

# Karl Seff

## List of Publications by Year in descending order

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#	ARTICLE	IF	CITATIONS
1	Using Crystallography and NMR to Count the Number of Three-Aluminum Six-Rings in Fully Zn <sup>2+</sup> -Exchanged Zeolite Y. These Six-Rings Concentrate at Single Six-Ring Positions. Journal of Physical Chemistry C, 2021, 125, 583-592.	1.5	4
2	Water Molecules in Zeolite Y Enhance the Photoluminescent Properties of Its Cesium Lead Bromide Quantum Dots, Na <sub>4</sub> Cs <sub>6</sub> PbBr <sub>48</sub> +. Journal of Physical Chemistry C, 2021, 125, 5904-5918.	1.5	3
3	Crystal Structure and Luminescence of Sn <sub>4</sub> I <sub>4</sub> Cs <sub>4</sub> Na <sup>+</sup> Y, a Lead-Free Zeolite Containing Tetrahedrally Distorted Cubes of Sn <sub>4</sub> I <sub>4</sub> 4+. Journal of Physical Chemistry C, 2021, 125, 15696-15710.	1.5	0
4	Quantum Dots of [Na <sub>4</sub> Cs <sub>6</sub> PbBr <sub>4</sub> ] <sup>8+</sup> , Water Stable in Zeolite X, Luminesce Sharply in the Green. Advanced Materials, 2020, 32, e2001868.	11.1	14
5	Crystal structure and X-ray luminescence of zeolite Y (Si/Al = 1.69) containing extraframework hafnium(IV). Microporous and Mesoporous Materials, 2019, 288, 109552.	2.2	3
6	Crystal structure of a hydrogen sulfide sorption complex of anhydrous Mn <sup>2+</sup> -exchanged zeolite Y (FAU, Si/Al = 1.56). Microporous and Mesoporous Materials, 2019, 279, 432-438.	2.2	7
7	Structure and luminescence of extraframework TiCl <sub>6</sub> <sup>2-</sup> in Cs <sup>+</sup> -containing zeolite LTA. Journal of Porous Materials, 2019, 26, 1079-1089.	1.3	2
8	Identification and structures of the X-ray induced luminescence centers in the zeolites Zr,X,Cs,Na-LTA, X = Cl, Br, and I. Microporous and Mesoporous Materials, 2019, 278, 443-454.	2.2	2
9	Crystal Structure of Zeolite LTA Containing Extraframework Tungsten(VI) Ions. Journal of Physical Chemistry C, 2018, 122, 6661-6668.	1.5	2
10	Structure of a cyclohexane sorption complex of partially dehydrated, fully Mn <sup>2+</sup> -exchanged zeolite Y (FAU, Si/Al = 1.56). Microporous and Mesoporous Materials, 2018, 264, 139-146.	2.2	3
11	Disproportionation of an Element in a Zeolite. III. Crystal Structure of a High-Temperature Sulfur Sorption Complex of Zeolite LTA Containing Two New Ions: Perthiosulfite, S <sub>4</sub> <sup>2-</sup> , and the Trisulfur Cation, S <sub>3</sub> <sup>2+</sup> . Journal of Physical Chemistry C, 2018, 122, 28133-28141.	1.5	7
12	Preparation, crystal structure, and luminescence of zeolite Ta,Cl,Cs,Na-A containing a cubic Cs <sub>11</sub> TaCl <sub>6</sub> 10+ continuum. Journal of Porous Materials, 2017, 24, 1117-1128.	1.3	2
13	Exchanging noble and seminoble cations into zeolites by oxygen vacancy ion exchange (OVIE). Microporous and Mesoporous Materials, 2017, 244, 47-49.	2.2	4
14	The Pentatin Cation in Zeolite Y: Thallous Ion Exchange and Crystal Structure of [Sn <sub>36</sub> Cl <sub>11</sub> ][Si <sub>128</sub> Al <sub>64</sub> O <sub>384</sub> ]-FAU Containing Sn <sub>5</sub> I <sub>2</sub> <sup>+</sup> , Sn <sub>2</sub> Cl <sub>3</sub> <sup>+</sup> , and Sn <sub>3</sub> Cl <sub>5</sub> <sup>+</sup> . Journal of Physical Chemistry C, 2017, 121, 471-480.	1.5	4
15	Progress toward Zeolite-Based Self-Luminous Sensors for Radioactive Isotopes such as <sup>201</sup> Tl and <sup>137</sup> Cs: Structures and Luminescence of Hf,Cl,Tl-A and Hf,Cl,Cs,Na-A. Journal of Physical Chemistry C, 2017, 121, 19619-19633.	1.5	9
16	Structures of the Subnanometer Clusters of Cadmium Sulfide Encapsulated in Zeolite Y: Cd <sub>4</sub> S <sub>6</sub> and Cd(SHCd) <sub>4</sub> <sup>6+</sup> . Journal of Physical Chemistry C, 2016, 120, 16722-16731.	1.5	13
17	Preparation, Crystal Structure, and Luminescence Properties of Zeolite LTA Containing Extraframework Tantalum(V), Tantalum(III), Thallium(I), and Chloride. Journal of Physical Chemistry C, 2016, 120, 12139-12148.	1.5	6
18	Encapsulating Luminescent Materials in Zeolites. III. Crystal Structure and Scintillation Properties of Cs,Na-LTA Treated with Zirconium Chloride Vapor. Journal of Physical Chemistry C, 2016, 120, 18682-18693.	1.5	7

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19	Surprising Intrazeolitic Chemistry of Silver. <i>Journal of Physical Chemistry C</i> , 2016, 120, 5277-5287.	1.5	18
20	Using the Thallous Ion Exchange Method to Exchange Tin into High Alumina Zeolites. I. Crystal Structure of $ \text{Sn}^{2+}_{5.3}\text{Sn}^{4+}_{0.8}\text{Cl}^{\ominus}_{1.8} [\text{Si}_{12}\text{Al}_{12}]$ . <i>Journal of Physical Chemistry C</i> , 2015, 119, 3244-3252.	1.5	12
21	Exchange of a Tetrapositive Cation into a Zeolite and a New Inorganic Scintillator. I. Crystal Structures and Scintillation Properties of Anhydrous $\text{Zr}_{1.7}\text{Ti}_{5.4}\text{Cl}_{1.7}$ and $\text{Zr}_{2.1}\text{Ti}_{1.6}\text{Cl}_{3.0}$ LTA. <i>Journal of Physical Chemistry C</i> , 2015, 119, 18226-18230.	1.5	13
22	Encapsulating Photoluminescent Materials in Zeolites. II. Crystal Structure of Fully Dehydrated $\text{Ce}_{21}\text{H}_{46}\text{O}_{18}$ Y (Si/Al = 1.69) Containing $\text{Ce}_4\text{O}_4$ , $\text{CeOH}_2$ , and $\text{H}_3$ . <i>Journal of Physical Chemistry C</i> , 2015, 119, 24501-24511.	1.5	16
23	The dependence of $\text{Co}^{2+}$ -exchange into zeolite FAU on its Si/Al ratio. <i>Journal of Porous Materials</i> , 2014, 21, 869-882.	1.3	6
24	First Successful Application of the Thallous Ion Exchange (TIE) Method. Preparation of Fully Indium-Exchanged Zeolite Y (FAU, Si/Al = 1.69). <i>Journal of Physical Chemistry C</i> , 2014, 118, 24655-24661.	1.5	10
25	Introducing copper ions into zeolite Y by the thallous ion exchange method: single crystal structure of $ \text{Cu}_{21.6}\text{Ti}_{39.2} [\text{Si}_{121}\text{Al}_{71}\text{O}_{384}]$ FAU. <i>Journal of Porous Materials</i> , 2014, 21, 321-330.	1.3	29
26	Encapsulating Photoluminescent Materials in Zeolites. Crystal Structure of Fully Dehydrated Zeolite Y (Si/Al = 1.69) Containing $\text{Eu}^{3+}$ . <i>Journal of Physical Chemistry C</i> , 2014, 118, 11014-11025.	1.5	13
27	Crystallographic Verification that Copper(II) Coordinates to Four of the Oxygen Atoms of Zeolite 6-Rings. Two Single-Crystal Structures of Fully Dehydrated, Largely $\text{Cu}^{2+}$ -Exchanged Zeolite Y (FAU, Si/Al = 1.56). <i>Journal of Physical Chemistry C</i> , 2012, 116, 963-974.	1.5	52
28	$\text{Li}^{+}$ Exchange into Zeolite $\text{Na}^{\ominus}$ Y (FAU) from Aqueous Methanol. Single-Crystal Structures of Fully Dehydrated $\text{LiNa}^{\ominus}$ Y. <i>Journal of Physical Chemistry C</i> , 2012, 116, 9009-9018.	1.5	39
29	Single-Crystal Structures of Fully and Partially Dehydrated Zeolite Y (FAU, Si/Al = 1.56) $\text{Ni}^{2+}$ Exchanged at a Low pH, 4.9. <i>Journal of Physical Chemistry C</i> , 2012, 116, 13985-13996.	1.5	45
30	Framework Sites Preferred by Aluminum in Zeolite ZSM-5. Structure of a Fully Dehydrated, Fully $\text{Cs}^{+}$ -Exchanged ZSM-5 Crystal (MFI, Si/Al = 24). <i>Journal of Physical Chemistry C</i> , 2011, 115, 24823-24838.	1.5	50
31	The Pentagallium Cation in Zeolite Y. Preparation and Crystal Structure of $\text{Ga}_{42}\text{Ti}_{9.3}\hat{\sim}\text{Si}_{121}\text{Al}_{71}\text{O}_{384}$ Containing $\text{Ga}_5$ , $\text{Ga}^{+}$ , $\text{Ga}^{2+}$ , $\text{Ga}^{3+}$ , and $\text{Ti}^{+}$ . <i>Journal of Physical Chemistry C</i> , 2011, 115, 2750-2760.	1.5	45
32	Single-crystal structures of highly -exchanged, fully deaminated, and fully $\text{Tl}^{+}$ -exchanged zeolite Y (FAU, Si/Al=1.56), all fully dehydrated. <i>Microporous and Mesoporous Materials</i> , 2010, 129, 11-21.	2.2	76
33	A General Method for the Ion Exchange of Zeolites Utilizing the Volatility of Thallous Compounds as Leaving Products. <i>Journal of Physical Chemistry C</i> , 2010, 114, 13295-13299.	1.5	18
34	Tetrahydroxytetraindium(III) Nanoclusters, $\text{In}_4(\text{OH})_4^{8+}$ , in Air-Oxidized Fully In-Exchanged Zeolite Y (FAU, Si/Al = 1.69). Preparation and Crystal Structures of $\text{In}^{\sim}$ Y and $\text{In}^{\sim}$ Y [ $\text{In}_4(\text{OH})_4$ ]. <i>Journal of Physical Chemistry C</i> , 2010, 114, 15741-15754.	1.5	46
35	Detailed Determination of the $\text{Tl}^{+}$ Positions in Zeolite $\text{Tl}^{\sim}$ ZSM-5. Single-Crystal Structures of Fully Dehydrated $\text{Tl}^{\sim}$ ZSM-5 and $\text{H}^{\sim}$ ZSM-5 (MFI, Si/Al = 29). Additional Evidence for a Nonrandom Distribution of Framework Aluminum. <i>Journal of Physical Chemistry C</i> , 2009, 113, 19937-19956.	1.5	25
36	Crystal Structures of Vacuum-Dehydrated $\text{Ni}^{2+}$ -Exchanged Zeolite Y (FAU, Si/Al = 1.69) Containing Three-Coordinate $\text{Ni}^{2+}$ , $\text{Ni}_8\text{O}_4$ , $\text{Ni}_4\text{O}_4$ , $\text{Ni}_4\text{O}_4$ Cores, and $\text{H}_3$ . <i>Journal of Physical Chemistry C</i> , 2009, 113, 5164-5181.	1.5	56

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37	Single Crystal Structure of Zeolite A (LTA) Containing Ag <sub>4</sub> Cl <sub>4</sub> Nanoclusters and Reduced 1,3,5-Tripyrylium Dimers with Remarkably Short 2.43 Å... Interplanar Spacings. <i>Journal of Physical Chemistry C</i> , 2008, 112, 11181-11193.	1.5	6
38	DOING CHEMISTRY IN A ONE-NANOMETER TEST TUBE (IN A ZEOLITE). <i>Comments on Inorganic Chemistry</i> , 2007, 28, 173-179.	3.0	2
39	Six Single-Crystal Structures Showing the Dehydration, Deamination, Dealumination, and Decomposition of NH <sub>4</sub> <sup>+</sup> -Exchanged Zeolite Y (FAU) with Increasing Evacuation Temperature. Identification of a Lewis Acid Site. <i>Journal of Physical Chemistry C</i> , 2007, 111, 18294-18306.	1.5	69
40	Crystal structures of the NO and NO <sub>2</sub> sorption complexes of fully dehydrated fully Mn <sup>2+</sup> -exchanged zeolite X (FAU). <i>Microporous and Mesoporous Materials</i> , 2006, 93, 12-22.	2.2	37
41	Single crystal structure of fully dehydrated fully K <sup>+</sup> -exchanged zeolite Y (FAU), K <sub>71</sub> Si <sub>121</sub> Al <sub>71</sub> O <sub>384</sub> . <i>Microporous and Mesoporous Materials</i> , 2006, 92, 234-242.	2.2	73
42	Single crystal structure of fully dehydrated fully Tl <sup>+</sup> -exchanged zeolite Y, $\alpha^{\text{Tl71}}[\text{Si}_{121}\text{Al}_{71}\text{O}_{384}]$ -FAU. <i>Microporous and Mesoporous Materials</i> , 2006, 94, 313-319.	2.2	54
43	Cationic zinc clusters with mean formula in the sodalite cavities of zeolite Y (FAU). <i>Microporous and Mesoporous Materials</i> , 2005, 85, 351-354.	2.2	50
44	Comment on "Synthesis of Fully Dehydrated Fully Zn <sup>2+</sup> -Exchanged Zeolite Y and Its Crystal Structure Determined by Pulsed-Neutron Diffraction": Cationic Zinc Clusters Formally Containing Zn(I) in the Sodalite Cavities of Zeolite Y (FAU). <i>Journal of Physical Chemistry B</i> , 2005, 109, 13840-13841.	1.2	11
45	Further crystallographic confirmation that Cs <sup>+</sup> ions can occupy sodalite cavities and double six-rings. Crystal structure of fully dehydrated partially Cs <sup>+</sup> -exchanged zeolite X, $[\text{Cs}_{45}\text{Na}_{47}][\text{Si}_{100}\text{Al}_{92}\text{O}_{384}]$ -FAU. <i>Microporous and Mesoporous Materials</i> , 2004, 71, 65-75.	2.2	49
46	Synthesis and Crystal Structure of Ag <sub>4</sub> I <sub>4</sub> Nanoclusters in the Sodalite Cavities of Fully K <sup>+</sup> -Exchanged Zeolite A. <i>Journal of Physical Chemistry B</i> , 2004, 108, 3168-3173.	1.2	18
47	Crystal structure of Mn <sub>46</sub> Si <sub>100</sub> Al <sub>92</sub> O <sub>384</sub> ·89H <sub>2</sub> S, a hydrogen sulfide sorption complex of fully dehydrated Mn <sup>2+</sup> -exchanged zeolite X. <i>Microporous and Mesoporous Materials</i> , 2003, 63, 21-31.	2.2	47
48	Two Crystal Structures of Fully Dehydrated, Fully Ag <sup>+</sup> -Exchanged Zeolite X. Dehydration in Oxygen Prevents Ag <sup>+</sup> -Reduction. Without Oxygen, Ag <sub>8n</sub> <sup>+</sup> (Td) and cyclo-Ag <sub>4m</sub> <sup>+</sup> (nearS <sub>4</sub> ) Form. <i>Journal of Physical Chemistry B</i> , 2003, 107, 6938-6945.	1.2	53
49	Spatially Ordered Quantum Dot Array of Indium Nanoclusters in Fully Indium-Exchanged Zeolite X. <i>Journal of Physical Chemistry B</i> , 2003, 107, 1120-1128.	1.2	60
50	Disproportionation of an Element in a Zeolite. I. Crystal Structure of a Sulfur Sorption Complex of Dehydrated, Fully Cd <sup>2+</sup> -Exchanged Zeolite X. Synthesis of Tetrahedral S <sub>44</sub> <sup>+</sup> and S <sub>42</sub> <sup>+</sup> , Two New Polyatomic Cations of Sulfur. <i>Journal of Physical Chemistry B</i> , 2003, 107, 3117-3123.	1.2	57
51	Disproportionation of an Element in a Zeolite. II. Crystal Structure of an Iodine Sorption Complex of Dehydrated Fully Cd <sup>2+</sup> -Exchanged Zeolite X Containing n-I <sub>5</sub> <sup>+</sup> and I <sub>3</sub> <sup>+</sup> and Square cyclo-I <sub>42</sub> <sup>+</sup> . <i>Journal of Physical Chemistry B</i> , 2003, 107, 10709-10714.	1.2	37
52	Reaction of Fully Indium-Exchanged Zeolite A with Hydrogen Sulfide. Crystal Structures of Indium-Exchanged Zeolite A Containing In <sub>2</sub> S, InSH, Sorbed H <sub>2</sub> S, and (In <sub>5</sub> ) <sub>7</sub> <sup>+</sup> . <i>Journal of Physical Chemistry B</i> , 2002, 106, 4578-4587.	1.2	31
53	Crystal Structure of a Mesitylene Sorption Complex of Dehydrated Fully Ca <sup>2+</sup> -Exchanged Zeolite X. Sorbed Mesitylene Appears to be Significantly Nonplanar. <i>Journal of Physical Chemistry B</i> , 2002, 106, 5827-5832.	1.2	46
54	Crystal Structure of a Cadmium Sorption Complex of Dehydrated Fully Cd <sup>2+</sup> -Exchanged Zeolite X Containing Cd <sup>2+</sup> , Cd <sup>+</sup> , and Cd <sup>0</sup> . <i>Journal of Physical Chemistry B</i> , 2002, 106, 7569-7573.	1.2	35

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55	Ronneburgite, $K_2MnV_4O_{12}$ , a new mineral from Ronneburg, Thuringia, Germany: Description and crystal structure. <i>American Mineralogist</i> , 2001, 86, 1081-1086.	0.9	13
56	Some chemical treatments diminish the long-range ordering in the aluminosilicate framework of zeolite X. <i>Microporous and Mesoporous Materials</i> , 2001, 42, 299-306.	2.2	22
57	Verification of linear $Na_{32+}$ clusters in zeolite X. <i>Microporous and Mesoporous Materials</i> , 2001, 46, 111-113.	2.2	6
58	Structure of a cyclopropane sorption complex of dehydrated fully $Mn^{2+}$ -exchanged zeolite X. <i>Microporous and Mesoporous Materials</i> , 2000, 40, 247-255.	2.2	37
59	Weak $Ag^+$ – $Ag^+$ bonding in zeolite X. Crystal structures of $Ag_{92}Si_{100}Al_{92}O_{384}$ hydrated and fully dehydrated in flowing oxygen. <i>Microporous and Mesoporous Materials</i> , 2000, 41, 49-59.	2.2	34
60	Structure of a cyclopropane sorption complex of dehydrated fully $Cd^{2+}$ -exchanged zeolite A. <i>Microporous and Mesoporous Materials</i> , 2000, 41, 61-68.	2.2	10
61	Cation Crowding in Zeolites. Reinvestigation of the Crystal Structure of Dehydrated Potassium-Exchanged Zeolite X. <i>Journal of Physical Chemistry B</i> , 2000, 104, 8946-8951.	1.2	58
62	Crystallographic Study of the Reaction of Zinc Vapor with Fully $Cd^{2+}$ -Exchanged Zeolite X. Complete Reduction of $Cd^{2+}$ by Zn, Extraction of $SiO_4$ and $AlO_4$ from the Zeolite Framework, and Reduction of $Si^{4+}$ to Si. <i>Journal of Physical Chemistry B</i> , 2000, 104, 9811-9816.	1.2	28
63	Crystal Structure of Partially $Pd^{2+}$ -Exchanged Zeolite X Dehydrated in Oxygen at 400 °C. Formation of Linear $Pd_2O_3$ Clusters Proposed To Be $HO^{\sim}PdIV^{\sim}O^{\sim}PdIV^{\sim}OH$ in $(Pd^{2+})_{14}(HOPdOH_4)_8(Na^+)_{32}Si_{100}Al_{92}O_{384}$ . <i>Journal of Physical Chemistry B</i> , 2000, 104, 2490-2494.	1.2	45
64	$Zn^{2+}$ Cations, Probable $Tl_4Zn_{12}$ and $Tl_6$ Clusters, and Zeolite Desilication (Less Likely Dealumination): Crystallographic Study of the Incomplete Reaction of Zn Vapor with $Tl^+$ -Exchanged Zeolite X. <i>Journal of Physical Chemistry B</i> , 2000, 104, 515-525.	1.2	46
65	A Cationic Rubidium Continuum in Zeolite X. <i>Journal of Physical Chemistry B</i> , 2000, 104, 11162-11167.	1.2	34
66	Crystal Structures of Fully Indium-Exchanged Zeolite X. <i>Journal of Physical Chemistry B</i> , 2000, 104, 8372-8381.	1.2	51
67	Crystal Structures of Fully $La^{3+}$ -Exchanged Zeolite X: An Intrazeolitic $La_2O_3$ Continuum, Hexagonal Planar and Trigonal Monocapped Trigonal Prismatic Coordination. <i>Journal of Physical Chemistry B</i> , 2000, 104, 2224-2236.	1.2	64
68	Crystal structure of an ammonia sorption complex of dehydrated fully $Ca^{2+}$ -exchanged zeolite X. <i>Microporous and Mesoporous Materials</i> , 1999, 28, 173-183.	2.2	50
69	Structures of cobalt(II)-exchanged zeolite X. <i>Microporous and Mesoporous Materials</i> , 1999, 33, 265-280.	2.2	103
70	Reinvestigation of the Crystal Structure of Dehydrated Sodium Zeolite X. <i>Journal of Physical Chemistry B</i> , 1999, 103, 9512-9518.	1.2	122
71	Structure of Dehydrated $Zn^{2+}$ -Exchanged Zeolite X. Overexchange, Framework Dealumination and Reorganization, Stoichiometric Retention of Monomeric Tetrahedral Aluminate. <i>Journal of Physical Chemistry B</i> , 1999, 103, 5631-5636.	1.2	61
72	Crystal Structure of a Zinc Sorption Complex of $Cd^{2+}$ -Exchanged Zeolite X Containing Tetrahedral $Cd_{2+4}(\frac{1}{4}ZnO_{Cd+Zn})_4$ Clusters. <i>Journal of Physical Chemistry B</i> , 1999, 103, 6493-6497.	1.2	53

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73	Hydronium Ions in Zeolites. 1. Structures of Partially and Fully Dehydrated Na,H <sub>3</sub> O <sup>+</sup> X by X-ray and Neutron Diffraction. Journal of Physical Chemistry B, 1999, 103, 10365-10372.	1.2	62
74	Crystal Structure of Anhydrous NH <sub>4</sub> <sup>+</sup> -Exchanged Zeolite X Partially Reacted with HgCl <sub>2</sub> Vapor. Cationic Chloromercuric Clusters, Regular Octahedral Hg(II), and Regular Trigonal Hg(II). Journal of Physical Chemistry B, 1999, 103, 10409-10416.	1.2	35
75	Crystal structure of a hydrogen sulfide sorption complex of fully Ca <sup>2+</sup> -exchanged zeolite X. Microporous and Mesoporous Materials, 1998, 23, 33-44.	2.2	19
76	Crystal structure of a carbon monoxide sorption complex of dehydrated fully manganese(II)-exchanged zeolite X. Microporous and Mesoporous Materials, 1998, 26, 101-107.	2.2	37
77	Crystal Structure of a Benzene Sorption Complex of Dehydrated Fully Ca <sup>2+</sup> -Exchanged Zeolite X. Journal of Physical Chemistry B, 1998, 102, 6071-6077.	1.2	59
78	Partial Structures of Fully Dehydrated Ni <sub>30</sub> Na <sub>7</sub> Cl <sub>12</sub> Si <sub>137</sub> Al <sub>55</sub> O <sub>384</sub> (Solid-State Nickel(II)-Exchanged) Tj ETQq0 0 0 rgBT /Overlock 10 Tf Chemistry B, 1998, 102, 2688-2695.	1.2	63
79	Crystal Structure of Indium-Exchanged Zeolite A Containing Sorbed Disulfur. Journal of Physical Chemistry B, 1998, 102, 17-23.	1.2	33
80	Crystal Structures of Dehydrated Fully Mn <sup>2+</sup> -Exchanged Zeolite X and of Its Ethylene Sorption Complex. Journal of Physical Chemistry B, 1997, 101, 9041-9045.	1.2	66
81	Crystal Structures of the Ethylene and Acetylene Sorption Complexes of Fully Ca <sup>2+</sup> -Exchanged Zeolite X. Journal of Physical Chemistry B, 1997, 101, 3091-3096.	1.2	58
82	Crystal Structure of an Ethylene Sorption Complex of Cd <sup>2+</sup> -Exchanged Zeolite X, Cd <sub>46</sub> Si <sub>100</sub> Al <sub>92</sub> O <sub>384</sub> ·29.5C <sub>2</sub> H <sub>4</sub> . Journal of Physical Chemistry B, 1997, 101, 2138-2142.	1.2	48
83	Crystal Structure of Zeolite X Exchanged with Pb(II) at pH 6.0 and Dehydrated: (Pb <sub>4+</sub> ) <sub>14</sub> (Pb <sub>2+</sub> ) <sub>18</sub> (Pb <sub>4O4</sub> ) <sub>8</sub> Si <sub>100</sub> Al <sub>92</sub> O <sub>384</sub> . Journal of Physical Chemistry B, 1997, 101, 5314-5318.	1.2	108
84	Crystal Structure of a Sodium Sorption Complex of Zeolite X Containing Linear Na <sub>32</sub> <sup>+</sup> Clusters. Journal of Physical Chemistry B, 1997, 101, 9022-9026.	1.2	39
85	Three Crystal Structures of Vacuum-Dehydrated Zeolite X, M <sub>46</sub> Si <sub>100</sub> Al <sub>92</sub> O <sub>384</sub> , M = Mg <sup>2+</sup> , Ca <sup>2+</sup> , and Ba <sup>2+</sup> . Journal of Physical Chemistry B, 1997, 101, 6914-6920.	1.2	97
86	Crystal structure of Zn <sub>4</sub> Na(OH) <sub>6</sub> SO <sub>4</sub> Cl·6H <sub>2</sub> O. Journal of Chemical Crystallography, 1997, 27, 325-329.	0.5	11
87	Crystal structure of fully dehydrated fully Ti <sup>+</sup> -exchanged zeolite X. Zeolites, 1997, 18, 325-333.	0.9	54
88	Crystal Structure of a Hydrogen Sulfide Sorption Complex of Dehydrated Partially Cobalt(II)-Exchanged Zeolite A. The Journal of Physical Chemistry, 1996, 100, 8373-8377.	2.9	24
89	Crystal structure of a hydrogen sulfide sorption complex of zeolite LTA. Zeolites, 1996, 17, 495-500.	0.9	13
90	Crystal Structures of Encapsulates within Zeolites. 2. Argon in Zeolite A. The Journal of Physical Chemistry, 1996, 100, 13725-13731.	2.9	10

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91	Two Anhydrous Zeolite X Crystal Structures, Cd <sub>46</sub> Si <sub>100</sub> Al <sub>92</sub> O <sub>384</sub> and Cd <sub>24</sub> Si <sub>143</sub> Al <sub>100</sub> O <sub>384</sub> . The Journal of Physical Chemistry, 1996, 100, 13720-13724.	2.9	67
92	Failure of ion exchange into zeolites A and X from four diverse nonaqueous solvents. Zeolites, 1995, 15, 377-381.	0.9	24
93	MOLECULES OF COPPER(II)-SPARTEINE DINITRATE ARE MIXED FOUR- AND FIVE-COORDINATE IN ONE CRYSTALLINE PHASE AND ONLY FOUR-COORDINATE IN ANOTHER. Journal of Coordination Chemistry, 1995, 34, 241-252.	0.8	26
94	Crystal Structures of Fully Dehydrated Cd(II)-Exchanged Zeolite A and of Its Cadmium Sorption Complex Containing Cd <sup>2+</sup> , Cd <sup>+</sup> , Cd <sub>2</sub> <sup>2+</sup> , and Cd <sub>20</sub> . The Journal of Physical Chemistry, 1994, 98, 3796-3800.	2.9	32
95	Structure of the tetrahedral sodium Na <sub>54</sub> <sup>+</sup> cluster in zeolite X. The Journal of Physical Chemistry, 1993, 97, 12663-12664.	2.9	56
96	Cesium Vapor Reacts with K <sup>+</sup> -Exchanged Zeolite A To Give Fully Cs <sup>+</sup> -Exchanged Zeolite A Containing (Cs <sub>4</sub> ) <sub>3</sub> <sup>+</sup> Clusters. ACS Symposium Series, 1988, , 177-193.	0.5	2
97	Preparation and structure of fully caesium exchanged zeolite A and of the linear (Cs <sub>4</sub> ) <sub>3</sub> <sup>+</sup> cation. Journal of the Chemical Society Chemical Communications, 1987, , 1225.	2.0	16
98	Reaction of dehydrated Na <sub>12</sub> -A with cesium. Synthesis and crystal structure of fully dehydrated, fully cesium ion-exchanged zeolite A. Journal of the American Chemical Society, 1987, 109, 7986-7992.	6.6	54
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105	Crystal and molecular structure of a diradical, 1,3-dinitro-4,6-di[3-(2,2,5,5-tetramethyl)-pyrrolidinyl-N-oxide]aminobenzene monohydrate. Journal of Crystal and Molecular Structure, 1976, 6, 87-100.	0.4	1
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107	Hydrated and dehydrated crystal structures of seven-twelfths cesium-exchanged zeolite A. The Journal of Physical Chemistry, 1975, 79, 2163-2167.	2.9	68