133, 135-176.	3.6	16
40	Ionic and molecular diffusion and the order-disorder phase transition in the thallium form of natrolite. <i>Journal of Structural Chemistry</i> , 1990, 31, 56-63.	1.0	15
41	Relationship Between Heat Capacity and Thermal Expansion Derived from the Lennard-Jones Potential. <i>Magyar Aprázavad Káplazeműnyek</i> , 2001, 65, 745-753.	1.4	15
42	Two mechanisms of thermal expansion in perovskite SrCo0.6Fe0.2Nb0.2O3-z. <i>Journal of Thermal Analysis and Calorimetry</i> , 2010, 100, 79-82.	3.6	15
43	Heat capacity of $\text{L}\pm\text{glycylglycine}$ in a temperature range of 6 to 440 K. <i>Journal of Thermal Analysis and Calorimetry</i> , 2006, 85, 485-490.	3.6	14
44	Dynamic Pseudo Jahn-Teller Effect and the Phase Transition Induced by Absorption of Molecules in Metal-Organic Nanotube Framework. <i>Journal of Physical Chemistry C</i> , 2008, 112, 5074-5077.	3.1	14
45	Phase transition at thermal dehydration in stilbite. <i>Journal of Thermal Analysis and Calorimetry</i> , 2012, 107, 1293-1299.	3.6	14
46	Polymorphic effects at the eutectic melting in the H <sub>2</sub> O-glycine system. <i>Journal of Thermal Analysis and Calorimetry</i> , 2013, 111, 2187-2194.	3.6	14
47	Measurements of Heat of Zeolite Dehydration by Scanning Heating. <i>Magyar Aprázavad Káplazeműnyek</i> , 1999, 58, 653-662.	1.4	13
48	Formation and properties of hydrosilicate liquids in the systems Na <sub>2</sub> O-Al <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> -H <sub>2</sub> O and granite-Na <sub>2</sub> O-SiO <sub>2</sub> -H <sub>2</sub> O at 600°C and 1.5 kbar. <i>Petrology</i> , 2014, 22, 293-309.	0.9	13
49	Experimental heat capacity of LiInS <sub>2</sub> , LiInSe <sub>2</sub> , LiGaS <sub>2</sub> , LiGaSe <sub>2</sub> , and LiGaTe <sub>2</sub> from 180 to 460 K. <i>Journal of Thermal Analysis and Calorimetry</i> , 2017, 129, 103-108.	3.6	13
50	Mechanochemical Synthesis of Nanocomposites of Drugs with Inorganic Oxides. <i>Materials and Manufacturing Processes</i> , 2009, 24, 1064-1071.	4.7	12
51	Physical properties and structure of bound water in collagen-type fibrillar proteins as studied by scanning calorimetry. <i>JETP Letters</i> , 2005, 82, 613-615.	1.4	11
52	Calibration coefficient of a heat flow DSC. <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 90, 289-298.	3.6	11
53	Thermophysical theory of DSC melting peak. <i>Journal of Thermal Analysis and Calorimetry</i> , 2012, 109, 545-553.	3.6	11
54	Natural specimen of triple solid solution ettringite-thaumasite-chromate-ettringite. <i>Journal of Thermal Analysis and Calorimetry</i> , 2013, 114, 777-783.	3.6	11

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55	Thermogravimetric investigation of the phase transition in the zeolite heulandite at dehydration. <i>Thermochimica Acta</i> , 1990, 159, 377-381.	2.7	10
56	CsLiB <sub>6</sub> O <sub>10</sub> crystals with Cs deficit: structure and properties. <i>Journal of Crystal Growth</i> , 2005, 282, 407-413.	1.5	10
57	Low-temperature heat capacity of magnesioferrite, MgFe <sub>2</sub> O <sub>4</sub> . <i>Journal of Thermal Analysis and Calorimetry</i> , 2008, 92, 717-721.	3.6	10
58	Universality of the emf of thermocouples. <i>Thermochimica Acta</i> , 2009, 496, 50-53.	2.7	10
59	Cooling rate “window” in the crystallization of metacetamol form II. <i>Journal of Thermal Analysis and Calorimetry</i> , 2017, 127, 1807-1814.	3.6	10
60	The stability of inclusion compounds under heating. <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 90, 463-467.	3.6	9
61	Low-temperature heat capacity of diglycylglycine. <i>Journal of Thermal Analysis and Calorimetry</i> , 2008, 93, 865-869.	3.6	9
62	Heat capacity of $\text{l}^2\text{-alanine}$ in a temperature range between 6 and 300 K. <i>Journal of Thermal Analysis and Calorimetry</i> , 2009, 98, 873-876.	3.6	9
63	Melting of orthorhombic betulin. <i>Journal of Thermal Analysis and Calorimetry</i> , 2013, 111, 2005-2008.	3.6	9
64	1A1 5T2 Spin transition in the solid phases of Fe <sub>x</sub> Ni <sub>1-x</sub> (ATr) <sub>3</sub> (NO <sub>3</sub> ) <sub>2</sub> (ATr = 4- amino- 1,2,4- triazole). <i>Journal of Structural Chemistry</i> , 1997, 38, 578-584.	1.0	8
65	Mechanism and modelling of formation of amorphous sulfur nuclei. <i>Mendeleev Communications</i> , 2003, 13, 37-38.	1.6	8
66	Low-temperature heat capacity of monoclinic enstatite. <i>Journal of Thermal Analysis and Calorimetry</i> , 2008, 94, 493-497.	3.6	8
67	Phase transition at 204–250 K in the crystals of $\text{l}^2\text{-alanine}$ : kinetically irreproduceable, or an artefact?. <i>Phase Transitions</i> , 2009, 82, 497-506.	1.3	8
68	The stability of inclusion compounds under heating. <i>Journal of Thermal Analysis and Calorimetry</i> , 2010, 100, 183-189.	3.6	8
69	New interpretation of heat effects in polymorphic transitions. <i>Journal of Thermal Analysis and Calorimetry</i> , 2013, 113, 419-424.	3.6	8
70	Decay of (Fe <sub>1-x</sub> Ni <sub>x</sub> ) <sub>0.96</sub> S DSC investigation. <i>Journal of Thermal Analysis</i> , 1997, 48, 727-734.	0.6	7
71	Changes in the heat capacity of Al <sub>2</sub> O <sub>3</sub> -Cr <sub>2</sub> O <sub>3</sub> solid solutions near the point of antiferromagnetic phase transition in Cr <sub>2</sub> O <sub>3</sub> . <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 90, 795-799.	3.6	7
72	Coefficients of thermal expansion of the potassium and rubidium halogenide plumbates. <i>Journal of Thermal Analysis and Calorimetry</i> , 2009, 95, 323-325.	3.6	6

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73	Heat capacity increases with pressure. <i>Journal of Thermal Analysis and Calorimetry</i> , 2009, 95, 313-317.	3.6	6
74	Cucurbituril-assisted transformation of nitronyl nitroxide into imino nitroxide in the solid state. <i>CrystEngComm</i> , 2011, 13, 3241.	2.6	6
75	Thermal transformations of the supramolecular compound of cucurbit[8]uril with cobalt(III) complex {trans-[Co(en)2Cl2]@CB[8]}Cl·17 H2O. <i>Journal of Thermal Analysis and Calorimetry</i> , 2011, 105, 103-106.	3.6	6
76	Thermal properties of betulin dipropionate and its mixtures with polymers. <i>Journal of Thermal Analysis and Calorimetry</i> , 2014, 115, 2521-2525.	3.6	6
77	Structural similarity and similarity in thermal properties of the polymorphs: melting and crystallization from the melt of tolbutamide and chlorpropamide. <i>Journal of Thermal Analysis and Calorimetry</i> , 2020, 142, 841-848.	3.6	6
78	Synthesis and characterization of sulfathiazole-pyridine solvate polymorphs. <i>Journal of Crystal Growth</i> , 2005, 274, 569-572.	1.5	5
79	Discrepancy in the low-temperature heat capacity of MgFe2O4 and related problems. <i>Journal of Thermal Analysis and Calorimetry</i> , 2014, 117, 443-446.	3.6	5
80	Probable metalâ€“insulator transition in Ag4SSe. <i>Journal of Alloys and Compounds</i> , 2015, 622, 236-242.	5.5	5
81	Refinement of NMR data on the structure of bound water in collagen using scanning calorimetry. <i>Journal of Structural Chemistry</i> , 2005, 46, 1131-1133.	1.0	4
82	Calorimetric search for the discontinuity in Fe0.96S-Ni0.96S solid solutions. <i>Journal of Thermal Analysis and Calorimetry</i> , 2007, 89, 303-307.	3.6	4
83	Synthesis and crystal structures of two polymorphs of sulfathiazole:pyridine (1:1) adducts. <i>Structural Chemistry</i> , 2007, 18, 449-456.	2.0	4
84	Approximation of the emf of a thermocouple. <i>Journal of Thermal Analysis and Calorimetry</i> , 2009, 96, 315-320.	3.6	4
85	Tuning of the nitronyl nitroxide radical magnetic and electronic properties by inclusion in cucurbit[n]urils. <i>Polyhedron</i> , 2011, 30, 3083-3087.	2.2	4
86	Concepts against mathematics: self-inconsistency in thermodynamic evaluations. <i>Journal of Thermal Analysis and Calorimetry</i> , 2011, 103, 753-759.	3.6	4
87	Thermocouples, their characteristic temperatures, and simple approximation of the emf vs. T. <i>Thermochimica Acta</i> , 2015, 603, 218-226.	2.7	4
88	Calorimetric measurements of sodium chloride dihydrate (hydrohalite). <i>Journal of Thermal Analysis and Calorimetry</i> , 2020, 140, 2555-2562.	3.6	4
89	Melting of PbBr2: A DSC investigation. <i>Magyar AprÃ³vad KÃ¶zlemÃ©nyek</i> , 1999, 57, 599-605.	1.4	3
90	Cation distribution in MgFe2O4 vs. pressure and temperature: Experiments in a "piston-cylinder" apparatus. <i>American Mineralogist</i> , 2005, 90, 764-767.	1.9	3

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91	Dechlorination of contaminated sediments of Ionian Sea. Journal of Thermal Analysis and Calorimetry, 2007, 90, 143-146.	3.6	3
92	From electrical analog to thermophysical modeling of DSC. Journal of Thermal Analysis and Calorimetry, 2011, 105, 495-500.	3.6	3
93	Microemulsion synthesis of powders of water-soluble energy-saturated salts. Russian Journal of Inorganic Chemistry, 2012, 57, 769-776.	1.3	3
94	Solid solutions in the MgO-Al <sub>2</sub> O <sub>3</sub> -Cr <sub>2</sub> O <sub>3</sub> system. Journal of Thermal Analysis and Calorimetry, 2009, 95, 81-86.	3.6	2
95	Model-free temperature scaling for heat capacity. Journal of Thermal Analysis and Calorimetry, 2017, 130, 5-13.	3.6	2
96	Title is missing!. Magyar AprÃ³vad KÃ¶zlemÃ©nyek, 1999, 56, 925-929.	1.4	1
97	Excess heat release during deuterium sorption-desorption by finely powdered palladium deuteride. Europhysics Letters, 2002, 58, 462-467.	2.0	1
98	Proton mobility in complex [RhL 4Cl <sub>2</sub> ]HSO <sub>4</sub> · nH <sub>2</sub> SO <sub>4</sub> · mH <sub>2</sub> O salts (L = Py, <sup>13</sup> -picoline). Russian Journal of Electrochemistry, 2011, 47, 631-636.	0.9	1
99	Response to the comments by Prof. Swendsen on "Concepts against mathematics: self-inconsistency in thermodynamic evaluations". Journal of Thermal Analysis and Calorimetry, 2012, 110, 1553-1554.	3.6	1
100	Fine structure of the thermoanalytical peak of the phase transition in quartz. Journal of Structural Chemistry, 1996, 37, 150-151.	1.0	0
101	Crystal-chemical analysis of the metal- to- sulfur ratio in ternary solid solutions of (Fe, Co, Ni) <sub>9</sub> S <sub>8</sub> (pentlandite). Journal of Structural Chemistry, 1997, 38, 682-684.	1.0	0
102	On the unusual arrangement of metal atoms in pentlandite. Journal of Structural Chemistry, 1998, 39, 791-793.	1.0	0
103	Response to the comments by Prof. Acree. Journal of Thermal Analysis and Calorimetry, 2014, 117, 1013-1014.	3.6	0
104	Comments on "Spectroscopic and thermographic study of Ni-Zn ferrites" by J.D. Baraliya and H.H. Joshi, J Therm Anal Calorim. 2015;119:85-90. Journal of Thermal Analysis and Calorimetry, 2015, 121, 543-544.	3.6	0