

# Yuheng Wang

## List of Publications by Year in descending order

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43  
papers

2,077  
citations

279798

23  
h-index

254184

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44  
docs citations

44  
times ranked

2449  
citing authors

#	ARTICLE	IF	CITATIONS
1	Rhizobacteria helps to explain the enhanced efficiency of phytoextraction strengthened by <i>Streptomyces pactum</i> . <i>Journal of Environmental Sciences</i> , 2023, 125, 73-81.	6.1	6
2	Accumulation, regional distribution, and environmental effects of Sb in the largest Hg-Sb mine area in Qinling Orogen, China. <i>Science of the Total Environment</i> , 2022, 804, 150218.	8.0	16
3	Insights on uranium removal by ion exchange columns: The deactivation mechanisms, and an overlooked biological pathway. <i>Chemical Engineering Journal</i> , 2022, 434, 134708.	12.7	43
4	Efficient removal of heavily oil-contaminated soil using a combination of fenton pre-oxidation with biostimulated iron and bioremediation. <i>Journal of Environmental Management</i> , 2022, 308, 114590.	7.8	8
5	Translocation of Foliar Absorbed Zn in Sunflower ( <i>Helianthus annuus</i> ) Leaves. <i>Frontiers in Plant Science</i> , 2022, 13, 757048.	3.6	2
6	Electrochemical removal and recovery of uranium: Effects of operation conditions, mechanisms, and implications. <i>Journal of Hazardous Materials</i> , 2022, 432, 128723.	12.4	24
7	Simultaneous removal of tetrachloroethylene and nitrate with a novel sulfur-packed biocathode system: The synergy between bioelectrocatalytic dechlorination and sulfur autotrophic denitrification. <i>Chemical Engineering Journal</i> , 2022, 439, 135793.	12.7	18
8	Impact of ZnSO <sub>4</sub> and ZnEDTA applications on wheat Zn biofortification, soil Zn fractions and bacterial community: Significance for public health and agroecological environment. <i>Applied Soil Ecology</i> , 2022, 176, 104484.	4.3	6
9	Coupled sulfur and electrode-driven autotrophic denitrification for significantly enhanced nitrate removal. <i>Water Research</i> , 2022, 220, 118675.	11.3	35
10	Weak electro-stimulation promotes microbial uranium removal: Efficacy and mechanisms. <i>Journal of Hazardous Materials</i> , 2022, 439, 129622.	12.4	18
11	Organic carbon mineralization and sequestration as affected by Zn availability in a calcareous loamy clay soil amended with wheat straw: a short-term case study. <i>Archives of Agronomy and Soil Science</i> , 2021, 67, 93-108.	2.6	4
12	Heavy metals in indoor dust: Spatial distribution, influencing factors, and potential health risks. <i>Science of the Total Environment</i> , 2021, 755, 142367.	8.0	56
13	Synergistic improvement of soil organic carbon storage and wheat grain zinc bioavailability by straw return in combination with Zn application on the Loess Plateau of China. <i>Catena</i> , 2021, 197, 104920.	5.0	16
14	Non-glandular trichomes of sunflower are important in the absorption and translocation of foliar-applied Zn. <i>Journal of Experimental Botany</i> , 2021, 72, 5079-5092.	4.8	15
15	Efficient cyclic oxidation of macro long-chain alkanes in soil using Fenton oxidation with recyclable Fe. <i>Journal of Hazardous Materials</i> , 2021, 417, 126026.	12.4	8
16	Reductive soil disinfestation attenuates antibiotic resistance genes in greenhouse vegetable soils. <i>Journal of Hazardous Materials</i> , 2021, 420, 126632.	12.4	9
17	Efficient and durable uranium extraction from uranium mine tailings seepage water via a photoelectrochemical method. <i>IScience</i> , 2021, 24, 103230.	4.1	16
18	Effect of Aging on the Stability of Microbially Reduced Uranium in Natural Sediment. <i>Environmental Science &amp; Technology</i> , 2020, 54, 613-620.	10.0	19

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19	Applications of anodized TiO <sub>2</sub> nanotube arrays on the removal of aqueous contaminants of emerging concern: A review. <i>Water Research</i> , 2020, 186, 116327.	11.3	84
20	Microbially Mediated Release of As from Mekong Delta Peat Sediments. <i>Environmental Science &amp; Technology</i> , 2019, 53, 10208-10217.	10.0	12
21	Arsenic Speciation in Mekong Delta Sediments Depends on Their Depositional Environment. <i>Environmental Science &amp; Technology</i> , 2018, 52, 3431-3439.	10.0	50
22	Diffusion- and pH-Dependent Reactivity of Layer-Type MnO <sub>2</sub> : Reactions at Particle Edges versus Vacancy Sites. <i>Environmental Science &amp; Technology</i> , 2018, 52, 3476-3485.	10.0	40
23	(Fe <sup>3+</sup> )-UVC-(aliphatic/phenolic carboxyl acids) systems for diethyl phthalate ester degradation: A density functional theory (DFT) and experimental study. <i>Applied Catalysis A: General</i> , 2018, 567, 20-27.	4.3	5
24	Products of in Situ Corrosion of Depleted Uranium Ammunition in Bosnia and Herzegovina Soils. <i>Environmental Science &amp; Technology</i> , 2016, 50, 12266-12274.	10.0	25
25	Uranium isotopes fingerprint biotic reduction. <i>Proceedings of the National Academy of Sciences of the United States of America</i> , 2015, 112, 5619-5624.	7.1	133
26	Arsenic(III) and Arsenic(V) Speciation during Transformation of Lepidocrocite to Magnetite. <i>Environmental Science &amp; Technology</i> , 2014, 48, 14282-14290.	10.0	66
27	Geochemical Control on Uranium(IV) Mobility in a Mining-Impacted Wetland. <i>Environmental Science &amp; Technology</i> , 2014, 48, 10062-10070.	10.0	41
28	Mobile uranium(IV)-bearing colloids in a mining-impacted wetland. <i>Nature Communications</i> , 2013, 4, 2942.	12.8	151
29	Structure and reactivity of As(III)- and As(V)-rich schwertmannites and amorphous ferric arsenate sulfate from the Carnoulès acid mine drainage, France: Comparison with biotic and abiotic model compounds and implications for As remediation. <i>Geochimica Et Cosmochimica Acta</i> , 2013, 104, 310-329.	3.9	86
30	Nanocrystalline Brookite with Enhanced Stability and Photocatalytic Activity: Influence of Lanthanum(III) Doping. <i>ACS Applied Materials &amp; Interfaces</i> , 2012, 4, 752-760.	8.0	26
31	Distinctive Arsenic(V) Trapping Modes by Magnetite Nanoparticles Induced by Different Sorption Processes. <i>Environmental Science &amp; Technology</i> , 2011, 45, 7258-7266.	10.0	94
32	Reactivity at (nano)particle-water interfaces, redox processes, and arsenic transport in the environment. <i>Comptes Rendus - Geoscience</i> , 2011, 343, 123-139.	1.2	58
33	New insight into the structure of nanocrystalline ferrihydrite: EXAFS evidence for tetrahedrally coordinated iron(III). <i>Geochimica Et Cosmochimica Acta</i> , 2011, 75, 2708-2720.	3.9	139
34	XANES Evidence for Rapid Arsenic(III) Oxidation at Magnetite and Ferrihydrite Surfaces by Dissolved O <sub>2</sub> via Fe <sup>2+</sup> -Mediated Reactions. <i>Environmental Science &amp; Technology</i> , 2010, 44, 5416-5422.	10.0	165
35	Evidence for Different Surface Speciation of Arsenite and Arsenate on Green Rust: An EXAFS and XANES Study. <i>Environmental Science &amp; Technology</i> , 2010, 44, 109-115.	10.0	98
36	EXAFS and HRTEM Evidence for As(III)-Containing Surface Precipitates on Nanocrystalline Magnetite: Implications for As Sequestration. <i>Langmuir</i> , 2009, 25, 9119-9128.	3.5	70

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37	Arsenite sequestration at the surface of nano-Fe(OH) <sub>2</sub> , ferrous-carbonate hydroxide, and green-rust after bioreduction of arsenic-sorbed lepidocrocite by <i>Shewanella putrefaciens</i> . <i>Geochimica Et Cosmochimica Acta</i> , 2009, 73, 1359-1381.	3.9	88
38	Extended X-ray Absorption Fine Structure Analysis of Arsenite and Arsenate Adsorption on Maghemite. <i>Environmental Science &amp; Technology</i> , 2008, 42, 2361-2366.	10.0	107
39	Arsenite sorption at the magnetite-water interface during aqueous precipitation of magnetite: EXAFS evidence for a new arsenite surface complex. <i>Geochimica Et Cosmochimica Acta</i> , 2008, 72, 2573-2586.	3.9	113
40	XAS Study of Arsenic Coordination in <i>Euglena gracilis</i> Exposed to Arsenite. <i>Environmental Science &amp; Technology</i> , 2008, 42, 5342-5347.	10.0	33
41	Biogenic vs. abiogenic magnetite nanoparticles: A XMCD study. <i>American Mineralogist</i> , 2008, 93, 880-885.	1.9	63
42	Synchrotron X-ray studies of heavy metal mineral-microbe interactions. <i>Mineralogical Magazine</i> , 2008, 72, 169-173.	1.4	2
43	Detection and phylogenetic identification of labeled prokaryotic cells on mineral surfaces using Scanning X-ray Microscopy. <i>Chemical Geology</i> , 2007, 240, 182-192.	3.3	9