## Hao Wu

## List of Publications by Year in descending order

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19	131	7	11
papers	citations	h-index	g-index
19	19	19	81
all docs	docs citations	times ranked	citing authors

#	Article	IF	CITATIONS
1	Synergistic adsorption behavior of a silica-based adsorbent toward palladium, molybdenum, and zirconium from simulated high-level liquid waste. Journal of Hazardous Materials, 2021, 411, 125136.	12.4	33
2	Effects of NH <sub>4</sub> <sup>+</sup> , K <sup>+</sup> , Mg <sup>2+</sup> , and Ca <sup>2+</sup> on the Cesium Adsorption/Desorption in Binding Sites of Vermiculitized Biotite. Environmental Science & Samp; Technology, 2017, 51, 13886-13894.	10.0	30
3	Extraction behaviors of platinum group metals in simulated high-level liquid waste by a hydrophobic ionic liquid bearing an amino moiety. Nuclear Engineering and Technology, 2021, 53, 1218-1223.	2.3	10
4	Selective separation of cadmium( <scp>ii</scp> ) from zinc( <scp>ii</scp> ) by a novel hydrophobic ionic liquid including an <i>N</i> , <i>N</i> , <i>N</i> , <ii>N,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N</io>,<io>N,<io>N,<io>N,<io>N,<io>N,<io>N,<io>N,<io< td=""><td>3.3</td><td>8</td></io<></io></io></io></io></io></io></io></io></ii>	3.3	8
5	Adsorption and separation behavior of palladium(II) from simulated high-level liquid waste using a novel silica-based adsorbents. Radiochimica Acta, 2021, 109, 367-375.	1.2	8
6	Extraction Behavior of Lanthanides by a Novel Ionic Liquid Including ⟨i>N⟨/i>,⟨i>N⟨/i>,⟨i>N⟨/i>′,⟨i>N⟨/i>′-Tetrakis(2-pyridylmethyl)-1,3-diaminopropane-2-amido Structure: A Softâ€"Hard Donor Combined Strategy. Chemistry Letters, 2018, 47, 732-735.	1.3	7
7	Adsorption and separation behavior of Pd(II) from simulated high-level liquid waste using ⟨i⟩N,N,N',N',⟨i⟩-tetra-2-ethylhexyl-thiodiglycolamide silica-based adsorbent. Separation Science and Technology, 2022, 57, 48-59.	2.5	7
8	Adsorption Performances of an Acid-stable 2D Covalent Organic Framework towards Palladium(II) in Simulated High-level Liquid Waste. Analytical Sciences, 2021, 37, 645-647.	1.6	7
9	Adsorption and separation behavior of palladium(II) on a silica-based hybrid donor adsorbent from simulated high-level liquid waste. Journal of Radioanalytical and Nuclear Chemistry, 2020, 326, 1323-1331.	1.5	6
10	Adsorption and separation behavior of strontium and yttrium using a silica-based bis(2-ethylhexyl) hydrogen phosphate adsorbent. Journal of Radioanalytical and Nuclear Chemistry, 2021, 329, 1001-1009.	1.5	5
11	Impregnation of covalent organic framework into porous silica support for the recovery of palladium ions from simulated high-level liquid waste. Journal of Radioanalytical and Nuclear Chemistry, 2021, 330, 1065-1074.	1.5	3
12	Adsorption Behaviors of Palladium Ion from Nitric Acid Solution by a Silica-based Hybrid Donor Adsorbent. Analytical Sciences, 2020, 36, 1541-1545.	1.6	2
13	Combination of N'-N'-di-n-hexyl-thiodiglycolamide and 2,2'-[(2-ethylhexyl)imino]bis[N,N-bis(2-ethylhexyl) acetamide] for the enhanced adsorption of palladium ions from simulated high-level liquid waste. Journal of Radioanalytical and Nuclear Chemistry, 2022, 331, 1731-1740.	1.5	2
14	Adsorption Behaviors of Palladium(II) in Simulated High-Level Liquid Waste Using 2,2'-[(2-ethylhexyl)imino]bis[ <i>N</i> N-bis(2-ethylhexyl)acetamide]-impregnated Adsorbent. Journal of Ion Exchange, 2021, 32, 8-14.	0.3	1
15	Effect of adding dodecanol as modifier to <i>N</i> , <i>N&lt;</i>	1.2	1
16	Adsorption Behavior of Sr and Ba Using TDS-Impregnated Microporous Silica-Based Adsorbents in Nitric Acid Solution. Journal of Ion Exchange, 2022, 33, 8-16.	0.3	1
17	Selective Separation of Trivalent Europium(III) from Americium(III) using <i>N,N′</i> Dimethyl- <i>N,N′</i> di-2-phenylethyl-diglycol Amide (MPEDGA) Extractant in Ionic Liquid. Solvent Extraction Research and Development, 2021, 28, 49-57.	0.4	0
18	Complexation Studies of Eu(III) by a Novel Soft N and Hard O Donor Combined Ligand Including <i>N,N,N',N'</i> -Tetrakis(2-pyridylmethyl)-1,3-diaminopropane-2-amide Structure: UV-vis Titration, X-ray Crystallography, EXAFS Spectroscopy Analysis. Solvent Extraction Research and Development, 2021, 28, 69-77.	0.4	0

#	Article	IF	CITATIONS
19	Adsorption Behaviors of a 2D Covalent Organic Framework Toward Pd(II) in Hydrochloric Acid Solution. Journal of Ion Exchange, 2022, 33, 32-35.	0.3	0