

# Laura M Cornaglia

## List of Publications by Year in descending order

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56  
papers

1,652  
citations

236925

25  
h-index

302126

39  
g-index

56  
all docs

56  
docs citations

56  
times ranked

1719  
citing authors

#	ARTICLE	IF	CITATIONS
1	Kinetics and reaction pathway of the CO <sub>2</sub> reforming of methane on Rh supported on lanthanum-based solid. <i>Journal of Catalysis</i> , 2007, 245, 25-34.	6.2	167
2	Quantitative determination of the number of surface active sites and the turnover frequency for methanol oxidation over bulk metal vanadates. <i>Catalysis Today</i> , 2003, 78, 257-268.	4.4	100
3	Kinetic and Stability Studies of Ru/La <sub>2</sub> O <sub>3</sub> Used in the Dry Reforming of Methane. <i>Topics in Catalysis</i> , 2008, 51, 98-106.	2.8	94
4	Novel PdAgCu ternary alloy: Hydrogen permeation and surface properties. <i>Applied Surface Science</i> , 2011, 257, 6626-6635.	6.1	60
5	Recent advances in catalysts, palladium alloys and high temperature WGS membrane reactors. <i>International Journal of Hydrogen Energy</i> , 2015, 40, 3423-3437.	7.1	59
6	XPS study of the surface properties and Ni particle size determination of Ni-supported catalysts. <i>Surface and Interface Analysis</i> , 2014, 46, 521-529.	1.8	57
7	Dry reforming of methane in membrane reactors using Pd and Pd-Ag composite membranes on a NaA zeolite modified porous stainless steel support. <i>Journal of Membrane Science</i> , 2010, 364, 17-26.	8.2	55
8	Surface characterization of Pd-Ag composite membranes after annealing at various temperatures. <i>Journal of Membrane Science</i> , 2011, 369, 267-276.	8.2	54
9	Pd-based binary and ternary alloy membranes: Morphological and perm-selective characterization in the presence of H <sub>2</sub> S. <i>Journal of Membrane Science</i> , 2014, 450, 299-307.	8.2	52
10	Activity and stability of a CuO/CeO <sub>2</sub> catalyst for methanol steam reforming. <i>International Journal of Hydrogen Energy</i> , 2015, 40, 13379-13387.	7.1	47
11	PdAgAu alloy with high resistance to corrosion by H <sub>2</sub> S. <i>International Journal of Hydrogen Energy</i> , 2012, 37, 18547-18555.	7.1	46
12	Novel PdAgCu ternary alloy as promising materials for hydrogen separation membranes: Synthesis and characterization. <i>Surface Science</i> , 2011, 605, 62-71.	1.9	44
13	PdAu membranes supported on top of vacuum-assisted ZrO <sub>2</sub> -modified porous stainless steel substrates. <i>Journal of Membrane Science</i> , 2013, 428, 1-10.	8.2	44
14	Kinetic Studies of the Dry Reforming of Methane over the Rh/La <sub>2</sub> O <sub>3</sub> -SiO <sub>2</sub> Catalyst. <i>Industrial &amp; Engineering Chemistry Research</i> , 2007, 46, 7543-7549.	3.7	41
15	Advances in hydrogen selective membranes based on palladium ternary alloys. <i>International Journal of Hydrogen Energy</i> , 2021, 46, 15572-15594.	7.1	40
16	Operando Raman spectroscopic studies of lithium zirconates during CO <sub>2</sub> capture at high temperature. <i>RSC Advances</i> , 2016, 6, 8222-8231.	3.6	37
17	Well-dispersed Rh nanoparticles with high activity for the dry reforming of methane. <i>International Journal of Hydrogen Energy</i> , 2017, 42, 16127-16138.	7.1	37
18	PdCuAu ternary alloy membranes: Hydrogen permeation properties in the presence of H <sub>2</sub> S. <i>Journal of Membrane Science</i> , 2015, 479, 246-255.	8.2	32

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19	Pd based membrane reactor for ultra pure hydrogen production through the dry reforming of methane. Experimental and modeling studies. <i>Applied Catalysis A: General</i> , 2011, 400, 185-194.	4.3	31
20	The effect of electroless plating time on the morphology, alloy formation and H <sub>2</sub> transport properties of Pd–Ag composite membranes. <i>International Journal of Hydrogen Energy</i> , 2011, 36, 4068-4078.	7.1	30
21	Hydrogen permeation and surface properties of PdAu and PdAgAu membranes in the presence of CO, CO <sub>2</sub> and H <sub>2</sub> S. <i>Journal of Membrane Science</i> , 2018, 563, 351-359.	8.2	30
22	Stability of Ni and Rh–Ni catalysts derived from hydrotalcite-like precursors for the partial oxidation of methane. <i>International Journal of Hydrogen Energy</i> , 2013, 38, 5616-5626.	7.1	29
23	NaA zeolite membranes synthesized on top of APTES-modified porous stainless steel substrates. <i>Journal of Membrane Science</i> , 2016, 512, 93-103.	8.2	29
24	Hydrogen production from ethylene glycol reforming catalyzed by Ni and Ni–Pt hydrotalcite-derived catalysts. <i>International Journal of Hydrogen Energy</i> , 2016, 41, 22000-22008.	7.1	28
25	Reactivity of rice husk-derived lithium silicates followed by in situ Raman spectroscopy. <i>Journal of Alloys and Compounds</i> , 2019, 778, 699-711.	5.5	26
26	Hydrogen production through CO <sub>2</sub> reforming of CH <sub>4</sub> over Pt/CeZrO <sub>2</sub> /Al <sub>2</sub> O <sub>3</sub> catalysts using a Pd–Ag membrane reactor. <i>Catalysis Today</i> , 2012, 193, 64-73.	4.4	25
27	Supported Rh nanoparticles on CaO–SiO <sub>2</sub> binary systems for the reforming of methane by carbon dioxide in membrane reactors. <i>Applied Catalysis A: General</i> , 2014, 474, 114-124.	4.3	24
28	Study of the performance of Rh/La <sub>2</sub> O <sub>3</sub> –SiO <sub>2</sub> and Rh/CeO <sub>2</sub> catalysts for SR of ethanol in a conventional fixed-bed reactor and a membrane reactor. <i>International Journal of Hydrogen Energy</i> , 2015, 40, 4154-4166.	7.1	24
29	Characterization of Pd–Ag membranes after exposure to hydrogen flux at high temperatures. <i>Journal of Membrane Science</i> , 2007, 306, 56-65.	8.2	23
30	Comparison of Ru/La <sub>2</sub> O <sub>2</sub> CO <sub>3</sub> performance in two different membrane reactors for hydrogen production. <i>Catalysis Today</i> , 2013, 213, 135-144.	4.4	23
31	A coke-resistant catalyst for the dry reforming of methane based on Ni nanoparticles confined within rice husk-derived mesoporous materials. <i>Catalysis Communications</i> , 2020, 135, 105898.	3.3	23
32	The effect of the Li:Na molar ratio on the structural and sorption properties of mixed zirconates for CO <sub>2</sub> capture at high temperature. <i>Journal of Environmental Chemical Engineering</i> , 2019, 7, 102927.	6.7	22
33	Optimization and characterization of electroless co-deposited PdRu membranes: Effect of the plating variables on morphology. <i>Journal of Membrane Science</i> , 2011, 382, 252-261.	8.2	20
34	Catalytic behavior of Ru nanoparticles supported on carbon fibers for the ethanol steam reforming reaction. <i>Catalysis Communications</i> , 2018, 114, 19-23.	3.3	19
35	Ni mesostructured catalysts obtained from rice husk ashes by microwave-assisted synthesis for CO <sub>2</sub> methanation. <i>Journal of CO<sub>2</sub> Utilization</i> , 2020, 42, 101328.	6.8	19
36	Surface composition of PdCuAu ternary alloys: a combined LEIS and XPS study. <i>Surface and Interface Analysis</i> , 2015, 47, 745-754.	1.8	16

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37	Pt encapsulated into NaA zeolite as catalyst for the WGS reaction. <i>Applied Catalysis A: General</i> , 2019, 572, 176-184.	4.3	15
38	Comparative study of lithium-based CO <sub>2</sub> sorbents at high temperature: Experimental and modeling kinetic analysis of the carbonation reaction. <i>Journal of Environmental Chemical Engineering</i> , 2020, 8, 104173.	6.7	15
39	Effect of the porous stainless steel substrate shape on the ZrO <sub>2</sub> deposition by vacuum assisted dip-coating. <i>International Journal of Hydrogen Energy</i> , 2017, 42, 7986-7996.	7.1	14
40	Formation of a solid solution of vanadium in TiO <sub>2</sub> (anatase) on vanadium-titanium solids with high vanadium content. <i>Journal of Materials Chemistry</i> , 1995, 5, 1443-1449.	6.7	12
41	Isolation of ibuprofen enantiomers and racemic esters through electrodialysis. <i>Journal of Membrane Science</i> , 2021, 618, 118714.	8.2	10
42	4.9 The Role of Acid-Base and Redox Features in the Catalytic Behavior of Vanadium-Phosphorous-Oxygen Formulations. <i>Studies in Surface Science and Catalysis</i> , 1994, 90, 429-440.	1.5	9
43	Determination of the Metal Dispersion of Supported Catalysts Using XPS. <i>Topics in Catalysis</i> , 2019, 62, 822-837.	2.8	9
44	Development of catalytic membranes over PdAu selective films for hydrogen production through the dry reforming of methane. <i>Molecular Catalysis</i> , 2020, 481, 100643.	2.0	8
45	K-doping effect in the kinetics of CO <sub>2</sub> capture at high temperature over lithium silicates obtained from rice husks: In situ/operando techniques. <i>Ceramics International</i> , 2021, 47, 1558-1570.	4.8	8
46	Study of the sorption properties of alkali zirconate-based sorbents at high temperature in the presence of water and low CO <sub>2</sub> concentration. <i>Journal of Alloys and Compounds</i> , 2022, 895, 162419.	5.5	7
47	The nature of the cobalt salt affects the catalytic properties of promoted VPO. <i>Studies in Surface Science and Catalysis</i> , 2000, 130, 1727-1732.	1.5	6
48	Influence of La incorporation on the catalytic activity of Ru/ETS-10 catalysts for hydrogen production. <i>Applied Catalysis A: General</i> , 2015, 504, 391-398.	4.3	6
49	Pure Hydrogen Production for Low Temperature Fuel Cells. <i>Catalysis Letters</i> , 2018, 148, 1015-1026.	2.6	5
50	PdAu and PdAuAg composite membranes with reduced film thickness using YSZ as a stainless-steel support modifier. <i>Journal of Alloys and Compounds</i> , 2021, 877, 160184.	5.5	5
51	Dissociation of perfluorinated ethers on Al <sub>2</sub> O <sub>3</sub> thin films. <i>Tribology Letters</i> , 1998, 4, 67-73.	2.6	3
52	Title is missing!. <i>Catalysis Letters</i> , 1999, 63, 131-133.	2.6	3
53	NaA zeolite membranes on modified porous stainless steel supports: a comparative study of different SiO <sub>2</sub> sources. <i>Brazilian Journal of Chemical Engineering</i> , 2020, 37, 383-397.	1.3	3
54	Synthesis of Pt-zeolite coated palladium alloys as catalytic membranes for hydrogen production. <i>International Journal of Hydrogen Energy</i> , 2021, 46, 2255-2268.	7.1	3

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55	Coupling of CO <sub>2</sub> capture and methanation processes using catalysts based on silica recovered from rice husks. <i>Fuel</i> , 2022, 324, 124604.	6.4	3
56	New PdNiAu ternary alloys as potential material for hydrogen separation processes. <i>International Journal of Hydrogen Energy</i> , 2022, 47, 11589-11600.	7.1	1