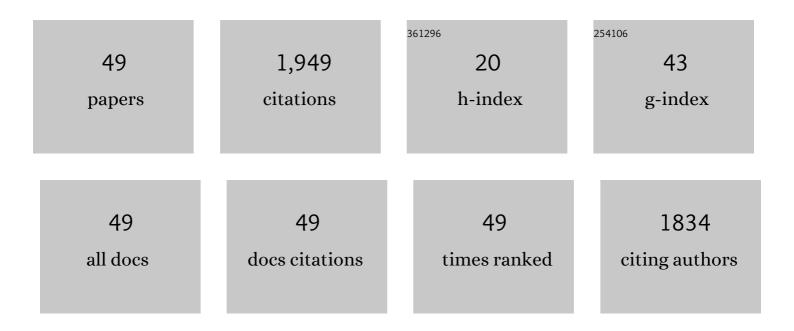
Qing-Hua Tian

List of Publications by Year in descending order

Source: https://exaly.com/author-pdf/332703/publications.pdf Version: 2024-02-01



Οινις-Ημα Τιανι

#	Article	IF	CITATIONS
1	Recovery of Cobalt from Secondary Resources: A Comprehensive Review. Mineral Processing and Extractive Metallurgy Review, 2022, 43, 679-700.	2.6	26
2	Copper and arsenic substance flow analysis of pyrometallurgical process for copper production. Transactions of Nonferrous Metals Society of China, 2022, 32, 364-376.	1.7	14
3	Recovery of gold from refractory gold ores: Effect of pyrite on the stability of the thiourea leaching system. International Journal of Minerals, Metallurgy and Materials, 2021, 28, 956-964.	2.4	9
4	Pyrometallurgical options for recycling spent lithium-ion batteries: A comprehensive review. Journal of Power Sources, 2021, 491, 229622.	4.0	399
5	Selective Extraction of Ni from Superalloy Scraps by Molten Mg-Zn. Metals, 2021, 11, 993.	1.0	2
6	Tungsten and arsenic substance flow analysis of a hydrometallurgical process for tungsten extracting from wolframite. Tungsten, 2021, 3, 348-360.	2.0	9
7	Dissolution behavior of nickel-based superalloy in molten zinc: Its mechanism and kinetics. Journal of Alloys and Compounds, 2021, 878, 160338.	2.8	3
8	Thermodynamic analysis and process optimization of zinc and lead recovery from copper smelting slag with chlorination roasting. Transactions of Nonferrous Metals Society of China, 2021, 31, 3905-3917.	1.7	14
9	Leaching kinetics of tellurium-bearing materials in alkaline sulfide solutions. Mineral Processing and Extractive Metallurgy Review, 2020, 41, 1-10.	2.6	12
10	Selective recovery of Sb and Te from the sodium sulfide leach solution of Te-bearing alkaline skimming slag by drop-wise H2O2 addition followed by Na2S–Na2SO3 precipitation. Hydrometallurgy, 2020, 191, 105219.	1.8	13
11	Comprehensive treatment of acid effluent containing antimony and arsenic by selective reduction and evaporative crystallization. Hydrometallurgy, 2020, 195, 105366.	1.8	8
12	Towards a circular metal additive manufacturing through recycling of materials: A mini review. Journal of Central South University, 2020, 27, 1134-1145.	1.2	16
13	Adsorption of Re(VII) by coated solvent-impregnated resins containing Alamine 304-1 from sulfuric acid solutions. Rare Metals, 2020, 39, 942-950.	3.6	4
14	Adsorption of Re(VII) from sulfuric acid solutions by coated impregnated resins containing TBP. Separation Science and Technology, 2020, 55, 3320-3328.	1.3	5
15	Optimization of Tellurium and Antimony Extraction from Residue Generated in Alkaline Sulfide Leaching of Tellurium-Bearing Alkaline Skimming Slag Using Central Composite Design. Mining, Metallurgy and Exploration, 2020, 37, 493-505.	0.4	2
16	A Novel Method for Making Co-Cr-Mo Alloy Spherical Powder by Granulation and Sintering. Jom, 2020, 72, 1279-1285.	0.9	9
17	Towards "zero waste―extraction of nickel from scrap nickel-based superalloy using magnesium. Journal of Cleaner Production, 2020, 262, 121275.	4.6	25
18	Stability constants of Sb5+ with Clâ^' and thermodynamics of Sbâ^'Sâ^'Clâ^'H2O system involving complex behavior of Sb with Cl. Transactions of Nonferrous Metals Society of China, 2020, 30, 3379-3389.	1.7	5

Qing-Hua Tian

#	Article	IF	CITATIONS
19	Review of the Effect of Oxygen on Titanium and Deoxygenation Technologies for Recycling of Titanium Metal. Jom, 2019, 71, 3209-3220.	0.9	30
20	Heavy metal redistribution mechanism assisted magnetic separation for highly-efficient removal of lead and cadmium from human blood. Journal of Colloid and Interface Science, 2019, 536, 563-574.	5.0	18
21	Absorption performance of DMSA modified Fe3O4@SiO2 core/shell magnetic nanocomposite for Pb2+ removal. Journal of Central South University, 2018, 25, 709-718.	1.2	16
22	Potential-controlled selective recovery of manganese and cobalt from cobalt slag leaching solution. Hydrometallurgy, 2017, 169, 201-206.	1.8	8
23	Nanostructured (Co, Mn)3O4 for High Capacitive Supercapacitor Applications. Nanoscale Research Letters, 2017, 12, 214.	3.1	22
24	Recovery of tellurium from high tellurium-bearing materials by alkaline sulfide leaching followed by sodium sulfite precipitation. Hydrometallurgy, 2017, 171, 355-361.	1.8	40
25	One pot controllable synthesis of AgCl nanocrystals with different morphology and their photocatalytic activity. Powder Technology, 2017, 308, 206-213.	2.1	15
26	Leaching kinetics of antimony-bearing complex sulfides ore in hydrochloric acid solution with ozone. Transactions of Nonferrous Metals Society of China, 2017, 27, 2073-2081.	1.7	10
27	Mineralogical characterization and pretreatment for antimony extraction by ozone of antimony-bearing refractory gold concentrates. Transactions of Nonferrous Metals Society of China, 2017, 27, 1888-1895.	1.7	28
28	Facile synthesis of Ag2Se quantum dots and their application in Dye/Ag2Se co-sensitized solar cells. Journal of Materials Science, 2017, 52, 12131-12140.	1.7	17
29	Application of pre-alloyed powders for diamond tools by ultrahigh pressure water atomization. Transactions of Nonferrous Metals Society of China, 2016, 26, 2665-2671.	1.7	19
30	Synthesis of size-controllable Fe3O4 magnetic submicroparticles and its biocompatible evaluation in vitro. Journal of Central South University, 2016, 23, 2784-2791.	1.2	7
31	Theoretical simulation and experimental study of hydrolysis separation of SbCl3 in complexation–precipitation system. Transactions of Nonferrous Metals Society of China, 2016, 26, 2746-2753.	1.7	21
32	Leaching behavior of metals from high-arsenic dust by NaOH–Na2S alkaline leaching. Transactions of Nonferrous Metals Society of China, 2016, 26, 575-580.	1.7	66
33	Ozonation leaching of a complex sulfidic antimony ore in hydrochloric acid solution. Hydrometallurgy, 2016, 159, 126-131.	1.8	39
34	Effect of selected parameters on stibnite concentrates leaching by ozone. Hydrometallurgy, 2016, 165, 295-299.	1.8	10
35	One-step synthesis of micro-sized hexagon silver sheets by the ascorbic acid reduction with the presence of H2SO4. Advanced Powder Technology, 2014, 25, 865-870.	2.0	14
36	Adsorption of Pb ²⁺ , Cu ²⁺ and Ni ²⁺ from aqueous solutions by novel garlic peel adsorbent. Desalination and Water Treatment, 2013, 51, 7166-7171.	1.0	33

Qing-Hua Tian

#	Article	IF	CITATIONS
37	Application of response surface methodology in optimizaing the sulfation-roasting-leaching process of nickel laterite. International Journal of Minerals, Metallurgy and Materials, 2012, 19, 199-204.	2.4	14
38	Extraction of valuable metals from manganese–silver ore. Hydrometallurgy, 2012, 119-120, 8-15.	1.8	36
39	Manufacturing microporous foam zinc materials with high porosity by electrodeposition. Journal Wuhan University of Technology, Materials Science Edition, 2011, 26, 843-846.	0.4	7
40	Adsorption of Pb2+ and Zn2+ from aqueous solutions by sulfured orange peel. Desalination, 2011, 275, 212-216.	4.0	84
41	Optionally ultra-fast synthesis of CoO/Co3O4 particles using CoCl2 solution via a versatile spray roasting method. Advanced Powder Technology, 2010, 21, 529-533.	2.0	26
42	Substance flow analysis of zinc in China. Resources, Conservation and Recycling, 2010, 54, 171-177.	5.3	55
43	Isotherms, kinetics and thermodynamic studies of adsorption of Cu2+ from aqueous solutions by Mg2+/K+ type orange peel adsorbents. Journal of Hazardous Materials, 2010, 174, 756-762.	6.5	287
44	Kinetics of oxidation-precipitation of cobalt(II) from solution by ozone. Transactions of Nonferrous Metals Society of China, 2010, 20, s42-s45.	1.7	13
45	Effective removal of heavy metals from aqueous solutions by orange peel xanthate. Transactions of Nonferrous Metals Society of China, 2010, 20, s187-s191.	1.7	75
46	Electroless copper plating on microcellular polyurethane foam. Transactions of Nonferrous Metals Society of China, 2010, 20, s283-s287.	1.7	30
47	Leaching behavior of metals from a limonitic nickel laterite using a sulfation–roasting–leaching process. Hydrometallurgy, 2009, 99, 144-150.	1.8	90
48	Application of orange peel xanthate for the adsorption of Pb2+ from aqueous solutions. Journal of Hazardous Materials, 2009, 170, 425-429.	6.5	143
49	Adsorption of Cu2+ and Cd2+ from aqueous solution by mercapto-acetic acid modified orange peel. Colloids and Surfaces B: Biointerfaces, 2009, 73, 10-14.	2.5	101