

Abdelaziz El Jazouli

List of Publications by Year in descending order

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Version: 2024-02-01

11

papers

249

citations

1307594

7

h-index

1281871

11

g-index

11

all docs

11

docs citations

11

times ranked

240

citing authors

#	ARTICLE	IF	CITATIONS
1	Structure and thermochemical study of strontium sodium phosphate glasses. <i>Journal of Non-Crystalline Solids</i> , 2016, 447, 59-65.	3.1	16
2	Structural characterization and calorimetric dissolution behavior of Na ₂ O CuO P ₂ O ₅ glasses. <i>Journal of Non-Crystalline Solids</i> , 2016, 452, 144-152.	3.1	16
3	Structural and thermochemical properties of sodium magnesium phosphate glasses. <i>Journal of Alloys and Compounds</i> , 2015, 632, 766-771.	5.5	47
4	Structural and thermochemical study of Na ₂ O-ZnO-P ₂ O ₅ glasses. <i>Journal of Non-Crystalline Solids</i> , 2014, 390, 5-12.	3.1	80
5	Structural investigations and calorimetric dissolution of manganese phosphate glasses. <i>Journal of Non-Crystalline Solids</i> , 2014, 389, 66-71.	3.1	41
6	Magnetic, Mössbauer and optical spectroscopic properties of the AFe ₃ O(PO ₄) ₃ (A=Ca, Sr, Pb) series of powder compounds. <i>Solid State Sciences</i> , 2014, 36, 52-61.	3.2	2
7	On the synthesis, characterization and magnetic properties of two new phases discovered in the PbO-Fe ₂ O ₃ -P ₂ O ₅ system. <i>Journal of Solid State Chemistry</i> , 2013, 202, 85-92.	2.9	4
8	Synthesis, crystal structure, and vibrational spectroscopic and UV-visible studies of Cs ₂ MnP ₂ O ₇ . <i>Journal of Solid State Chemistry</i> , 2013, 198, 379-385.	2.9	25
9	Crystal Growth and Structure of the New Ferrimagnetic Oxyphosphate PbFe ₃ O(PO ₄) ₃ . <i>European Journal of Inorganic Chemistry</i> , 2011, 2011, 5486-5495.	2.0	7
10	Jahn-Teller phase transition in Cu _{0.50} TiO(PO ₄): Powder structural characterization of the $\tilde{\Gamma}^2$ -variety and thermal study. <i>Solid State Sciences</i> , 2007, 9, 258-266.	3.2	8
11	Synthesis and crystal structure of Na _{3.5} Cr _{1.5} Co _{0.5} (PO ₄) ₃ phosphate. <i>Powder Diffraction</i> , 2006, 21, 210-213.	0.2	3