

# Nicolas Finck

## List of Publications by Year in descending order

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36  
papers

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687363

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677142

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times ranked

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#	ARTICLE	IF	CITATIONS
1	First Principle Investigation of the Incorporation of Trivalent Lanthanides and Actinides in Hydroxycarbonate and Hydroxychloride Green Rust. <i>Journal of Physical Chemistry C</i> , 2022, 126, 8016-8028.	3.1	2
2	Retention of Iodide and Chloride by Formation of a Green Rust Solid Solution GR-ClI: A Multiscale Approach. <i>Inorganic Chemistry</i> , 2021, 60, 10585-10595.	4.0	3
3	Unexpected behavior of sodium sulfate observed in experimental freezing and corrosion studies. <i>Journal of Raman Spectroscopy</i> , 2021, 52, 1499-1506.	2.5	1
4	Adsorption of Strontium onto Synthetic Iron(III) Oxide up to High Ionic Strength Systems. <i>Minerals (Basel, Switzerland)</i> , 2021, 11, 1093.	2.0	7
5	Reactive Transport Modelling of the Long-Term Interaction between Carbon Steel and MX-80 Bentonite at 25 °C. <i>Minerals (Basel, Switzerland)</i> , 2021, 11, 1272.	2.0	11
6	Sorption of americium / europium onto magnetite under saline conditions: Batch experiments, surface complexation modelling and X-ray absorption spectroscopy study. <i>Journal of Colloid and Interface Science</i> , 2020, 561, 708-718.	9.4	11
7	Iron speciation in Opalinus clay minerals. <i>Applied Clay Science</i> , 2020, 193, 105679.	5.2	1
8	Adsorption of arsenic(V) onto single sheet iron oxide: X-ray absorption fine structure and surface complexation. <i>Journal of Colloid and Interface Science</i> , 2019, 554, 433-443.	9.4	20
9	Structural iron in smectites with different charge locations. <i>Physics and Chemistry of Minerals</i> , 2019, 46, 639-661.	0.8	8
10	Treatment of temperature dependence of interfacial speciation by speciation codes and temperature congruence of oxide surface charge. <i>Applied Geochemistry</i> , 2019, 102, 26-33.	3.0	9
11	Fate of Lu(III) sorbed on 2-line ferrihydrite at pH 5.7 and aged for 12 years at room temperature. I: insights from ICP-OES, XRD, ESEM, AsFFFF/ICP-MS, and EXAFS spectroscopy. <i>Environmental Science and Pollution Research</i> , 2019, 26, 5238-5250.	5.3	4
12	Fate of Lu(III) sorbed on 2-line ferrihydrite at pH 5.7 and aged for 12 years at room temperature. II: insights from STEM-EDXS and DFT calculations. <i>Environmental Science and Pollution Research</i> , 2019, 26, 5282-5293.	5.3	4
13	Uranium Redox Transformations after U(VI) Coprecipitation with Magnetite Nanoparticles. <i>Environmental Science &amp; Technology</i> , 2017, 51, 2217-2225.	10.0	112
14	Tetrahedral charge and Fe content in dioctahedral smectites. <i>Clay Minerals</i> , 2017, 52, 51-65.	0.6	15
15	Adsorption of Selenium and Strontium on Goethite: EXAFS Study and Surface Complexation Modeling of the Ternary Systems. <i>Environmental Science &amp; Technology</i> , 2017, 51, 3751-3758.	10.0	62
16	Yttrium co-precipitation with smectite: A polarized XAS and AsFFFF study. <i>Applied Clay Science</i> , 2017, 137, 11-21.	5.2	9
17	XAS signatures of Am(III) adsorbed onto magnetite and maghemite. <i>Journal of Physics: Conference Series</i> , 2016, 712, 012085.	0.4	7
18	Trivalent Actinide Uptake by Iron (Hydr)oxides. <i>Environmental Science &amp; Technology</i> , 2016, 50, 10428-10436.	10.0	15

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19	Aqueous U(VI) interaction with magnetite nanoparticles in a mixed flow reactor system: HR-XANES study. <i>Journal of Physics: Conference Series</i> , 2016, 712, 012086.	0.4	8
20	Interaction of selenite with reduced Fe and/or S species: An XRD and XAS study. <i>Journal of Contaminant Hydrology</i> , 2016, 188, 44-51.	3.3	9
21	Macroscopic and spectroscopic investigations on Eu(III) and Cm(III) sorption onto bayerite ( $\hat{I}^2\text{-Al}(\text{OH})_3$ ) and corundum ( $\hat{I}^\pm\text{-Al}_2\text{O}_3$ ). <i>Journal of Colloid and Interface Science</i> , 2016, 461, 215-224.	9.4	30
22	Structural iron in dioctahedral and trioctahedral smectites: a polarized XAS study. <i>Physics and Chemistry of Minerals</i> , 2015, 42, 847-859.	0.8	16
23	Synthetic Smectite Colloids: Characterization of Nanoparticles after Co-Precipitation in the Presence of Lanthanides and Tetravalent Elements (Zr, Th). <i>Chromatography (Basel)</i> , 2015, 2, 545-566.	1.2	2
24	Am(III) coprecipitation with and adsorption on the smectite hectorite. <i>Chemical Geology</i> , 2015, 409, 12-19.	3.3	15
25	Fluorescence X-ray Absorption Study of $\text{ScCl}_3$ -Doped Sodium Alanate. <i>Journal of Physical Chemistry C</i> , 2015, 119, 15810-15815.	3.1	3
26	Montmorillonite colloids: I. Characterization and stability of dispersions with different size fractions. <i>Applied Clay Science</i> , 2015, 114, 179-189.	5.2	26
27	Characterization and radionuclide retention properties of heat-treated concrete. <i>Physics and Chemistry of the Earth</i> , 2014, 70-71, 45-52.	2.9	5
28	Characterization of Eu(III) co-precipitated with and adsorbed on hectorite: from macroscopic crystallites to nanoparticles. <i>Mineralogical Magazine</i> , 2012, 76, 2723-2740.	1.4	7
29	Selenide Retention by Mackinawite. <i>Environmental Science &amp; Technology</i> , 2012, 46, 10004-10011.	10.0	18
30	Flow field-flow fractionation (FFF) coupled to sensitive detection techniques: a way to examine radionuclide interactions with nanoparticles. <i>Mineralogical Magazine</i> , 2012, 76, 2709-2721.	1.4	8
31	Chemical status of U(VI) in cemented waste forms under saline conditions. <i>Radiochimica Acta</i> , 2010, 98, 674-683.	1.2	11
32	High level nuclear waste glass corrosion in synthetic clay pore solution and retention of actinides in secondary phases. <i>Journal of Nuclear Materials</i> , 2009, 385, 456-460.	2.7	16
33	Sites of Lu(III) Sorbed to and Coprecipitated with Hectorite. <i>Environmental Science &amp; Technology</i> , 2009, 43, 8807-8812.	10.0	22
34	TRLFS characterization of Eu(III)-doped synthetic organo-hectorite. <i>Journal of Contaminant Hydrology</i> , 2008, 102, 253-262.	3.3	10
35	Temperature effects on the interaction mechanisms between U(VI) and Eu(III) and $\text{Zr}_2\text{O}_7$ : experiment and modelling. <i>Radiochimica Acta</i> , 2008, 96, 11-21.	1.2	12
36	Temperature effects on the surface acidity properties of zirconium diphosphate. <i>Journal of Colloid and Interface Science</i> , 2007, 312, 230-236.	9.4	16